

**ENHANCED BURN WOUND HEALING THROUGH CONTROLLED AND
SUSTAINED DELIVERY OF BIOACTIVE INSULIN
FROM ALGINATE SPONGE DRESSINGS**

By

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Abstract

Skin is a dynamic and complex organ that relies on the interaction of different cell types, biomacromolecules and signaling molecules. Upon injury, a cascade of events occurs to quickly restore the skin's integrity. Depending on the size and severity of the wound, a dressing is used to provide a temporary barrier to protect from dehydration, microorganisms and debris. Current wound dressings however, cannot accelerate wound healing beyond the natural rate, require frequent dressing changes, and cannot be easily removed without triggering additional pain or tissue destruction. Insulin, a peptide used to treat Type 1 diabetes, has been reported to improve the recovery of severe burn wounds. Yet, no one has successfully demonstrated a convenient and effective insulin delivery vehicle that can be used to accelerate burn wound healing.

Poly(lactic-co-glycolic acid) microparticles, were shown to release bioactive insulin for a period of 25 days, stimulating human keratinocyte migration *in vitro*. A wound dressing made from poly(ethylene glycol) and alginate was formulated incorporating the insulin-loaded poly(lactic-co-glycolic acid) microparticles. Bioactive insulin release was achieved for nearly 3 weeks, along with favourable water handling and physical properties conducive for wound healing. Finally, *in vivo* testing confirmed that a constant dose of insulin from alginate-PEG sponge dressings loaded with 0.125mg, or 0.04mg/cm² insulin, with dressing changes every 3 days, was sufficient to significantly improve wound healing by 25%, as compared to an alginate-PEG sponge dressing without insulin. Insulin releasing alginate-PEG sponge dressings are therefore, an effective method of improving burn wound healing and may serve as a delivery vehicle platform to incorporate other therapeutic molecules in the future.

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Chapter 1

INTRODUCTION

1.0 Burn Injuries and Treatment

Burn injuries are one of the most common types of skin injuries that an individual can experience throughout their lifetime. Burns often result from excessive exposure to heat, caustic chemicals, or solar radiation, and predominantly occur in the home, workplace, or in traffic accidents.[1,2] Depending on the size, and the depth, a burn wound can take several weeks before the skin's integrity is restored, and months before the skin returns to its original state.[3] In some cases, burn injuries can be so severe, that skin grafts, reconstructive surgery, amputation, or loss of life can result.[4] Rehabilitation following severe burns is difficult, painful, time consuming and costly. Consequently, burn injuries place a significant burden on healthcare providers and medical resources.

Presently, burn treatments encompass a wide range of approaches depending on the condition of the burn. Most treatments combine the use of gauze bandages, hydrocolloids and ointments (white petrolatum) as gold standards of treatment.[5] These dressings help to prevent the entry of foreign debris, maintain a moist wound healing environment and absorb excess blood and exudate from the wound.[6,7] Gauze bandages, ointments, and hydrocolloids are also affordable, widely available, convenient, and remain stable for extended periods.[8] All of these dressings however, have no inherent capacity to promote wound healing beyond the natural rate. Removal and reapplication can also cause significant damage to the wound bed, resulting in impaired wound healing.[9] Infections caused by infrequent dressing changes can also lead to

the development of chronic wounds. Therefore, a new approach in wound dressing technologies must be explored to enhance patient recovery and comfort.

Delivery of growth factors to a wound is one approach that would enhance current burn wound treatments. Growth factors are signaling molecules that stimulate skin cells to migrate and proliferate when applied to a wound.[10] Epidermal growth factor (EGF) and keratinocyte growth factor (KGF) are two examples of growth factors which have been tested in the past.[11,12] Unfortunately, growth factors are expensive, unstable and require suitable delivery vehicles to be used effectively.[13] Insulin on the other hand, has been reported to promote wound healing over the past century. Insulin is affordable, available in a highly pure crystalline form, and is compatible with most common biomaterials used in wound dressings and drug delivery devices.

Currently, no group has demonstrated the ability to deliver bioactive insulin from a burn wound dressing. Consequently, the focus of this research was to design and test a delivery vehicle that released bioactive insulin in a sustainable and controllable manner for a period of 1 month. Crystalline insulin stabilized within poly(lactic-co-glycolic acid) (PLGA) microparticles was determined to be the best approach to achieve extended controlled release, and subsequent integration into an alginate-poly(ethylene glycol) (PEG) foam dressing. The dressing would enable uniform application of the insulin and protect the wound from the outside environment. Rodent testing revealed up to a 25% improvement in wound re-epithelialization, and improved scar tissue morphology. Extended insulin release from insulin-loaded PLGA microparticles, incorporated into alginate-PEG dressings, is therefore an affordable and potentially effective method of improving burn wound recovery.

Chapter 2

LITERATURE REVIEW

2.0 Human Skin

Human skin is a sophisticated and complex organ, responsible for protecting internal tissues from the external environment. Skin as the largest organ, collectively weighs an average of 4 kg, and covers a surface area of nearly 2 m². [14,15,16] Embedded within the skin, is a dense network of nerve endings that detect heat, pain and pressure, providing feedback to the brain about the environment. [17] This feedback mechanism alerts the body to excessive heat, or mechanical stress which may rupture the integrity of the skin. The skin also defends against pathogens by using a combination of cytokines, antibodies, complement proteins and immune effector cells that rapidly detect and deter infectious agents. [18] The skin also synthesizes vitamin D, [19] produces glandular secretions, [19,20] facilitates water transfer [21] and CO₂ gas exchange to facilitate homeostasis. [22] The skin therefore plays an integral role in supporting organ systems and protecting the body.

Consequently, when the skin becomes injured, its integrity must be restored as quickly as possible. The process of wound healing is a coordinated event that utilizes signaling molecules and various cell types to repair the defect. [23] To better understand this process, the following sections outline the normal physiology and anatomy of the skin, followed by a description of the wound healing process.

2.1 Skin – Physiology and Anatomy

Physiologically, the skin is comprised of several specialized layers, and components, as illustrated in Figure 1. The epidermis is the outermost layer followed by the dermis, skin appendages and subcutaneous fat. Each layer is comprised of specific cell types which play a role in the formation and regeneration of the skin after injury.

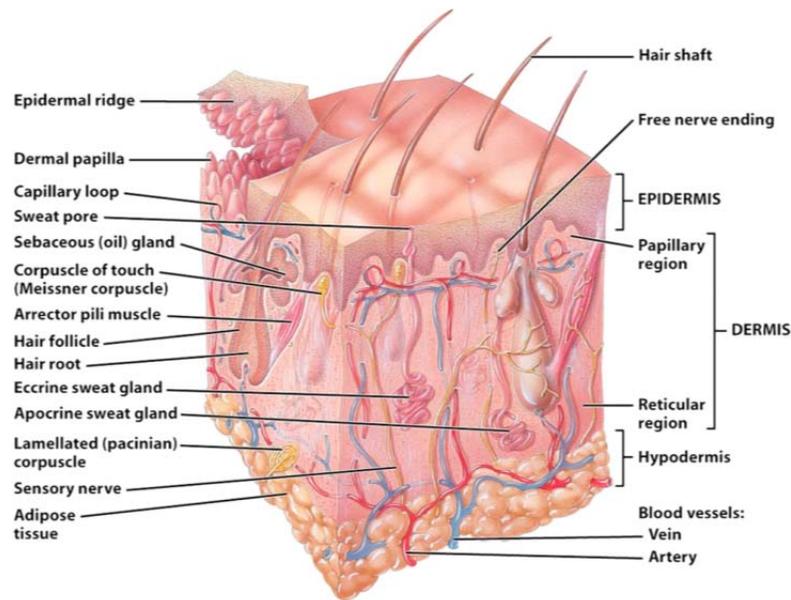


Figure 1. An illustration of healthy human skin comprised of the epidermis, dermis, skin appendages and subcutaneous fat.[9]

2.2 Epidermis

The epidermis is the outermost layer and originates from the embryonic ectoderm. It is composed of four layers and normally replaces itself about every 4 weeks.[17] The five major cell types found within the epidermis are keratinocytes, melanocytes, Merkel cells, Langerhans cells and fibroblasts. These cells help make up the four different epidermal layers including the stratum basale, the stratum spinosum, the stratum granulosum and the stratum corneum.

Connecting the epidermis to the dermis are rete ridges, small projections that protrude from the

epidermis into the dermis, securing the two major layers in place.[24] The epidermis is therefore a complex structure that protects all underlying tissues.

2.2.1 Keratinocytes

Keratinocytes are stratified squamous epithelial cells that constitute nearly 90-95% of the total cells present in the epidermis (Figure 2).[25] Keratinocytes originate in the stratum basale, and migrate to the surface of the epidermis where they are shed. During this process, keratinocytes migrate and differentiate within the stratum spinosum. This process takes approximately 40-56 days and plays an important role in host defense, and wound healing.[26] Keratinocytes are mainly responsible for producing keratin, a filamentous protein embedded within the amorphous matrix that contributes to the skin's elasticity, and resistance to chemical attack.[27,28] Keratinocytes also secrete a discrete cytokine profile during sessile and activated states in response to the absence, or presence of pathogens.[29] Sessile keratinocytes produce high levels of interleukin-1 (IL-1) within the epidermis, signaling a healthy state to adjacent immune effector cells. When in contact with antigen, keratinocytes function as a signal transducer and begin to secrete IL-6, IL-8, granulocyte-macrophage colony-stimulating factor (GM-CSF) and macrophage colony-stimulating factor (M-CSF), and monocyte chemotactic factors to mount an immune response and inflammation.[30]

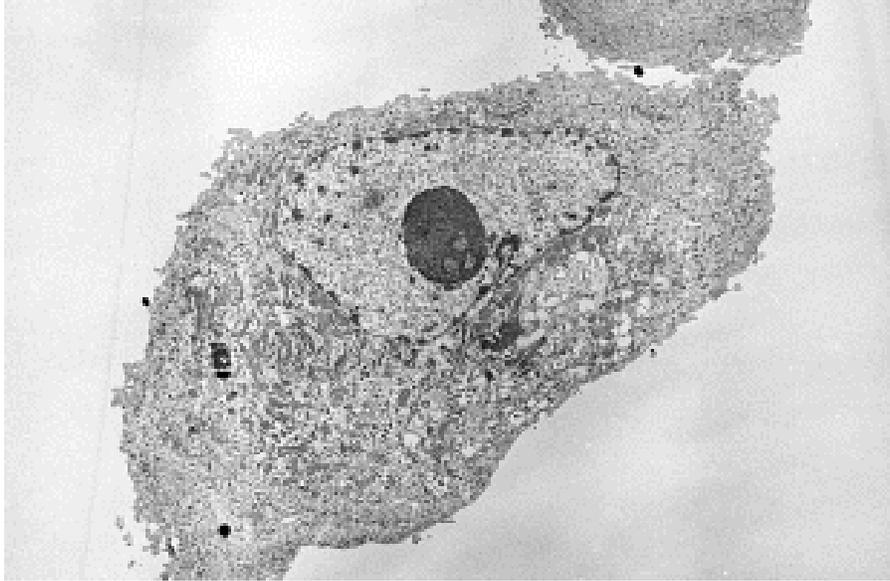


Figure 2. An example of a healthy keratinocyte.[28]

2.2.2 Melanocytes

Melanocytes are located in the epidermis, hair follicles and eyes, and are the primary producers of melanin, a pigment responsible for skin colour.[27] Morphologically, melanocytes are highly dendritic and reside in fewer numbers than keratinocytes, with approximately 1 melanocyte for every 10 keratinocytes.[31] Melanin is produced by specialized membrane-bound organelles called melanosomes. The purpose of melanin production is to protect the skin from DNA-damage induced by solar ultraviolet (UV) radiation. Upon stimulation by solar radiation, melanosomes translocate within the dendrites, positioning above the keratinocyte nuclei.[32] Almost instantaneously, melanin secretion occurs and blocks UV induced DNA-damage by photo-oxidizing and delayed tanning, through a mechanism that has not been clearly elucidated.[33] With chronic UV exposure however, uneven photo-ageing with uneven pigment distribution can occur. More recently, melanin has also been implicated in host immune function as well. Melanin has been shown to also interact with enzymes and serve as a cation chelator, thereby neutralizing oxidants, and inhibiting bacterial enzymes.[34] Melanocytes are therefore

important for protecting keratinocytes from UV radiation and help in maintaining host immune defenses.

2.2.3 Merkel Cells

Merkel cells reside in the undersurface of the epidermis, oral mucosa and are relatively scarce in numbers. Located within the basal layer of the epidermis, Merkel cells resemble nerve fibers and are believed to help perceive touch.[31] Localized concentrations of Merkel cells have been identified in the outer root sheaths of hair follicles and epidermal ridges.[35] Merkel cells are also implicated in the stimulation of the proliferation of keratinocytes leading to the three-dimensional development of the epidermal anlagen (a point of foundation), the maintenance of normal differentiation of keratinocytes, and the release of bioactive substances to subepidermal spaces.[36] Without Merkel cells, epidermal health cannot be maintained and sensation of the outside environment is hindered.

2.2.4 Langerhans Cells

Langerhans cells are immature dendritic cells of the skin. These cells are considered sentinel cells because they are phagocytic, and actively sample antigen present within the skin.[37] They are located in concentrations of 460-1000 cells/mm², making up nearly 3-8% of the total cells present in the epidermis.[31] They originate from the bone marrow and reside in the epidermis in an immature state.[38] Upon stimulation, the pathogen is engulfed, processed and presented once the Langerhans cell migrates via the lymphatic system to local lymphoid tissue.[39] At this point, the Langerhans cell matures into a dendritic cell responsible for presenting the processed antigen via major histocompatibility complexes (MHC) class I or II.[37] This facilitates the presentation of truncated antigen fragments to naïve T cells in conjunction with a cytokine profile to stimulate the adaptive arm of the immune system. These cytokines

include IL-1 α , IL-6, IL-7, IL-10, IL-12, IL-13, IL-15, IL-18, tumour necrosis factor- α (TNF- α), and transforming growth factor- β (TGF- β).[40] These cytokines help to attract, prime, and differentiate naïve T cells to become an effector T cell. Langerhans cells also play a role in rejecting foreign xenogenic, or allogenic skin grafts due to MHC class II mismatch between donor and host.[41] Figure 3 shows an illustration of the process by which Langerhans cells detect, engulf, transport and display antigen in local lymph nodes. Langerhans cells are an important cell type that detects foreign pathogens and alerts the immune system to an attack.

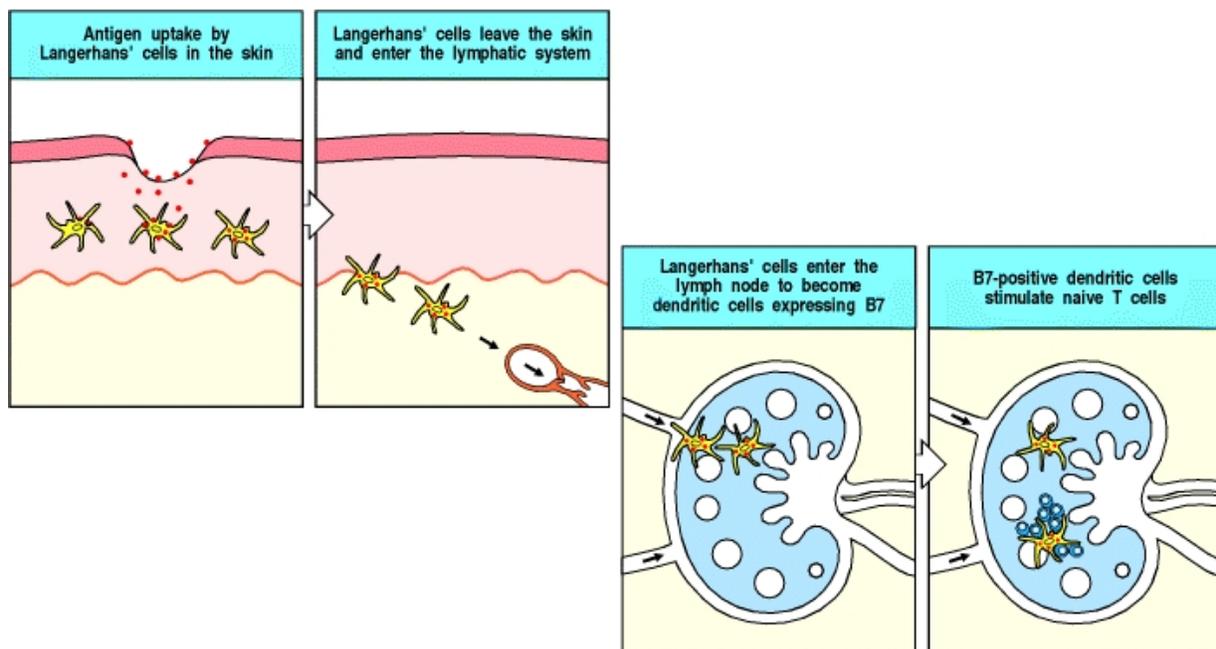


Figure 3. An illustration of Langerhans cells sampling and engulfing foreign antigen. The antigen is then internalized and processed while the Langerhans cells migrate through the local lymphatic ducts to a lymph node. While in the lymph node, the antigen is presented to naïve T cells where a release of cytokines, prime and activate the T cell into a mature effector T cell. The effector T cell is then free to migrate away from the lymph node and search for the foreign pathogen bearing the antigen.[37]

2.2.5 Fibroblasts

Dermal fibroblasts are a member of the connective-tissue family, which include cartilage and bone cells. Fibroblasts originate as differentiated mesenchymal cells and constitute 40-60% of the total cell populations found in the skin, lung and gingiva.[42] Morphologically, fibroblasts appear as spindle-shaped cells, which contain well-developed endoplasmic reticulum and are pertinent to the normal production and organization of the extracellular matrix (ECM). Fibroblasts are chiefly responsible for the secretion of types I and III collagen,[43] elastin and laminin, and assist with epidermal differentiation.[44] In addition to biomacromolecules, fibroblasts also secrete various growth factors such as insulin growth factor (IGF), keratinocyte growth factor (KGF), platelet derived growth factor (PDGF), transforming growth factor (TGF) and vascular endothelial growth factor (VEGF) which helps stimulate and bridge communication between various cell types found within the skin.[44,45] Upon injury, fibroblasts are one of the first cell populations to arrive and help rebuild the ECM by laying down new collagen to repair the defect.

2.2.6 Stratum Basale

The stratum basale, also commonly referred to as the stratum germinativum is a continuous layer that is typically one-cell layer thick, with occasionally two or three layers found.[19,46] The basal cells consist of non-serated (cuboidal) and serated basal keratinocytes (projections).[47] These cells are considered to be the “stem cells” of the stratum basale and help to continuously renew the epidermis approximately every 14 days.[19,24,48] This rate of renewal can increase if injury or inflammation are present, ensuring that there is a continuous supply of keratinocytes to heal the wound.

2.2.7 Stratum Spinosum

The stratum spinosum is located next to the stratum basale and contains keratinocytes undergoing active differentiation. This region is characterized by spines, or intercellular bridges that extend between keratinocytes.[19] The keratin is produced by keratinocytes which linked together by the help of desmosomes which anchor adjacent cells together. The keratinocytes in this layer appear “spiky” and progressively transition from polygonal to a larger and flatter shape as they move towards the surface of the skin.[49]

2.2.8 Stratum Granulosum

Adjacent to the stratum spinosum, the stratum granulosum is a region where keratinocytes continue to differentiate, acquire additional keratin, and progressively become flatter in morphology.[19] At this point, the keratinocytes also begin to acquire distinctive dark granules that are composed of keratohyalin, consisting of two proteins rich in proline and cysteine amino acid residues.[50] Profilaggrin, the precursor to filaggrin is associated with the aggregation of keratin fibers in the stratum corneum.[14] Involucrin is the second protein and plays a role in the formation of the cell envelope of cells located within the stratum corneum.[51] The granular cells within the stratum granulosum also contain lamellar granules containing a variety of biomacromolecules including polysaccharides, glycoproteins, and lipids which are secreted into the intracellular spaces.[19] These biomacromolecules help anchor cells within the stratum corneum, and prevent moisture loss from the skin.[52] As a result, the stratum granulosum is responsible for providing additional support to the epidermis and helping to shape the stratum corneum.

2.2.9 Stratum Corneum

The stratum corneum is composed of cells which are rich in keratin and stacked in approximately 15 to 25 layers.[19] The soles of the feet and palms of the hands can have up to 100 layers, providing additional support and protection. This arrangement facilitates the formation of a semi-permeable layer which is highly effective in preventing moisture loss and chemical penetration from occurring. Morphologically, the cells within the stratum corneum appear large, flat, and polyhedral, and are held together by a lipid rich glue. The primary purpose of the epidermis is to produce the stratum corneum, the major physical barrier of the skin.[53] Figure 4 illustrates the entire composition of the epidermis, with all four sub-layers.

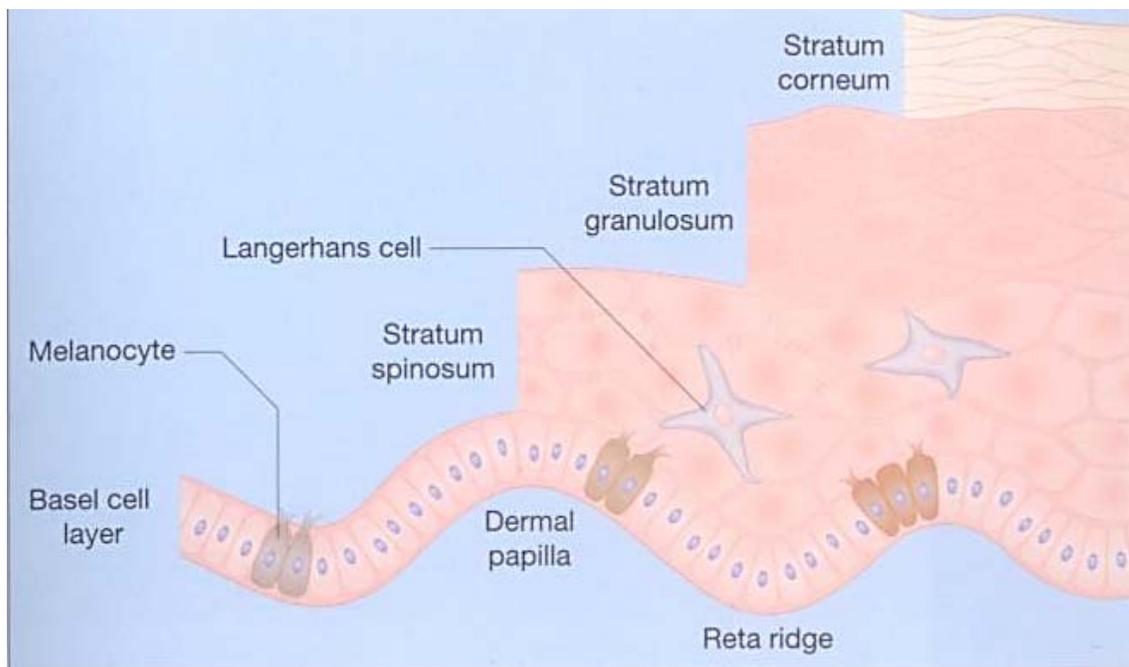


Figure 4. The epidermis including the stratum basale, stratum spinosum, stratum granulosum and stratum corneum. Note the position of melanocytes and Langerhans cells located within the deeper layers of the epidermis.[16]

2.3 Dermal-Epidermal Junction

The dermal-epidermal junction is also commonly referred to as the basement membrane zone and marks the interface between the two major layers of the skin. The basement membrane is a complex mixture of biomacromolecules that are thin, continuous layers, containing type IV and VII collagen, laminin, nidogen, and heparin sulfate proteoglycans.[54] Four distinct layers are contained within the dermal-epidermal junction including keratin filaments, lamina lucida, lamina densa, and anchoring fibrils.[19] Keratin filaments in the basal keratinocytes attach to hemidesmosomes which connect to adjacent anchoring filaments found in the lamina lucida.[55] The lamina lucida is a lucent space with a thickness of approximately 26-49 nm and contains structures such as collagen fibrils, anchoring fibrils, or microfibrils.[56] The lamina densa is rich in type IV collagen, laminins and perlecan, all synthesized by the keratinocytes located in the stratum basale.[54] Anchoring fibrils resemble thick fibers which are predominantly composed of type VII collagen and located in the sublamina densa region of the papillary dermis.[19] These four layers in turn, help to connect the epidermis and dermis through the use of biomacromolecules which form a glue between the layers.

2.4 Dermis

The dermis is a robust elastic structure that supports the normal health and function of the epidermis located above. The dermis is approximately 1-4 mm thick and has three main functions including facilitating nutrient exchange, providing sensory input, and draining lymph from the skin.[19] Nutrient exchange is achieved through a dense, microcirculatory bed that is concentrated around the dermal papillae, hair follicles and eccrine sweat glands. The capillaries form loops with a diameter of 2-3 μm to increase surface area and support metabolically active areas with sufficient gas, nutrient and waste exchange.[57] Neuronal communication within the

dermis is achieved by both sensory and autonomic arms which provide information and reaction to the outside environment. Free or specialized nerve endings closely situated around blood vessels, sweat glands and hair follicles, provide sensory input, whereas autonomic fibers provide homeostatic control of blood vessels, sweat glands and erector pili muscles near hair follicles.[58] The dermal lymphatic system, similar to the blood capillary bed, forms a dense network which shuttles water, proteins, macromolecules, cell fragments, and mobile inflammatory and tumour cells from the interstitial spaces to regional lymph nodes located away from the skin.[59] This arrangement ensures that immune effector cells can quickly access the skin and remove a potentially harmful pathogen from invading the host.

Structurally, the dermis is composed of fibrous and non-fibrous biomacromolecules which give the skin elasticity and tensile strength. Fibrous elements are the most predominant component of the dermis and include over 75-80% type I collagen and 25-20% type III collagen. Throughout life, the ratio of type I and type III collagen remains fairly constant, however the proportion of type III collagen begins to increase with advancing age.[60] Collagen, along with elastin, helps to dissipate mechanical energy during movement and allows the skin to return to its original shape after deformation.[61] Non-fibrous components of the dermis include finely filamentous glycoproteins, glycosaminoglycans, and proteoglycans (i.e., hyaluronic acid, chondroitin sulphate, dermatan sulphate, versican, and decorin).[62] These biomacromolecules help to guide collagen fiber formation, organization and interaction, as well as provide turgor and resilience.[63] Fibroblasts synthesize collagen and help to maintain the supply of collagen during normal growth and repair after injury.[64] The dermis is therefore an integral component of the architecture of the skin which provides strength and support for the overlying epidermis.

2.5 Skin Appendages

The skin also contains several skin appendages that help the body maintain thermal regulation and mechanical function. These components include eccrine glands, apocrine sweat glands, hair follicles and sebaceous glands and nails. The skin appendages excluding nails, originate from the epidermis, but become situated within the dermis over time.[19] Each of the elements comprising the skin appendages will be described in the following sections.

2.5.1 Eccrine Sweat Glands

The eccrine sweat glands are one of the most significant skin appendages as they are responsible for releasing sweat upon thermal or emotional stimuli. The release of sweat is chiefly responsible for regulating body temperature through evaporation from the skin surface. The secretory portion of the gland consists of a coiled tubule which is located deep within the dermis and leads to the surface via a sweat duct through the epidermis.[19] There are approximately 3-4 million eccrine sweat glands distributed across the skin releasing up to 10 L per day.[65] Sweat is similar in composition to blood plasma, but becomes hypotonic as it transitions through the sweat duct nearing the surface of the skin.[19] The eccrine gland is therefore an important skin appendage necessary for thermal regulation.

2.5.2 Apocrine Sweat Glands

Apocrine sweat glands are another skin appendage concentrated mostly in the armpit and anogenital regions of the body. Similar to the eccrine sweat glands, apocrine glands consist of coiled tubules, but terminate at the distal portion of the hair follicle.[66] This results in sweat being carried up the length of the hair follicle until it reaches the surface. During development, apocrine sweat glands require stimulation via androgens to develop, and serve little function in thermal regulation later in life.[19]

2.5.3 Sebaceous Sweat Glands

Sebaceous sweat glands are skin appendages that are physically fused together with hair follicles throughout the body, except for the soles and palms. Sebaceous glands are responsible for the production of sebum, a lipid-rich liquid consisting of cholesterol, squalene, waxes, and inorganic salts. The purpose of sebum is to bath the hair shaft to prevent it from drying out, giving it an oily appearance when not washed for several days. The greatest concentration of sebaceous glands are found on the scalp, forehead, cheeks and chin.[67] Harkey (1993), estimates that a 10 cm² area of the scalp is capable of producing 0.7-2.4 mg of sebum within a 4 h period. Sebaceous glands, especially during adolescence, are implicated in the development of acne vulgaris, when the sebaceous gland becomes clogged with sebum.

2.5.4 Hair Follicles

Hair follicles produce and secure hair deep within the dermis in almost every body surface except for the palms and soles of the feet. In animals, hair growth is necessary to retain heat, provide camouflage and provide protection from environmental elements. In humans, hair helps with thermoregulation by providing insulation and a large surface area for sweat to evaporate and cool the skin.[19] Within the hair follicle, three different cell types including epithelial, mesenchymal, and neuroectodermal cells are present to help establish and synthesize building blocks used to form the hair shaft.[68] Overall, the hair follicle is divided into three distinct zones, all with specific functions pertinent to the formation and anchoring of the hair shaft.[67] Approximately 3-4 mm below the surface of the skin, the innermost region near the bulb of the hair root, is the area associated with the biological synthesis of hair cells. Nearby this area are small smooth muscle bundles originating in the dermis and function to raise the hair in order to retain additional heat.[69] Immediately above the bulb, in the second region is the

keratogenous zone where the hairs harden and solidify. The third zone is a fully hardened hair shaft consisting of dehydrated, cornified cells shaped into fibrils that fuse with an intercellular binding material.[67] Alongside the hair shaft are two types of glands which excrete sweat down the length of the hair follicle. In the armpit and anogenital areas, the apocrine sweat gland drains into the hair follicle. In all other areas of the body, the sebaceous sweat gland is fused next to the hair follicle and is also responsible for releasing sweat.[19] In combination with the eccrine glands, the hair helps to trap the liquid sweat and spread it over a larger surface area, thereby enabling more efficient cooling whenever the body is exposed to hot environments. Once the hair has exited the hair follicle, the hair can be classified according to its physical appearance and length. Short, fine and light coloured hair is called vellus hair, whereas thick, long and dark hair is classified as terminal hair.[66]

2.5.5 Nails

The human nail is a specialized skin appendage which grows from the ends of the toes and fingers to facilitate grasping, manipulation, scratching, and prying.[70] Physiologically, the nail unit is composed of the nail plate, nail bed, hyponychium, nail matrix, nail folds, cuticle, anchoring ligament and distal phalangeal bones (Figure 5).[71] The nail is produced by the matrix through continuous differentiation of basal cells that lose their nuclei, flatten, cornify and migrate to form the nail plate.[71] The average growth rate of the nail is approximately 0.1 mm/day, or 3 mm/month in a healthy human being and can be influenced by the overall health of the individual as well as specific medications.[72]

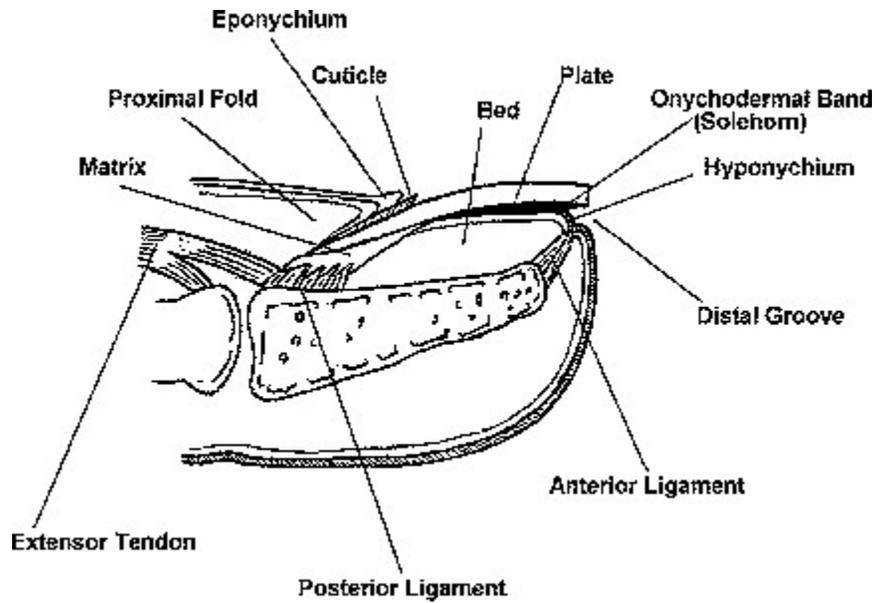


Figure 5. The nail unit comprised of the nail plate, nail bed, hyponychium, nail matrix, nail folds, cuticle, anchoring ligament and distal phalangeal bones.[71]

2.6 Wounds

When skin is exposed to excess thermal energy, reactive chemicals, or mechanical force, a wound typically results.[1] Wounds come in several forms and can be as minor as a small cut in the skin or as severe as a full thickness burn extending down to the underlying bone. Depending on the extent of the wound, healing can be slow, impair the quality of life and take up substantial healthcare resources.[73] Consequently, wound healing is an active and dynamic process, designed to regenerate lost skin layers.[74,75]

Wounds are frequently classified as either acute or chronic, depending on the time frame required to establish complete wound healing. According to Whitney (2005), all wounds at the time of trauma can be considered acute, regardless if they are the result of accident, trauma, burn, or surgery.[76] What differentiates the two categories is the length of time required for healing take place. Acute wounds generally heal within a period of 8-12 weeks, while chronic wounds

extend well beyond this time frame.[77] Burn wounds for example, are classified as a form of acute injury, and vary significantly in healing time depending on the type of burn (first-, second-, or third-degree), treatment provided (wound dressing selection, fluid replacement, surgery, etc.), and age of the patient.[78] Figure 6 demonstrates how the burn depth is used to classify the severity of burn wound.

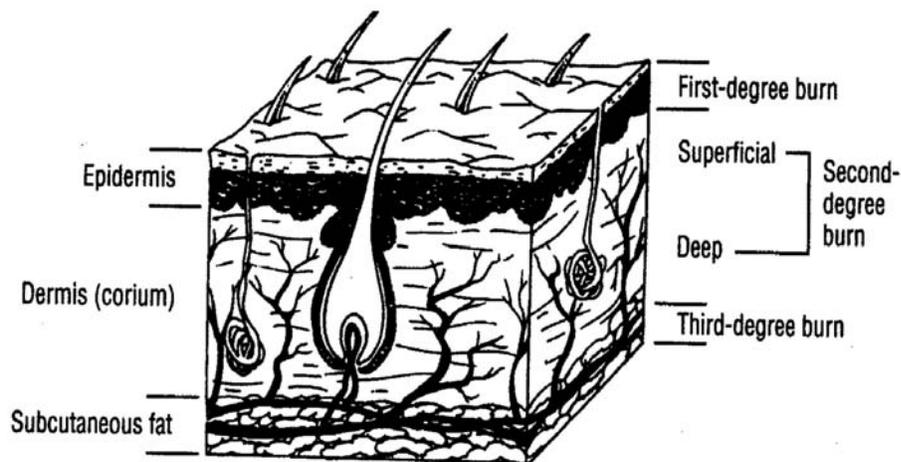


Figure 6. A cross-section of the skin and the classification of first-, second- and third-degree burns. Note that first-degree burns are marginalized to the epidermis, while second- and third-degree burns can reach the dermis and subcutaneous fat.[78]

The danger posed by acute wounds is the risk of bacterial infection if a wound remains open for too long. When infection occurs, the body's adaptive immune response can lead to an intense inflammatory response resulting in localized tissue hypoxia and poor wound healing characteristics.[79] Excessive tissue damage, wound tension and inadequate hemostasis may also result from bacterial infection.[80] Therefore, a dressing can function as a temporary protective barrier, isolating the wound from the external environment and improve healing.

Chronic wounds fail to heal within the normal 8-12 week period.[1] Chronic wounds can develop as a result of persistent infections, tissue hypoxia, necrosis, diabetes, tumours, excess exudate, malnutrition, old age and excess levels of inflammatory cytokines, delaying the normal

progression of healing.[77] According to Fonder *et al.* (2008), venous ulcers, arterial ulcers, diabetic foot ulcers, vasculitis, and pressure ulcers are all forms of chronic wounds that result from one or more of the factors listed above.[81] As a result, chronic wounds require longer periods of care, and more complex treatment regimens.

The overall goal in the treatment of burn wounds therefore must be to stimulate the natural regenerative mechanisms at the earliest possible opportunity, while also providing an optimal wound healing environment to facilitate the restoration of damaged tissues. This in turn stimulates cell migration, differentiation and re-epithelialization of the wound.

2.6.1 Burns

Burns are one form of thermal injury that occurs when the skin is subjected to high temperatures for extended periods time, distorting the epidermis and/or dermis.[23] Depending on the size, depth and extent of the burn, significant medical attention can be required. The process of burn wound healing is therefore often lengthy, susceptible to infection, painful and has the potential to lead to death.[82]

2.6.2 Burn Epidemiology

The Canadian Burn Foundation estimates that a child is burned every 7.5 minutes at the home by a hot liquid and that thousands of Canadians are seriously burned each year.[83] Spinks *et al.*, (2008) studied pediatric burn cases over a ten-year period between 1994-2003 and determined that nearly 500 children died from burn injuries and that over 10,000 were admitted to Canadian hospitals for burn treatment.[84] Scalds were the most significant etiological factor, accounting for about 50% of all hospital admissions across Canada. In the United States, the American Burn Association (2011 Fact Sheet), has estimated that nearly 450,000 burn injuries required medical attention, while 3,500 individuals died directly from fire or burns alone.[2]

Table 1 summarizes the demographics for 2001-2010 burn admissions to burn centers as reported by the American Burn Association, showing that most burns originate from fire in the home, and involve a higher number of males as compared to females.

Table 1. Selected statistics from the American Burn Association 2001-2010, for burn admissions to burn centers.

Category	Percentage
Survival Rate	96.1
Gender	70 Male
	30 Female
Ethnicity	60 Caucasian
	19 African-American
	15 Hispanic
	6 Other
Admission Cause	44 Fire/Flame
	33 Scald
	9 Contact
	4 Electrical
	3 Chemical
	7 Other
Place of Occurrence	68 Home
	10 Occupational
	7 Street/Highway
	15 Other

2.6.3 Burn Etiology

Burns originate whenever sources of heat, chemicals, electricity, or radiation are applied to the skin in excessive amounts, causing a disruption in the epidermis and/or dermis. Burns stemming from heat are classified as either originating from wet or dry heat. Wet heat is most frequently associated with scalds, resulting in skin loss. Boiling water, superheated steam and hot grease only require seconds to cause damage across large surface areas.[3] Figure 7 shows an example of a common scald injury to the foot of a pediatric patient. Consequently, scalds are one of the most common types of burn injuries particularly in young children.[23] Dry heat injuries can result from direct contact with a flame, hot surface or molten metal. Dry heat burns

are the second most common form of burn injury and are usually concentrated to a small area, but can extend deeper into underlying tissues.[3]



Figure 7. A pediatric patient with a scald injury to the foot resulting in a superficial partial thickness burn. British Association of Plastic and Reconstructive and Aesthetic Surgeons, 2012.

Chemical burns are much more variable in nature due to reactivity of the compound, concentration and contact time. Cement is one material that contains large quantities of lime that when left on the skin over extended periods, results in a chemical burn. Dust during cement mixing and application can penetrate clothing, enter work boots and come into direct contact with skin. Over a period of time, lime begins to react with the skin, resulting in a chemical burn.[85] Figure 8 shows a photograph of a male patient with alkaline burns to the face and upper chest. Phenol and hydrofluoric acid used in metal processing and cleaning are other workplace chemicals that can cause severe burns, long after exposure has occurred.[3] In all cases, chemical residue must be removed to prevent further damage once a wound has been dressed.



Figure 8. A clinical photograph of a male patient with severe alkaline chemical burns to the left cheek and neck.[85]

Electrical burns occur when the skin comes into direct contact with a high energy source. These sources may include frayed electrical cords, high-voltage supplies and improperly grounded industrial electrical equipment. The passage of current from the source will frequently result in a small punctate that extends deeply into the underlying tissue.[3] Figure 9 shows an example of an electrical burn. In some cases, current may pass across the chest resulting in cardiac arrhythmias, or through limbs causing severe destruction of muscle and nerve bundles.[23,86]

Over exposure to solar radiation is another form of burn injury that commonly affects all age demographics. Sunburns are often localized to the superficial layers of the skin, but can cover extensive surface areas resulting in excessive pain and discomfort.[3] Figure 10 illustrates a sunburn caused by excessive exposure to solar radiation. More rarely are burns associated from ionizing radiation, which typically appear in patients receiving radiotherapy.[86]



Figure 9. A patient with electrical burn to the foot. The location of the injury represents an exit wound whereby the electrical current would have flown out of the body to a ground source. Dr. Ken Harrison, NSW Institute of Trauma and Injury Management.



Figure 10. A male patient with a severe sunburn affecting the back. The uneven appearance of the sunburn is a result of uneven sunscreen application prior to exposure.[86]

2.6.4 Burn Scoring

Burn scoring is the process of determining the severity of a burn by measuring both the depth of the injury and the total body surface area affected. Classically, burn depth has been ranked as first, second and third degree depending on the appearance of the injury. More recently, this ranking method has been replaced with a method which describes the anatomical thickness of the skin layers affected by the injury.[18] Burn wounds are now classified as being superficial, partial-thickness and full-thickness.[4]

2.6.5 First Degree/Superficial Burns

First degree, or superficial burns are injuries that only affect the epidermis. According to Barrett-Nerin *et al.*, (2006), superficial burns are not included in the total body surface area assessment and are typically caused by UV radiation from the sun. These types of burns are minor in severity and typically manifest with the appearance of reddened skin, painful to tactile touch, and occasionally the formation of small blisters.[87] Healing occurs typically within 1 week and is achieved through the sloughing off of keratinocytes. No medical attention is needed, however the application of topical moisturizing creams and anaesthetics provide relief from dry and tender skin.

2.6.6 Second Degree/Partial-Thickness Burns

Second degree, or partial-thickness burns extend beyond the epidermis and into the dermis. Depending on the depth, partial-thickness burns can be further classified as superficial or deep.[18] In superficial, partial-thickness burns, the injury extends to the superficial papillary dermis, causing blistering and extreme pain. The skin will also appear bright pink and blanches whenever tactile pressure is applied.[4] Healing from this type of wound usually occurs via the proliferation of keratinocytes from the margins of the wound and from hair follicles which are still intact. Healing time takes approximately three weeks to complete and will usually not result in the formation of a scar.

Deep partial-thickness burns extend into the reticular dermis and require more time to heal. Hair follicles, skin appendages and nerve endings are often damaged resulting in longer recovery times beyond three weeks. Characteristically, deep partial-thickness burns appear waxy, and white-pink and typically are less painful than superficial partial-thickness burns. Recovery can involve surgery to remove dead tissue and may require skin grafting to accelerate recovery.

Increased scarring and a higher susceptibility to infection are common complications that occur following recovery from deep partial-thickness burns.[4]

2.6.7 Third Degree/Full-Thickness Burns

Full-thickness burns extend throughout the entire depth of the dermis and into the subcutaneous fat. This is the most severe, debilitating and life threatening burn injury that can be sustained. In children, full-thickness burns occur more easily due to thinner skin as opposed to adult skin which requires significantly more thermal energy, and are more difficult to stabilize.⁸⁸ Full-thickness burns appear yellow, brown, or black, along with severe edema and lack of blistering. Since the integrity of the skin is lost, there is a serious risk of infection and sepsis.[18] Drastic physiological shifts in fluid balance and metabolic activity frequently follow full-thickness burns, resulting in impaired organ function and body system failure. Recovery is often painful, difficult and requires significant medical resources. Table 2 summarizes the three degrees of burn injuries and their causes.

Table 2. Classification of the three degrees of burn injuries with (-) meaning "absence of" and (+) meaning "presence of." [87]

Burn Depth	Colour	Appearance	Pain	Treatment
First Degree/ Superficial	Pink	(-) Blisters (-) Eschar	Mild	Moisturizers Analgesic
Superficial Partial Thickness	Red	(+) Blisters (-) Eschar	Moderate to severe	Topical Antibiotic
Deep Partial Thickness	Red, Cream, Yellow	(+/-) Blisters (+/-) Eschar	Moderate	Topical Antibiotic Possible Skin Graft
Third Degree/ Full Thickness	Brown, Black	(-) Blisters (+) Eschar	None	Topical Antibiotic Skin Graft Required

2.6.8 Total Body Surface Area (TBSA)

Total body surface area is a measurement used to estimate the percentage of the skin surface area that has been burned. Adult TBSA is calculated using the rule of nines, estimating the surface area of the arms, torso, head and legs as multiples of nines. In burns whereby the surface area only affects a portion of the limb or torso, the patient's palm is usually used to represent 1% of the TBSA.[18] In children and infants, since the rule of nine is less accurate, the Lund-Browder chart for children is used.[88] Figure 11 illustrates both the rule of nine and Lund-Browder estimates for TBSA coverage. The purpose of measuring TBSA and burn depth is to calculate how much fluid resuscitation is needed to restore the fluid balance in the patient after experiencing a severe burn injury.

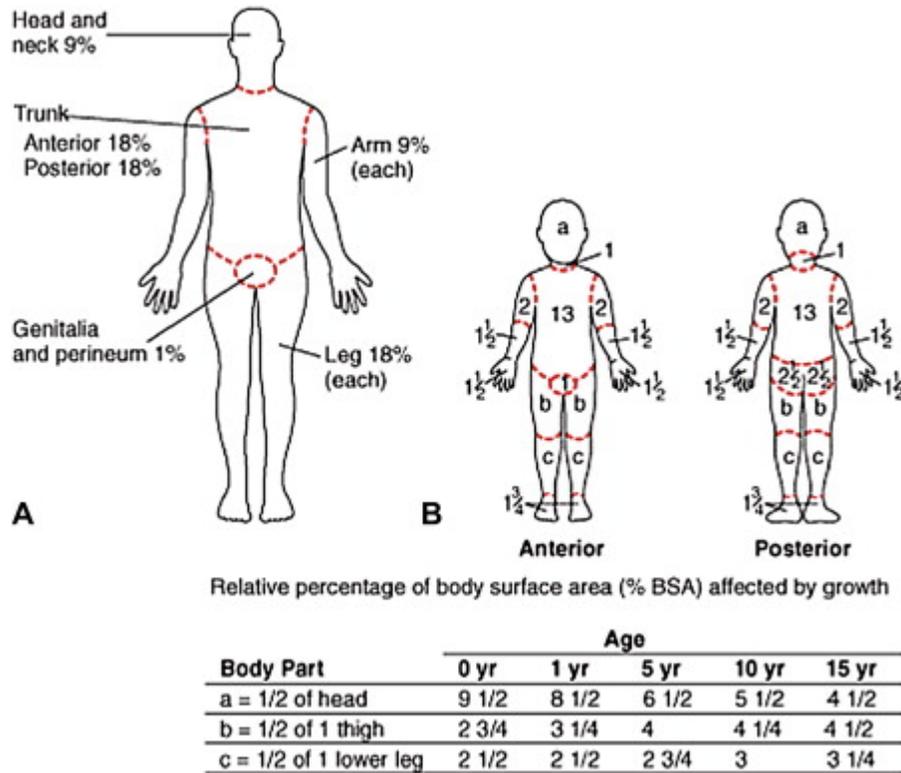


Figure 11. The rule of nines in adults (A), and the Lund-Browder estimations for calculating TBSA in children (B).[88]

2.7 Wound Healing

Wound healing is a dynamic process that relies on the interaction of different cell types and signalling molecules, and on the communication and cooperation of resident cells in the surrounding area of the wound (e.g., fibroblasts) and peripheral circulating immune cells (e.g., macrophages, neutrophils, monocytes and lymphocytes).[89] As soon as a wound forms, a four part wound healing process begins. To understand this process, the following sections outline the phases of wound healing and the mediators involved.

2.7.1 Haemostasis

Haemostasis is the process whereby vascular constriction and the formation of a fibrin clot take place to prevent excessive blood loss from the injury. A fibrin clot is necessary to initiate platelet aggregation and the release of proinflammatory cytokines and growth factors initiating the wound healing response.[90] After hemostasis has been achieved, the wound begins to secrete exudate, which is essentially blood, from which most of the red blood cells and platelets have been removed. Exudates continue to be released throughout the wound healing process.[91] Exudates function to keep the wound moist and continually irrigate the region. Burns in particular are highly destructive and destroy the semipermeable membrane associated with the lipoprotein layer in the stratum corneum, leading to excessive levels of exudates.[92,93] Table 3 lists the measured amount of water vapour lost from burns of various degrees. Hemostasis is designed to control and stop loss of blood, while also priming and signalling immune effector cells to the wound region.

Table 3. Evaporative water loss and skin temperature values for different burn wound types.[93]

Wound Condition	Evaporative Water Loss (g/m²/24 h)	Surface Temperature (°C)
Normal Skin	204 ± 12	35.8 ± 0.2
First Degree	278 ± 26	35.3 ± 0.1
Second Degree	4274 ± 132	35.3 ± 0.4
Third Degree	3437 ± 108	34.5 ± 0.4
Granulating Wound	5138 ± 202	34.7 ± 0.2
Donor Site	3590 ± 180	35.3 ± 0.2

2.7.2 Inflammatory Phase

Twenty-four hours after platelets have coagulated and growth factors have been released, immune effector cells begin to arrive at the wound site.[94] Inflammatory cells such as neutrophils, lymphocytes, monocytes and macrophages migrate through the blood vessels and surrounding tissues toward the wound to protect the area from infection.[23] Figure 12 illustrates a timeline at which different cell types appear within the wound during the healing process.

These cells also function to clean and remove cellular debris through phagocytosis, to prepare the site for tissue repair. During this stage, a second influx of cytokines and low-molecular weight signalling molecules such as interleukins, thromboxanes, leukotrienes, kinins, serotonin, and catecholamines are released and fill the wound to promote regional inflammation and tissue repair.[95] Likewise, reactive oxygen species (ROS) released by inflammatory cells, destroy any foreign material that may have entered the wound to maintain a clean wound environment.[23,96] After approximately 3 days, the wound transitions from an inflammatory phase into a proliferative phase.

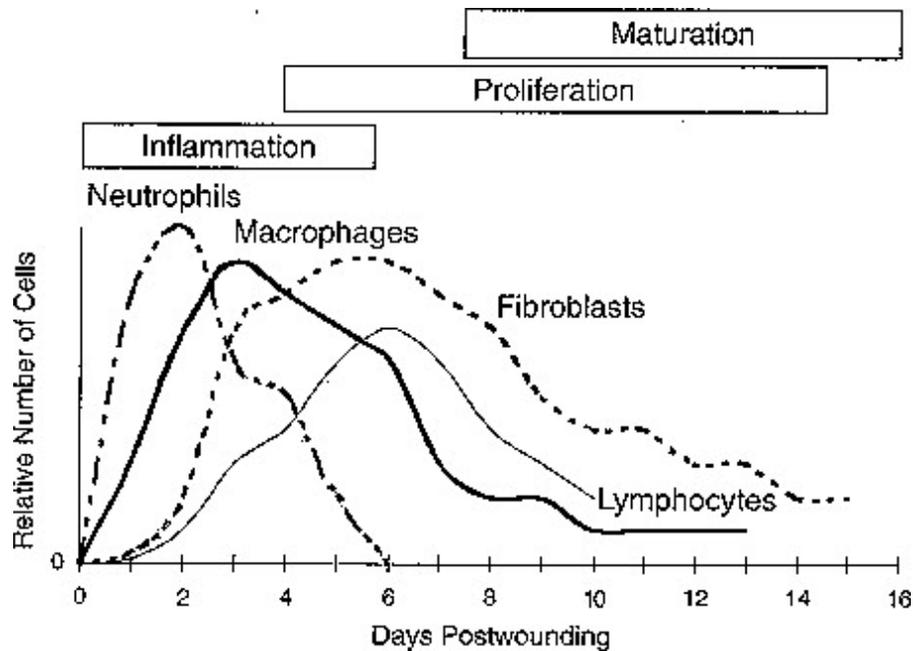


Figure 12. A timeline plotting the appearance of neutrophils, macrophages, fibroblasts during different phases of wound healing.[102]

2.7.3 Proliferative Phase

The proliferative phase is the third stage of wound healing responsible for replacing tissue and repairing vascular damage. Angiogenesis, collagen deposition, granulation tissue formation, and re-epithelialization are the major events of this stage of wound healing.[97] Once the wound site has been cleaned and liberated of cellular debris, fibroblasts migrate to the wound following a gradient of cytokines and growth factors such as transforming growth factor- α (TGF- α), platelet-derived growth factor (PDGF), EGF, KGF-1 and fibroblast growth factor (FGF).[98,99] Many of these growth factors result in the differentiation of fibroblasts into wound fibroblasts which begin to lay down a new extracellular matrix (ECM).[100] This new ECM consists of granulation tissue which is a mixture of collagen types I and III, tenascin, fibronectin, proteoglycans and in some cases, the inclusion of a new network of blood vessels.[100] Figure 13 illustrates a timeline of the appearance of different ECM components during the wound healing process. As the fibroblasts invade and differentiate, they produce

another influx of messenger molecules and peptides such as plasminogen activator, lymphokines and hyaluronan.[101] As the fibroblasts establish a new ECM and become inhibited by increasing concentrations of glycosaminoglycan, another set of cells migrate toward the center of the wound.[102] Basal keratinocytes at the wound edges and hair follicles begin to migrate toward the inner portion of the wound,[102] stimulated by the increasing levels of fibrin, fibronectin and collagen type I.[103] Keratinocytes continue to proliferate, transform and phagocytose any remaining cellular debris, resulting in re-epithelialization, with closure of the open wound achieved by about the 12th day from the time of injury.[23] It is this phase of wound healing that is of strong interest to most burn researchers because of the high number of cellular and signaling events taking place. Consequently, the delivery of additional growth factors to the wound site may modulate the proliferative phase and accelerate the wound healing. Once keratinocytes have entered the wound, the proliferative phase is complete and wound remodeling can take place.

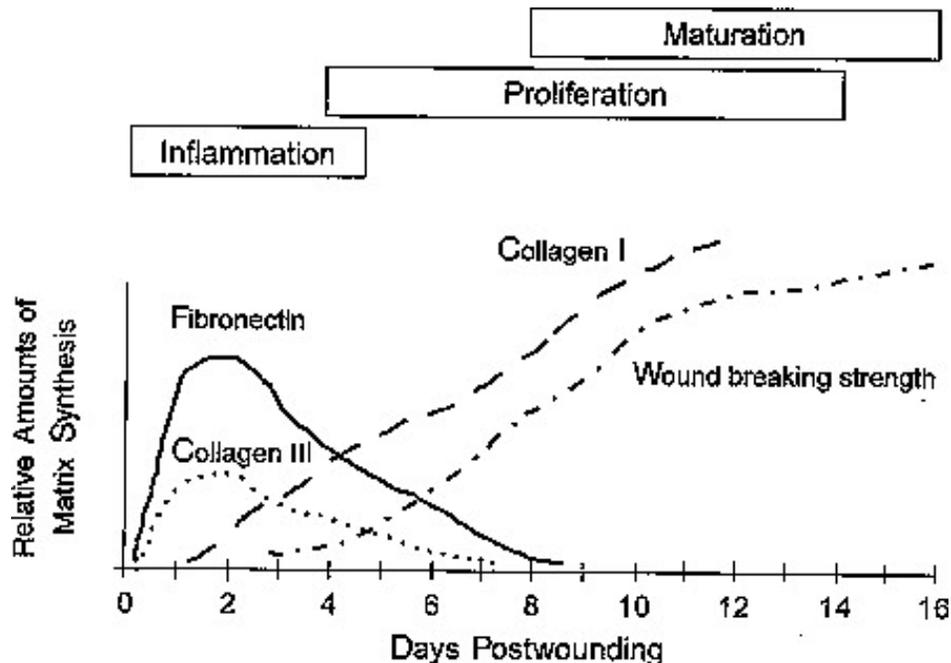


Figure 13. Appearance of fibronectin, and types I and III collagen during the wound healing process.[102]

2.7.4 Remodeling Phase

Remodelling is the final phase of wound healing taking several weeks to complete.[76]. The goal of wound remodeling is to replace weaker components of the ECM with more robust biological molecules that provide greater strength to the re-epithelialized wound. An example of this process is the exchange of hyaluronan with sulphated proteoglycans such as decorin, biglycan and versican produced by mature scar fibroblasts to provide the tissue with greater resilience.[104] In addition, collagen replenishment continues along with more complex interwoven molecular cross-linking until about 7 weeks after wound healing.[104] The outcome of this event is a proportion of activated fibroblasts transforming into myofibroblasts that pull normal dermal and adipose tissue into the wound defect.[105] As a result, a scar begins to form and wound contraction results. Although this remodeling is an important step in wound healing, it serves little purpose when trying to accelerate wound healing once re-epithelialization has occurred in the proliferative phase of healing. However, all four phases are necessary components in effective and efficient wound healing. Figure 14 illustrates the exchange of chemotactic and cellular communication that takes place during the wound healing process.

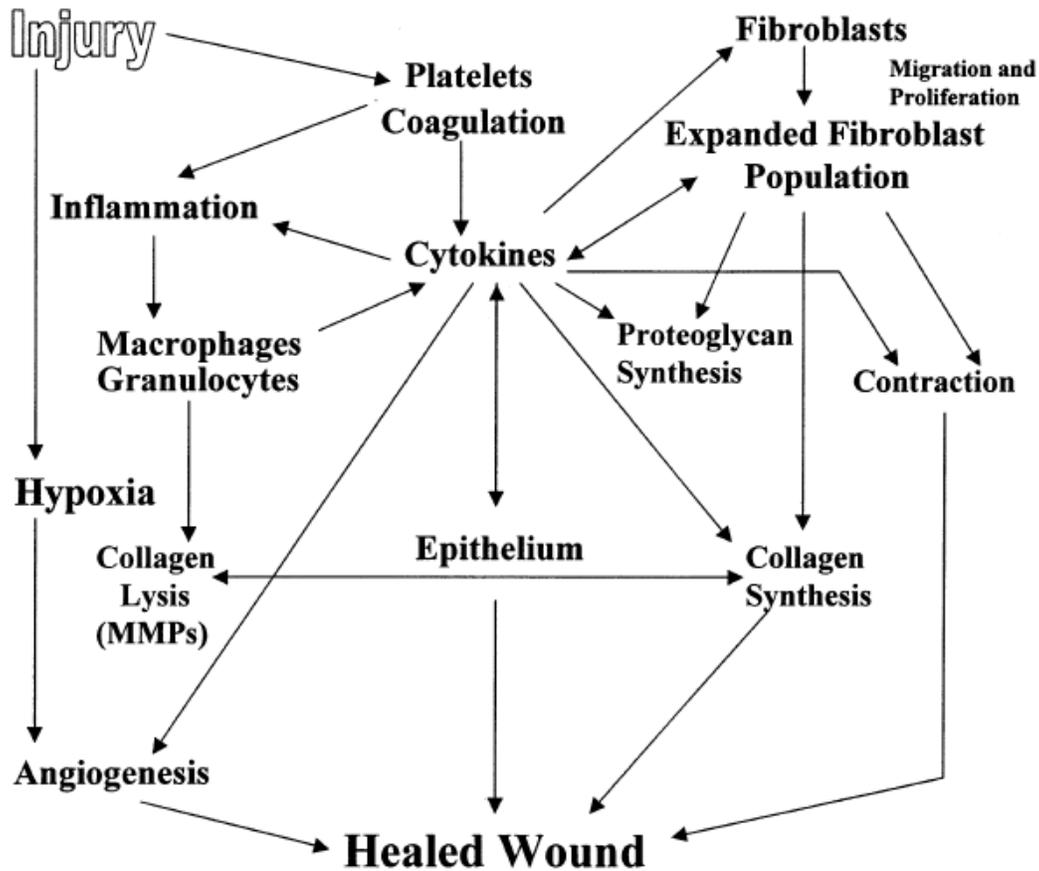


Figure 14. The complex wound healing cascade illustrating the importance of cytokines that drive the transition from one phase of wound healing to the other. The phases of hemostasis (coagulation), inflammation, migration, proliferation and remodeling are the necessary components for successful wound healing.[81]

2.8 Dressing Requirements

Skin wounds inflicted by mechanical injury, and burn wounds heal in an almost identical fashion to those observed in other large wounds of the skin.[7] One point of difference however, is that burn wounds lose large amounts of fluid through evaporation and exudation, compared to deep lacerations. Elevated fluid loss from the surface of the skin, and heat loss due to evaporation is increased leading to a drop in core body temperature. Dressings designed for burns must therefore be able to absorb fluid, control water vapour transfer rate at approximately

2-2.5 kg/m²/ 24 h, allow oxygen and carbon dioxide permeability, prevent bacterial growth and be easily applied and removed.[106] Table 4 illustrates the elements required for an effective wound dressing.[7]

Table 4. A list of transparent film wound dressing products.[7]

Absorbency	Ease of Application and Removal
Adherence	Elasticity
Bacterial Barrier	Gaseous Exchange
Comfort	Haemostatic
Comformability	Nonantigenic and Nontoxic
Drug/Peptide Delivery	Sterilizability
Durability	Water Vapour Transmissability

Dressings are applied to the wound surface to provide a protective barrier, relieve pain and accelerate wound healing.[7] In early times, dressings made of honey, fats, and plant fibres were used to prevent the entry of debris into the wound.[77,107] Although the primary function was to wick away moisture and prevent infection, these primitive dressings could hinder the wound healing process.[91] Modern dressings help to maintain a moist wound healing environment, proven to facilitate the diffusion of growth factors, and more effective cell-to-cell communication.[108]

Wound dressings come in a variety of materials including biomaterials and synthetic polymers woven into different shapes and sizes. Materials such as natural (cotton, chitosan, etc.) or synthetic polymers (polyurethanes, etc.) are commonly used because of cost-effectiveness and wide availability.[23] Ultimately, the purpose of the wound dressing is to offer the highest rate of healing, but also facilitate the best aesthetic repair of the wound. Therefore, the design of a wound dressing suitable for all types of wounds is challenging and may not be effective in all cases.

Table 5 lists the various dressing types organized based on compatibility with wounds releasing different levels of exudates. According to Seaman (2002), wound dressings may include: absorptive wound fillers, foams, collagens, gauze, hydrocolloids, hydrogels and transparent films.[109] Consequently, depending on the type of wound, an appropriate wound dressing must be selected.

Table 5. A classification chart matching exudate levels with different dressing examples.[109]

Wound Characteristic	Goal of Therapy	Dressing Examples
High Exudate	Exudate Absorption	Calcium Alginate (Wound Filler & Hydrogel) Foams (high moisture vapour transmission rate) Gauze
Low Exudate	Maintain Moisture	Hydrocolloids Foams (low moisture vapour transmission rate) Transparent Films
No Exudate	Add Moisture	Hydrogels Transparent Films Hydrocolloids

2.8.1 Calcium Alginate Dressings

Calcium alginate wound dressings have been used since the Second World War and have been recognized for excellent haemostatic properties and biocompatibility. [110][111]

Commonly, calcium alginates are combined with starch copolymers to make pads and packings with a variety of shapes and sizes.[112] Alginates are polysaccharides harvested from brown seaweed consisting of linear chains of (1,4)- β -D-mannuronic acid (M unit) and (1,3)- α -L-guluronic acid (G unit).[113] Alginates are block copolymers of M and G units organized in a manner such that divalent cations such as Ca^{2+} cooperatively bind between the blocks of G units in adjacent alginate chains.[114] The crosslinked alginate polymer branches have been referred to as the "eggbox model," because of the arrangement of the G units around the calcium ion.

Depending on the type of alginic acid used and the ratio of sodium to calcium in the dressing, the gelling and structural properties of the dressing can be altered.

Alginates also have several chemical and bioactive properties that make them useful in wound dressings. Alginates are generally well tolerated by the body and are broken down into simpler glucose type residues which later become fully absorbed.[115] This allows wounds to be completely covered, or packed by alginate without rejection by the body. In a porcine wound model, Barnett *et al.* (1987) observed that calcium alginate dressings could be left on a wound for up to 14 days with little ill effect.[116] Calcium alginates are therefore, well tolerated and are highly biocompatible as wound dressings.

Alginates also have several bioactive properties that promote wound healing. According to Thomas *et al.* (2000), alginates have the capacity to induce cytokine production by human monocytes, as well as activate macrophages at the site of the wound.[117] Furthermore, the slow release of calcium ions from the dressing are also used in the clotting cascade to provide superior hemostatic control in the wound.[118] Both of these features allow for control over severe bleeding, but also stimulate the inflammatory phase at an earlier opportunity to promote wound healing.

Alginate dressings also have the capacity to absorb up to 20 times the dressing weight in water and can conform to irregular surfaces.[119] The benefit of having such a large water absorptive capacity is that it minimizes pain and additional trauma stemming from over packing, or frequent dressing removal from heavily exudating wounds. Finally, according to Porter (1991), alginate dressings can be applied and left in place until it separates spontaneously from the wound which can lower the cost of constantly replacing medical resources.[118] However,

alginate dressings usually require a secondary outer barrier to hold the dressing in place. Table 6 lists some common, commercially available alginate wound dressings on the market.

Table 6. Alginate wound dressing products.[111]

Product	Company
Sorbsan™	Maersk Medical (UK)
Algosteril™	Laboratory Brothier (Fr)
Kaltostat™	Convatec (UK)
Urgosorb™	Urgo (Fr)
Tegagel™	3M (USA)
Tegagen HG™	3M (USA)
Curasorb™	Kendall Healthcare (USA)

Calcium alginate dressings can be manufactured using a variety of methods, forming electrospun mats and sponges, and can be combined with silk fibroin, gelatin, poly(ethylene glycol), poly(vinyl alcohol) (PVA) and chitosan to control physical and chemical properties.[120,121,122,123] By controlling the physical and chemical properties of alginate dressings, factors such as water uptake, gelling, and evaporation can be further attenuated to minimize the risk of wound drying. Alginates have also been used in the microencapsulation of insulin and functions as an effective drug delivery material for diabetes treatment, making it a highly amenable biomaterial.[124]

2.8.2 Foam Dressings

Foam dressings are categorized as an occlusive wound dressings, used on partial thickness dermal injuries, or as coverings for hydrogels, hydrophilic pastes (alginates), or powders.[125] They are available as pads of varying thickness, adhesive and non-adhesive forms, and can be laminated with a vapour barrier. According to Seaman (2002), moisture transmission rates can range between 0.8 to 5 kg/m²/day based on the thickness and chemical composition of the material.[109] This wide range of variations allows foam dressings to be

used in combination with ointments, creams and wound fillers that may synergistically promote healing.

Foams are typically fabricated from a wide variety of synthetic polymers including PVA, polyurethane or poly(siloxane).[126,127] Similar to calcium alginate dressings, foams have the ability to absorb large amounts of wound exudate due to a porous microstructure and polymer composition. Thus most foam wound dressings are reserved for wounds that are classified as high exudating because of a large capacity to absorb moisture, and not recommended for drier wounds.[128] Table 7 is a list of common foam dressings available on the market.

Table 7. Foam wound dressing products.[128]

Product	Company
Allevyn™	Smith & Nephew
Curafoam™	Tyco Kendall
PolyMem™	Ferris Mfg. Corp.
Flexan™	Bertek Pharmaceuticals
Lyof foam™	ConvaTec
Mepilex™	Medline Industries
Tielle Hydropolymer™	Johnson & Johnson

2.8.3 Gauze

Gauze dressings are one of the oldest and most commonly used wound dressings. It is a standard in wound care and can be used to wipe debris away from a wound, absorb blood and protect the wound during healing.[23] Gauze dressings are made from materials such as lint, cotton and cellulose, and are available in either woven, or non-woven formats.[129,130] Woven gauze is a fabric-like material composed of fibers spun into thread and woven into a ribbon or pad. According to Seaman (2002), woven gauze is predominantly used for packing wounds due to its lower water absorption capacity. This can be useful in injuries such as puncture wounds, or deep gashes that need to be packed with material to help stop bleeding.[23] One pitfall however,

is that woven gauze has a higher tendency to shed lint which can become entrapped within the wound bed and may impede wound healing.[131]

Non-woven gauze is made by a blending of materials instead of weaving.[132] This type of gauze is especially effective when used to help stop heavily exudating or bleeding wounds and does not shed as much lint. In addition, non-woven gauze can be combined with white petrolatum and other additives to help maintain moisture and control bacterial growth.

Although gauze dressings are widely available and effective first aid tools, gauze dressings can also be problematic. Gauze dressings require attention and maintenance to prevent a condition known as wet-to-dry, where the gauze dressing becomes saturated with exudate, or is pre-soaked in saline by a physician, applied to the wound and accidentally left to dry. The dry gauze adheres to the newly healed tissue, resulting in a difficult, destructive and painful removal of the gauze dressing.[133] Gauze wound dressings are not effective for moist wound treatment, but useful for first aid and wound preparation. Table 8 lists common gauze dressings available on the market.

Table 8. Gauze wound dressing products.[129]

Product	Company
Scarlet Red™	Kendall (USA)
Vaseline Gauze™	Kendall USA)
Xeroform™	Kendall (USA)
Telfa™	Kendall (USA)
Adaptic™	Johnson & Johnson (USA)

2.8.4 Hydrocolloids

Hydrocolloids are a type of adhesive wound dressing commonly used for the treatment of burns, and chronic or low exudating wounds. Hydrocolloids consist of materials such as

poly(isobutylene) sodium carboxymethylcellulose, gelatin and pectin, that swell upon contact with wound exudate.[134] Hydrophilic materials help to absorb and maintain a moist wound environment when placed over the wound. Often, hydrocolloids are processed into sheets and layered with a thin impermeable membrane to prevent water vapour loss and the entry of bacteria.[135] Table 9 lists commercially available hydrocolloid dressings.

Clinically, hydrocolloids are an attractive wound dressing because of low cost and ease of use. Hydrocolloids may be applied for up to 7 days without changing, thereby reducing the burden on healthcare providers and the use of resources.[136] Hydrocolloids are often described as comfortable to wear and can be removed easily with little trauma to the wound bed. The disadvantage in using hydrocolloid dressings is that frequent dressing changes are needed to prevent bacterial growth.[137] Overall, hydrocolloids are an effective wound dressing for chronic wounds, but should be avoided for heavy exudating wounds.

Table 9. Hydrocolloid wound dressing products.[109]

Product	Company
Tegasorb™	3M
Comfeel™	Coloplast Corp.
DuoDerm™	ConvaTec
RepliCare™	Smith & Nephew
Restore™	Hollister Inc.

2.8.5 Hydrogels

Hydrogels are water-based products that contain 60-90% water and are used in a variety of tissue engineering, wound healing and biomaterials applications.[138] Hydrogels are typically used on wounds that are very dry, or have low levels of exudate whereby the dressing can keep the wound environment moist.[81] Due to the moistening effect, the dressings are non-adherent and are comfortable to apply and remove.[139] Hydrogels are made from natural and

synthetic polymers such as dextran, collagen, chitosan, Pluronic F-127, poly(vinyl alcohol) (PVA) and poly(ethylene glycol) (PEG).[140,141,142] According to Drury (2003), synthetic hydrogels have the ability to be easily reproduced with a high level of control over chemical properties such as molecular weight, block structure, crosslinking density and degradable linkages. These properties influence the gelling characteristics that can be adjusted depending on the desired properties. Natural polymers on the other hand, more closely mimic the chemical and physical properties of native extracellular matrix components. They can be used in a natural form, or chemically modified to become semi-synthetic and have added properties such as photocrosslinkability. Hydrogels therefore, can be easily attenuated to suit the needs of the wound healing application and offer a wide range of flexibility. Table 10 lists typical hydrogel wound dressings which are available on the market.[139]

Linear PEG hydrogels are of particular interest in wound dressings for several reasons. PEGs are generally non-toxic, non-immunogenic, do not harm active proteins or cells, and are easily cleared by the body.[143] Hydrophilicity of PEG also helps to preserve the bioactivity of hydrophilic biomacromolecules and depending on the molecular weight and content, can attenuate drug or peptide release kinetics.[144] In this case, incorporation into alginate dressings would allow for control over drug delivery release kinetics, without affecting the biocompatibility of the dressing.

Table 10. Hydrogel wound dressing products.[139]

Product	Company
Aquaform	Robert Bailey
Granugel	Convatec
Intrasite Gel	Smith & Nephew
Nu-Gel	Johnson & Johnson
Purilon Gel	Coloplast
Vigilon PEG Hydrogel	Bard Inc.

2.8.6 Transparent Films

Transparent films are thin, clear polyurethane dressings with an adhesive layer that sticks to the skin, but not to a moist wound.[145] Transparent films function by covering superficial wounds, or by acting as secondary dressings to protect wounds which have been packed with absorptive fillers. Thinness and conformability lead to a dressing which is unable to absorb large amounts of exudate, but permit the passage of a limited amount of water vapour and exudate.[146] Typically, films will have water vapour transfer rates of 0.3 to 0.8 kg/m²/24 h.[147,148] As a result of this low permeability, films are more resistant to permitting bacteria and other pathogens from entering the wound. Flexibility and thinness also allows for easy conformability around a variety of shapes which makes for easy application to uneven body surfaces, or around I.V. lines, or drainage tubes.

Films can also be left on for extended periods of time ranging between 5 to 7 days allowing the underlying wound to heal without being disturbed. [147] On the other hand, films have limited capacity to absorb exudate and if used improperly, can cause underlying fluid to pool leading to maceration. Table 11 lists several different types of films available in the clinic. For superficial wounds or as secondary dressings, films are a primary choice, but not in the healing of severe exudating wounds.

Table 11. Transparent film wound dressing products.[213]

Product	Company
Bioclusive™	Johnson & Johnson (USA)
Opsite IV3000™	Smith & Nephew (UK)
Tegaderm™	3M (USA)

2.8.7 Biologicals

Biologicals are a class of recombinant medicines including monoclonal antibodies, nucleic acids, and small or large molecular weight proteins.[149,150,151] Approximately 150 recombinant biopharmaceuticals have been approved by the Food and Drug Administration (FDA) and the list of submissions and approvals continues to grow. [152] Recombinant peptides are an increasingly important therapeutic intervention against a variety of medical conditions including diabetic, oncologic, cardiovascular, immunosuppressive and gastroenterological diseases.[153,154,155,156,157]

Growth factors are polypeptides with the capability of stimulating, or inhibiting cellular division, differentiation, migration or gene expression, depending on the cells involved.[158] Growth factors can act in an autocrine, paracrine or endocrine fashion depending on the target receptor.[159] Once released, growth factors will either bind to their respective receptor, or become consumed by proteolytic enzymes inactivating the polypeptide. The action of growth factors is therefore strong and usually short-lived.

Biomedical researchers have examined the potential use of growth factors, especially in the field of wound healing.[160] Exhaustively investigated growth factors have included epidermal growth factor (EGF),[161] transforming growth factor beta (TGF- β),[162] and platelet derived growth factor (PDGF).[163] As a result, growth factors are one method of ameliorating the wound healing process by providing an earlier onset of the body's own regenerative mechanisms.

Unfortunately, the high cost of producing highly purified growth factors has prevented their application within wound dressings. Beginning in the 1970's, growth factors were first harvested by processing tissue and/or blood samples from animals and humans.[164] Tissues

would undergo mechanical and chemical treatments to yield small quantities of semi-pure growth factors. This approach was laborious and ineffective in sustaining large scale production. As molecular techniques advanced, recombinant technologies utilizing bacteria and yeast made it easier and safer to increase production.[165] As a result, the mean cost of growth factors such as EGF, TGF- β and other highly specialized growth factors can be upwards of \$1,500 - 10,000 USD per mg.[166] Growth factor application in wound healing technologies has therefore been confined largely to experimental settings, and not commercial markets.

2.8.7.1 Epidermal Growth Factor

Epidermal growth factor (EGF) is a peptide and biological responsible for regulating cell proliferation and differentiation in animals and humans. Structurally, EGF is a peptide with a molecular weight of 5400 Da and contains a primary amino acid sequence of 53 amino acid residues.[167] Within the peptide chain are six strategically placed cysteine residues forming three disulfide bonds upon protein folding.[168] EGF binds to the EGF receptor (EGFR), a tyrosine kinase transmembrane receptor found on the surface of keratinocytes, throughout the basal layer of the epidermis.[169] Once two EGF molecules bind to the receptor, the EGFR dimerizes resulting in autophosphorylation triggering a downstream signaling cascade.[12]

According to Barrientos *et al.* (2008), EGF has been one of the most heavily studied growth factors in the field of wound healing, due to its important role in re-epithelialization of acute wounds. Originally, Cohen (1962) reported isolating a polypeptide from the submaxillary gland of male mice that accelerated eyelid opening and incisor eruption in newborn animals. Subsequent testing on mice showed that proliferation and keratinization of the epidermis was possible and that faster growth of chick embryos *in vitro*, could be achieved.[170] EGF was later report to be secreted by platelets, macrophages and fibroblasts in order to stimulate keratinocytes

during the wound healing process.[12] Hardwicke *et al.*, 2008, reported that EGF stimulates fibronectin synthesis, angiogenesis, fibroplasia, and collagenase activity. These functions help re-epithelialize the wound after the initial clot has formed.[11] Franklin and Lynch (1979) tested the effect of topical EGF application on wounds inflicted to the anterior surface of the rabbit ear down to the cartilage. Every 12 hours, a topical ointment of 0.83 µg/mL mouse EGF was applied for up to 20 days, along with a blank topical ointment for comparison. After 20 days, the EGF-treated group had wounds which had healed with a thicker and denser epithelium, along with less wound contracture versus control. Wound healing was faster and resulted in a more naturally appearing scar than the untreated group.[171]

In vivo studies performed by Nanney (1990) compared the healing rates associated with topical EGF treatment and untreated dermatomed pigs. Using a semi-occlusive bandage in conjunction with an application 10 µg/mL/day was optimal in achieving re-epithelialized skin within 3 days versus 4 days in the control group.[172] Tanaka *et al.*, (2005) examined the *in vivo* efficiency of gelatin sheets loaded with 10 mg/mL human recombinant EGF on the repair of partial-thickness wounds on the dorsal side of hairless dogs. The sheets were replaced daily for 14 days and resulted in a significant reduction in the wound area by days 9 and 12 versus the blank control group. By day 9, more than 50% of the wound area had been repaired, as opposed to only approximately 20% in the control.[161] EGF is therefore a potent mitogen of keratinocyte migration and proliferation.

2.8.7.2 Transforming Growth Factor Beta

Transforming growth factor beta (TGF-β) is a polypeptide implicated in several aspects of wound healing including cellular proliferation, differentiation and matrix metabolism.[158] Human TGF-β is a dimer polypeptide with a molecular weight of 25 kDa and yields two

identical chains of 122 amino acid residues in length.[173] The two chains are held together with a disulfide bond required for ligand-TGF- β receptor (TGFR) binding to take place successfully.[173] TGF- β is secreted by platelets, activated macrophages, T-lymphocytes, endothelial cells and fibroblasts where active wound repair is present.[162] The TGFR is part of the cell surface, transmembrane, serine-threonine kinase receptor family and is found on nearly all cells including epithelial, mesenchymal, and lymphocytic.[12] Activation of the TGFR results in a downstream signaling cascade surmounting in anchorage-independent growth and proliferation of fibroblasts, as well as inhibiting the proliferation of mitogen stimulated B and T cells, and keratinocytes.[12,174] The activation of fibroblasts enables matrix formation to take place with the deposition of type I and III collagen, which when uncontrolled, can lead to significant fibrosis and scar tissue formation.[158,175] As a result, the release of TGF- β occurs in a highly controlled and timely manner during the wound healing process.

TGF- β has been used in wound healing applications and has demonstrated some beneficial effects. Puolakkainen *et al.*, (1995) reported the use of TGF- β incorporated into 4 drug delivery vehicles including phosphate-buffered saline, polyoxamer gel, DuoDerm hydroactive paste and a poly(ethylene oxide) gel. The various topicals were applied to full-thickness wounds in rats and wound healing measurements were recorded. A single TGF- β dose of 1 $\mu\text{g}/\text{wound}$ was administered and resulted in enhanced wound healing between days 2-5 when compared to the blank vehicle controls.[162] Beyond day 7, the blank controls and TGF- β treatment groups did not differ in wound closure rate. Pandit *et al.*, (1999) examined the wound healing response of rabbits with full-thickness wounds treated with collagen scaffolds loaded with 2 $\mu\text{g}/\text{cm}^2$ TGF- β . All wounds were covered with a secondary polyurethane dressing monitored at 1, 2 and 3 week intervals.[176] By week 3, nearly $95 \pm 47\%$, $51 \pm 14\%$ and $44 \pm$

25% wound re-epithelialization was recovered in the TGF- β , control and collagen only study groups respectively. No significant differences were reported in mechanical strength of repaired tissue. Overall, TGF- β delivery did help to enhance wound healing, but showed little difference in mechanical strength. Brown and colleagues (1988) studied the effects of a single dose of TGF- β and EGF delivered separately by a single injection into rats given a single incision. The wound tissue was harvested 10 days later, revealing that EGF caused wound tensile strength to increase by 200%, whereas TGF- β increased wound tensile strength by only 50% when compared to controls.[177] TGF- β showed strong potential in enhancing wound healing, but only if delivered immediately after injury.

2.8.7.3 Platelet Derived Growth Factor

Platelet derived growth factor (PDGF) is a polypeptide released from platelet granules upon injury. It belongs to a family of homo and heterodimeric growth factors and is produced primarily by platelets, but in smaller quantities by macrophages, vascular endothelium, fibroblasts, and keratinocytes.[12] The polypeptide consists of an A and B chain held together with two disulfide bonds, resulting in a molecular weight of 30 kDa.[158] When platelets are exposed to subendothelial basement or collagen, the platelets degranulate causing PDGF to be released at the site of the injury.[178] PDGF then binds to the PDGF receptor (PGFR), a transmembrane, tyrosine kinase receptor that autophosphorylates upon activation, located on fibroblasts, and lymphocytes.[179] The elevated levels of PDGF cause a series of chain reactions beginning with the attraction of neutrophils, macrophages and fibroblasts to the site of the wound. Once activated, the macrophages and fibroblasts begin to proliferate, while also secreting fibronectin, hyaluronin and collagen at the injury site.[180] PDGF is therefore a

powerful stimulator of fibroblasts which enable the replacement and production of lost ECM components.

Delivery of PDGF exogenously through wound dressing devices has been explored as one method of ameliorating the wound healing process. Li and colleagues (2009) reported the effects of using a PDGF loaded two-component polyurethane scaffold to heal rat skin inflicted with an excisional wound. A range of loadings included 0-11.5 $\mu\text{g/mL}$ of scaffold were tested, along with blank scaffold controls. By the 14th day post injury, PDGF treated animals demonstrated nearly 100% re-epithelialization, while the blank control group achieved only 70% tissue recovery.[181] Jin *et al.*, (2008) tested PDGF loaded poly(lactic-co-glycolic acid) (PLGA) microparticles incorporated into nanofibrous scaffolds as a strategy to repair incisions made on the dorsal regions of rats. Loadings ranging between 2.5-25 $\mu\text{g/wound}$ were implanted, and monitored over a period of 21 days. Treatment groups showed increased angiogenesis and tissue neogenesis versus the blank control groups.[182] Sun *et al.*, (2007) used collagen membranes loaded with PDGF to accelerate wound healing in rabbit dermal ischemic ulcer models. Using electrospinning techniques, collagen was spun to form sheets loaded with 0.64 nM PDGF resulting in a mean re-epithelialization of approximately 55%, versus 30% in the blank control after 14 days. PDGF delivery is therefore one method of accelerating wound healing of cutaneous wounds.

2.8.7.4 Additional Growth Factors

Wound healing includes additional growth factors, released during different periods of the wound healing process. Many of the growth factors listed in Table 12 have been tested in a laboratory setting, with few making it to the clinic.

Table 12. Growth factors released during wound healing, along with the target cell types and cost of obtaining pure growth factor.

Growth Factor	Cells Involved	Function	Format	Cost CAD/mg
EGF	Platelets Macrophages Fibroblasts	Re-epithelialization	Lyophilized Powder	\$987[183]
TGF- β	Platelets Keratinocytes Macrophages Lymphocytes Fibroblasts	Inflammation Granulation tissue formation Re-epithelialization Matrix formation	Lyophilized Powder	\$9800[184]
PDGF	Platelets Keratinocytes Macrophages Endothelial Cells Fibroblasts	Inflammation Granulation tissue formation Matrix formation and remodeling	Lyophilized Powder	\$4300[185]
FGF-2 (Fibroblast Growth Factor-2)	Keratinocytes Mast Cells Fibroblasts Endothelial Cells Smooth Muscle Cells Chondrocytes	Granulation tissue formation Re-epithelialization Matrix formation and remodeling	Lyophilized Powder	\$5700[13]
VEGF (Vascular Endothelial Growth Factor)	Platelets Neutrophils Macrophages Endothelial Cells Smooth Muscle Cells Fibroblasts	Granulation tissue formation	Lyophilized Powder	\$4300[186]
IL-1 (Interleukin-1)	Neutrophils Monocytes Macrophages Keratinocytes	Inflammation Re-epithelialization	Lyophilized Powder	\$5200[187]
IL-6 (Interleukin-6)	Neutrophils Macrophages	Inflammation Re-epithelialization	Lyophilized Powder [188]	\$3000[188]
TNF- α (Tumour Necrosis Factor- α)	Neutrophils Macrophages	Inflammation Re-epithelialization	Lyophilized Powder	\$1500[189]
Insulin	Keratinocytes Fibroblasts Endothelial Cells	Re-epithelialization Angiogenesis	Crystalline Powder	\$1[190]

2.9 Insulin and Wound Healing

Insulin is a peptide and growth factor with several physiological roles. It is primarily known as a hormone that regulates circulating blood glucose levels within the body. Over the past 100 years however, insulin has played a quieter role in the field of wound healing. As an effective and affordable peptide, which can help restore the integrity of the skin, insulin is becoming an increasingly useful peptide of interest in the field of wound care.

2.9.1 Discovery of Insulin

The discovery of insulin was one of the greatest achievements of the 20th century. Prior to the early 1900's, medicine offered little relief to patients who were diagnosed with juvenile diabetes, and little was known about the physiology and function of the pancreas. In almost all cases, juvenile diabetics resulted in poor health outcomes and typically underwent one of two approaches. The first treatment included a starvation diet developed by Frederick Allen, which limited the patient's intake of carbohydrates and reduced the incidence of glycosuria.[191] The second and more morbid therapeutic option was to provide no medical intervention.[192] In both cases, young patients would often suffer and die from cachexia, or diabetic ketoacidosis resulting from impaired metabolic regulation.[193]

Frederick Banting and Charles Best successfully isolated insulin in 1921 from processed porcine pancreases.[194] Systemic administration of insulin isolates demonstrated successful lowering of blood glucose levels in dogs, and signified the introduction of a new therapy for juvenile diabetes.

Soon after the discovery of insulin, stability issues became apparent. Without access to fresh insulin, the peptide would degrade and fail in controlling serum glucose levels. In an attempt to correct this problem, John Abel developed a method in 1926, of co-crystalizing insulin

with zinc salts to form a stable hexamer.[195] Jørgen Schlichtkrull refined the formulation conditions, and identified a reliable and reproducible method of crystallizing insulin. This was an important step because a crystalline form enabled longer periods of storage without loss of bioactivity. Schlichtkrull also discovered that hexameric insulin favourably complexed with either two, or four zinc atoms stabilizing insulin for extended periods,[193] allowing Fred Sanfer to determine the primary amino acid sequence of the peptide nearly three decades later in 1955,[196] and Don Steiner to identify the biosynthetic pathway in 1967, making it subsequently possible to partially synthesize insulin. Steiner discovered that insulin is first synthesized as a proinsulin precursor, requiring post-translational modification to become fully active.[197] Two years later, Dorothy Crowfoot Hodgkin and colleagues described the three-dimensional structure of insulin using X-ray crystallography, allowing for structure and function relationships to be completed.[198] These discoveries culminated with the introduction of recombinant technologies, which permitted industrial-scale production, resulting in reduced cost and increased clinical availability.[199] Today, insulin is prepared in a number of highly stable and pure forms for about \$1 CAD/mg.

2.9.2 Biosynthesis and Structure

Insulin is one of the most highly conserved peptides found in vertebrates and generally differs by 1-3 amino acids between species.[200] Naturally, insulin is biosynthesized by β -cells of the islets of Langerhans, clustered within the pancreas. The islets are supported by a highly vascularized and densely innervated network that provides accurate sensing, and control of blood glucose levels.[201] Normally, the pancreas is capable of producing up to 200 U (8mg) insulin per day, of which only 50 U (2mg) is usually secreted.[202] Within the islets, insulin is synthesized as a single-chain precursor, known as preproinsulin,[203] comprised of several

segments including: an N-terminal signal sequence of 24 amino acids, followed by the insulin B-chain, an Arg-Arg sequence, a connecting C-peptide of 31 amino acids, a Lys-Arg sequence and the insulin A-chain.[201] These signalling sequences are necessary to direct the secretion of the preproinsulin through the endoplasmic reticulum and Golgi bodies so that further post-translational modifications, such as the cleavage of the C-peptide, can take place.[204,205] Figure 15 depicts the biosynthesis of preproinsulin, modification and release of insulin, while Figure 16 illustrates the structure of proinsulin.

Post-translationally modified insulin is approximately 5800 Da and consists of an A-chain with 21 and a B-chain with 30 amino acid residues, held together by 2 disulfide bonds. Figure 17 shows the structure of proinsulin undergoing proteolytic cleavage of the C-chain. After modification, insulin complexes with zinc and forms crystalline granules within β -cells of the pancreas and becomes sequestered until the time of release.[206] The crystallization process is a key facet in ensuring that insulin can be preserved for extended periods of time within the body, but released when needed.

Once released into systemic circulation, insulin is degraded by four different mechanisms, including degradation by the insulin-degrading enzymes (IDE), and by being bound and internalized by the insulin receptor (IR).[207,208,209,210] These mechanisms ensure that prolonged insulin stimulation can be avoided and tight regulatory control enabled.

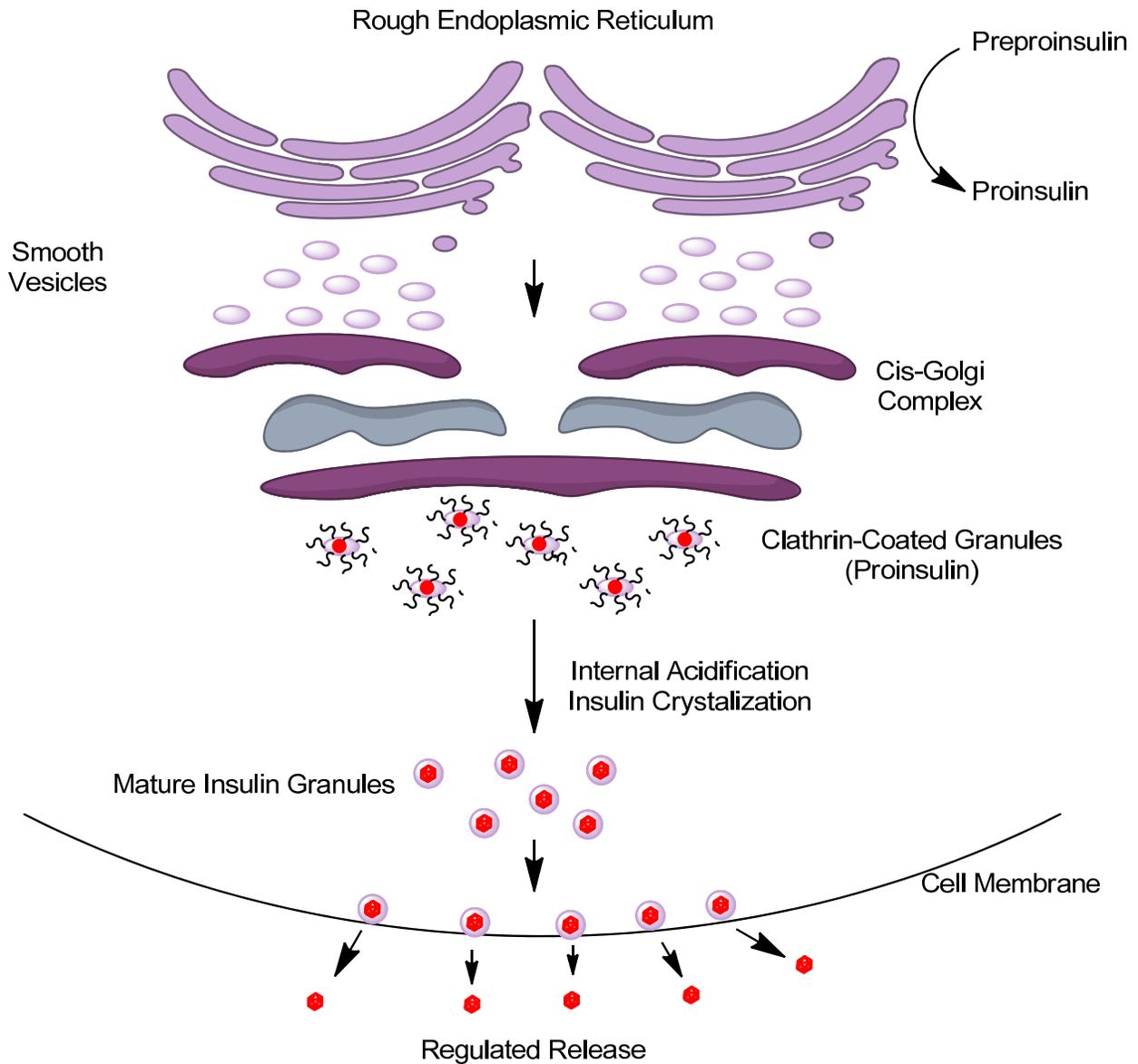


Figure 15. The biosynthetic pathway of insulin in its initial, unmodified form as preproinsulin, through post-translational modification and release of mature insulin within the pancreatic β cells.[204]

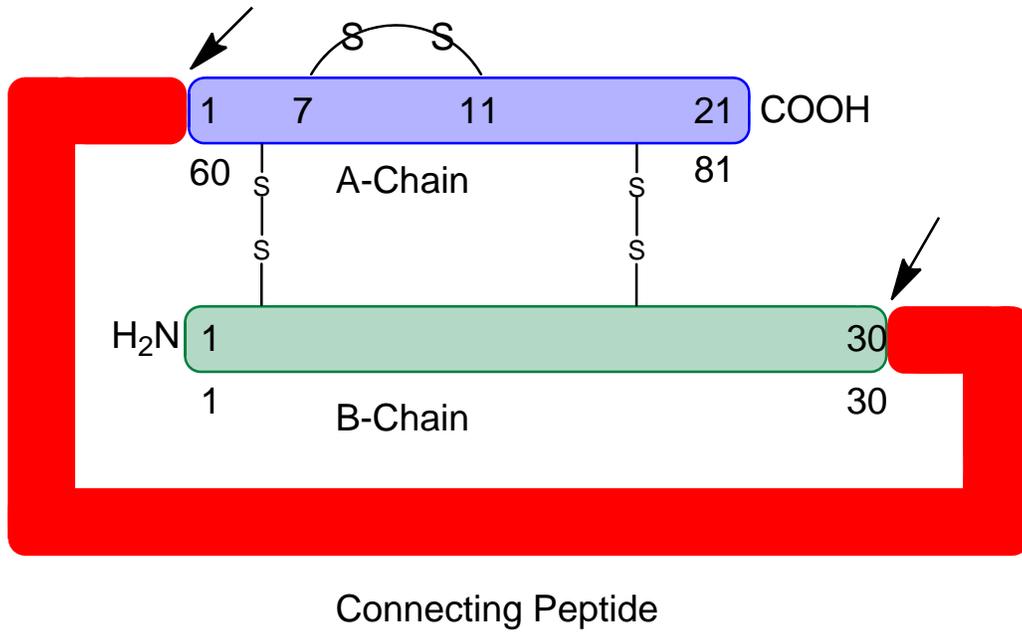


Figure 16. Proinsulin prior to having the connecting peptide (C-peptide) proteolytically cleaved. The arrows indicate the cleavage sites, while -S-S- indicate the presence of disulphide bonds.[202]

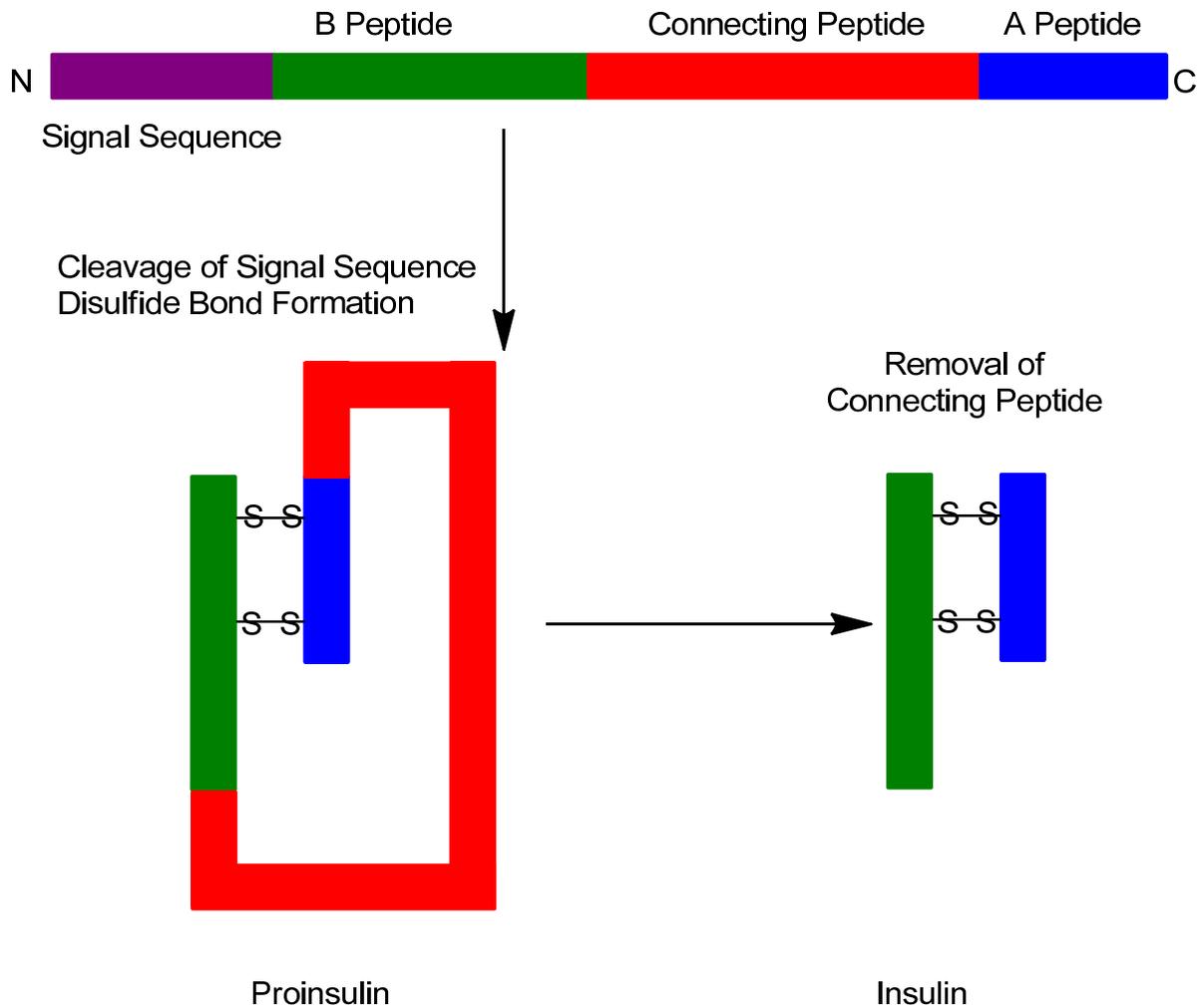


Figure 17. The proteolytic processing of preproinsulin to proinsulin, then to insulin. Note the three key steps in insulin’s biosynthetic pathway, 1) cleavage of the signal sequence, 2) formation of the disulphide bond and 3) the removal of the connecting (C-peptide) polypeptide.[205]

2.9.3 Recombinant Synthesis and Crystallization

Industrially, insulin is produced using recombinant DNA technologies with genetically modified yeast and bacteria.[211] A recombinant approach alleviates the need for animal sourced organs, safeguarding against the transmission of animal-to-human pathogens, and is a cost-effective approach in large-scale production.[212] Purified insulin is crystallized at pH 5.8-6.0, by adding zinc salts.[213] Insulin also complexes and precipitates with cobalt and cadmium

salts, however these preparations are not intended for clinical use.[214] The crystallization process is therefore a key step in stabilizing the peptide.

Crystallization of insulin also retards the dissolution rate under physiological conditions. For example, four-zinc hexameric insulin dissolves slower than non-complexed monomeric insulin, thereby allowing diabetic patients to maintain near constant serum levels of injected insulin for prolonged periods.[215] Regulation of blood sugar levels is then enabled through sleep periods by avoiding multiple injections. The rate limiting step in the dissolution mechanism is liberation of the zinc atom, which in turn releases insulin monomers that then bind with the insulin receptor (IR).[213]

2.9.4 Insulin Receptor

The insulin receptor (IR) is amongst the most ubiquitous cell surface receptors in the body. The IR belongs to a family of receptor tyrosine kinase (RTK) transmembrane signaling proteins, which include the insulin-like growth factor-1 (IGF-1), EGF and PDGF receptors.[216] Insulin receptors are found on myocytes, erythrocytes, adipocytes, granulocytes, lymphocytes and in brain tissue at nearly 40 to 300×10^3 binding sites per cell.[217,218,219,220,221] IR binding sites have also been identified on both human skin fibroblasts and keratinocytes in separate high and low affinity binding populations. [222,223] Fibroblasts have approximately 1,100 high affinity and 30,000 low affinity insulin binding sites per cell.[224] Keratinocytes have nearly 6,000 high affinity and 80,000 low affinity binding sites per cell. Specific to keratinocytes, the IR is responsible for inducing the expression of keratins 1 and 10 during calcium-induced differentiation and conversely inhibits the activation of the IGF-1 receptor which has an inhibitory effect.[225] The presence of high and low affinity binding sites permits

IR to respond differently to variable insulin concentrations in the blood stream.[226] Insulin receptors are a key mediator of signal transduction in homeostasis and tissue repair.

Structurally, the receptor is initially synthesized as a proreceptor and cleaved to form a 350-400 kDa heterotetrameric receptor composed of two α -subunits (135 kDa each) and two β -subunits (95 kDa each) joined by disulphide bonds in a β - α - α - β configuration (Figure 18).[227] The proreceptor is approximately 1340-1382 residues in length and can be spliced in more than one way.[228,229] The α -subunits are located near the N-terminal domain and constitute the extracellular[230], whereas the β -subunit contains a transmembrane domain, an intracellular region which contains the tyrosine phosphokinase, ATP-binding site and a tyrosine autophosphorylation site.[231] Other key regions of the IR include a leucine-rich repeat domain, a cysteine-rich region, a second leucine-rich repeat domain, and three fibronectin type III domains which help to maintain the conformational shape of the receptor.[232]

During receptor activation, insulin binds to one of two binding sites on each of the $\alpha\beta$ monomer called, site 1 and site 2.[233] When insulin first binds to site 1, a high-affinity cross-linking reaction takes place, allowing 2 to become accessible for ligand binding to activate the receptor.[234,235] This binding regime means that IR demonstrates a negative co-operativity, meaning that as one ligand binds to the receptor, the IR affinity for a second ligand will decrease (Figure 19). Once bound, the α -subunit which normally prevents autophosphorylation of the β -subunit, is inhibited causing the kinase activity of the β -subunit to transphosphorylate, change conformation and increase its own kinase activity.[236] Interestingly, IR and IGF-1 receptors form functional hybrids which help elucidate why IGF-1 and insulin can act on each other's respective receptors.[236]

After insulin binds to the IR, specific biochemical events take place downstream of the binding domain. Specifically, tyrosine residues in the β -subunit become phosphorylated, thus forming docking sites for receptor substrates found within the cytosol which contain SH2 (Src-homology-2) domains.[237,238,239] Insulin receptor substrates (IRS) are then permitted to interact with the docking sites and result in a wide array of downstream signalling cascades, including activation of PI(3)-kinase and downstream PtdIns(3,4,5)P₃-dependent protein kinases, ras and the MAP kinase cascade, and Cbl/CAP, glucose transporter (GLUT4) and the activation of TC10.[237,240,241] Other tyrosines found within the kinase domain begin to regulate the catalytic activity of the β -subunit.[238] The culmination activating IRS results in coordinated regulation of vesicle trafficking, protein synthesis, enzyme activation and inactivation, and gene expression, aspects which are necessary for normal metabolism and tissue repair.

Soon after activation, the IR is quickly inactivated through two major pathways. The first includes internalization of the receptor which becomes sequestered from the plasma membrane and concentrated into heteratogeneous, non-lysosomal tubulovesicular structures called endosomal apparatuses.[242] Once within the endosomal apparatus, the IR undergoes receptor sorting, recycling and degradation.[242] The second method includes protein tyrosine phosphatases that desphosphorylate the receptor and its substrates, resulting in receptor inactivation.[243] IR inactivation can be variable depending on the nature of the stimulus.

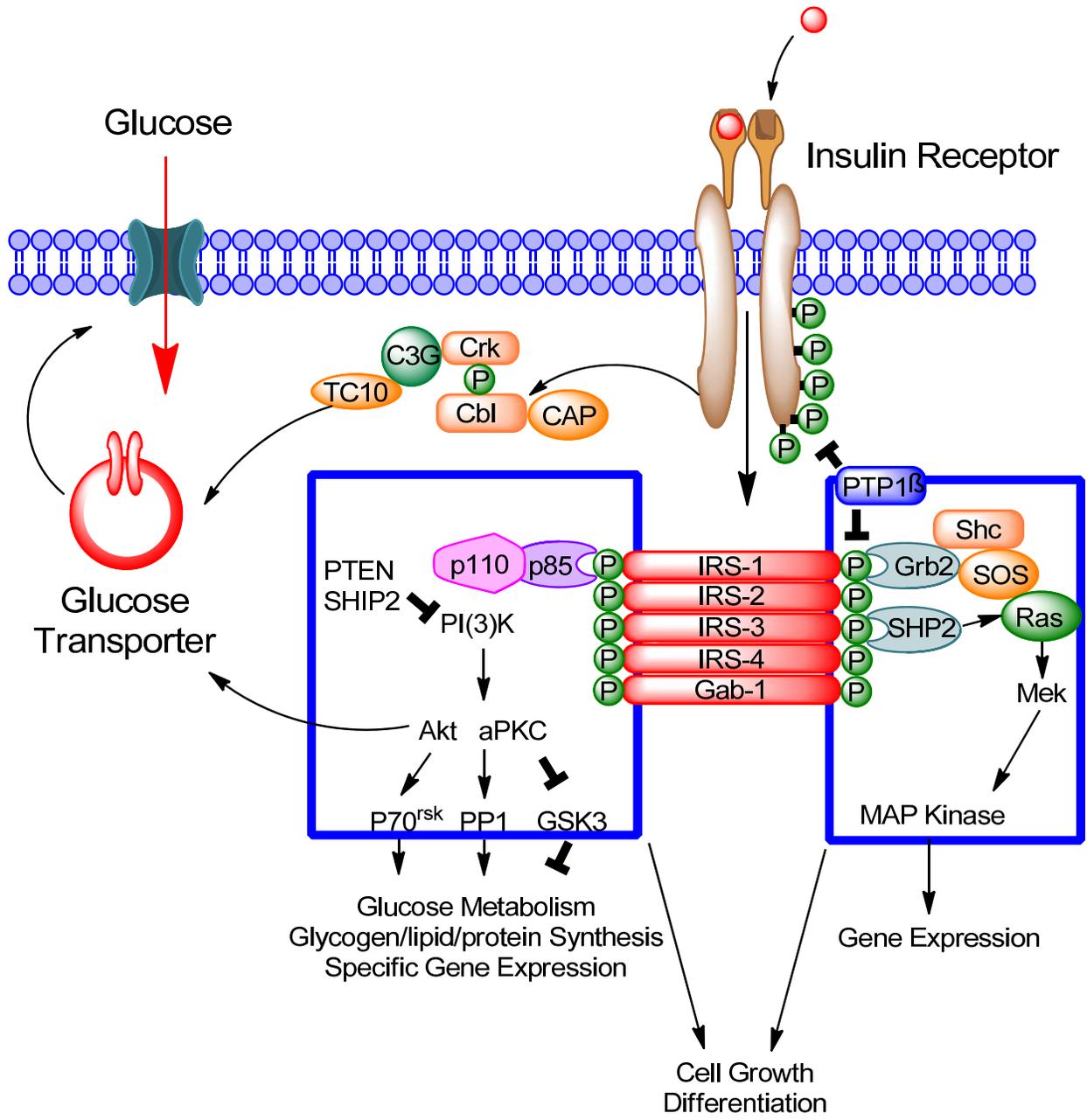


Figure 18. An illustration of the insulin receptor situated on the surface of the plasma membrane.[237]

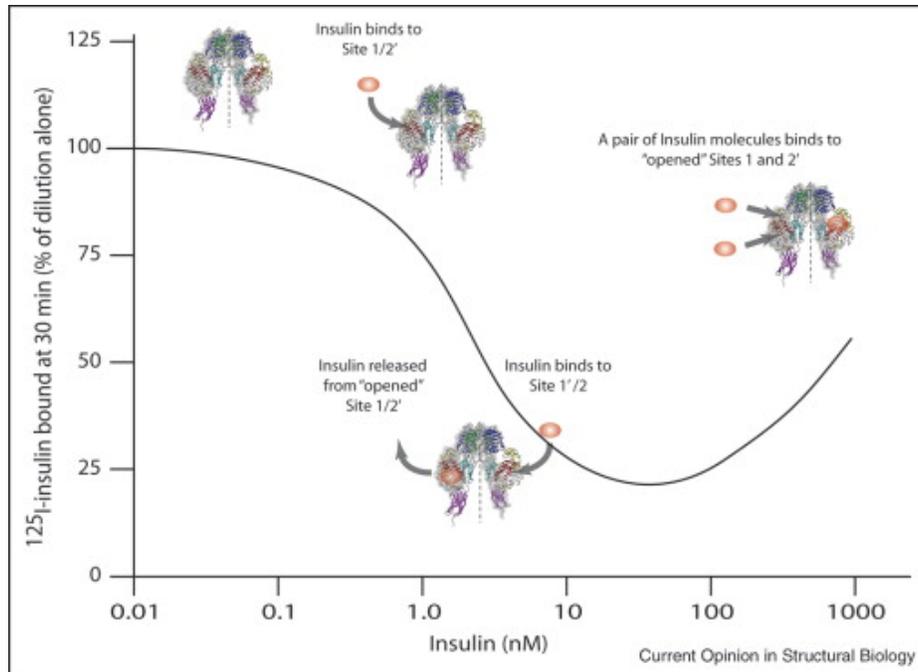


Figure 19. A plot of insulin concentration over time, along with the proposed structural changes which occur as insulin binds with Sites 1 & 2.[232]

2.9.5 Beginning of Insulin-Wound Healing

Insulin's role in wound healing can be traced back to the early 20th century to surgeons who observed differences in healing rates amongst diabetics and non-diabetics recovering from surgery.[244] In diabetics, wounds would fail to re-epithelialize normally, exposing raw flesh to open environment, resulting in infections. As the infection spread to the blood, death would soon follow. After the discovery of insulin in 1921, William Thalhimer was the first to report the application of insulin to treat postoperative (nondiabetic) acidosis. At the time, Thalhimer had been monitoring three patients who had undergone surgery and were suffering from severe vomiting and post-operative metabolic imbalance. Thalhimer reported that there was strong evidence linking acidosis with the development of severe infection and death. After injecting three patients with glucose and 10 units of insulin, the symptoms of acidosis were cleared, and the metabolic balance restored.[245] Improved recovery for the patients was evidence for the

role of insulin regulating tissue repair. Nellis Foster in 1925 was also attempting to treat similar cases of diabetic acidosis by administering 30-75 units of insulin.[246] The mortality rate from infection in diabetics was reduced from 40 to 12% in those patients receiving insulin therapy intravenously amongst 20 patients. The Thalimer and Foster observations provide early indications of insulin's ability to improve wound healing, and thus enhance postoperative health in diabetics.

Schazzilo and Ksendowsky used radiographic and histological techniques in 1928 to investigate the effects of insulin on bone fracture repair. Using a rodent animal model, 1 unit of crystalline insulin was injected into test animals, resulting in faster healing compared to blank control specimens of 5 animals each.[247] Soon after, Barnet Joseph (1930) used insulin therapy to treat cardiac patients with non-diabetic bed sores. Joseph administered 10 units of insulin once daily to 5 patients, and observed improvements to the wound of 50 to nearly 100% after 14 days of therapy.[248] Rabinowitch (1932) later studied 250 consecutive surgical cases of gallbladder disease, which included a subset of 50 diabetic patients. Rates of death in both diabetic (4%) and non-diabetic (5.5%) patient pools showed that insulin had reduced the risk of post-operative wound infection to a level comparable with normal patient recovery.[249] Walter Stuck (1932) examined the effect of exogenous administration of 1.5 units insulin intravenously to promote faster bone healing in rabbits with fractured left fibulas. Faster calcification of the callus in the presence of insulin was reported between 14 and 28 days versus the control group, and while bone morphology between the insulin and control groups did not differ, those treated with insulin had healed faster.[250]

2.9.6 Insulin-Wound Healing Research: 1940s-1980s

Much of the research during the early part of this period was directed toward bone fracture healing before refocusing on soft tissue injuries. Lawrence *et al.* (1954) studied the effects of insulin on nitrogen balance in hypophysectomized rats, postulating that insulin was responsible for nitrogen retention related to the uptake of amino acids, synthesis of proteins and inhibition of protein catabolism. These three factors are important to tissue repair following damage or injury. The study compared hypophysectomized rats treated with, and without insulin, along with non-hypophysectomized rats. Hypophysectomized rats receiving insulin responded positively to the insulin treatment and demonstrated a linear relationship in stimulating protein synthesis with dose.[251] Lawrence concluded that insulin could therefore be a potent growth factor in the absence of pituitary signaling molecules, and potentially in tissue repair. Later, Stunkle and Wray (1965) examined the effect of long-acting insulin on the rate and quality of recovery from fractured tibias in non-diabetic and diabetic rats. This was one of the first studies to utilize long-acting insulin, contained two arginine amino acids added to the C-terminal of the B chain and a substitution of asparagine 21 on the A chain with glycine.[252] The principle was based on lowering the isoelectric point to pH 4, so that upon injection, insulin would remain crystalline under normal physiological pH and dissolve more slowly with time. The study concluded that there was no significant difference in tensile strength of repaired tibias between treated and control animals.[253] Unfortunately, the study did not present findings related to the time to heal in control and treatment groups. In the same year, Gregory (1965) tested the effect of insulin on bone healing, by performing a simulated bone fracture on male albino rats. A hole was drilled through the lateral aspect of the femur with a dental bur, cutting into the marrow. Following the injury, 40 units/kg of protamine-zinc insulin (intermediate-acting insulin) were

injected every 24 hours in 31 test animals and compared to 31 control subjects. The application of insulin showed no effect on bone strength, weight gain or histology.[254] Like the previous study, this study was not designed to examine the time to heal the defect in both control and insulin treated groups, only the break strength after insulin therapy. At this point, insulin therapy for bone regeneration in rabbit and rat models did not appear to improve bone strength after recovery and provided contradictory results to studies performed in the late 20's and 30's.

In the late 1960's, interest in using insulin for soft tissue wound healing was gaining popularity in both laboratory, veterinary and some human studies. In 1966, Paul described a case report on a 56-year-old diabetic woman who had developed gangrene which was spreading throughout the rest of her foot. After amputation, the stump became infected with *Staphylococcus pyogenes*, releasing puss and eventually becoming necrotic. After repeated dosing with cloxacillin and erythromycin, no improvement in the infection was noted and the wound continued to heal poorly. Gauze soaked with 20 units of soluble insulin was then applied to the wound and covered. The dressing was changed twice daily and the injury healed within 3-4 day.[255] Paul concluded that insulin may have helped to promote glucose utilization in tissues, resulting in improved healing and immediate clearing of the infection. Following this study, Rosenthal (1968) examined the acceleration of primary wound healing by insulin in male albino Wistar rats. A 4 cm vertical incision through the musculofascial layer, including the peritoneum was made on the left side of the upper part of the abdomen. Protamine zinc insulin suspensions were given daily for 2 to 3 days prior to the surgery, and then daily. Insulin treated animals showed a significant increase in wound tensile strength of nearly 15% over that of controls, 7 days after surgery, and weight gain of nearly 7 fold.[256] Rosenthal concluded that the insulin must have been stimulating protein synthesis increasing the wound tensile strength in

animals receiving insulin treatment. Belfield *et al.* (1970) developed a cream containing 10 units of ulcerin, which was applied to surgical wounds, fresh bites and lacerations on cats and dogs. In non-infected wounds, no noticeable wound healing response was observed, however animals with abscesses, ulcers, or infected surgical wounds had a strong response to ulcerin in as little as 24 hours in some cases.[257] After nearly 100 animal cases, the cream was moved into the clinic for testing. Ulcerin strength was adjusted from 10 to 80 units of insulin and was applied topically twice daily. Healing of ulcerated wounds began to appear within 48 h along with new tissue growth in the treatment of 20 animals. The authors suggested that clinically, ulcerin was helping to normalize cell permeability, increase vascularization, reduce exudation, arrest bacterial growth, enhance phagocytosis, stimulate proliferation, decrease local tissue hypoxia, eliminate edema and increase wound contracture.

Udupa *et al.* (1971) examined the biochemical and histological aspects of insulin-mediated wound healing. Rats were dosed daily with 0.02 U/g of body weight, administered subcutaneously. The experiments included 25 insulin treated rats and 25 controls, and continued for approximately 3 weeks in which insulin therapy ceased 5 days before sacrificing. Linear musculoperitoneal wounds measuring 5 cm in length were made and sutured closed. Protamine zinc insulin significantly increased the bursting strength of experimental abdominal wounds in rats, along with an earlier appearance of collagen fibers which were dense and better oriented compared to control animals.[258] There was however, no difference in the appearance of collagen fibers, and no data dealing with healing times was presented. Goodson *et al.* (1978) examined wire mesh cylinders implanted subcutaneously into diabetic rats. Rats were given insulin, either immediately or in intervals after the implantation procedure. Wound healing was assessed by measuring the concentration of hydroxyproline in the tissue which had grown into

the cylinders after 21 days. Without insulin therapy in diabetic rats, wound healing was impaired and insulin was needed in the earlier phases of wound healing.[259] Weringer *et al.* (1982) continued to study the effect of insulin on wound healing in diabetic mice. Diabetic and non-treated C57B1/6 mice received small dermal wounds in the ear up to 40 h after injection of insulin, or placebo. The wound tissue was excised 8 h later and the presence of capillaries, fibroblasts, and collagen, studied microscopically.[260] Diabetic mice showed impaired healing, whereas diabetic mice provided with insulin responded similarly to non-diabetic controls by reducing the mean level of hyperglycemia experienced after wounding. No detectable difference in the duration of healing was observed between insulin-treated and non-insulin treated diabetic mice. The authors concluded that the mild reduction in hyperglycemia does support the hypothesis that insulin is necessary in an adequate wound healing response.

2.9.7 Animal and Human Wound Healing Studies – 1980s to Present

Sakurai *et al.* (1992) from the Wolfe/Herndon group at the Shriner's Burn Center in Galveston, Texas published one of the most significant reports on insulin-mediated wound healing. The effectiveness of long term insulin infusion in burn patients to alter protein kinetics in skeletal muscle in severely injured burn patients was studied. The hypothesis tested was that protein anabolism could be altered by an increased rate of transmembrane amino acid transport upon stimulation by insulin infusion.[261] Nine patients with severe burns were studied and given 7 days of high-dose insulin randomly. Exogenous insulin delivery was found to stimulate protein synthesis by 50% in the wound, as compared to controls, however varied greatly between patients. Three years later, Sakurai's group conducted a similar experiment whereby six patients with burns to greater than 40% of total body surface area received insulin plus glucose infusions for 7 days.[262] The aim was to show that limb protein anabolism and improved wound healing

could be achieved. Patients were given 25 to 49 U/h along with dextrose to maintain euglycemia, and burn wound donor-sites were biopsied and analyzed. Insulin, along with glucose was shown to reduce healing times from 6.5 +/- 1.0 days to 4.7 +/- 1.2 days during insulin infusion.

Laminin staining of the biopsied tissue revealed intense staining along the basal lamina and blood vessels, together with an increase in collagen type IV after combined insulin therapy. The study proved to be clinically significant because it highlighted that a combined insulin-glucose therapy can be used to greatly improve the quality of wound matrix formation, and reduce the healing time by nearly 30%.

Wolfe and colleagues in 1999 continued to examine wound healing in animal models to determine if insulin or growth hormone could stimulate protein anabolism in male New Zealand white rabbits. Four groups including a control (n=10), a low-dose insulin (n=5), a high-dose insulin (n=5) and a growth hormone group (n=5) were subjected to a partial thickness, scald burn to the ear. A L-[ring-¹³C₁₆]phenylalanine trace was infused to measure muscle protein synthesis, while insulin was infused at either 0.6, or 2.3-3.4 mU/kg/min, or 2 mg/kg/d of growth hormone. Low or high dose insulin inhibited protein proteolysis, while growth hormone had no net effect.[263] Using the same experimental protocol, the effects of co-administration of amino acids, including radio-labeled phenylalanine and proline, and insulin was tested to determine if the net protein balance in skin wounds could be improved. Rabbits were injured with a partial thickness scald to the ear and treated with 2.5 mU/kg/min insulin and a combination of lysine, valine, phenylalanine, histidine, threonine, methionine, tryptophan, alanine, arginine, glycine, proline, serine and tyrosine at doses ranging from 40 mg to 2.07 g.[264] The combination of both insulin and amino acid therapy caused a shift from protein catabolism, to anabolism in the skin wound from -6.5 +/- 4.5 to 1.4 +/- 5.2 μ mol/100 g/h, compared to insulin, amino acids alone,

or without any treatment at all.[264] The authors concluded that there is an interactive effect of insulin and sufficient amino acid supply on protein metabolism in skin wound and that it is better when co-administered, rather than individually infused into animals. This study was instrumental in the design of human trials conducted by Gore and colleagues (2002) examining six severely burned male patients with burns on greater than 60% of their total body surface area. The amino acid and insulin therapy did not mimic the anabolic effects of insulin and amino acid therapy in human patients as seen in the previous studies by Wolfe and colleagues.[265]

Wolfe's group in 2007 examined the effects of localized insulin delivery versus intravenous administration as reported in previous studies. Local insulin-zinc was injected into the backs of dermatomed New Zealand white rabbits, simulating a partial thickness wound to the skin. [266] The purpose was to determine if localized insulin delivery could accelerate wound healing as effectively as intravenous infusion, but without the effects of hypoglycemia and hypokalemia. Injections of 0.25 units insulin were directly placed into 5 sites around the wound, approximately 1 to 1.5 cm from the wound margin every other day. The treatment resulted in wounds healing in 11.2 +/- 2.3 days as compared to controls that healed on average 15.1 +/- 4.1 days.[267] Using a similar experimental protocol, Zhang and colleagues determined that DNA synthesis was also increased by nearly 50% in wound tissue as compared to control animals.[266] Wolfe's research showed that although infusion of insulin is one method of improving wound healing in animal models, localized insulin delivery can be just as effective in accelerating wound closure. Zhang *et al.* (2009) revisited the insulin infusion model to determine if DNA and protein synthesis would increase with the presence of insulin. Protein and DNA synthesis rates were up 9.0 +/- 1.2 and 5.4 +/- 0.5 % per day, versus the controls which had rates of 6.4 +/- 0.5 and 4.0 +/- 0.6 % per day respectively.[268] As a result, the Wolfe/Herndon group was able to

demonstrate that intravenous and topical insulin delivery are capable of reducing wound healing times, and increasing both protein and DNA synthesis by wound tissue.

Wound repair of corneal tissues was also examined in diabetic rats to study insulin's role in improving re-epithelialization. Zagon *et al.* (2006) investigated whether intensive treatment with insulin could be used to maintain blood glucose levels at near normal levels in diabetic rats to facilitate delayed corneal wound healing. Corneal abrasions were inflicted and followed with insulin injections to maintain blood glucose near normal levels after 9 and 11 weeks post-induction of diabetes. Healing and DNA synthesis in diabetic rats treated with insulin exhibited similar healing as control animals, suggesting that intensive insulin therapy prevents delayed wound healing on the ocular surface of diabetic rats.[269] Zagon in 2007, reported similar results whereby insulin was topically applied to injured corneal surfaces. Eye drops containing 1, 2 and 5 U of insulin applied 4 times daily for 7 days resulted in diabetic rats healing at a similar rate to that of non-diabetic test animals.[270]

Recently, four clinical studies between 2009 and 2011 were designed to test insulin therapy in pediatric and adult wound cases. Rezvani *et al.* (2009) conducted a randomized, double-blind, placebo-controlled trial to determine the effects of topical insulin on wound healing in 45 patients with non-infected acute and chronic extremity wounds. Topical application of a 0.1mL/cm² of 0.9% saline spray, or 10 U of crystalline insulin twice daily were used. The endpoint was complete wound closure.[271] There were no significant differences in wound healing rates between control and treatment groups which could have been related to the experimental protocol where the solution was permitted to dry on the surface of the wound. This would have prevented insulin from penetrating and diffusing into the wound, resulting in no improvement to wound healing. Wilson *et al.* (2008) reported on the topical use of insulin to

promote healing of problematic surgical wounds. An 80 year-old female patient with a chronic non-healing wound following a laparotomy procedure, failed to heal, even after 3 weeks of negative-pressure vacuum pump dressing application. The patient underwent insulin therapy as a last resort method. The dressings were replaced with daily irrigation with 20 mL of saline containing 2 U of soluble insulin for 7 days.[272] At the end of the therapy, visible improvement was observed, along with no metabolic side effects. Fram *et al.* (2010) conducted a randomized trial on 20 pediatric burn cases with burns to more than 40% of total body surface area. Blood glucose levels of one group were maintained between 80 to 110 mg/dL, and approximately 215 mg/dL for a second group, with intravenous infusions for 10-14 days. The findings concluded that intensive insulin therapy in pediatric patients can be performed safely with the ability to improve metabolic balance and to prevent insulin resistance after acute trauma.[273] Scimeca *et al.* (2010), combined insulin chemotherapy with negative pressure wound therapy (NPWT) in treating complex wounds. NPWT is used to assist in the development of granulation tissue by improving wound contracture, moisture and re-epithelialization. A 71 year old male diabetic patient who had received an emergency amputation, received NPWT in conjunction with an infusion drip of insulin at a rate of 40 mL/h following a dose similarly used by Wilson *et al.* (2008).[274] Within 48 h, the wound base was 90% granular with neo-epithelialization visibly present. Tuvdendorj *et al.* (2011) also explored the use of intensive insulin-therapy in enhancing the healing of skin graft donor sites in burn patients by increasing the fractional synthesis rate of wound protein. The randomized trial followed 13 control and 10 insulin treated pediatric patients with burns to greater than 38% body surface area, requiring skin grafting. Insulin was provided at a rate of 0.1 U/kg/h and the blood glucose level maintained at 80-110 mg/dL. The metabolic response 2-4 (early period) and 5-6

(late period) days postoperatively was examined. In the early period, intensive insulin therapy increased the fractional synthesis rate of wound proteins as determined by radiotracers administered during the insulin therapy to measure protein turnover rates.[275] The significance of this study reveals that in skin graft donor sites, insulin therapy administered as early as possible, is able to increase fractional synthesis of proteins more effectively.

2.9.8 Cell Based Wound Healing Studies – 1990s to Present

While human and animal trials were being conducted by the Wolfe group, insulin therapy was also being explored in the animal and tissue-based studies to further examine the molecular events behind the role of insulin in enhancing the healing process. Yano *et al.* (1996) evaluated the effect of insulin on fracture and wound repair of bone at both the tissue and cellular level using parietal bones removed from 20 day old Sprague-Dawley rat fetuses and cultured *in vitro*. After fracturing, specimens were placed in either medium with 10^{-6} mol/L insulin, or medium without insulin, and studied 12 days. Bones receiving insulin treatment showed complete bone gap restoration after the 12th day, along with a thick osteoid layer and several layers of osteoblasts. No osteoclasts were detected. In the control group, 6 bones did not bridge the gap, while the remaining 5 indicated some marginal bone growth. The frequency of bone bridging (gap restoration) was statistically significant in the insulin treated group, compared to the insulin-free controls. Yano concluded that insulin had promoted bone repair through the stimulation of bone-forming cells and inhibition of bone-resorbing processes. Huang *et al.* (1999) studied the effect of extracellular calcium, in combination with zinc, insulin and insulin-like growth factor 1 (IGF-1) on DNA synthesis in 3T3 fibroblasts cultured in serum containing 1.8 mM Ca^{+2} , stimulated with $0.75\text{-}2.0 \text{ mM Ca}^{+2}$, along with the addition of zinc (15-60 mM), insulin and

IGF-1. Insulin, along with zinc had prolonged Ca^{+2} induced mitogen-activated protein (MAP) kinase activity and p70 S6 kinases, thereby promoting improved fibroblast growth.[276]

Shanley *et al.* (2004) investigated the effects of insulin and leptin on *in vitro* wound healing of transformed human corneal epithelial cell monolayer. The aim was to identify cellular and intracellular signaling pathways involved in either proliferation, or migration events. Using a monolayer of human corneal epithelial cells (HCEC), a single scratch was made into the monolayer of cells. The scratched monolayers were exposed to insulin and leptin and monitored every hour for 8 h. Western blot analysis of phosphoinositide 3-kinase (PI3-kinase) and MAP-kinase activation was conducted. Inhibitors such as Ly 294002 and wortmanin were used to prevent PI3-kinase and ERK 1/2 insulin induced activation as well. Shanley concluded that insulin alone facilitated wound closure of HCEC cell monolayers through the phosphorylation of ERK 1/2 and Akt and that their inhibition severely reduced insulin-induced wound healing.[277] Insulin facilitated cell migration instead of proliferation through the lack of BrDU uptake as the scratch area became filled over the 8 h period. The importance of this study is that the authors showed that for small wounds, cell migration is the key driving force in wound healing whereas in larger wounds, proliferation plays a larger role. Musselmann *et al.* (2005) observed the effect of keratocytes (mature corneal stroma cells) co-cultured with three different growth factors including fibroblast growth factor-2 (FGF-2), insulin and platelet derived growth factor-BB (PDGF) on keratocyte proliferation and the maintenance of the keratocyte markers over 7-days at low and high seeding densities in serum-free conditions. Proliferation was measured using [^3H]thymidine incorporation and by DNA content of the cultures. The expression of cytosolic aldehyde dehydrogenase and keratocan determined by Western blot, measured the degree of keratocyte phenotype expression during the course of the experiment. All growth factors

stimulated proliferation and increased the DNA content of the cultures over the 6-day period. Insulin however, stimulated thymidine incorporation in a more consistent fashion, and attained the greatest accumulation of DNA across all cultures. Likewise, insulin maintained keratocyte morphology at the end of the study period more effectively than the other growth factors. The relevance of this study indicates that insulin may be an important element in maintaining the phenotype of proliferating cells in culture.

The cellular and molecular mechanisms of wound healing and vascular repair within the skin were examined by the Martins-Green group. Liu *et al.* (2008) studied insulin's angiogenic effects using a combination of *in vitro* and *in vivo* techniques. C57BL/6J mice were injected with 0.03 units of insulin per 20 μ L saline in treatment groups and saline only in control groups every 24 hours for 5 days. On day 6, skin samples were removed and histologically analyzed. Insulin injections stimulated the formation of microvessels, with significantly more branching as compared to control animals receiving saline injections. Cellular migration and DNA synthesis was measured using human microvascular endothelial cells (HMEC) treated with insulin concentrations ranging from 10^{-5} to 10^{-8} M. As well, insulin stimulated endothelial cell migration and tube formation independently of the VEGF/VEGFR receptor pathway. The molecular mechanism of action was found to be driven by the PI3-K, Akt, sterol regulatory element binding protein 1 (SREBP-1) and Rac 1 signaling proteins. Inhibition of any of these pathways resulted in the elimination of endothelial cell migration, tube formation and the development of microvessels.[278] Liu *et al.* (2009) explored the cell and molecular mechanisms of keratinocyte function in similar *in vivo* and *in vitro* studies.[153] Using C57BL/6J mice, 7 mm excision wounds were made and treated with a topical application of 0.03 units of insulin. In addition, human keratinocytes co-incubated with insulin, were assayed using

a cell scratch assay. Both studies demonstrated that topical insulin application to excision wounds and in human cell culture would result in the migration and differentiation of keratinocytes. Wounds in mice showed accelerated wound re-epithelialization more rapidly than controls. The tissue response in culture depended on the activation of the PI3-K and Akt pathway, followed by the activation of Rac1 and the integrin $\alpha 3$ and the ECM molecule laminin 332. The authors reported that this study helps to identify the critical pathways involved in skin wound healing and the importance that insulin could have in future biopharmaceutical wound healing therapies.[153]

Topical insulin delivery has therefore been shown to be beneficial in stimulating re-epithelialization, angiogenesis and in the recovery of both acute and chronic wounds *in vitro* and *in vivo*. Wound healing technologies that could provide sustained and controlled delivery of bioactive insulin would be beneficial.

2.9.9 Peptide Delivery

The use of recombinant polypeptides in the clinic has been continually growing for the past two decades. Whilst growth is rapid, instability and the need for a sustained delivery vehicle remain. Unlike small molecular weight drugs less than 1 kDa, polypeptides are complex molecules with molecular weights ranging from 5-200 kDa.[279] Maintaining amino acid sequence and unique three-dimensional structure is essential for preserving polypeptide bioactivity and avoiding undesirable immunological reactions.[280,281] Furthermore, polypeptide delivery has often been accomplished via intravenous infusion, or injection as is practiced with insulin .[94,282] Unfortunately, these routes of delivery offer little to no protection from enzymatic degradation, and require frequent dosing to maintain therapeutic levels within systemic circulation. As a result, careful selection of the delivery vehicle and

biomaterial are necessary in designing a peptide delivery system which can stabilize a sensitive polypeptide.

Microparticles, are an example of a peptide delivery vehicle, used to deliver a variety of biologicals.[94] Microparticles can be manufactured from several biocompatible biopolymers and techniques such as emulsification or electrospray. Insulin for example, has been encapsulated in a variety of delivery vehicles including aqueous insulin solutions, nano/microparticles and hydrogels.[283,284,285] Various delivery vehicles and biomaterials that can be used to deliver insulin and other growth factors topically are described in the following sections.

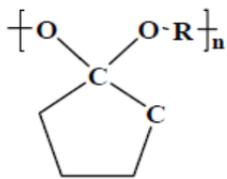
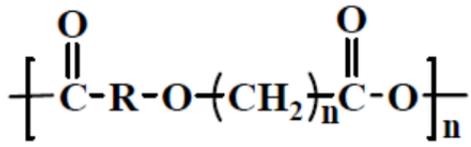
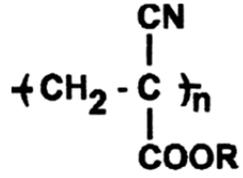
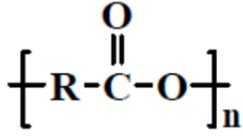
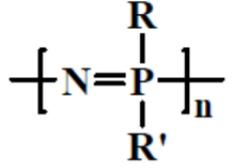
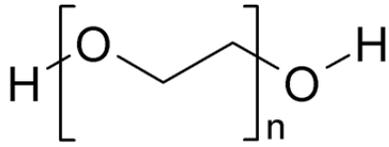
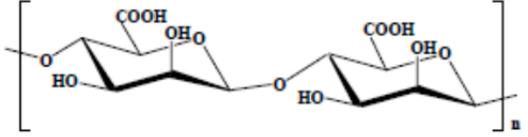
2.9.9.1 Microencapsulation

Microencapsulation is a technique widely used in pharmaceuticals to stabilize and release peptides and/or drugs. Unlike small molecular weight drugs, recombinant peptides and proteins are sensitive to enzymatic degradation, shear stress, and changes in temperature and pH. [286,287] In some cases, microencapsulation has also been used to deliver plasmid DNA, antisense oligonucleotides, and synthetic double-stranded DNA for gene therapy and other vaccine applications.[288,289,290] Consequently, microencapsulation utilizes carefully selected biocompatible polymers and manufacturing techniques to take into account the molecular weight of the peptide, biological half-life, immunogenicity, conformational stability, dose requirement, pharmacokinetics and pharmacodynamics, in order to successfully deliver the therapeutic molecule.[290,291] Microencapsulation is therefore a useful drug delivery technique to overcome physiological and stability issues associated with the delivery of a recombinant peptide.

Polymeric microparticles are one of the most popular methods to microencapsulate recombinant peptides, although liposomes and implants such as rods, disks and pellets are

sometimes used as well.[292] Microparticles are advantageous because they are small in size (<1-500 μm), easy to manufacture and can be delivered virtually anywhere within the body through a single injection, or spread across large surface area.[293] In addition, microparticles can be prepared from several different types of polymers which are either synthetic or natural. Characteristically, the polymers used in microencapsulation are biocompatible, demonstrate low cytotoxicity, biodegrade within a desired time frame, and are usually approved by the FDA for use in the human body.[294] Since most microparticles are injected within the body, many formulations use polyesters and other biodegradable polymers to slowly erode and release the peptide over time. This ensures that after delivery, the microparticles will not remain in the body risking the chance of becoming lodged in small capillaries or orifices. Some common biopolymers that are used for peptide delivery include poly(lactic-co-glycolic acid) PLGA, poly(lactic acid) (PLA), poly(ethylene glycol) PEG, and poly(caprolactones) (PCL).[94,283,295,296] Table 13 lists common biocompatible polymers used to prepare microparticles. Applying these polymers, microparticles can be produced using techniques such as spray drying, emulsion-solvent evaporation, extrusion and electrospraying.[297,298,299] Depending on the method used, the porosity, size distribution and distribution of peptide within the polymer matrix can be tailored to suit drug release profiles. Polymeric microparticles are therefore, effective drug delivery platforms that are well suited to provide sustained, controlled delivery of peptides such as insulin.

Table 13. Biocompatible polymers commonly used in the production of microparticles used in peptide delivery.[290]

Class	Type	
Synthetic	Polyorthoesters	 $\left[\begin{array}{c} \text{O} \quad \text{O-R} \\ \diagdown \quad / \\ \text{C} \\ / \quad \backslash \\ \text{C} \\ \diagup \quad \diagdown \\ \text{O} \end{array} \right]_n$
Synthetic	Polyanhydrides	 $\left[\text{C}(=\text{O})-\text{R}-\text{O}-(\text{CH}_2)_n-\text{C}(=\text{O})-\text{O} \right]_n$
Synthetic	Polyalkylcyanoacrylates	 $\left[\text{CH}_2-\text{C} \begin{array}{c} \text{CN} \\ \\ \text{COOR} \end{array} \right]_n$
Synthetic	Polyesters	 $\left[\text{R}-\text{C}(=\text{O})-\text{O} \right]_n$
Synthetic	Polyphosphazenes	 $\left[\text{N}=\text{P} \begin{array}{c} \text{R} \\ \\ \text{R}' \end{array} \right]_n$
Synthetic	Polyethers	 $\text{H} \left[\text{O}-\text{CH}_2-\text{CH}_2 \right]_n \text{O}-\text{H}$
Natural	Polysaccharides	 $\left[\begin{array}{c} \text{COOH} \quad \text{OH} \\ \quad \\ \text{HO} \quad \text{HO} \end{array} \right]_n$

2.9.9.2 Poly(lactic-co-glycolic acid)

Poly(lactide-co-glycolic acid) (PLGA) is a biocompatible polymer commonly used in sustained, controlled release applications. Originally, PLGA was developed as a polymer for bioabsorbable sutures, however the polymer's degradation kinetics made it more ideal for extended drug delivery.[300] Soon after, PLGA became a popular choice for microencapsulating recombinant proteins,[301] nucleic acids,[302] and small molecular weight drugs for nearly 30 years.[303] PLGA is approved by the FDA for use in medical devices and has been incorporated into several pharmaceutical products throughout the world (Table 14).[300,304]

Table 14. Pharmaceuticals which employ the use of PLGA in the drug delivery vehicle to provide sustained and controlled release.

Pharmaceutical/Biological	Manufacturer
Lupron Depot	TAP Pharmaceutical Products Inc.
Nutropin Depot	Genentech Inc.
Sandostatin LAR	Novartiz
Trelstar Depot	Watson
Zoladex	Astrazeneca

Chemically, PLGA is a copolymer of poly(lactic acid) (PLA) and poly(glycolic acid) (PGA), linked together with an ester bond,[305] as illustrated in Figure 20. The ester bond hydrolyzes in water, resulting in polymer erosion.[94] According to Houchin and Topp (2008), the PLGA hydrolysis mechanism proceeds via a tetrahedral intermediate, which produces both a primary alcohol and a carboxylic acid. The ester functional groups are susceptible to hydrolysis because the oxygen atoms provide an electron withdrawing effect, which results in a partially positive ester carbon. Eventually, the accumulation of carboxylic acid hydrolysis products

within the PLGA matrix causes additional autocatalysis and the erosion of the matrix. Figure 21 illustrates the reaction mechanism of PLGA hydrolysis. As the PLGA polymer matrix begins to erode away, small channels form throughout the matrix allowing water to penetrate deep within the polymer network. The formation of microchannels helps to slowly release the drug or biological entrapped within the matrix over a slow, and near constant rate. As the matrix begins to further break down, acidic byproducts accumulate within the central core, eventually resulting in complete breakdown and fragmentation of the delivery vehicle.[306] Often, this process results in a large burst of the drug toward the end of the release period, resulting in a biphasic drug release profile.[307] In some cases, the acidic environment can be harmful to peptides, resulting in protein degradation and loss of bioactivity. Figure 22 depicts the bulk degradation process of PLGA microparticles.

PLGA can also be chemically tailored to meet specific physico-chemical properties by altering the ratio of lactic and glycolic acid, adjusting the molecular weight and changing the end-cap groups. Characteristics such as degradation rate, glass transition temperatures (T_g), hydrophobicity, and mechanical strength are all properties that can be adjusted by changing the ratio of PLA to PGA.[308] By altering the proportions between 25:75, 50:50, 75:50 lactic:glycolic acid, PLGA can be tailored to provide shorter term-or longer term drug release by varying the degree of crystallinity within the polymer chain, or molecular weight.[308] Similarly, the polymer end groups can be capped with either ester or carboxylic acid, resulting in a change to the overall hydrophobicity of the polymer.[309] As more carboxylic end groups are attached, the hydrophobicity can be lowered to allow greater water uptake and faster hydrolysis of the peptide.[310] PLGA can therefore be highly tuned to achieve a range of drug release profiles.

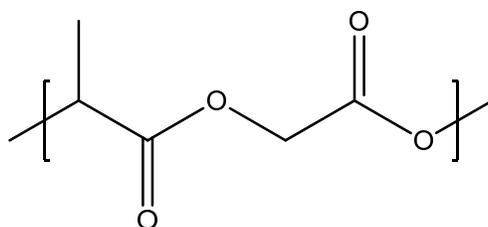


Figure 20. The chemical structure of PLGA. Note the ester bond which links the lactic acid moiety to the glycolic acid moiety. The ester bond can be broken by hydrolysis causing the erosion of PLGA with time.

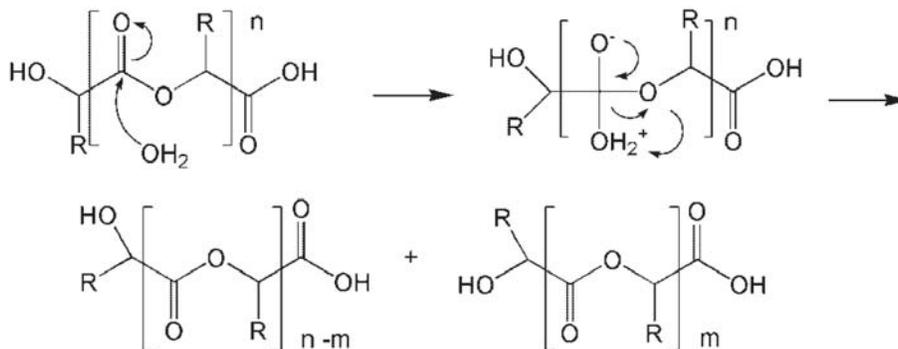


Figure 21. The reaction mechanism illustrating the hydrolysis of PLGA. The R group represents a methyl group for the lactide and H for the glycolide. The sequence can be initiated by the deprotonation from the carboxylic acid by the alcohol, or by the protonation of the alcohol and deprotonation of the carboxylic acid by water.[309]

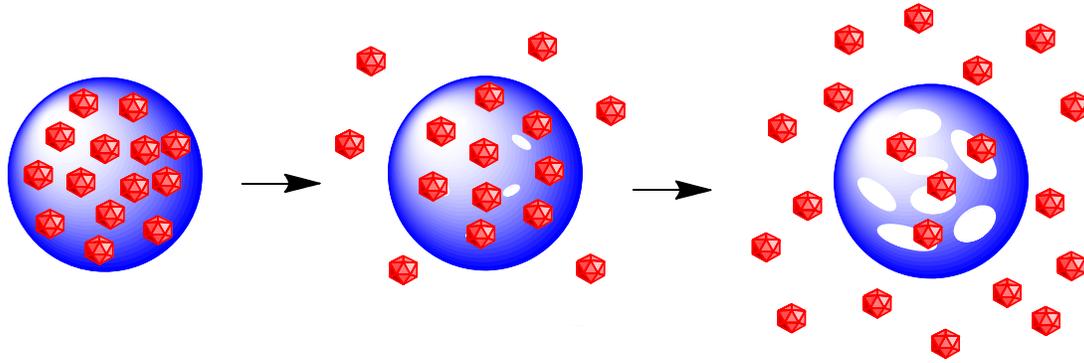


Figure 22. The bulk degradation process of a PLGA microparticle (blue), loaded with a drug (orange). As pores on the surface of the microparticle begin to erode, small microchannels are formed throughout the matrix. This process results in pockets within the microparticle where water and carboxylic acids begin to accumulate as a result of the hydrolysis mechanism. Once the structure of the matrix has been compromised, the entire vehicle collapses and fragments.[305]

PLGA can be formulated into a number of shapes including slabs, films and microparticles. Commonly though, PLGA microparticles are prepared because they have a number of favourable characteristics such as having lubricious, smooth, permeable, biodegradable, and biocompatible qualities.[94,140] For topical insulin delivery, microparticles are a favourable approach because they can be spread across a large surface area effectively. Conventionally, PLGA microparticles are formed using techniques such as the water-in-oil-in-water (W/O/W) emulsion, solvent evaporation method. This approach is advantageous due to its relative ease in preparation, but does pose a problem in preserving peptide activity.[311] Factors such as heat and shear forces associated with mixing, and interfacial tension can lead to unfolding and deactivation of polypeptides.[312,313,314,315] Likewise, polypeptides such as insulin can be exposed to highly acidic microenvironments within PLGA microparticles, causing peptide denaturation during the bulk erosion process [316,317,318] To prevent degradation from occurring, the incorporation of poly(ethylene glycol) (PEG) can also help to preserve peptide bioactivity and attenuate release kinetics.[144] PLGA microparticles are therefore an effective

method of providing peptide delivery when used in combination with other polymers such as PEG.

2.9.9.3 Summary

Burn wounds are destructive, painful and potentially life threatening injuries that can require significant medical care, time and inflict great amounts of pain. Currently, wound dressings such as gauze, hydrocolloids and alginate dressings provide a sufficient temporary occlusive barrier that enables a debris-free, moist wound healing environment. Unfortunately, these dressings do not provide any inherent method of accelerating wound closure beyond the natural rate of healing. Consequently, some investigators have reported that the topical application of growth factors such as EGF, TGF- β , PDGF and others, are capable of stimulating angiogenesis, re-epithelialization and a quicker recovery overall. The cost of these growth factors (\$1000-9800 per mg) however, subsequently limits widespread use. Insulin on the other hand, is an affordable peptide (\$1.00 per mg) that has been demonstrated by Rosenthal, Wolfe and Martins-Green, to be a potent inducer of rapid wound healing. To date, no one has successfully demonstrated a convenient and effective insulin delivery vehicle which can be used to increase the rate of wound healing. As a result, a biodegradable, insulin-eluting wound dressing that could provide sustained, controlled release of insulin from biocompatible materials would be valuable.

Chapter 3

RESEARCH OBJECTIVES

Burns are destructive injuries that result in severe pain and tissue disruption, and thus require immediate medical attention. A dressing subsequently provides a temporary barrier that facilitates healing, prevents the entry of microorganisms and debris, absorbs exudate and maintains a moist wound healing environment conducive for wound healing. Current wound dressings however, cannot accelerate wound healing beyond the natural rate, often require frequent dressing changes and cannot be removed easily without triggering additional pain or tissue destruction. Incorporation of growth factors into wound dressings is also technically challenging, expensive and difficult to preserve over extended periods of time.

One approach would be to formulate a composite wound dressing capable of releasing insulin from a natural polymer that absorbs water and forms a stable hydrogel. Alginate foam wound dressings are of interest because of beneficial healing properties, excellent water handling characteristics, ease of use, and ability to deliver peptides. Incorporation of PLGA microparticles loaded with human recombinant crystalline insulin is proposed as a mechanism to enable sustained, controlled insulin delivery over an extended period of time and across a large surface area. In order to develop a composite wound dressing capable of insulin release, the following research objectives were developed.

1. Identify a microparticle formulation method that can provide release of bioactive insulin for up to 21 days at a controlled and adjustable rate to stimulate wound re-epithelialization.
2. Characterize the release profiles and bioactivity of insulin in a physiologically relevant *in vitro* model.
3. Develop a wound dressing in which insulin loaded microparticles can be embedded and monitored for the insulin release and bioactivity.
4. Demonstrate accelerated wound healing in a rodent burn model and characterize the effect of topical insulin delivery on metabolic and wound healing responses.

Chapter 4

MATERIALS AND METHODS

4.1 Materials

PLGA (*D,L*;50:50, 5-15kDa), human recombinant crystalline insulin (hRcI), poly(ethylene glycol) (PEG) (Avg. MW 1450 Da and 10 kDa), poly(vinyl alcohol) (PVA) (Avg MW 13-23 kDa), CaCl₂, NaCl, NaOH, and sodium dodecyl sulphate (SDS) were supplied by Sigma Aldrich (Oakville, Canada). Purasorb 5002 PLGA (*D,L*;50:50, avg. 17 kDa) was obtained from Purac Biomaterials (Lincolnshire, USA). Protanal LF10/60 (65-75% guluronic/25-35% mannuronic) (G) and LF10/60 LS (35-45% guluronic/55-65% mannuronic) (M) were obtained from FMC Biopolymer (Philadelphia, USA). Bovine serum albumin (BSA) and micro BCA protein assay kit were obtained through ThermoFisher (Ottawa, Canada). Dichloromethane (DCM) was supplied by Caledon Labs (Georgetown, Canada). Rat L6 myoblasts were supplied by ATCC (Manassas, USA). Human keratinocytes (HaCaT) were purchased from Cell Line Service (Eppelheim, Germany). Dulbecco's modified Eagle's medium (DMEM), Dulbecco's phosphate buffered saline (DPBS), and 0.25% trypsin with EDTA were obtained through Gibco (Carlsbad, USA). Fetal bovine serum (FBS) and penicillin-streptomycin were purchased from Hyclone (Fair Lawn, USA). Fast Activated Cell-based ELISA (FACE) AKT ELISA assay kit was obtained through Active Motif (Carlsbad, USA). Human insulin ELISA kit was supplied by Merckodia (Winston Salem, USA).

4.2 Methods

4.2.1 Microparticle Preparation

Human recombinant crystalline insulin (hRcI) was microencapsulated by solvent evaporation method from a solid (insulin crystals)-in-oil-in-water (S/O/W) multiple phase emulsion method. Microparticles were prepared with 0 (blank), 2.5, 5 and 10% w/w hRcI by dissolving 90-100 mg of PLGA in 1 mL of dichloromethane (DCM) and combining with 0, 2.5, 5 or 10 mg of hRcI. The suspension was aspirated several times with glass Pasteur pipette to disperse the crystals, then transferred drop-wise into 30 mL of a chilled aqueous solution of 5% PVA. The resulting S/O/W emulsion was stirred for 1 min at 430 rpm with a high shear impeller mixer (Caframo; Warton, Canada), then the mixture was poured into a beaker filled with 400 mL distilled water chilled to 5-7°C, and stirred overnight in an ice bath at 200 rpm with a propeller mixer. The microparticles were collected by vacuum filtration, rinsed 3 times with ice cold distilled water, and then transferred to a -20°C freezer overnight. The microparticles were then refrigerated at 4°C for 12 h prior to being placed under vacuum for another 12 h to remove residual DCM. Microparticles were stored at -20°C when not in use.

4.2.2 Encapsulation Efficiency

Microparticles (8-10 mg) were dissolved in 3 mL DCM. DCM was evaporated off, then the residue was dissolved in 10 mL aqueous 5% SDS, 0.1 M NaOH with stirring at 80 rpm for 24 h at 37°C. Protein concentration was determined by micro BCA protein assay and read spectrophotometrically at 562 nm. Encapsulation efficiencies (E.E.%) were calculated by taking the ratio of residual insulin versus the initial insulin added during microparticle formulation. All samples were measured in triplicate.

4.2.3 Particle Size and Morphology

Mean sphere and pore sizes were determined by measuring at least 1000 microparticles by hand from SEM images (JEOL 840, USA). To examine the internal morphology, microparticles were embedded into LR white resin, sectioned and mounted onto copper grids. TEM images (JEOL 1200EX, USA) were later digitally captured and then analyzed for presence and location of insulin crystals.

4.2.4 DSC Analysis

Approximately 4 mg of microparticles were placed into aluminum pans that were hermetically sealed. Both empty and sample pans were placed into a DSC (TA Instruments, USA), cooled to -30°C and equilibrated for 1 min. Samples were then heated to 110°C and cooled back to -30°C at a rate of 10°C per min.

4.2.5 *In vitro* Insulin Release Kinetics from Microparticles

Microparticles (4-6 mg) were added to 2 mL of 20 mM phosphate buffer at pH 7.4 and mixed on an orbital shaker at 60 rpm and 37°C. Insulin in supernatants was measured by a micro BCA assay compared to insulin standards. Supernatants were replaced with fresh buffer after each sampling. All samples were tested in triplicate.

4.2.6 Cell Culture of Rat L6 Myoblasts

Rat L6 myoblasts were cultured in T75 flasks containing 15 mL complete DMEM medium supplemented with 20% FBS and 1% pen-strep. Cells were incubated at 37°C and maintained in an atmosphere of 5% CO₂. When cells reached 80-90% confluence, the medium was aspirated and cells washed with 15 mL PBS. Cells were then detached with 5 mL of a 0.25% trypsin and EDTA solution and incubated for 5 min. The trypsin was neutralized with complete medium, and cells centrifuged at 200 x g for 5 min, then resuspended in 10 mL of fresh complete medium. Cells were counted by removing 10 µL of cell suspension and combining it with an equal volume of trypan blue solution. This mixture was vortexed and a sample loaded into a hemacytometer chamber (Fisher; Pittsburgh, USA) and cells counted under an inverted microscope (Nikon; Melville, USA).

4.2.7 Cell Culture of HaCaT Cells

HaCaT cells were cultured in T75 flasks containing 15 mL complete DMEM medium supplemented with 10% FBS and 1% pen-strep. Cells were incubated at 37°C and maintained in an atmosphere of 5% CO₂. Cell passaging and counting were performed as described above for the L6 myoblasts.

4.2.8 Insulin Bioactivity: FACE AKT ELISA

L6 myoblast cell suspensions were diluted to 112,500 cells/mL in complete medium and seeded onto 96 well plates (Costar; Lowell, USA) to a final well volume of 200 µL. Cells reached 80% confluence in 24 h, at which time the cells were washed twice in the wells with PBS. Then, 200 µL of starvation medium consisting of 99% DMEM and 1% pen-strep were placed into the wells 6 h prior to stimulation. Cells were then stimulated with starvation medium used to dilute (i) fresh insulin dissolved in starvation medium to 150 nM; (ii) supernatant

containing 150 nM insulin released from microparticles; or (iii) supernatant from placebo microparticles. Supernatant samples from release experiments were taken from days 1, 7, 14 and 23. After 20 min stimulation, the medium was aspirated off and 100 μ L of 4% formaldehyde in PBS was added. The wells were then processed according to the Fast Activated Cell-based ELISA (FACE) AKT kit protocol, measuring total AKT and phosphorylated AKT (phospho-AKT). The level of AKT phosphorylation was normalized for cell density measured by crystal violet staining. Six replicates were used for each sample.

4.2.9 Insulin Bioactivity by HaCaT Cell Scratch Assay

HaCaT cell suspensions were diluted to 250,000 cells/mL in complete medium and seeded onto 24 well plates (Costar; Lowell, USA) to a final well volume of 1 mL. Cells formed a confluent layer 48 h later, at which time a single scratch was made using a 10 μ L pipette tip. All wells were washed twice with 1 mL PBS to remove cellular debris. Then, 500 μ L of starvation medium consisting of DMEM, 2% FBS and 1% pen-strep were used to dilute (i) fresh insulin to 10^{-7} M; (ii) supernatant containing 10^{-7} M insulin released from microparticles; or (iii) supernatant from placebo microparticles. Supernatant samples from release experiments were taken from days 1, 7, 14 and 23. Scratches were photographed and width measured using a bright field inverted microscope (Nikon; Melville, USA) at 0, 4, 24 and 48 h. All samples were tested in triplicate.

Using the same method above, supernatants obtained from alginate sponge dressing (ASD) insulin release studies were also tested and compared to the same fresh insulin control. In some cases, supernatants contained very little insulin making it difficult to achieve the 10^{-7} M concentration. The samples which have been affected are indicated in the figure caption. Cell

migration across the scratch was measured under a phase contrast inverted microscope (VWR; Mississauga, Canada) at 0, 4, 24 and 48 h.

4.3 Alginate Sponge Dressing (ASD)

Aqueous alginate solutions (2 and 4% w/v) were prepared by slowly adding alginate M (high mannuronic content) or G (high guluronic content) powder into 100 mL distilled water, under high shear mixing. Various quantities of PEG (0.1-10% w/w) with differing molecular weights (1.45 or 10 kDa) were added to 2% M alginate solutions during the mixing process. Once all of the constituents were fully dissolved, 0.9 mL were transferred into 24 well plates (Costar; Lowell, USA). Insulin-loaded PLGA microparticles (5 mg) were then directly deposited into the wells and mixed briefly. The plates were frozen at -80°C for 1 h, then freeze-dried (Virtis; Gardiner, USA). Circular alginate sponge dressings (ASD) made from high guluronic alginate (G-ASD), high mannuronic alginate (M-ASD), and PEG-high mannuronic alginate (P/M-ASD) were then removed from the wells and stored at -20°C until use.

4.3.1 Scanning Electron Microscopy

External and internal morphologies of the various ASD formulations were characterized using SEM (JEOL 840, USA). Cross sections were prepared by cutting the circular dressings using a razor blade and mounting sections on aluminum stubs covered with double-sided carbon tape. All stubs were gold sputter coated and analyzed.

4.3.2 Physical Properties: Dressing Density

The density of the various ASD was calculated by measuring the dimensions of the circular sample and the mass (Mettler; Mississauga, Canada). The density was expressed as the ratio between the mass and its volume.

4.3.3 Physical Properties: Tensile Testing

The tensile strength of the various ASD was measured by cutting samples into 5 x 20 mm rectangular strips. The dimensions were measured using a digital micrometer prior to attachment to the instrument. The maximum stress (MPa), Young's modulus (MPa) and elongation at fracture were obtained using a texture analyzer (TA Instruments, New Castle, USA) with a strain of 2 mm/min.

4.3.4 Physical Properties: Water Vapour Transmission Rate

The water vapour transmission rate (WVTR) was determined using a method described in ASTM E96.[319,320] ASD were attached using silicone sealant (GE, Huntsville, USA) to the underside of an inverted netwell insert (Corning; Lowell, USA), adhered to a polystyrene petri dish lid. The chamber within the netwell insert was filled with 8 mL distilled water and the entire apparatus placed into an incubator at 35°C to mimic human skin temperature and a relative humidity of 33%. The mass of the apparatus was weighed at the beginning of the experiment, and then 24 h later to record the amount of water lost. $WVTR = W / (A \times t)$, where W is the weight lost through the membrane, A is the area of the membrane and t is fixed at 24 h.

4.3.5 Physical Properties: Water Absorption Capacity

The water absorption capacity (WAC) of the ASD was tested using a method modified from that described in the British Pharmacopoeia (BP) monograph for alginate wound dressings and packings.[321] Individual ASD were weighed (W_1) and placed in plastic weigh boats filled with solution A (142 mM NaCl, 2.5 mM CaCl₂), equal to 40 times the mass of the sample. The ASD were transferred into an incubator at 35°C for 30 min, then removed by lifting the corner using forceps, and held in suspension for 30 s to remove excess liquid before weighing (W_2). The water absorption capacity was calculated using $WAC = (W_2 - W_1) / W_1$.

4.3.6 Insulin Release Kinetics from PLGA Microparticles within ASD

ASD were individually placed inside netwell insert supports, prior to being loaded into 6 well plates filled with 1.5 mL British Pharmacopoeia solution A as the wound exudate simulant.[321] All plates were covered and sealed with parafilm prior to being placed on an orbital shaker rotating at 60 rpm at 35°C for up to 21 days. Insulin supernatants were measured using a human ELISA assay kit using human recombinant insulin as calibration standard. A cumulative release plot was constructed by adding the quantity of insulin released at each time point and expressing it as a ratio of the total insulin released over time versus the amount of insulin initially encapsulated within the microparticles. Supernatants were replaced with fresh solution A after each sampling.

4.4 Insulin Stability

A 40 mg/L insulin solution was prepared by dissolving crystalline insulin into solution A at room temperature. Approximately 40 mL insulin solution was transferred into a 50 mL conical vial (Becton Dickinson; Oakville, Canada) and placed on an orbital shaker rotating at 30 rpm, in an incubator set at 37°C. Sampling was performed daily by removing a 2 mL aliquot and storing it at -80°C until testing. After 5 days of sampling, the aliquots were quickly thawed from storage and assayed using a micro BCA and human insulin ELISA as per manufacturer's instructions.

4.5 Alginate Microbial Contamination Assay

A 2% alginate solution was prepared by thoroughly mixing high M alginate and distilled water using an impeller mixer. Using a sterile technique, 0.4 mL alginate solution, or distilled water were spread across 3 tryptase agar plates, and 3 blood agar plates using a glass rod. All plates were transferred to the incubator set at 37°C, inverted and colonies counted 48 h later.

4.6 Animal Burn Model

Female Sprague Dawley rats (200-240g) were obtained from the University of California Riverside vivarium and housed under a 12 hour light/dark cycle. All experimental protocols were approved by the UCR Institutional Animal Care and Use Committee, and animals were provided with fresh water and standard rat chow. Rats were anesthetized with a single injection of ketamine (40mg/kg) and xylazine (12mg/kg) intraperitoneally (IP). The depth of anesthesia prior to the burn, was performed by testing for a lack of responses to tail clamping and foot pinching. Once anesthetized, hair was removed from the back of the animal's neck with the aid of electric clippers and the use of Nair™ (Church & Dwight; Riverside, USA) to remove excess stubble. A burn with approximately 20-30% TBSA was performed by heating a brass cylinder with a surface area of 3.1 cm² in 80°C for approximately 3 minutes, and contacting the bare skin for 6 s. Immediately after the burn, an opioid analgesic, buprenorphine 0.05-0.1 mg/kg was administered to prevent post-burn pain. Each test group was comprised of 4 test (ASD containing insulin loaded microparticles) and 4 control animals (ASD containing blank microparticles).

4.6.1 Dressing Application and Monitoring

ASD consisting of 1% PEG 1450 and 2% high M alginate was prepared and sterilized under UV light for 20 min each side prior to application. 100 µL of sterile saline solution was applied to the contact surface of the ASD and trimmed to fit over the burn. A tegaderm patch (3M; London, Canada) was applied to secure the ASD in place over the burn. Dressings were changed every 2-3 days and wounds cleaned with sterile saline to prevent bacterial growth. Wound traces were performed by placing a cellophane sheet over the wound and tracing the margins of the wound until re-epithelialization was achieved. Photographs of the wound were taken using a digital camera at a distance of 4 cm. Blood sugar was monitored using an

electronic glucose meter (Bayer; Tarrytown, USA) and animal weight measured using a digital electronic scale (Mettler Toledo; Columbus, USA).

4.6.2 Tissue Histology

At the conclusion of the animal study, rats were sacrificed using CO₂ asphyxiation. Wound tissue was excised by cutting the skin around the edge of the wound including an area of 5mm of the surrounding unwounded area. Tissues were fixed in 4% paraformaldehyde for 2 h and incubated in 0.1 M glycine/PBS for 1 h followed by 15% and 30% sucrose prior to embedding in OCT (Tissue-Tek; Torrance, USA), freezing in a mixture of ethanol/dry ice, and stored at -80°C. The migration of keratinocytes with the tissue samples was measured in sections stained with Hematoxylin and Eosin. The length of the migration tongue was determined by the distance from the margin of the wound to the tip of the migrating keratinocytes.

4.7 Statistical Analysis

Statistical analysis was conducted with GraphPad Prism 5 and JMPTM. A one-way ANOVA, followed by a pairwise comparison was performed wherever appropriate with a p= 0.05.

Chapter 5

RESULTS AND DISCUSSION

The aim of the present study was to develop an effective method of accelerating burn wound healing by topical delivery of therapeutic levels of bioactive insulin. This study represents the first report of a wound dressing designed to provide sustained and controlled delivery of a bioactive therapeutic, in this case insulin, over a three week period, a time critical for tissue repair. The investigation was organized into three phases aimed at developing, characterizing and testing an effective insulin-eluting wound dressing.

Phase 1 was designed to develop a formulation approach, to achieve three weeks of continuous release of bioactive insulin from polymeric microparticles, at therapeutic levels. Scanning electron microscopy (SEM), differential scanning calorimetry (DSC), micro BCA and cell-based assays were used to characterize microparticle morphologies and insulin bioactivity.

The second phase of the study was designed to incorporate the polymeric microparticles into a dressing material, facilitating application to the surface of a wound, and enabling direct delivery of bioactive insulin to the wound site. Water (exudate) handling characteristics and insulin release kinetics were considered over a 3 week period, along with cell-based assays to assess insulin bioactivity. After an optimal dressing formulation was selected, it was advanced into the third phase of development.

In the final phase of the study, the efficacy of the designed wound dressing material, releasing bioactive insulin, was tested on burn wound healing, using a partial-thickness, rodent burn model. Wound healing was monitored using wound surface area measurements, photography and histology, while animal weight and blood sugar provided feedback on animal

health and insulin tolerance. This aspect of the study was conducted at the laboratory of Prof. Manuela Martins-Green in the Department of Cell Biology and Neuroscience, University of California, Riverside.

5.1 Phase 1 – Microparticle Development

Poly(lactic-co-glycolic acid) (PLGA) was selected as the encapsulation polymer, because of known slow drug release characteristics and biocompatibility.[322] The study began by testing different molecular weights of PLGA (5-75 kDa) and batch sizes (25-100 mg) to identify optimal combinations which would reliably emulsify and produce microparticles (Table 15). An oil-in-water (O/W) emulsion, solvent evaporation technique was used to formulate microparticles made from 5-15 kDa PLGA. Batch sizes of 25 mg and 50 mg resulted in significant microparticle loss during transferring and washing, providing a recovery of approximately 40%, whereas a batch size of 100 mg resulted in higher product recovery of approximately 95%. The study was repeated a second time using PLGA with a molecular weight of 40-75 kDa. All formulations failed to produce discrete microparticles and high molecular weight PLGA was discontinued from further development. During emulsification, the high molecular weight PLGA would stick to the impeller mixer, failing to disperse into the external aqueous phase. A final formulation consisting of 100 mg, 5-15 kDa PLGA was found to form stable and discrete microparticles with a recovery of nearly 95%. The formulation was therefore selected and carried forward into subsequent formulation studies with insulin.

Table 15. PLGA molecular weight and batch size combinations and outcomes.

PLGA Molecular Weight (kDa)	Batch Size (mg)	Outcome
5-15	25	Microparticles formed, significant product loss, Approximately <25% remaining
5-15	50	Microparticles formed, product loss noted, <40% remaining
5-15	100	Microparticles formed, little product loss, up to 90-95% remaining
40-75	25	No microparticles formed, thick polymer gel adhering to impeller
40-75	50	No microparticles formed, thick polymer gel adhering to impeller
40-75	100	No microparticles formed, thick polymer gel adhering to impeller

Insulin was then combined with PLGA in quantities ranging from 0, 2.5, 5 and 10% per 100mg batch. Initially, aqueous solutions of insulin (Novolin ge; Novo Nordisk, Canada) were added directly into the PLGA polymer/solvent mixture and emulsified. Repeated encapsulation attempts resulted in efficiencies of less 4%. Most of the insulin was being lost during the emulsification and solvent evaporation steps into the surrounding aqueous phase, a common challenge associated with PLGA microencapsulation of hydrophilic molecules.[323] To overcome this challenge, bovine and human insulin crystals were encapsulated resulting in encapsulation efficiencies of 78-100%. Lowering the temperature of the aqueous outer phase was found to help harden microparticles during the solvent evaporation phase, stabilizing the crystalline insulin entrapped within. Figure 23 shows the temperature profile of the PVA solution during the solvent evaporation phase. A low temperature within the first hour is critical in stabilizing the microparticles and helped to prevent coalescence from taking place. A final formulation methodology was established and used for all subsequent microparticle batches as shown in Figure 24.

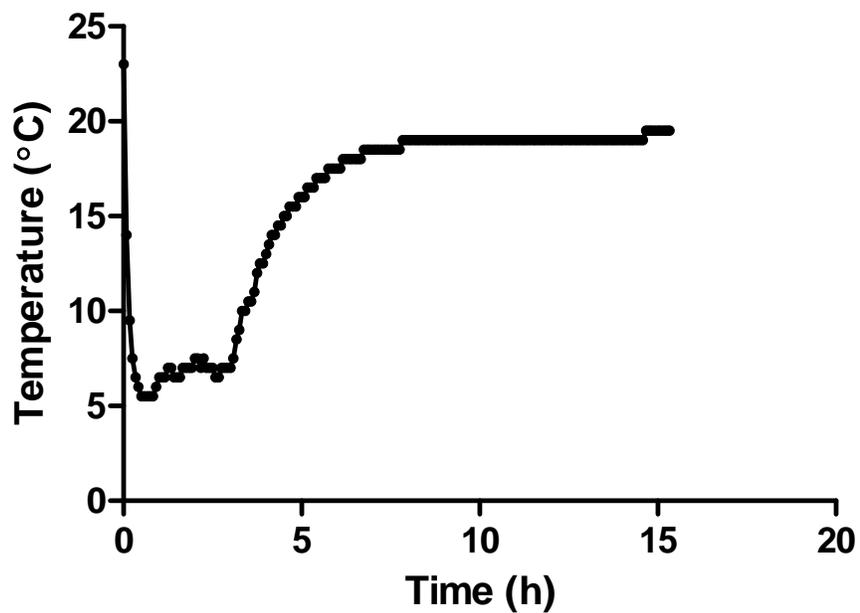


Figure 23. A temperature profile of the aqueous phase containing microparticles undergoing solvent evaporation under gentle stirring.

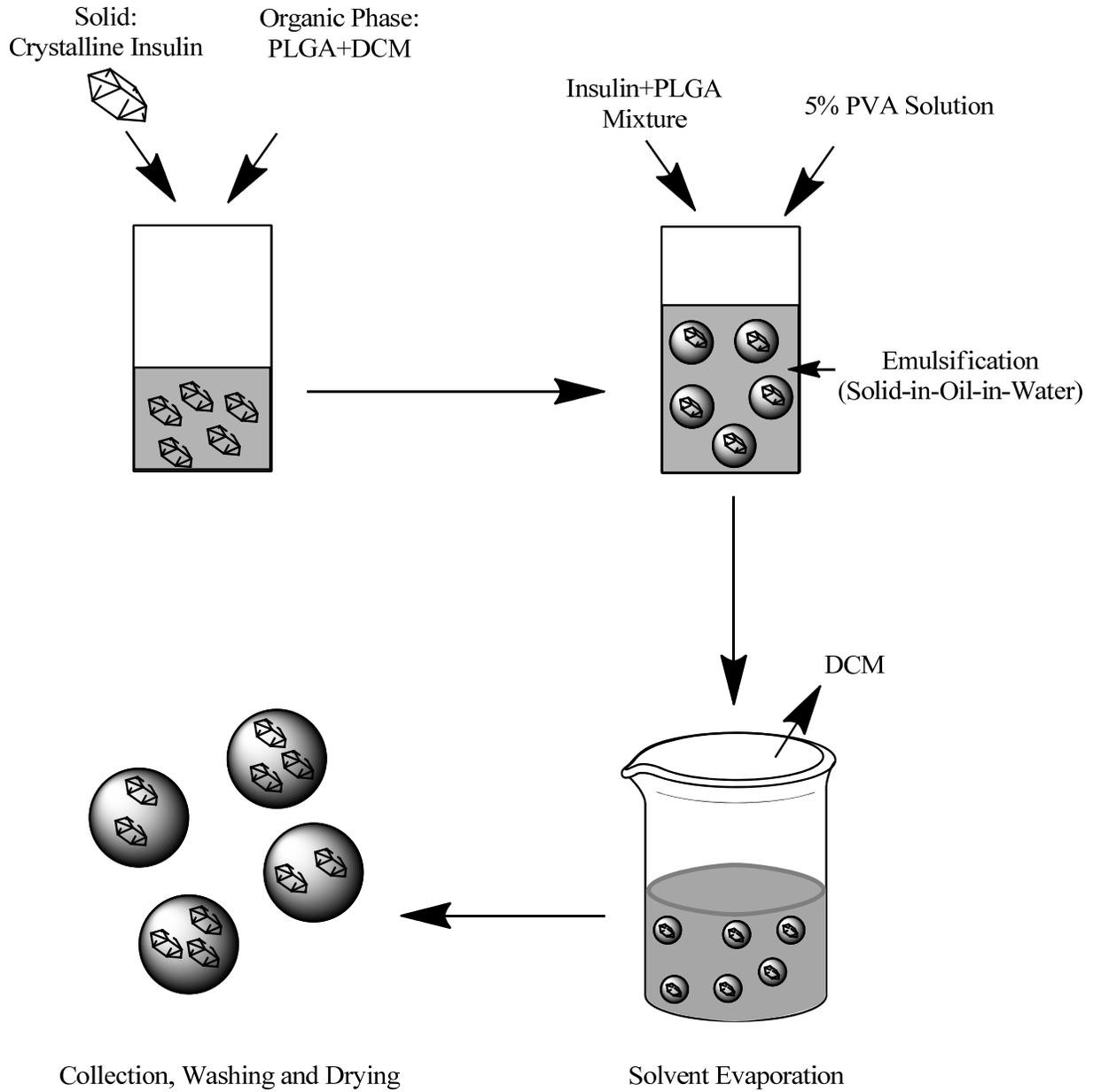


Figure 24. Preparation of insulin-loaded PLGA microparticles using a S/O/W emulsification, solvent evaporation technique. Dichloromethane (DCM) and poly(vinyl alcohol) (PVA).

5.1.1 Insulin Content in PLGA Microparticles

Microencapsulation of human recombinant crystalline insulin was performed using the S/O/W emulsion, solvent evaporation technique illustrated in Figure 24. Encapsulation efficiency was 99 ± 10 , 83 ± 8 and $78 \pm 1\%$ for loadings of 2.5, 5 and 10% insulin per batch respectively, showing that at the lowest loading, insulin was essentially fully entrapped.

5.1.2 Particle Size and Morphology

Morphologically, microparticles were discrete with spherical geometries as seen in Figures 25A-D. The surface appeared highly porous as seen in Figure 25E, and varying amounts of insulin crystals may be seen protruding from the surface of the microparticles in Figures 25B, C and D. The mean microparticle diameters for formulations containing 0, 2.5, 5 and 10% crystalline insulin were 154 ± 57 , 131 ± 92 , 153 ± 53 , and $137 \pm 52\mu\text{m}$ respectively. Mean pore diameters were 318 ± 121 , 282 ± 106 , 314 ± 126 and $387 \pm 145\text{nm}$ for microparticles with 0, 2.5, 5 and 10% insulin, respectively. According to Yang *et al.*, the high degree of porosity exhibited on the surface of the PLGA microparticles is a function of the low temperatures at which the microparticles are made. The authors reported that solvent diffusion occurs at a much slower rate from the hardening of microparticles at reduced temperatures.[298] This causes the polymer to remain semi-solid for a longer period of time, thereby allowing water from the outer aqueous phase to interact at the particle interface. The small water droplets become entrapped within the outer layer of the microparticle and either leave a porous structure toward the outer surface of the microparticle, or migrate into the microparticle as hardening occurs. Insulin crystals seen in Figure 25F with sharp and distinct edges and composed of various sizes. The mean crystal size was $4 \pm 5\mu\text{m}$.

TEM imaging revealed insulin crystals embedded within the polymer matrix of the microparticle, as well as at the periphery as seen in Figure 26A. A magnified view of the outer edge of the microparticles revealed a porous structure going well into the matrix, as illustrated in Figure 26B.

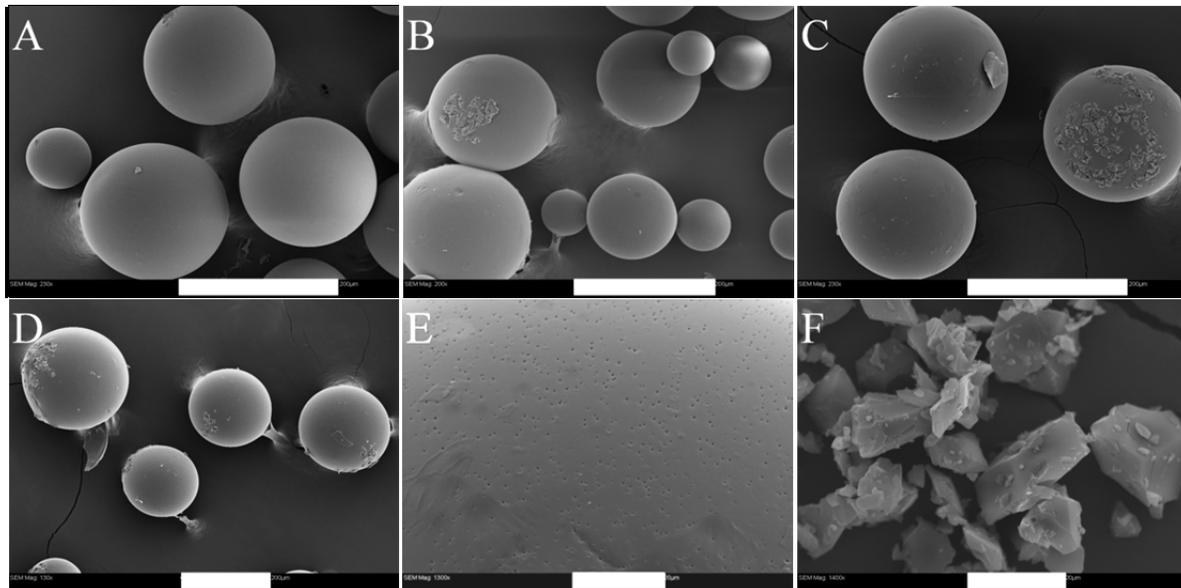


Figure 25. Shape and surface morphology of microparticles with A) 0%, B) 2.5%, C) 5%, and D) 10% insulin crystal loadings. A highly porous surface structure is visible in panel (E). Recombinant human insulin crystals are shown in (F). White size bar found at the bottom of each image represents 200 μ m in A-D, 20 μ m in E and F.

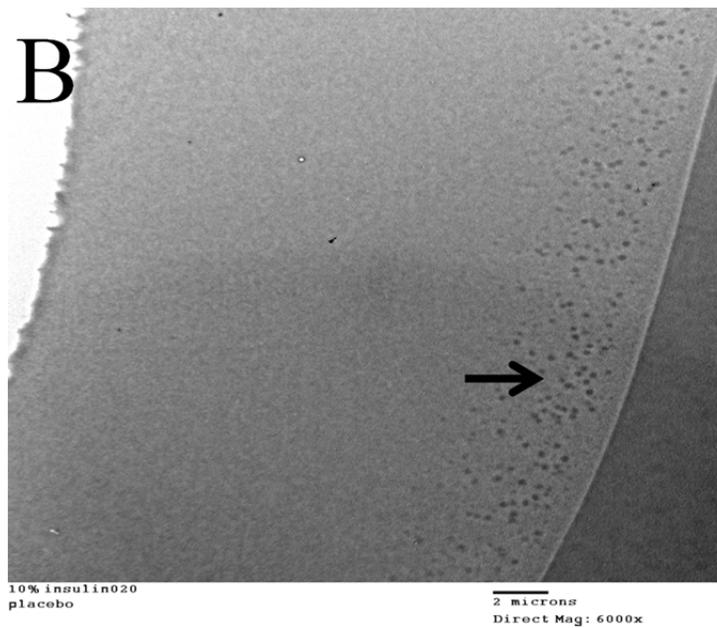
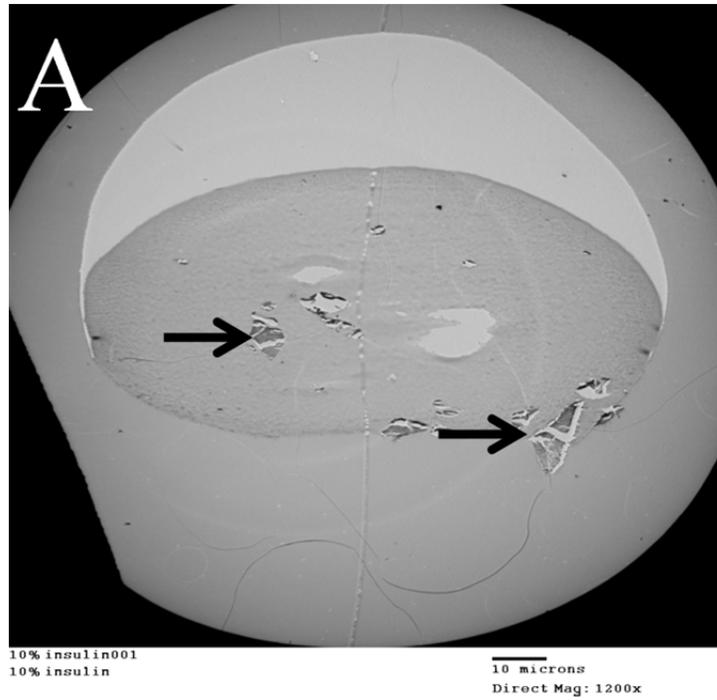


Figure 26. TEM cross sectional images of PLGA microparticles with 10% w/w crystalline insulin entrapment. Arrows point to insulin crystals embedded within, and extending out from the surface of PLGA microparticles (A). The highly porous outer region of the PLGA microparticle is highlighted by an arrow (B). Size bars represent 10 μ m in A and 2 μ m in B.

5.1.3 DSC Analysis

DSC analysis was performed to determine the effect of increasing insulin crystal content on the glass transition temperature (T_g) of fabricated PLGA microparticles. Thermograms in Figure 27 illustrate T_g of 13.2°C, 19.4°C, 20.8°C, and 24.0°C for microparticles with 0, 2.5, 5 and 10% insulin respectively. During formulation, PLGA microparticles with highest crystalline insulin content were noticeably much more rigid when being handled at room temperature. Bouissou *et al.* [324] reported T_g of W/O/W formulated PLGA microparticles of 20-29°C depending on the addition of surfactants in the formulation. The authors also reported that an added peptide acted like an anti-plasticizing agent, raising the T_g . Higher levels of crystalline insulin in the present study may have the same effect. Okada *et al.* (1994) also reported an increase in T_g with an increase in drug content.[325] Using leuprorelin acetate, a peptide-like analogue, the authors suggested that the ionic interaction between basic amino acids and the terminal carboxylic acid groups of the polymer result in stronger molecular interactions that elevate the T_g . The relationship between T_g and drug content provides insight into approaches toward manufacturing, handling and storage of microparticles to prevent aggregation and irreversible coalescence.

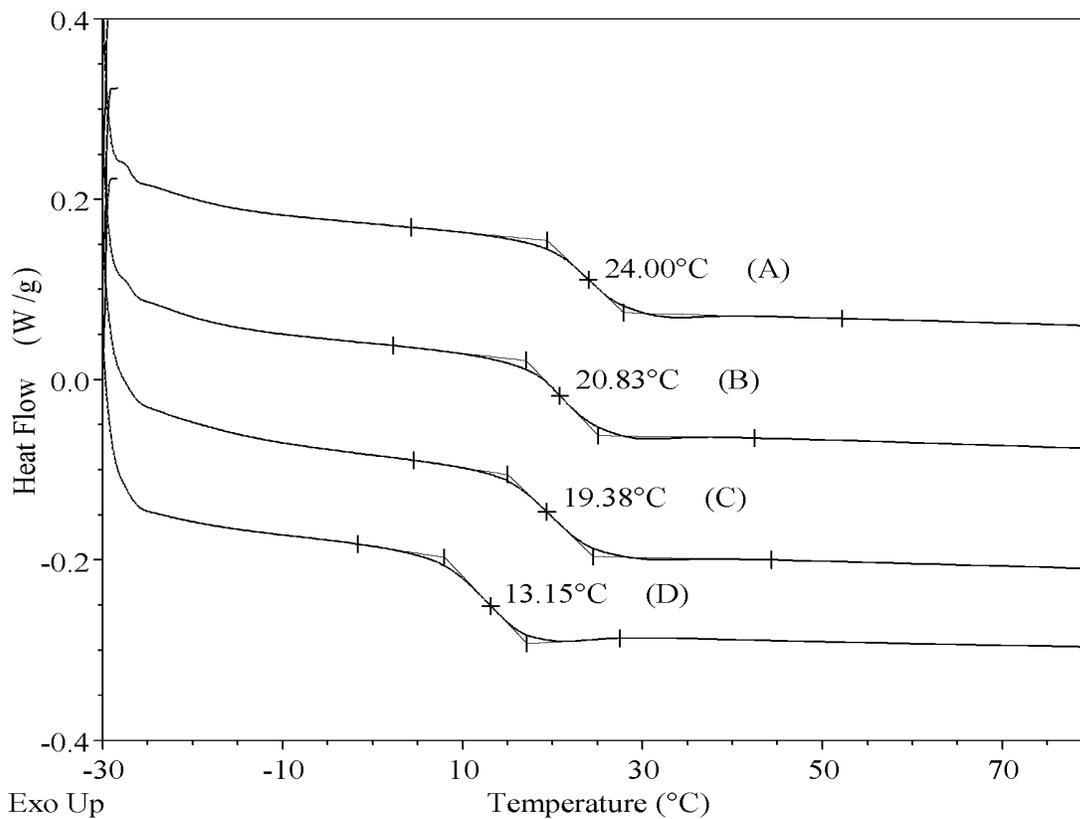


Figure 27. DSC thermograms for PLGA microparticles loaded with A) 10%, B) 5%, C) 2.5%, and D) 0% crystalline insulin.

5.1.4 Insulin Dissolution

Insulin crystals dissolved instantaneously within 37°C phosphate buffer, and the insulin solution was then used to construct a standard curve using a micro BCA assay as shown in Figure 28. This calibration curve was subsequently used to assay for insulin release from microparticles.

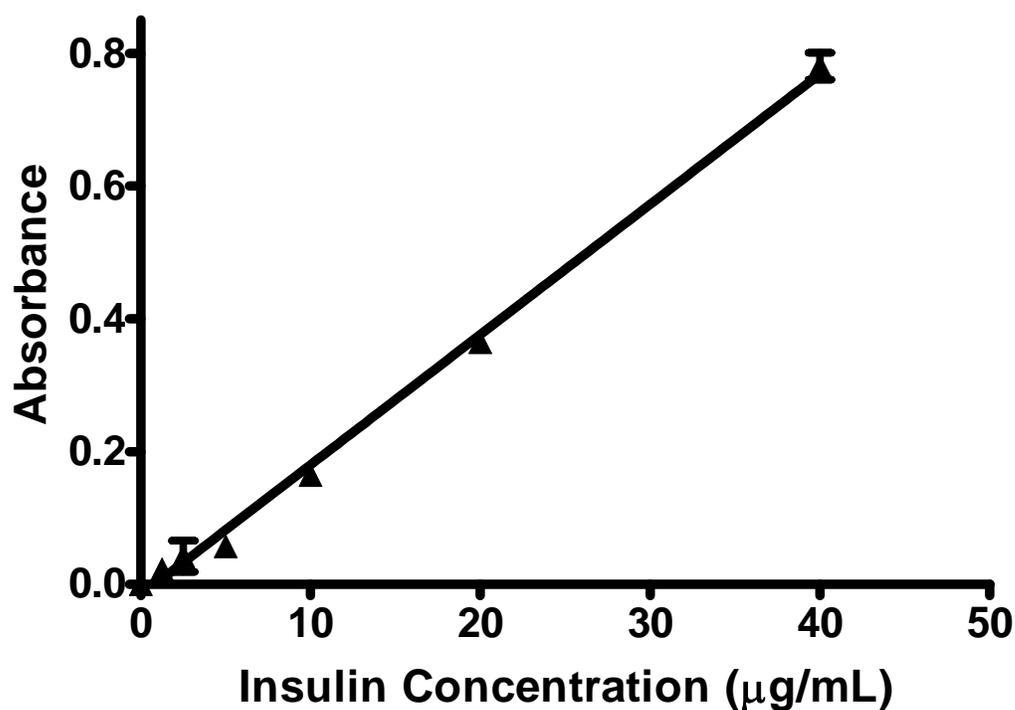


Figure 28. A micro BCA assay calibration curve for insulin standards prepared between 1.25-40 µg/mL.

5.1.5 Insulin Release Kinetics

Insulin release kinetics revealed that all microparticle formulations achieved nearly 100% insulin release over a 25 day period as seen in Figure 29. A burst release of 28-46% was observed within the first 24h of the assay, followed by a period of near zero order release kinetics for approximately 10 days thereafter. The rapid dissolution of insulin from crystals exposed on the surface of the microparticles causes an initial rapid release of insulin into solution, resulting in a burst effect in the release profile. Figure 25 confirms the presence of large insulin crystals embedded on the surface of the microparticles, making it possible for rapid dissolution to take place.

After the burst phase, the formation of pores and channels within the microparticle matrix caused by PLGA hydrolysis, resulted in a constant and near-zero order release of insulin for about 10 days. This process is governed by insulin diffusion through pores and channels in the microparticle and erosion controlled release by polymer hydrolysis.[326] By days 12-14, all formulations exhibited a rapid increase in release of approximately 27-29% before reaching a plateau by day 25. Thus the overall release behavior appeared to be biphasic. Li *et al.* 2008,[327] reported a similar secondary burst effect of methotrexate loaded PLGA microparticles and attribute the secondary burst phase to changes in the physicochemical properties of the PLGA matrix. During the later stages of PLGA degradation, the porosity and formation of internal channels increases from PLGA polymer weight loss attributed to polymer hydrolysis. [327,328,329] This causes drug diffusion to suddenly increase, resulting in a secondary burst to occur toward the end of the release kinetics profile.[330]

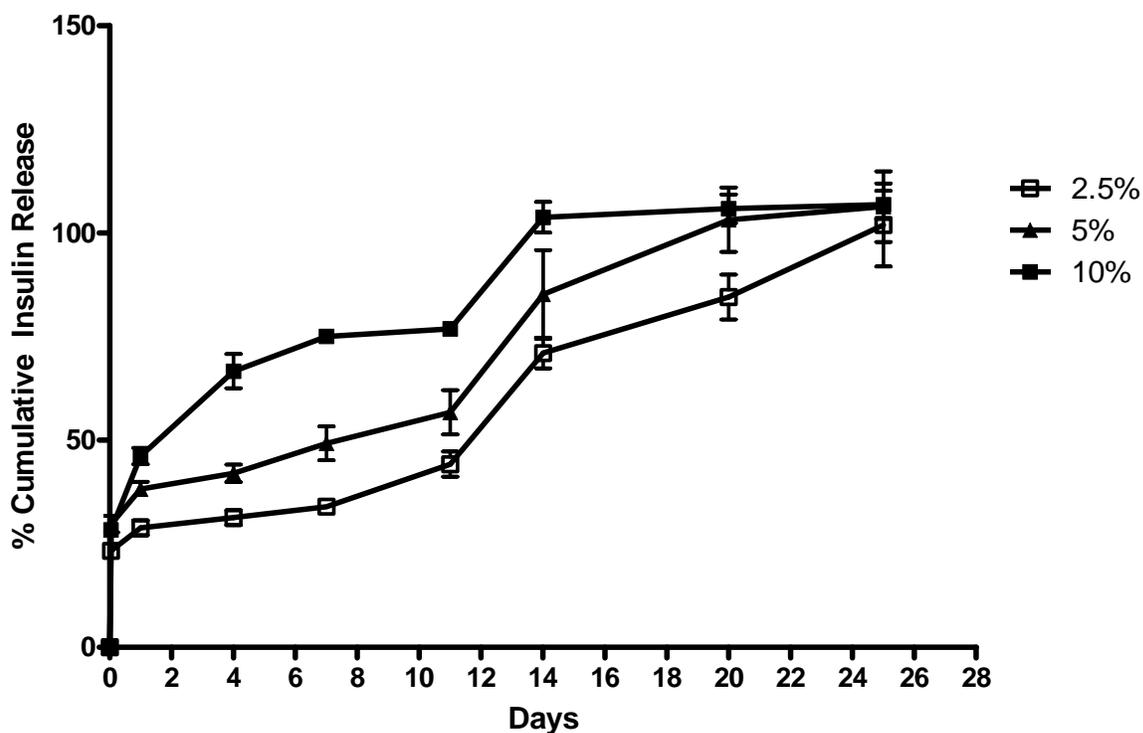


Figure 29. Insulin release kinetics from PLGA microparticles containing 2.5-10% insulin.

5.1.6 Insulin Bioactivity: Cellular AKT Phosphorylation Assay

Bioactivity of released insulin was assessed by measuring the induction of the cellular insulin receptor signaling pathway in rat L-6 myoblasts using a fast activated cell-based AKT ELISA assay. When insulin binds to insulin-receptor 1, the receptor dimerizes and autophosphorylates, causing downstream signaling molecules such as AKT to also become phosphorylated.[331] AKT is an important downstream signaling molecule because it is responsible for inducing cellular events such as cellular migration, proliferation and other survival responses to occur.[153,332,333] Unlike chemical structural assay methods such as HPLC-MS, and UV circular dichroism which provide evidence of primary/secondary structural integrity, or the presence of degradation products, [310,334,335] ELISA assays directly assess

the induction of the insulin receptor signal transduction pathway, which is only achievable through a biologically active peptide molecule.

By comparing the levels of AKT phosphorylation relative to the activity of a fresh insulin control at the same protein concentration, insulin bioactivity could be monitored over the 23 day release period as shown in Figure 30. Released insulin on day 1 was fully active for all formulations, demonstrating that insulin biological activity was fully retained during and following the encapsulation protocol. The activity of released insulin then dropped gradually over the 23 day release period, yet activity remained to the end of 23 days. In general, microparticles with the lowest loading of insulin showed the highest level of released activity to 14 days. A pairwise comparison for all bioactivities showed that they were statistically significant ($p < 0.05$) when compared to placebo microparticle supernatant at time points up to day 23. A Welch test of means, which accounts for non-equal variances, concluded a significant difference between the sample and placebo groups. By day 23, the activity of released insulin ranged from 25- 37%, depending on the formulation. Loss of insulin activity may be attributed to the acidic microenvironment within the particle due to PLGA hydrolysis [316].

AKT phosphorylation not only indicates the direct bioactivity of insulin, but sheds light on the potential for wound healing applications. The use of a novel encapsulation technique incorporating crystalline insulin seems to suggest that the conditions are much milder than more conventional W/O/W preparation techniques and could be beneficial for other protein therapeutic formulations. Overall, active insulin continues to be released beyond a three week period, indicating that long term PLGA mediated insulin delivery is a viable option.

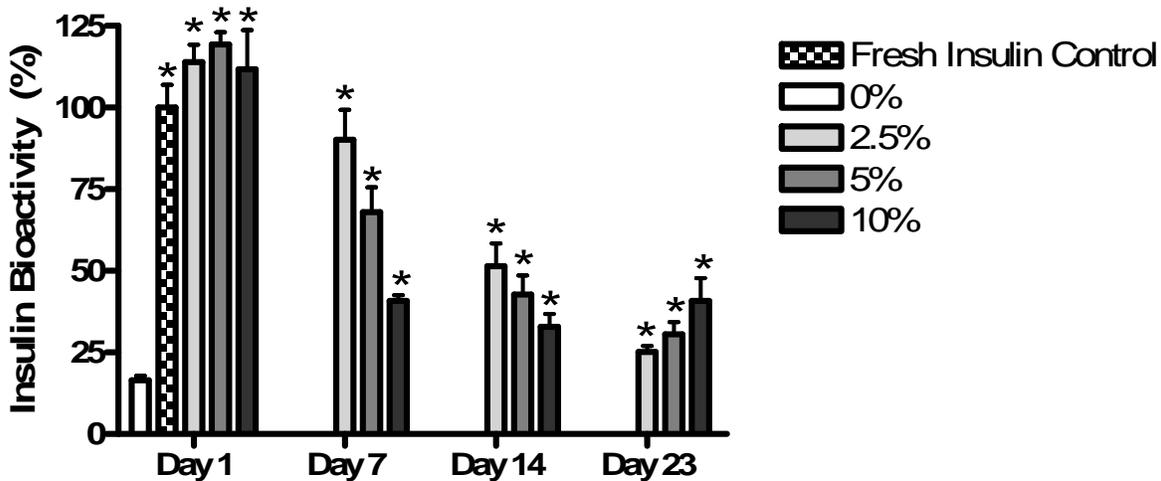


Figure 30. Insulin bioactivity measurements for PLGA microparticle formulations with 0-10% w/w crystalline insulin. Asterisks indicate statistical significance between placebo and sample ($p=0.05$).

5.1.7 Insulin Bioactivity: Scratch Assay

Scratch repair assays were performed on human HaCaT cells to further investigate the bioactivity of released insulin. The approximate initial width of the applied scratch was 0.5 mm, which filled rapidly due to cell migration, stimulated by applied insulin as may be seen in Figure 31. Unlike the AKT ELISA assay that provides information about the ability to induce an insulin stimulated signal cascade, the scratch assay provides a measure of cell migration, a process critical for wound healing. Using this model, it was possible to observe whether insulin bioactivity changes, or remains stable over time of release by comparing microparticle supernatants collected from days 1, 7, 14 and 23. The amount of insulin-protein applied to the tissue scratch at the beginning of the assay was 10^{-7} M for all conditions, with the exception of supernatant from blank microparticles (0% insulin control), in which case protein free supernatant was assayed to determine if there was a tissue response to PLGA degradation products. Figure 31A, B and C represent assays for insulin released from 2.5, 5 and 10% insulin

loaded microparticles, respectively. Assay of supernatant from 2.5% insulin loaded microparticles was most problematic because of the very small quantities of insulin being released on individual days, which may explain the irregular response as was observed for example, with insulin sampled on day 14. Clearer and more consistent trends are evident with insulin released from 5 and 10% insulin microparticles where standardization of insulin-protein was more accurately conducted. Figure 32 shows photomicrographs taken of the migrating HaCaT cells after being exposed to insulin released into supernatants obtained from 10% insulin loaded PLGA microparticles.

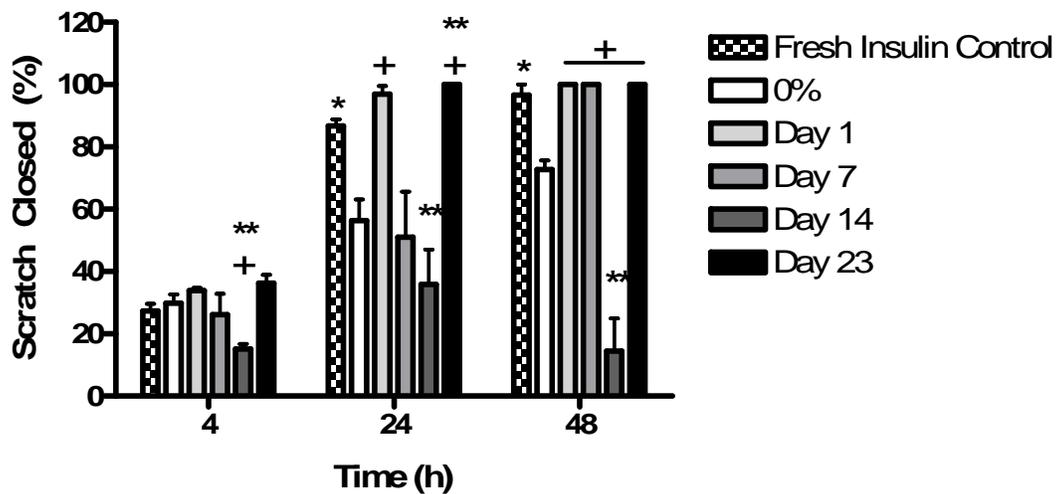
The most rapid cell migration in the scratch assay took place between 4 and 24h for cells exposed to 10^{-7} M fresh insulin, with cells approaching confluence at 24h, and fully confluent at 48 h. Cell migration was also evident in the absence of insulin, but the differences between insulin exposed cells and cells growing in absence of insulin were large, and significantly different based on a one-way ANOVA test followed by a pairwise comparison, beyond 4h of stimulation. For example, at 24h, percent scratch closure was 84% for cells exposed to fresh insulin in comparison to 46% for cells migrating in absence of insulin. At 48h, the scratch was fully closed in the presence of fresh insulin, in comparison to 67% for cells growing in the absence of insulin. Thus it is clear that insulin strongly stimulates growth and migration of HaCaT cells.

A comparison may then be drawn between cell migration in presence of insulin released from PLGA microparticles over the 23 day period, to that of fresh insulin. By 48h, the scratch was fully closed for all but one supernatant insulin sample. In comparison to the fresh insulin control, there was no significant difference in tissue response at 4, 24 or 48h for all remaining supernatant samples. It would then appear that supernatant insulin released to 23 days from all

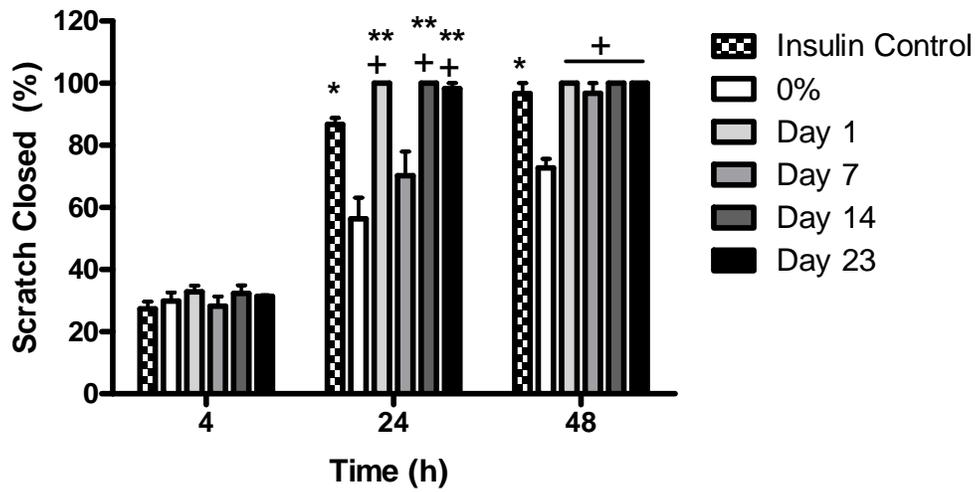
formulations, was fully active, in comparison to fresh insulin when evaluated in terms of tissue migration response.

Overall, most insulin containing supernatants resulted in full scratch closure within 48 h, and at a much faster rate than in the absence of insulin, as was observed with supernatants from blank microparticles. This indicates that released insulin was bioactive, and PLGA degradation products did not have a stimulatory effect. Therefore, we conclude that the cellular response was due to the addition of insulin alone. Thus these results show that insulin released over a 23 day period stimulates the same cell migration response as fresh insulin, indicating a high level of sustained bioactivity.

A



B



C

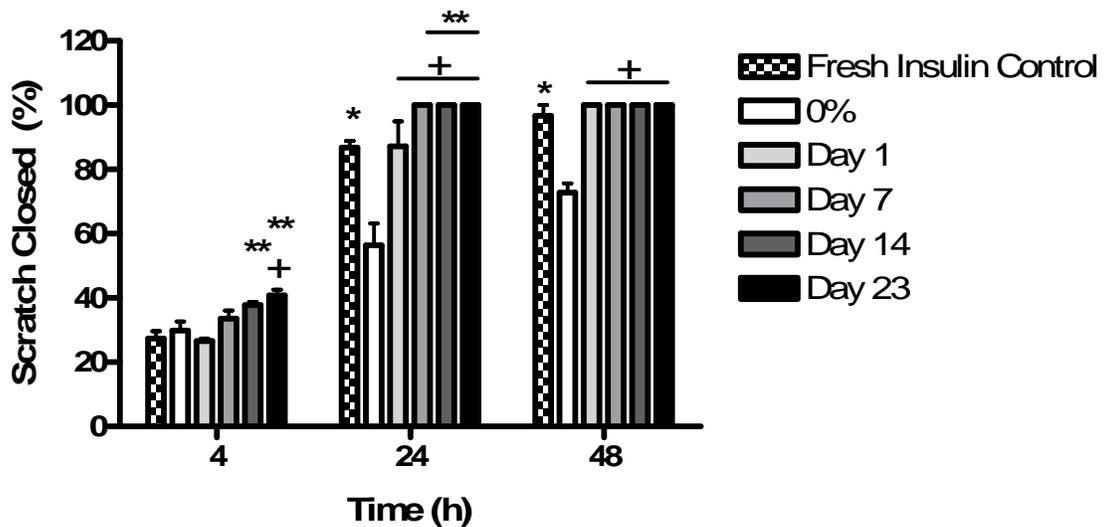
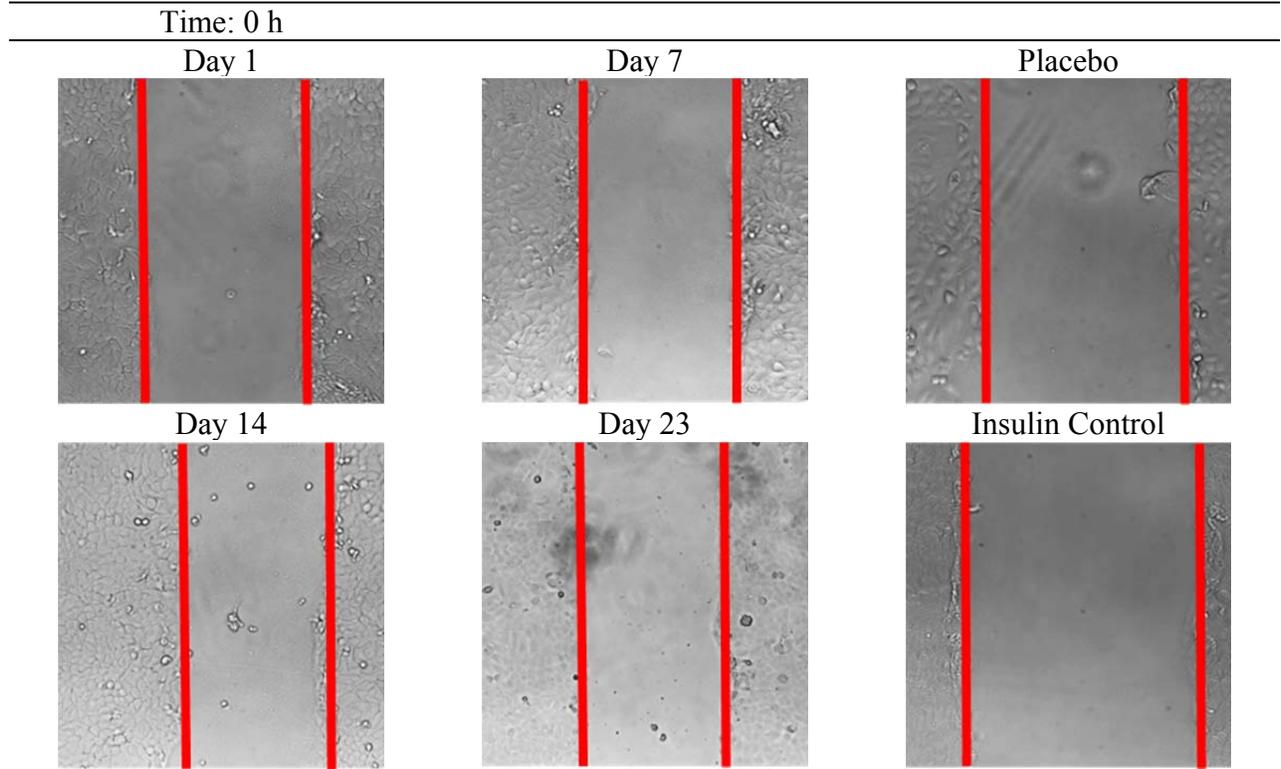
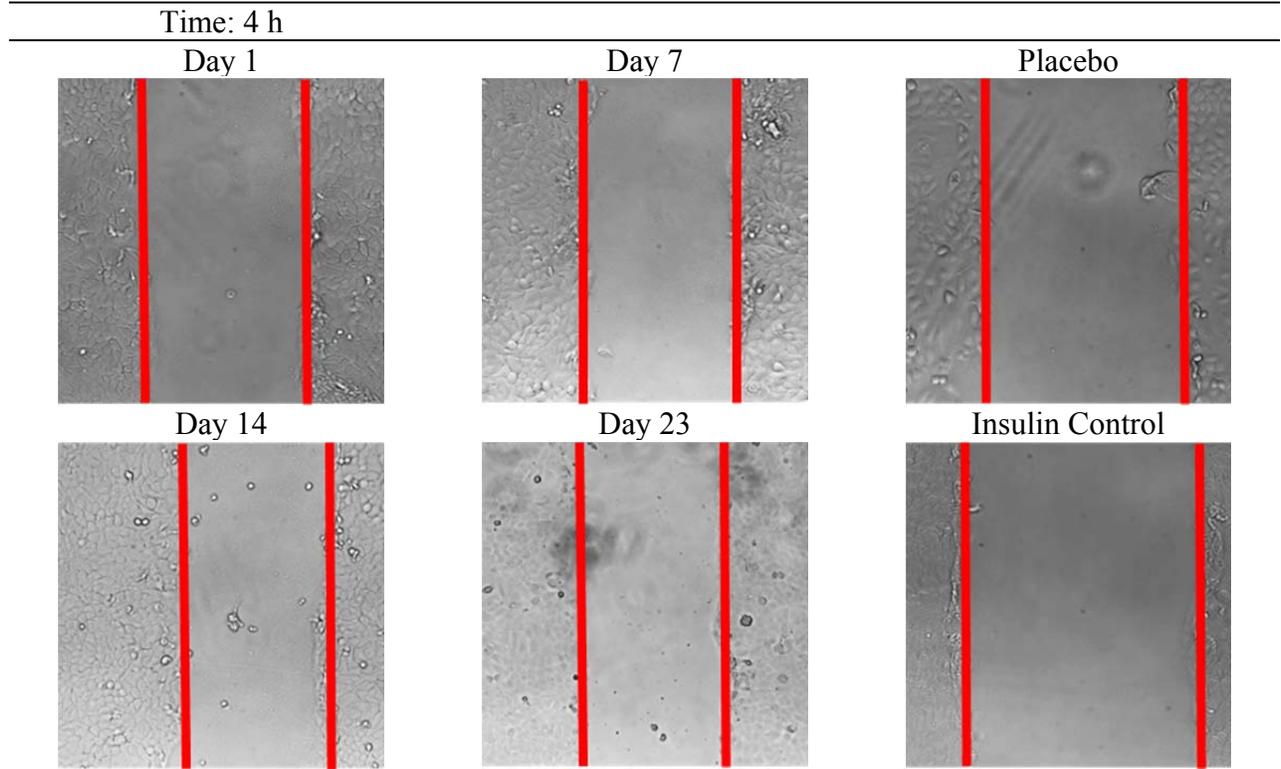


Figure 31. HaCaT cell scratch repair assays evaluating bioactivity of insulin released from 2.5 (A), 5 (B), and 10% (C) insulin loaded PLGA microparticles into supernatants sampled on days 1, 7, 14 and 23 days of sustained release. Single asterisks indicate statistical significance between fresh insulin control and insulin free PLGA supernatant, double asterisk indicates difference between insulin released to supernatant, and fresh insulin control, and plus sign indicates difference between insulin released to supernatant and insulin free PLGA supernatant ($p=0.05$).

A



B



C

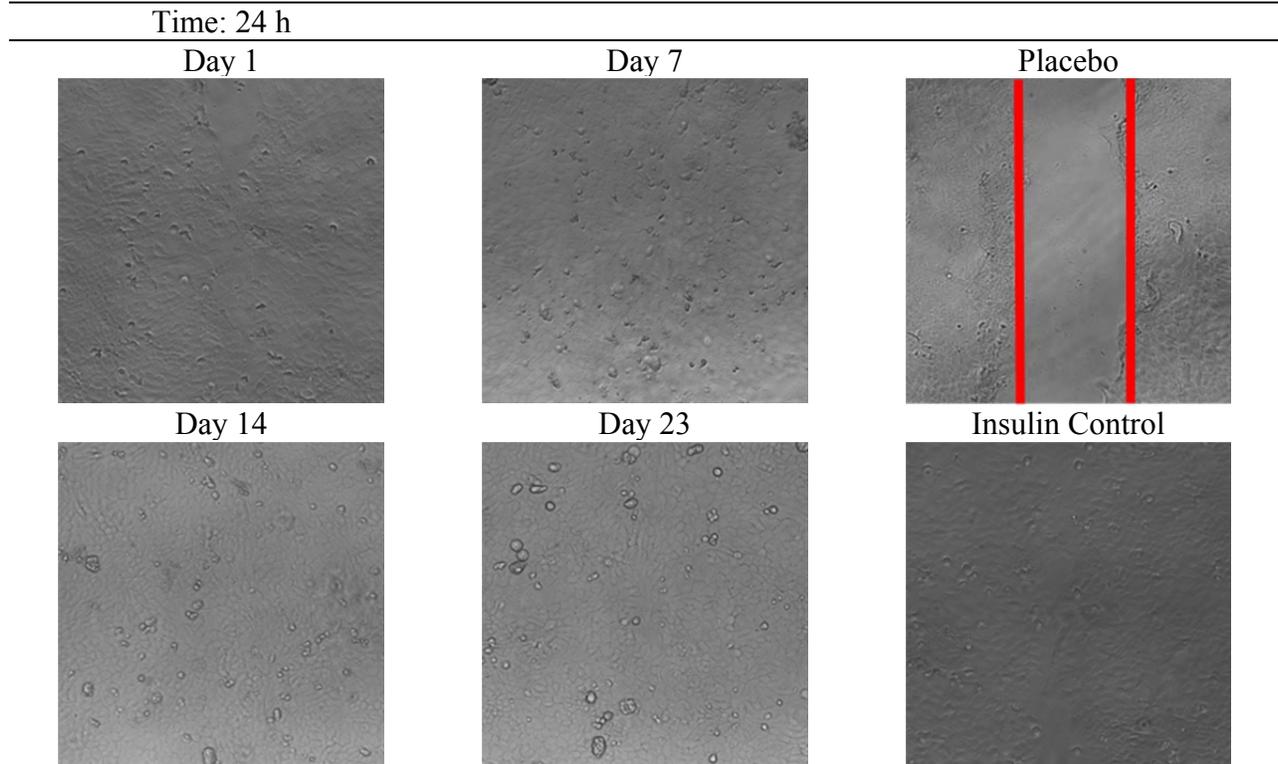


Figure 32. Phase contrast micrographs of HaCaT scratch wounds with supernatant from 10% insulin-loaded microparticles from days 1, 7, 14 and 23, placebo and fresh insulin control solutions after A) 0 h, B) 4 h and C) 24 h. The vertical lines indicate the margins of the scratch.

5.2 Phase 2 - Alginate Sponge Dressing (ASD) Development

Previously, crystalline insulin was encapsulated into PLGA microparticles which functioned as a drug depot device, having the capability to slowly and controllably deliver bioactive insulin over a period of 1 month. Microparticles alone are difficult to homogeneously spread across a wound and secure in position for extended periods. Microparticles also have no capacity to function as a protective barrier, or to provide an optimal, moist wound healing environment. A polymer matrix with the ability to protect the wound, absorb exudate, provide a moist healing environment, and permit insulin diffusion was needed.

A candidate composite polymer blend consisting of alginate and poly(ethylene glycol) (PEG) was selected as the constituents of the matrix formulation. These two polymers were selected based upon their inert chemical properties, suitable biocompatibility, and hygroscopic characteristics.

Formulation development began by defining the process parameters necessary to produce a robust matrix which could support the incorporation of insulin-loaded microparticles. During development, design criteria were established to help guide formulation design. Some of the characteristics included the ability for the dressing to be stored dry, and upon hydration, for a hydrogel that would facilitate insulin diffusion, and PLGA degradation with time. The matrix also had to endure ambient temperature storage conditions, resist significant mechanical stress and conform around uneven skin surfaces. Optimization of the polymer concentrations, manufacturing techniques and process parameters would enable the formulation of a basic alginate wound dressing.

An alginate sponge dressing (ASD) made from freeze-dried solutions of alginate and PEG were selected as the matrix material to support incorporation of insulin-loaded

microparticles. ASD was formulated by preparing alginate solutions of 0.3-2.5% and filling small plastic petri dishes with approximately 5-15 mL of the various alginate solutions. ASDs were then produced by lyophilizing the solution, resulting in circular ASD as seen in Figure 33. ASD prepared from fill volumes of ≥ 10 mL and alginate concentrations of $\geq 1\%$, were flexible, uniform and resistant to damage. ASD prepared from alginate solutions $< 1\%$ and fill volumes < 10 mL were non-uniform in morphology and significantly more fragile. In some cases, large cracks and holes were formed in the center of the ASD prepared with 5 mL 0.3% alginate as seen in Figure 34.

Dilute alginate solutions (i.e., < 5 mL) and low fill volumes (i.e., 0.3%) caused a meniscus to form within the petri dish as seen in Figure 34. As the solutions froze, the shape of the meniscus was retained after lyophilisation. The thinner central portion was delicate and broke with ease, suggesting that an optimal fill volume and concentration was necessary to avoid this problem. A fill volume of at least 10 mL and an alginate concentration of 1% consistently prevented thinning from occurring.

Water absorption was tested by placing ASD into distilled water, to observe the behaviour of the ASD over several hours. All samples absorbed water rapidly and remained intact for approximately 4-5h. Beyond 5h, ASD began to slowly dissolve and leave behind small fragments. With time, the hydrogel network dissociated and resulted in the complete loss of the ASD. Cross-linking the alginate hydrogel was tested in the subsequent study, to determine if this approach would help retain the alginate hydrogel structure over a period of time exceeding 5h.



Figure 33. ASD prepared from alginate concentrations 0.3-2.5%, and fill volumes ranging from 5-15 mL. ASD prepared with 0.3% alginate solutions and low fill volumes of less than 10 mL were often delicate and contained numerous large cracks.



Figure 34. An ASD prepared with 5 mL 0.3% alginate solution. Note the thinning of the center of the ASD, resulting from the formation of a meniscus during the pouring and lyophilizing process.

ASD prepared from 1-4mL, 0.25, 05, 1 and 2% alginate solutions were cross-linked by rapidly dipping ASD into a solution of calcium chloride, followed by re-lyophilization. Figure 35 shows the dry ASD prior to being crosslinked, and Figure 36 shows the appearance of ASD after cross-linking with calcium and re-lyophilization. All ASD shown in Figure 37 after removal from the molds, became brilliant white after cross-linking with calcium, and were significantly more brittle and stiffer, as opposed to non-cross-linked ASD. Fill volumes and alginate concentration had no immediate effect on the ASD, other than changing the dimensions of the ASD. Porosity was completely lost in all samples upon closer examination. The loss of porosity prevented ASD from swelling, even after sitting in distilled water for 4 h. Cross-linking with calcium chloride resulted in an ASD which was stiff, unable to swell to form a hydrogel and brittle. Cross-linking ASD with calcium was determined not to benefit the ASD formulation and was dropped from further investigation.



Figure 35. Dry ASD prior to being cross-linked with calcium. Samples 1-4, 5-6, 9-12 and 13-16 were prepared from 2, 1, 0.5 and 0.25% alginate solutions and fill volumes of 1-4mL respectively.



Figure 36. Dry ASD cross-linked with calcium. Samples 1-4, 5-6, 9-12 and 13-16 were prepared from 2, 1, 0.5 and 0.25% alginate solutions and fill volumes of 1-4mL respectively.



Figure 37. Calcium cross-linked ASD prepared from 2% alginate and fill volumes ranging from 1-4mL (left to right). Note the brilliant white appearance and warped morphology after being cross-linked and lyophilized a second time.

The last approach combined PEG with alginate. PEG with a molecular weight of 1-10 kDa and concentrations of 0.1-10% was tested. The sodium alginate (Sigma-Aldrich) previously used, was also substituted for high mannuronic (high M) and high guluronic (high G) alginates (FMC Biopolymer) because of its purity and different chemical compositions. ASD was produced using a similar method as described before. Alginate, water, PEG and insulin-loaded PLGA microparticles were combined together and poured into multi-well plates. All samples were frozen and lyophilized using a method described in Figure 38. Each sample was then tested to measure the effects of formulation characteristics on insulin release kinetics and bioactivity.

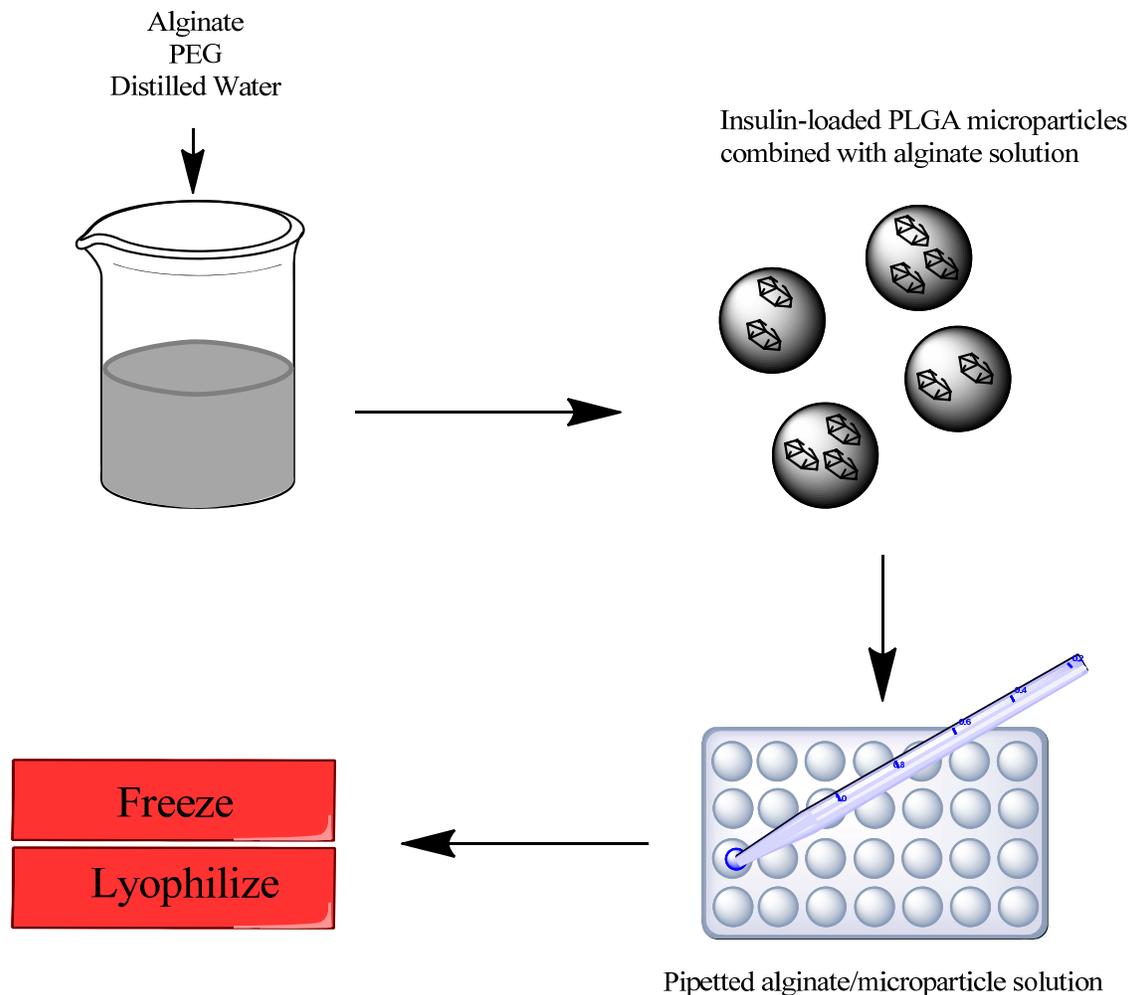


Figure 38. Alginate, PEG, distilled water and insulin-loaded PLGA microparticles are transferred into a 24 well-plate. The plate is then frozen at -80°C and lyophilized overnight to yield circular ASD disks.

5.2.1 Encapsulation Efficiency

Prior to incorporating insulin-loaded PLGA microparticles into the ASD, the amount of insulin loaded was measured. Three microparticle batches were prepared and the mean crystalline insulin loading of 0.04mg/mg of microparticle determined, representing an $80 \pm 7\%$ encapsulation efficiency.

5.2.2 Alginate Sponge Dressing (ASD) Morphology

Strength is a critical component of wound dressing and can be influenced by the morphology and chemical composition of a material. Scanning electron micrographs in Figure 39A-D, revealed a complex, porous microstructure in both high M ASD (M-ASD) and high G (G-ASD) with clusters of microparticles embedded within the lamellar structure. Distinct differences in morphology were observed between M-ASD (Figures 39A-B) and G-ASD (Figures 39C-D). M-ASD revealed a highly corrugated and sharp lamellar structure with pores having an internal diameter ranging from approximately 6-105 μm . G-ASD on the other hand showed smoother, fold-like lamellar structures with the absence of distinct open porous spaces.

ASD prepared with 0.1-10 % PEG at molecular weight 1.45 kDa (Figure 40A-C), or 10 kDa (Figure 40D-F) were also examined using SEM. The dressings with lowest concentration of 1.45 kDa PEG displayed almost no morphological difference to PEG free M-ASD, revealing little bulking effect by PEG incorporation. As the PEG concentration increased to 1% (Figure 40B), the corrugated surfaces of the sharp lamellar internal structures became increasingly larger, and an overall bulking effect of the foam network was observed. As the 1.45 kDa PEG content was increased to 10% (Figure 40C), the porous lamellar sponge network became covered by a dense, outer layer of PEG, filling in much of the porous internal spaces.

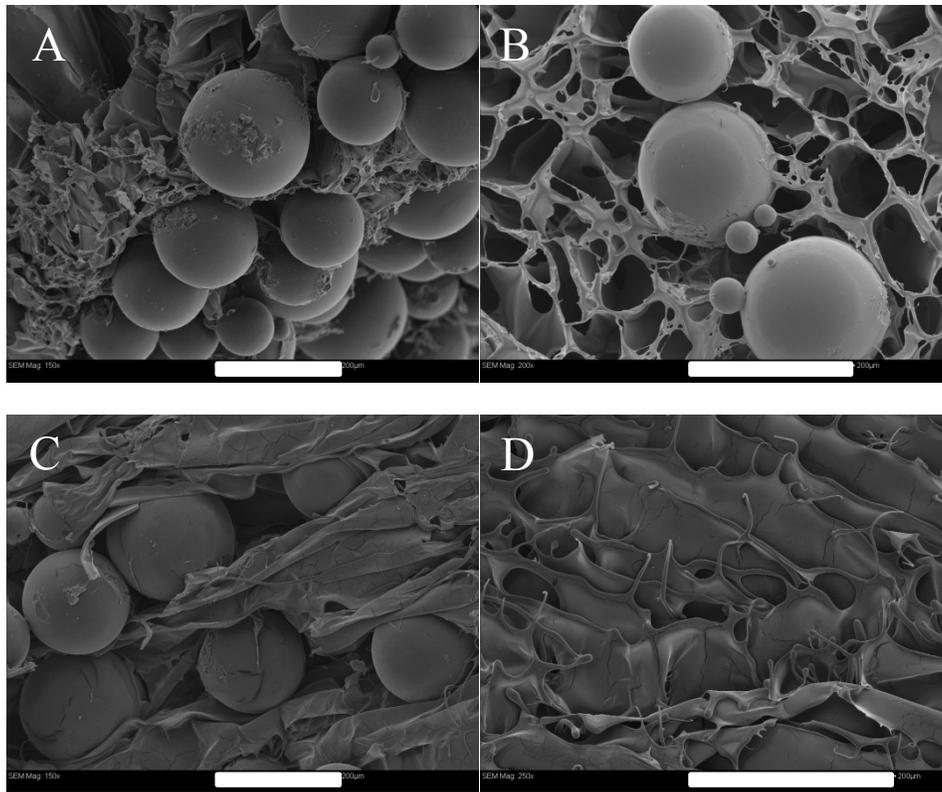


Figure 39. Cross-sectional SEM images of 2% M- (A), 4% M- (B), 2% G- (C), and 4% G- (D) ASD loaded with insulin-PLGA microparticles. White scale bars represent 200 μ m.

Figures 40D-F show the effects of incorporating 0.1-10% PEG 10 kDa into high M alginate sponge dressing (P/M-ASD). An immediate bulking effect was apparent with the incorporation of higher molecular weight 0.1% 10 kDa PEG (Figure 40D). The internal lamellar structure supporting the pores appeared bulkier and smoother, with less of a corrugated arrangement than PEG-free M-ASD. As the concentration of 10 kDa PEG increased to 1%, the open cell internal structure of the P/M-ASD transitioned into aligned and thick, layered sheets of alginate and PEG (Figure 40E). Finally, as the concentration of 10 kDa PEG increased to 10% (Figure 40F), the internal open cell structures were completely lost and the P/M-ASD became a solid mass.

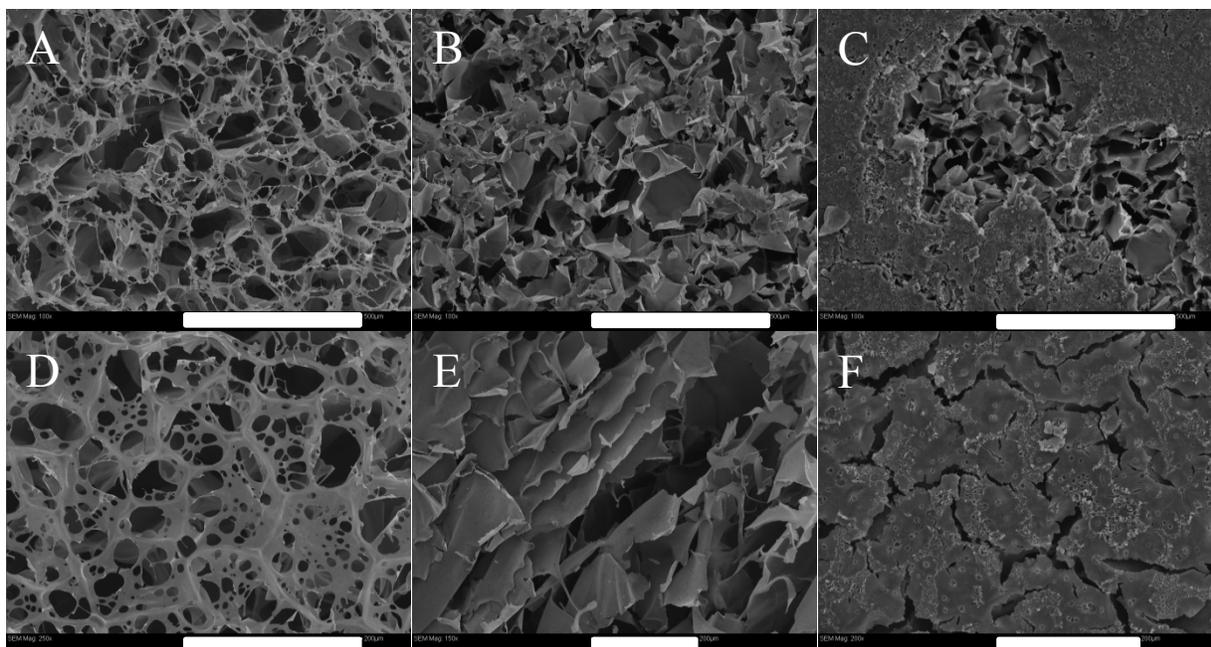


Figure 40. Cross sectional SEM images of P/M-ASD prepared with 1.45 kDa PEG at concentrations of 0.1 (A), 1 (B) and 10% (C). Images of P/M-ASD with 10 kDa PEG at concentrations of 0.1 (D), 2 (E) and 10% (F). White scale bars in images A-C and D-F represent 500 and 200 μm respectively.

Scanning electron microscopy revealed that for the most part, both M- and G-ASD had similar, thin reticulating pore microstructures. As more alginate and PEG were added to the formulation, the pore walls became noticeably bulkier in size. Shapiro *et al.* (1997) reported that as the overall polymer content of a porous material increases, the additional polymer adds to the bulkiness of the pore walls.[336] Alginate and PEG content therefore played a role in contributing to the morphology of the pore walls within ASD.

5.2.3 Density of Sponge Dressings

Densities of ASD were measured and are summarized in Table 16. In both M- and G-ASP, the density doubled in response to a doubling in initial alginate solution concentration from 2-4% in the preparation of the sponge dressing. The effect of adding various concentrations of PEG with different molecular weights, into G-ASD (formed with 2% alginate), also had an effect

on final ASD density as shown in Table 17. As PEG concentration increased in P/M-ASD, the density did so as well. PEG molecular weight did not significantly affect the density of the P/M-ASD between 1.45 to 10kDa.

Table 16. Densities of M- and G-ASD.

	2% M-ASD	4% M-ASD	2% G-ASD	4% G-ASD
<i>Density (mg/cm³)</i>	26 ± 3	45 ± 3	24 ± 1	43 ± 0.2

Table 17. Density P/M-ASD with 0.1-10%, 1.45 kDa or 10kDa PEG.

<i>Percent w/v PEG</i>	PEG 1.45 kDa			PEG 10 kDa		
	0.1	1	10	0.1	1	10
<i>Density (mg/cm³)</i>	26 ± 1	32 ± 4	110 ± 7	27 ± 2	36 ± 4	122 ± 4

5.2.4 Tensile Strength of Sponge Dressings

Tensile testing was performed on ASD to determine the influence of alginate type and PEG content on dressing stiffness, an important property that determines how well a dressing can stretch around body contours. Young’s modulus (MPa), maximum stress at fracture (MPa), and the percent elongation at fracture of M- and G-ASD made from 2 and 4% alginate solutions are summarized in Table 18. In both M- and G-ASD, the Young’s modulus increased with alginate concentration, indicating a stiffer material. On the other hand, the Young’s modulus showed no significant difference between M- and G-ASD suggesting that alginate concentration plays more of a role in controlling ASD stiffness, than alginate chemistry. As alginate concentration was increased from 2-4%, the tolerated stress before failure increased by about 4-fold. These trends were similar in both M- and G-ASD, indicating that the alginate chemistry played less of a role in governing tensile strength, as opposed to the density which was a more influencing factor. On the other hand, the percent elongation at fracture remained nearly constant across all four of the

different formulations. Considering density and tensile strength, denser ASD proved to be much stiffer, and was not as influenced by the type of alginate used (M or G).

Table 18. Tensile strength of M- and G-ASD.

<i>Alginate Concentration (w/w)</i>	M-ASD		G-ASD	
	2%	4%	2%	4%
<i>Young's Modulus (MPa)</i>	1.04 ± 0.24	9.40 ± 5.27	1.04 ± 0.73	4.29 ± 2.81
<i>Max Stress (MPa)</i>	0.06 ± 0.01	0.25 ± 0.13	0.04 ± 0.04	0.16 ± 0.04
<i>Elongation at Fracture (%)</i>	5.31 ± 0.72	4.52 ± 0.67	5.36 ± 1.14	6.16 ± 2.08

Alternatively, the incorporation of low and high molecular weight PEG at different concentrations was further investigated to determine if dressing stiffness could be controlled. The effect of PEG molecular weight and concentration on tensile strength is illustrated in Table 19. When 1.45 kDa PEG was introduced to M-ASD, an increase of PEG from 0.1% to 1.0% reduced the Young's modulus from 5.0 to 2.2 MPa, almost half the original value, indicating that PEG had initially increased the Young's modulus in comparison to PEG-free ASD, but then dropped as PEG concentrations surpassed 0.1%. With PEG concentrations increasing beyond 0.1%, the reduction in stiffness of the sponge was a result of PEG's plasticizing effect which increases flexibility and reduces brittleness.[337] Beyond 1% 1.45 kDa PEG, the ASD became increasingly PEG-based and too fragile to handle.

Table 19. Tensile strength of P/M-ASD prepared from 2% M alginate solutions containing 0.1-10% PEG, 1.45 kDa and 10 kDa.

<i>PEG (% w/v)</i>	PEG 1.45 kDa		PEG 10 kDa	
	0.1	1	0.1	1
<i>Young's Modulus (MPa)</i>	5.0 ± 2.1	2.2 ± 0.7	0.3 ± 0.3	0.6 ± 0.3
<i>Max Stress (MPa)</i>	0.13	0.03	0.01	0.02
<i>Elongation at Fracture (%)</i>	5.1 ± 1.3	2.41 ± 0.5	12.0 ± 3.0	4.5 ± 0.9

The experiment was repeated using P/M-ASD made from 0.1-1%, 10 kDa PEG to determine if increasing the molecular weight of PEG could have a more significant effect on reducing the Young's modulus. The result was an immediate reduction in the Young's modulus for preparations containing 0.1% and 1% 10 kDa PEG. The dressings however were so delicate, that the calculated moduli were nearly baseline, indicating that as the molecular weight of PEG increases significantly in P/M-ASD, the material becomes too fragile to be physically stretched. The maximum stress achieved by P/M-ASD containing 10 kDa PEG was nearly zero, indicating that no matter the PEG content, high molecular weight PEG limits any significant amount of stretching. At 10% PEG, P/M-ASD was too brittle to test.

5.2.5 Water Vapour Transmission Rate and Water Absorption Capacity of ASD

Water vapour transmission rate (WVTR) and water absorption capacity (WAC) were determined for 2-4% M- and G-ASD as summarized in Table 20. M- and G-ASD prepared from solutions of 2% M or G alginates, transmitted between 17-18mg/cm²/h vapour with no significant difference between preparations. WAC of M- and G-ASD measured between 16-27 mg H₂O/mg ASD showing a greater capacity to absorb water in ASD prepared with 2% alginate solutions versus 4%. WVTR and WAC were also measured on ASD prepared from solutions of 2% M alginate, combined with 0.1-10% 1.45 and 10 kDa PEG, as reported in Table 20. When 1.45 kDa PEG was introduced into the M-ASD, the WVTR jumped to a range of 23-29mg/cm²/h, which was slightly higher than ASD without PEG. PEG as a hygroscopic material may be acting as a desiccant, absorbing as opposed to transmitting water vapour through the dressing material. To determine if PEG molecular weight had an effect, samples with 0.1-10% 10 kDa PEG were tested. A similar range of WVTR of 24-29mg/cm²/h was measured, indicating that increasing the PEG molecular weight had little effect on WVTR.

Table 20. WVTR and WAC for 2-4% M- and G-ASD.

	2% M-ASD	4% M-ASD	2% G-ASD	4% G-ASD
WVTR ($mg/cm^2/h$)	17.1 ± 0.8	16.7 ± 0.4	17.9 ± 0.4	18.3 ± 0.8
WAC ($mg H_2O/mg ASD$)	23.8 ± 4.9	16.3 ± 1.0	27.2 ± 3.5	18.1 ± 0.8

Since PEG increased WVTR, it was also important to determine if WAC could be controlled by varying the amount and molecular weight of PEG. M-ASDs were prepared with 0.1-1.0% 1.45 or 10 kDa PEG, and resulted in a WAC ranging from 15-19mg H₂O/mg ASD (Table 21). When comparing to PEG-free ASD formulation, PEG reduced the absorptive capacity of the ASD slightly. This effect is due to the reduction in porosity and surface area as the ASD becomes denser with higher levels of PEG being added to the formulation. Data was unavailable for dressings with 10% PEG because the dressing material was unstable, dissolving in the assay solution.

Table 21. WVTR and WAC for ASD prepared from 2% M alginate solutions and combined with 0.1-10% PEG 1.45 kDa and 10 kDa.

Percent PEG (w/v)	PEG 1.45 kDa			PEG 10 kDa		
	0.1	1	10	0.1	1	10
WVTR ($mg/cm^2/h$)	26.3 ± 2.1	28.8 ± 2.1	22.9 ± 0.8	28.8 ± 3.8	26.7 ± 3.3	23.8 ± 0.8
WAC ($mg H_2O/mg ASD$)	19.0 ± 3.9	16.1 ± 0.8	-	19.1 ± 5.2	15.4 ± 1.3	-

Although flexibility and strength are important wound dressing attributes, water absorption and vapour transfer are equally as important. Since human burn wounds can release exudate at a rate of 20mg/cm²/h, the ability of a wound dressing to absorb and release moisture through evaporation ensures that the wound is kept moist, but not wet.[338,339] Moisture is also an important driving force in PLGA erosion and the diffusion of insulin, so being able to adjust the formulation to maintain a moist environment is essential. One method of controlling water absorption capacity was by changing the internal sponge surface area. Lower density M- and G-

ASD had highly open, complex porous networks which absorbed up to 30 times its mass in water, whereas denser materials absorbed much less water. Although this capability is important, especially in wounds with heavy exudate, retaining such large amounts of moisture can also promote bacterial growth and impede healing. Quinn *et al.* (1985) recommended that water vapour transfer rates for burn wound dressings should range between 8-10mg/cm²/h to sustain healthy healing.[7] In the present study, M-, G- and P/M-ASD exhibited WVTR values that were nearly double and triple the recommended range. Although this may suggest that the wound could dry out, a secondary dressing used to secure the ASD in place over the wound could also function to reduce the rate of moisture loss.[109] The formulation studies have therefore demonstrated that polymer content is an important contributor to sponge density, which in turn influences flexibility, strength and water handling properties.

5.2.6 Insulin Release Kinetics

Insulin release kinetics were measured by placing ASD containing insulin loaded microparticles, into netwell inserts, held at the air-water interface to mimic the surface of a wound as illustrated in Figure 41. A calibration curve was constructed using human insulin ELISA assay (Figure 42) and used to measure insulin release from M- and G-ASD containing 5 mg of insulin-loaded PLGA microparticles as shown in Figure 43. Over 21 days of assay, M-ASD released a cumulative 6-12% of the entrapped insulin, whereas G-ASD released 3-5% of the encapsulated insulin. After the initial release, M- and G-ASD provided insulin sustained release over the following 21 days. M-ASD appeared to maintain near zero order release kinetics over the assay period, more so than G-ASD which plateaued earlier.

Figure 44 is a plot of insulin released from P/M-ASD-prepared with 0.1-10%, 1.45 or 10 kDa PEG. All P/M-ASD released approximately 20-30% of the total entrapped insulin over a 21

day period. Almost no immediate release of insulin was observed in any of the P/M-ASD, as opposed to the PEG-free ASD (Figure 43). Near zero order release kinetics were maintained until day 14, after which the release kinetics gradually plateaued. Differences in PEG molecular weight and content did not appear to play a role in P/M-ASD release beyond 14 days, however significantly improved the amount of insulin detected over the entire study period when compared to PEG-free ASD. This effect is likely to be the result of PEG's ability to stabilize proteins in solution,[340] and also promote a shift in equilibrium towards the liberation of monomeric forms of the insulin from multimeric configurations,[341] which are stoichiometrically quantifiable with the ELISA assay. Insulin stability in a wound exudate simulant solution (British Pharmacopoeia solution A) was verified by sampling a stock insulin solution over a period of 5 days. A micro BCA and ELISA assay both confirmed that insulin reactivity remained steady over a 5 day period, indicating that insulin does not undergo degradation in solution A. These data support the conclusion that PEG may have had a role in facilitating the release of monomeric insulin from the multimeric configuration, attributed by the larger amount of insulin measured from P/M-ASD as compared to PEG-free ASD.

Insulin release kinetics were considered over a 21 day period, a time frame critical for wound healing. ASD with and without PEG released nearly equivalent amounts of insulin within the first 24 hours. Since all formulations contained the same amount of insulin-PLGA microparticles, it is possible that the surface bound insulin was first to dissolve in equivalent proportions. After 5 days, differences in the release kinetics began to appear in PEG-free ASD which resulted in less insulin being released over the same period of 25 days. Although the acidic degradation products of PLGA were considered as potentially responsible for reduced insulin release, it was demonstrated earlier that biologically active insulin was released from

PLGA for an extended period of 25 days as verified through two assays. One assay showed high activity of released insulin through the stimulation of rat L9 myoblasts, phosphorylating Akt (Figure 30). The second assay (Figure 31), involved measuring HaCaT cell migration following an applied injury to a cell monolayer. Biologically active insulin is therefore still capable of being released over a month-long period, a period critical for wound healing.

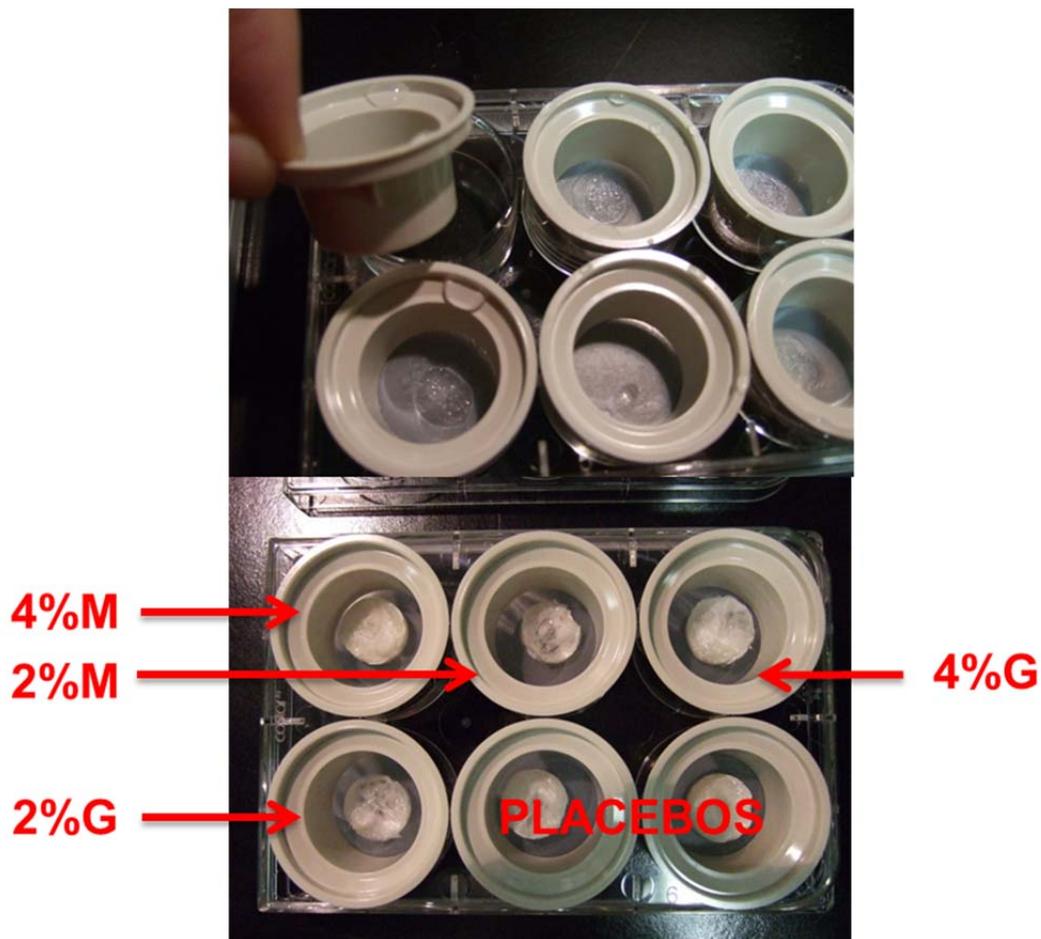


Figure 41. Placement of ASD into netwell inserts, held within 6 well tissue culture plates. The positioning of the netwell membrane was strategically placed at the air-water interface to mimic the surface of a fresh wound.

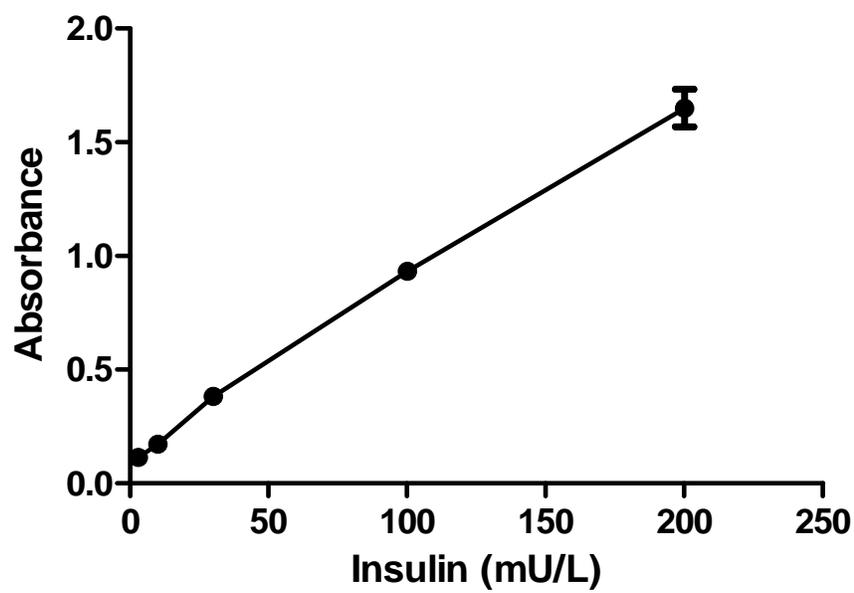


Figure 42. A representative human insulin ELISA assay calibration curve from 3-200 mU/L.

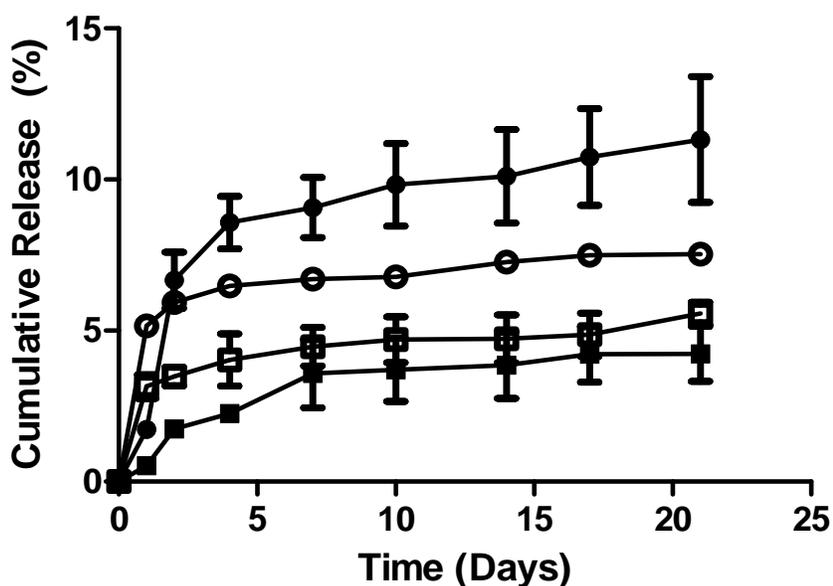


Figure 43. Insulin release kinetics from ASD prepared from 2% (○) and 4% (●) M and 2% (□) and 4% (■) G alginates. All samples were performed in triplicate and bars around the mean represent the standard error.

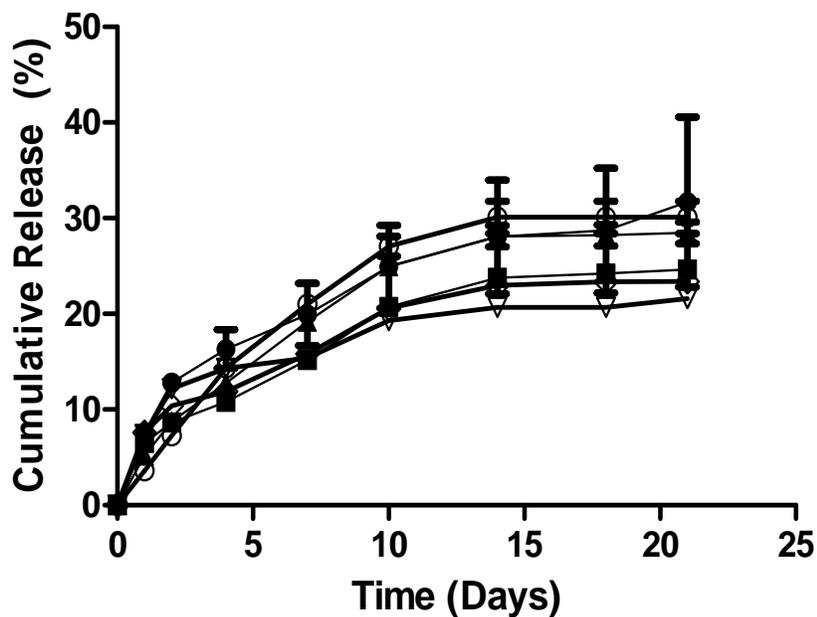


Figure 44. Insulin release kinetics from ASD prepared from 2% M alginate solution with 0.1% (●), 1% (■) and 10% (▲) 1.45 kDa PEG and 0.1% (▼), 1% (◆) and 10% (○) 10 kDa PEG. All samples were performed in triplicate and bars around the mean represent the standard error.

5.2.7 Bioactivity of Released Insulin: HaCaT Scratch Assay

Wounding experiments were simulated using a cell monolayer scratch assay with a human keratinocyte cell line (HaCaT) to assess the bioactivity of released insulin. A scratch of approximately 500 μm in width was made on a confluent layer of HaCaT cells and monitored for the progression of cell migration upon stimulation with insulin. Figure 45 shows a representation of (A) HaCaT cells after being freshly scratched and (B) recovery after 48h. Figures 46 and 47 show the extent of HaCaT cell migration over time to close the scratch, after stimulation. In cases where ASD were formulated with alginate and PEG, ASD with only 1% 1.45 and 10 kDa PEG were tested.

The greatest stimulation of cell migration took place between 4 and 24h for cells that were exposed to 10^{-7} M fresh insulin versus the blank control. By 24h, the scratch was 70% closed and fully confluent after 48h, whereas the blank insulin-free negative control failed to achieve full scratch closure in the same period of time. Although some cell migration was observed, cells treated with blank supernatant only reached 15% and 48% scratch recovery at 24 and 48h respectively. A one-way ANOVA followed by a pairwise comparison indicated a significant difference between the fresh insulin control and blank alginate control beyond hour 4, suggesting that insulin has a strong stimulatory effect on the migration of HaCaT cells.

A comparison between insulin released from ASD over a 21 day period and fresh insulin is also presented in Figures 43 and 44. In all preparations, insulin released after the first day had the same bioactivity as that of the fresh insulin control and did not differ in performance statistically, over the 48h scratch assay. When examining the bioactivity of insulin released from M- and G-ASD (Figure 46A-B), both had reduced bioactivity by day 7 of release. In either case, supernatants obtained from day 7 only led to a scratch recovery of 16% and 29% for M- and G-

ASD respectively. By day 21, the bioactivity of the insulin released from M- and G-ASD had similar activity to the blank, only recovering 34% and 56% of the scratch respectively. A comparison between M- and G-ASD revealed that over the course of the scratch assay, insulin bioactivity was similar for both formulations.

ASD prepared from initial solutions of 2% M ASD with 1% 1.45 or 10 kDa PEG had prolonged bioactivity of released insulin for at least 10 days, longer than that of insulin from M- or G-ASD alone (Figure 47A-B). Insulin bioactivity measured on the first day of the assay was nearly identical for 1.45 and 10 kDa P/M-ASD and comparable to that of the fresh insulin control. On day 10 however, P/M-ASD supernatants were better able to match the insulin bioactivity comparable to that of the fresh insulin control, as opposed to M- and G-ASD containing no PEG. By day 21, insulin bioactivity had diminished and was inactive as compared to the fresh insulin control.

Overall, insulin bioactivity was maintained for at least 10 days. ASD prepared from 2% M alginate solutions with 1% PEG 1.45 and 10 kDa helped to maintain insulin bioactivity at levels similar to that of fresh insulin, versus insulin free ASD. No differences were observed in activity between P/M-ASD prepared with either 1.45 or 10 kDa PEG.

A cell scratch assay to test insulin for bioactivity is based on the ability to promote changes in cell migration, thus serving as an *in vitro* model of repair in human epidermal cells. Insulin released from both M- and G-ASD had strong stimulatory effects after the first day of release, similar to that of the fresh insulin control. However, by day 7 and beyond, the bioactivity was reduced relative to the fresh insulin control. Since low quantities of insulin were being detected during the latter half of the assay, there were often insufficient levels of insulin to reach target concentrations to fully stimulate HaCaT migration. Furthermore this low

concentration of insulin in the supernatants also resulted in HaCaT cells being almost entirely exposed to release solution A, which lacks many of the nutrients normally obtained from the starvation media to facilitate migration and normal growth. On the other hand, the presence of PEG in P/M-ASD sustained insulin bioactivity for up to 10 days. PEG hydrogels have been shown to protect and preserve peptides in harsh environments.[342] It also appeared that PEG molecular weight within the dressing does not influence insulin bioactivity while its presence is necessary to preserve dressing integrity and insulin bioactivity.

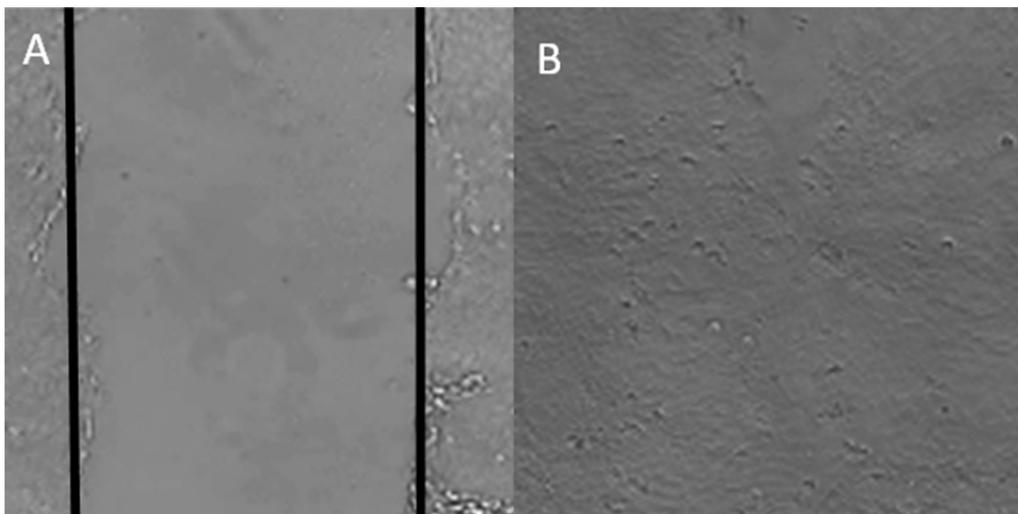
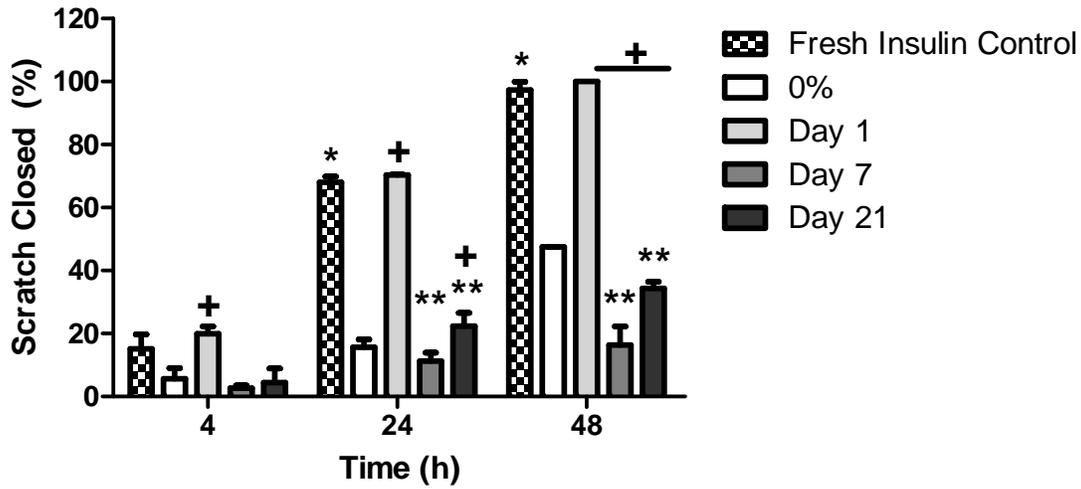


Figure 45. A representation of HaCaT cells after being (A) freshly scratched by a pipette tip and (B) after treatment with insulin for 48 hours as seen through an inverted contrast microscope. The vertical bars in (A) indicate the margins of the scratch approximately 500 μ m in width, with the space between the bars representing the “scratch wound.”

A 2% M-ASD



B 2% G-ASD

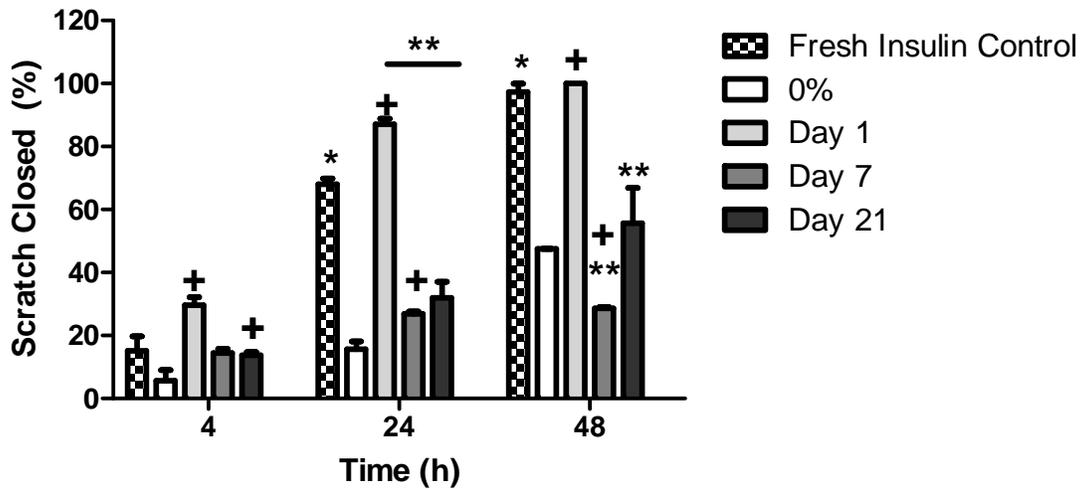
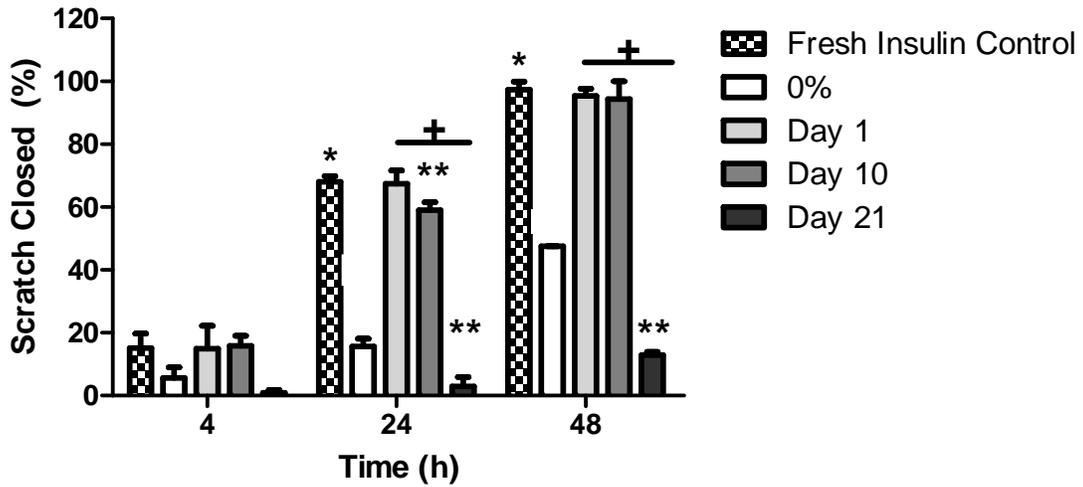


Figure 46. HaCaT cell scratch assays evaluating bioactivity of insulin released from 2% M- (A), and 2% G- (B) ASD from days 1, 7, and 21. * indicate statistical significance between fresh insulin control and insulin free PLGA supernatant, ** indicates difference between insulin released to supernatant, and fresh insulin control, and + indicates difference between insulin released to supernatant and insulin free PLGA supernatant ($p= 0.05$). Supernatants from days 7 and 21 in (A) and (B) represent samples in which 10^{-7} M insulin was difficult to achieve.

A P/M-ASD with 1.45 kDa PEG



B P/M-ASD with 10 kDa PEG

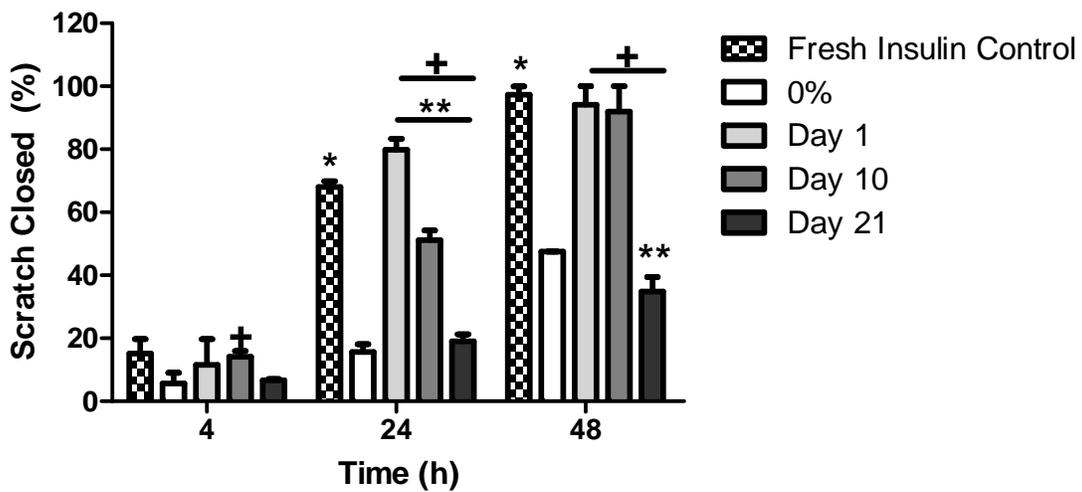


Figure 47. HaCaT cell scratch assays evaluating bioactivity of insulin released from 2% M- with 1% PEG 1.45 kDa (A), and 2% M- with 1% PEG 10 kDa (B) ASD from days 1, 10, and 21. * indicate statistical significance between fresh insulin control and insulin free PLGA supernatant, ** indicates difference between insulin released to supernatant, and fresh insulin control, and + indicates difference between insulin released to supernatant and insulin free PLGA supernatant (p= 0.05). Supernatant from day 21 in (A), and (B) represent samples in which 10^{-7} M insulin was difficult to achieve.

5.2.8 Alginate Microbial Contamination

Alginate and distilled water were tested for microbial contamination to determine if bacteria were present in the raw materials used to formulate ASD. Alginates, unlike PEG, are obtained from natural sources such as kelp and brown algae.[343] During harvesting and processing, bacteria from the sea can become entrapped and remain in the raw material where it becomes passed on to the final product.[344] This presents a concern for wounds which may become infected, but also for the insulin which is entrapped within the alginate sponge matrix. Bacteria such as *Escherichia coli* can secrete insulin degrading enzymes (IDE), also known as protease III, which are capable of hydrolyzing insulin causing the insulin to lose its therapeutic properties.[207]

Distilled water and 2% M-alginate solutions were tested by plating samples on to tryptase and blood agar plates and counting colonies reported in Figure 48. Control plates did not produce any colonies. Only 1 colony was observed on 1 out of 3 plates, on both tryptase and the blood agar plates, treated with distilled water, meaning that the distilled water had very low contamination. Aqueous M-alginate solutions plated on tryptase and blood agar resulted in 100 ± 43 and 53 ± 16 colonies/mL respectively. M-alginates used in ASD, are sourced from marine brown algae and considered pharmaceutical grade.[345] Prior to formulating the ASD, the alginate was not filtered or sterilized, thereby allowing bacteria to remain in the ASD. Leinfeld *et al.* (2003) reported that when raw alginate was placed into complete growth medium with 10% bovine serum albumin and 1% human growth serum, bacterial growth was detected, even after rigorous post-purification. ASD sterilization will have to be completed, prior to being placed on a wound to prevent infection.

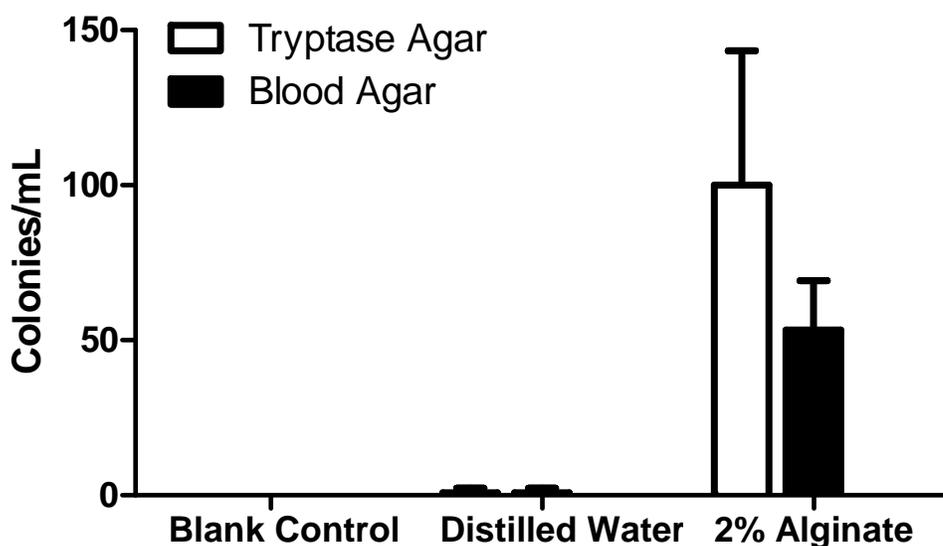


Figure 48. Microbial contamination measured in colonies/mL for distilled water and 2% alginate solutions plated on tryptase and blood agar.

5.2.9 ASD Summary

ASD formulated with 2% M-alginate and 1% 1.45 kDa PEG yielded porous sponges with favourable flexibility and water handling characteristics necessary for wound dressings. Insulin was released from these dressings for up to 21 days, while maintaining bioactivity for up to 10 days and potentially longer, with the addition of a small quantity of PEG. In conclusion, an M-alginate-PEG sponge dressing (P/M-ASD) containing PLGA insulin-loaded microparticles is capable of delivering bioactive insulin in a sustained and controlled manner over a prolonged period of time. As a result, this dressing will serve as a candidate ASD in animal trials in the next phase of testing.

5.3 Phase 3 – Animal Studies

A 2% M-alginate and 1% 1.45 kDa PEG candidate dressing was selected to advance into animal trials using a rodent, partial-thickness burn model. The goal of this study section was to test the biocompatibility, determine the therapeutic insulin dosing window and assess the physiological response to insulin delivery through blood glucose monitoring, wound measurements and histology. Using this information, the effectiveness of alginate sponge dressings (ASD) was evaluated. *In vivo* studies were conducted in the laboratory and under the supervision of Prof. Manuela Martins-Green in the Department of Cell Biology and Neuroscience at the University of California, Riverside. The animal care protocol established at the University of California was followed. The initial experiments establishing the test protocols that were subsequently followed were conducted jointly by Michael Hrynyk with Dr. Yan Liu (burn surgeon) under the direct supervision of Prof. Martins-Green. For subsequent experiments, dressings were prepared by Michael Hrynyk, and the wound testing was conducted either by Dr. Yan Liu, or by Sandeep Dhall under the direct supervision of Prof. Martins-Green. The responsibilities associated with each experiment are noted in the appropriate sections.

5.3.1 Microparticle Biocompatibility

PLGA microparticles loaded with 10% crystalline insulin were provided to Prof. Martins-Green who tested for biocompatibility by applying a small, 10mg sample to 7 mm punch wounds made on the dorsal side of a single Sprague Dawley rat. Changes in wound morphology, increased swelling, redness, and animal discomfort were monitored as indicators of microparticle incompatibility. During the first 72h, no signs or symptoms of distress were visible in the animal after the microparticles had been applied. Shortly after 72h, the rat died from hypoglycemia due to a sudden release of insulin from the microparticles. The sudden release of insulin was a result

of the burst effect from the microparticles and high dose of insulin used. Although this result ended in death, it gave an early indication that the insulin being released was bioactive, and that a dose of $26\mu\text{g}/\text{mm}^2$ (a 1 mg insulin dose spread equally over a 7 mm diameter punch wound) is beyond what a rat can tolerate. The study also demonstrated that applying microparticles in a saline suspension was difficult. Some microparticles adhered to the administering syringe, thus resulting in a smaller portion of the sample being transferred to the wound. A matrix material (dressing) would consequently have to be designed to secure the microparticles uniformly in place, enabling the diffusion of insulin to the wound.

5.3.2 ASD Biocompatibility

Insulin-free M-ASD pre-treated with and without cross-linking with calcium were provided to Prof. Martins-Green, who tested for biocompatibility using a similar, 4 circular punch wound model previously discussed. Four wounds were made on the dorsal side of a single rat, approximately 1-2 cm apart as seen in Figure 49. Five days later, the control wound in Figure 50A, was compared to wounds treated with uncross-linked M-ASD in position C, and M-ASD treated with 100 (B) and 200 mM CaCl_2 (D). Wound morphology and signs of inflammation were examined for any signs of M-ASD incompatibility, or infection. The control wound (Figure 50A) maintained all adhesive tissue connections around the circumference of the wound, showing no separation of subcutaneous tissues from the wound margins, or infection. In contrast, wounds in Figure 50B-D had all become detached, resulting in a skin pocket forming underneath the margins of the wound. Inflammation and necrosis are also apparent in wounds B and D, suggesting the presence of an infection. A magnified view shown in Figure 51A shows the margins of the control wound in comparison to Figure 51B which shows the appearance a wound treated with the M-ASD alone, and the loss of tissue adhesion. In surgical punch wounds,

like the ones used in this study, loss of tissue adhesion, or dehiscence, often result from wound infections, or high levels of tension applied to the surrounding skin.[346][347] Since no tension was applied to the wound, the cause of the dehiscence was most likely attributed to bacteria which had not been killed during the sterilization process. Ultraviolet (UV) sterilization for this study was only performed for a maximum of 5 min, and may not have been sufficient to sterilize the M-ASD. UV sterilization was increased to 20 min/side for all subsequent studies. No advantage was observed in pre-treating M-ASD with solutions of CaCl_2 as well. The additional fluid retained by the M-ASD made it more difficult to secure, resulting in tearing and loss of the sample. The M-ASD would have to be applied with a minimal amount of moisture, prior to application.

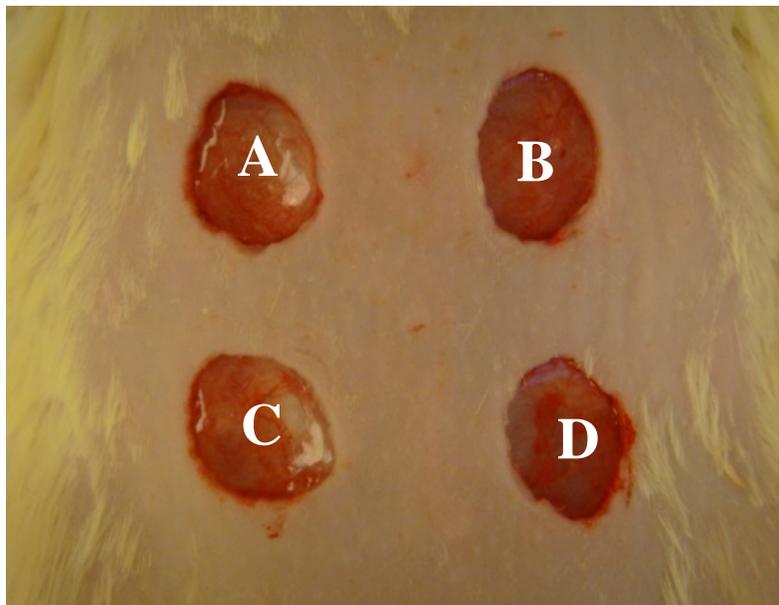


Figure 49. Four punch wounds formed on the dorsal side of a Sprague Dawley rat. Wound A) was the blank control, B) 100 mM CaCl_2 pre-treatment, C) M-ASD, and D) 200 mM CaCl_2 .

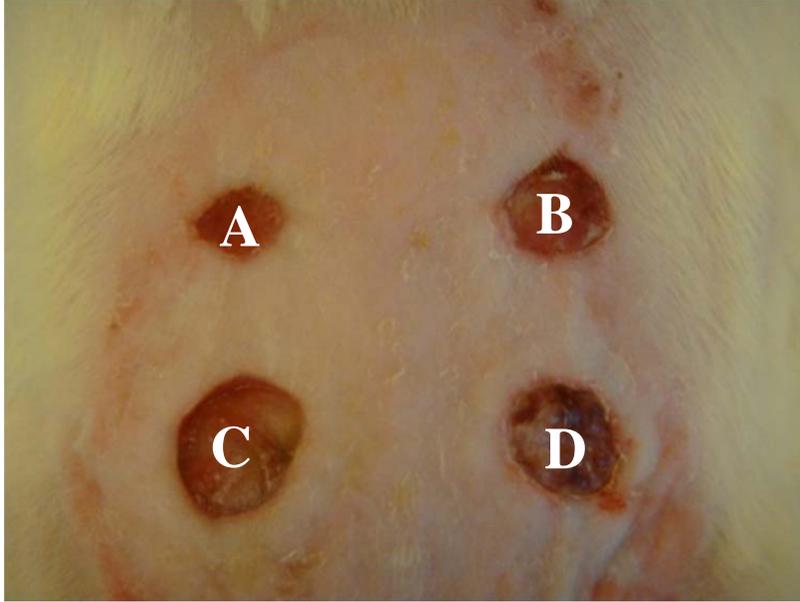


Figure 50. Wounds A) blank control, B) M-ASD 100 mM CaCl₂ pre-treatment, C) M-ASD, and D) M-ASD 200 mM CaCl₂ pre-treatment 5 days post injury.

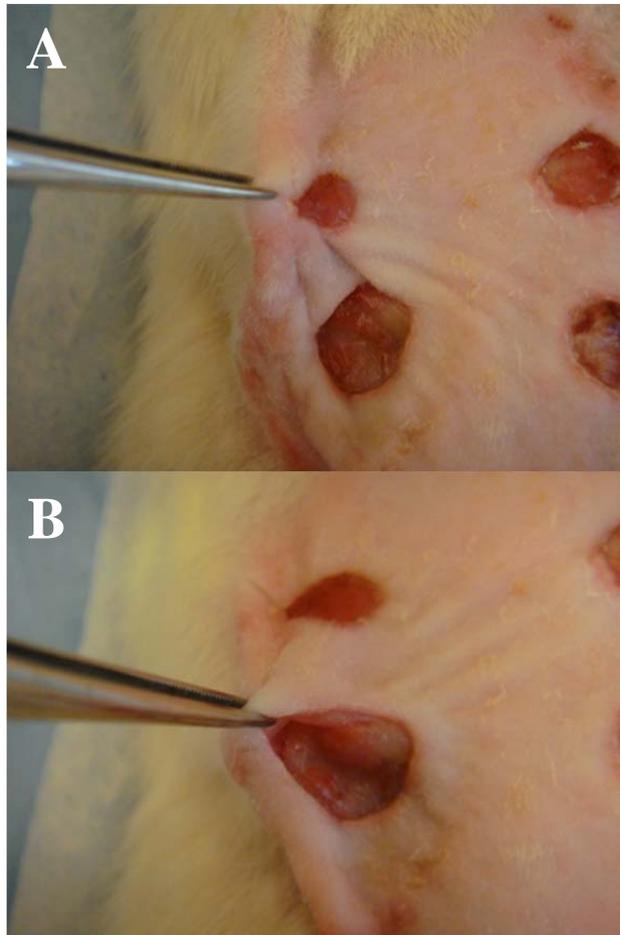


Figure 51. Wounds 5 days post-injury showing in panel A) a strong connection between subcutaneous tissues and wound margins, in comparison with panel B) showing a separation between the skin and underlying tissues treated with M-ASD.

5.3.3 ASD Pilot Study

The study described in the following sections 5.3.3 - 5.4.3, was conducted by Michael Hrynyk and burn surgeon Dr. Yan Lin, under the supervision of Prof. Martins-Green. A partial-thickness scald wound as seen in Figure 52, was produced on the dorsal side of two anesthetised rats by pressing a heated brass cylinder for 6 sec on the surface of the skin. Prior to placement, the dressings were sterilized for 20min/side using UV light and secured over the wound using a single tegaderm patch as seen in Figure 53A-B. The two types of dressings tested, included a P/M-ASD loaded with 0.25mg insulin and a blank control P/M-ASD.



Figure 52. A partial-thickness scald burn, formed after the application of a brass cylinder held firmly over the skin for 6 sec.

Blood glucose levels were recorded every hour for 7h, followed by two single measurements taken at 10 and 23h as shown in Figure 54. Baseline blood glucose levels measured for control and insulin treated rats were 255 and 249mg/dL respectively. Elevated blood sugar levels are common after anaesthesia, and are attributed to eating prior to surgery. Saha *et al.*, (2005), reported blood glucose levels of 300mg/dL after providing food and anaesthetics to rats undergoing minor surgery. The blood glucose levels remained elevated for up to ten hours, before returning back to 100-168mg/dL without any harm to the animals.[348]

At 23h, the experiment was stopped and the animal was re-evaluated. The blank animal control showed positive signs of recovery and was unaffected by the burn, or dressing. The insulin treated animal on the other hand, appeared lethargic and had a drop in blood glucose levels near 99mg/dL. This was significantly lower than the 168mg/dL measured in the control animal. This combination of lethargy and low blood glucose indicated that the insulin dose was too high for the animal, resulting in an adverse reaction. An ASD with a lower insulin dose would have to be used to avoid such reactions. Another observation revealed that repeated tail-

vein bleeding resulted in tail atrophy and added animal stress. The blood sampling method was discontinued, and moved to other parts of the body such as the paws to minimize localized vein atrophy.

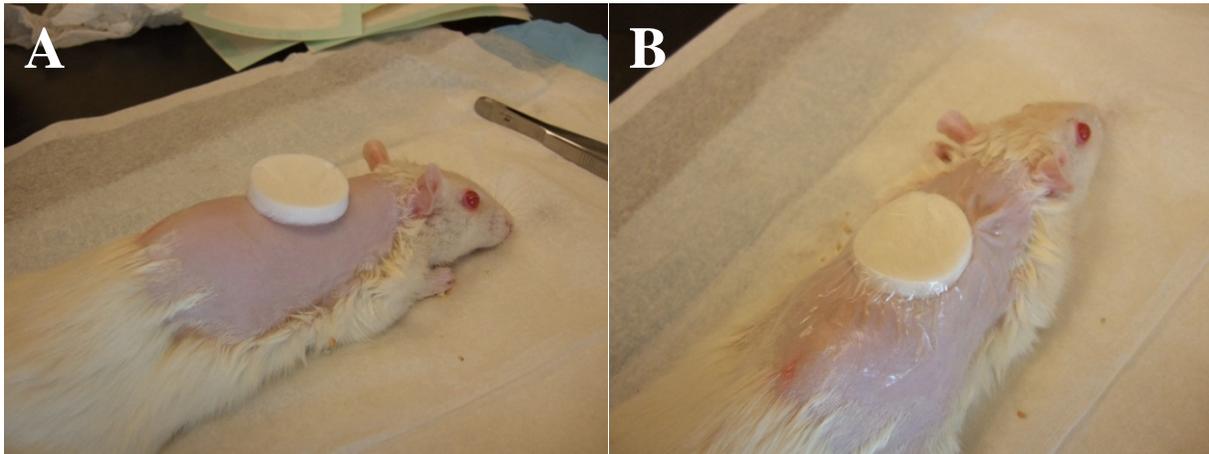


Figure 53. A P/M-ASD placed over a burn wound covered without a tegaderm patch (A) and with a tegaderm patch (B).

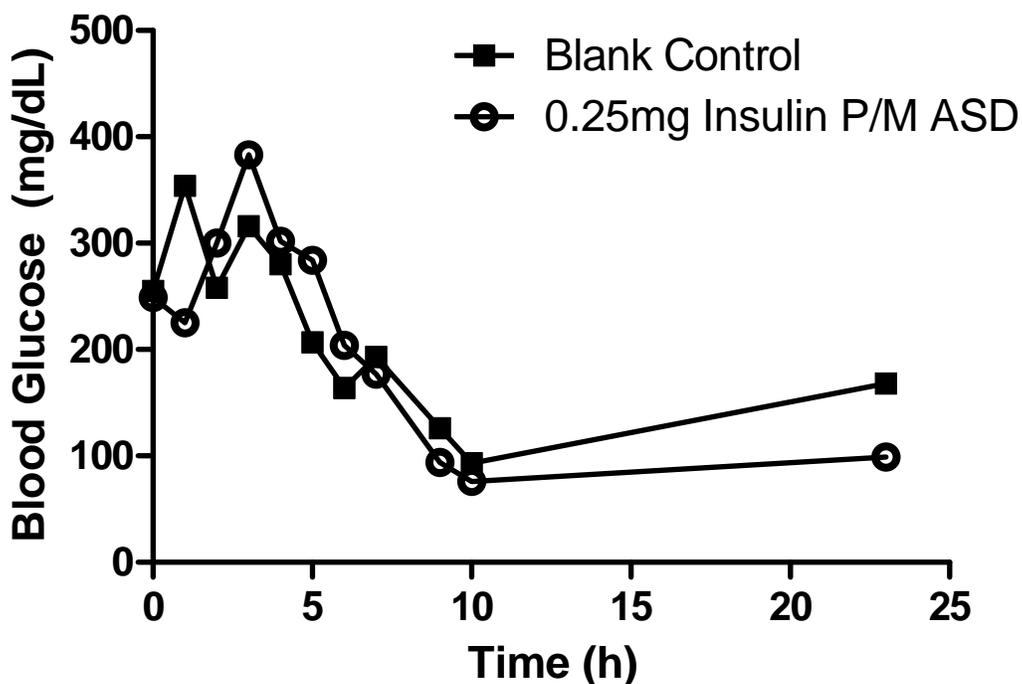


Figure 54. Blood glucose levels for control and insulin treatment group over a 23h period. Basal blood glucose levels were recorded at time 0, and continuously measured post P/M-ASD application.

5.3.4 ASD Dosing

The previous results showed that an insulin dose of 0.25mg was too high and needed to be reduced to prevent hypoglycemia. A P/M-ASD with 0.25mg insulin was tested again, but only allowed to cover 50% and 75% of the total wound surface area. A control rat received a blank P/M-ASD applied to the entire wound surface area. Blood glucose was monitored for 3 days and plotted in Figure 55. All three rats had a basal blood glucose level ranging from 190-240mg/dL after being given anaesthesia. Blood glucose levels in all animals rose to 300mg/dL within 2h after dressing application and remained steady for several hours. A sharp decrease in blood glucose levels to 100mg/dL was observed 8h later in all rats, followed by a return to basal levels 54h later. A dressing change performed on all animals after 2 days revealed no signs of infection,

or inflammation suggesting that the sterilization method used, was sufficient to prevent contamination of the wound. The dressing did however appear dry and nearly completely free of exudate or blood. Rat skin produces no exudate following a burn injury, unlike human skin which produces large amounts of exudate to cleanse the wound.[349] Without any exudate to facilitate the formation of a hydrogel, PLGA degradation and insulin release was limited. A method of pre-wetting the P/M-ASD would therefore be needed to facilitate hydrogel formation and insulin diffusion.

Consequently, the blank and 50% P/M-ASD animal subjects were discontinued, while the rat receiving 75% ASD treatment remained. A dressing change was performed, replacing the current P/M-ASD with a lower insulin dose (0.125 mg), pre-treated with 75 μ L sterile saline. This was performed to provide additional fluid which would help form a hydrogel and facilitate insulin release from the PLGA microparticles. Blood sugar levels measured 7h after dressing reapplication had decreased to 100mg/dL, 7h later, suggesting that insulin was diffusing out of the dressing as intended. A photo of the wound 24h (Day 5 post injury) after applying the wet dressing showed no signs of infection and only slight inflammation around the margins of the wound as seen in Figure 56. The experiment was stopped after 144h (6 days), with no obvious signs of infection or rejection. Moistening the P/M-ASD prior to placement prevented the rat from removing the dressing overnight and prevented itching. Therefore, a 0.125mg insulin dressing, pre-moistened with sterile saline was determined to be the best treatment to be carried forward.

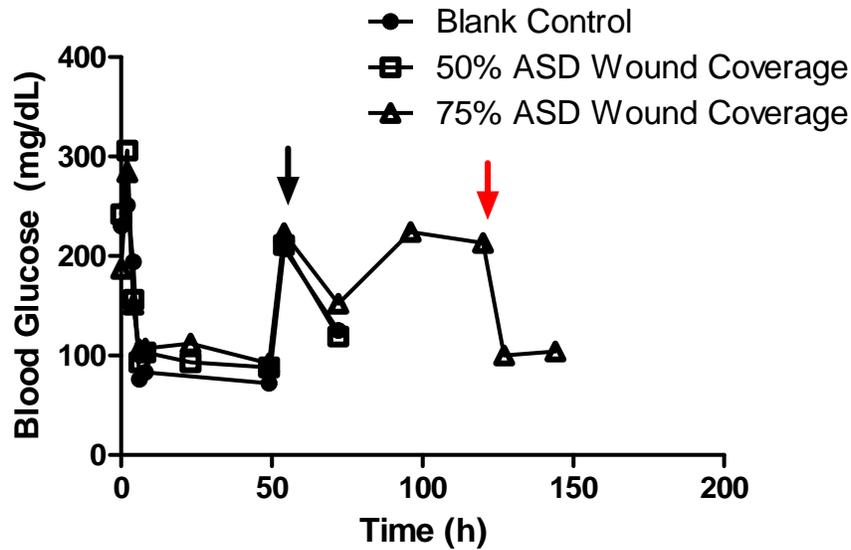


Figure 55. Blood glucose levels measured from control and treatment rats over a 72h period. All P/M-ASD formulations resulted in similar blood glucose profiles, with no significant deviations in rats treated with insulin or blank controls. The black arrow indicates the point at which dressings were changed without addition of saline. The red arrow indicates a dressing change along with 75 μ L of sterile saline.



Figure 56. A photograph of a partial-thickness scald burn, 5 days post injury after being 75% covered with a 0.25mg insulin P/M-ASD. A red ring of inflammation around the margins of the wound is visible as part of the normal healing response. No infection is visible in the wound.

5.4 Saline Moistened ASD Application with 0.125mg Insulin Dose

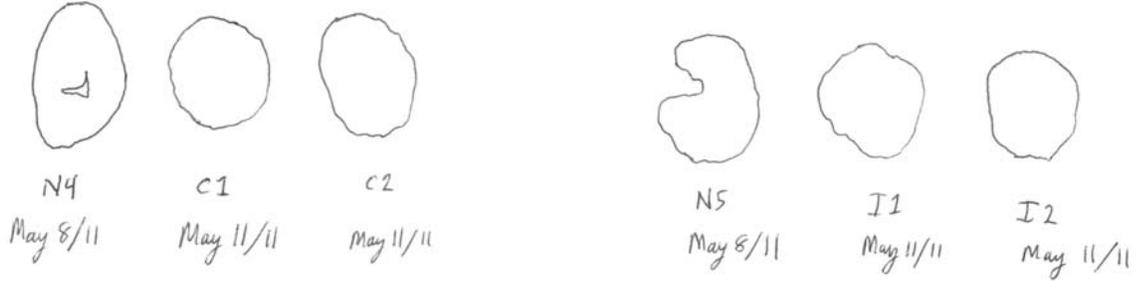
5.4.1 Wound Measurements

P/M-ASD loaded with and without 0.125mg insulin were sterilized, pre-moistened with 100 μ L sterile saline and applied to three rats in two separate groups. Dressings were changed every 3 days, until wounds had completely re-epithelialized. Direct wound traces obtained during dressing changes were compiled into Figures 57A-D and used to plot re-epithelialization over time in Figure 58. Both control and insulin treated animal groups maintained a steady rate of re-epithelialization for the first 9 days of healing. On day 12, the insulin treated animals maintained a steady rate of wound closure of approximately 7% per day until full re-epithelialization on day 15. This was significantly different than the control animals which exhibited a period of stagnation between days 12 and 15, resulting in full re-epithelialization by day 16. Photographs of the wound were taken from animals N4 (control) and N5 (insulin treated) and shown in Figure 59. Animal I2 in the insulin treatment group did develop an infection on day 12, which persisted until day 23. The infection resulted in a slower mean re-epithelialization rate than in the other 2 test animals, which had healed by day 15. Topical insulin delivery enhanced wound healing and reduced healing times by a full day.

A

Legend: N4, C1, C2 - placebo
N5, I1, I2 - insulin

Day 0 (Zero)



Day 3 (Three)



Day 6 (Six)



B

Legend: N4, C1, C2 - Placebo
N5, I1, I2 - Insulin

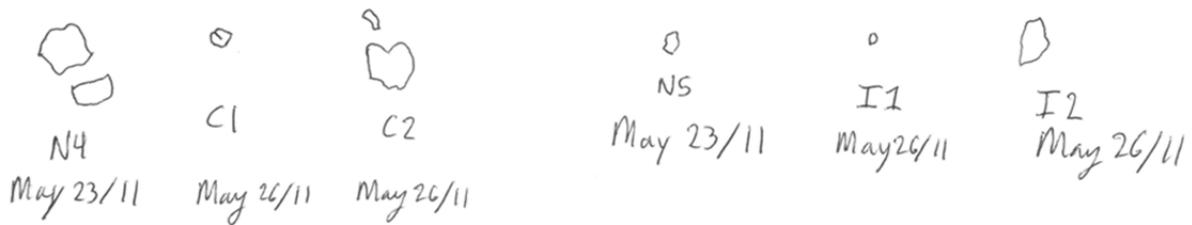
Day 9 (nine)



Day 12 (twelve)



Day 15 (fifteen)



C

Legend: N4, C1, C2 - Placebo
NS, I1, I2 - Insulin

Day 16 (sixteen)

N4	^o C1	C2	NS	I1	I2
No data recorded	May 27/11	No data recorded	Healed	Healed	No data

Day 17 (seventeen)

- May 28/11
- no data: all samples

Day 18 (eighteen)

N4	C1	C2	Healed	Healed	^o I2
No data recorded	Healed	^o May 29/11	NS	I1	May 29/11

Day 19 (nineteen)

- May 30, 2011
- no data: all samples

Day 20 (twenty)

N4	C1	C2	NS	I1	I2
Healed	Healed	No data recorded	Healed	Healed	No data recorded

May 31/11

D

Legend: N4, C1, C2 - Placebo
NS, I1, I2 - Insulin

Day 21 (twenty-one)

N4	C1	C2	NS	I1	I2
Healed	Healed	Healed	Healed	Healed	Healed
		June 1/11			Healed June 1/11

Day 22 (twenty-two)

- June 2/11
- no data: all samples

Day 23 (twenty-three)

- N4 - C1 - C2	- NS - I1 - I2
[Healed]	[Healed] Healed I2
	June 3/11

Figure 57A-D. Wound traces obtained from control group rats treated with P/M-ASD and treatment rats receiving 0.125mg P/M-ASD, following dressing changes every 3 days. Animal I2 had developed a wound infection by day 15, slowing re-epithelialization.

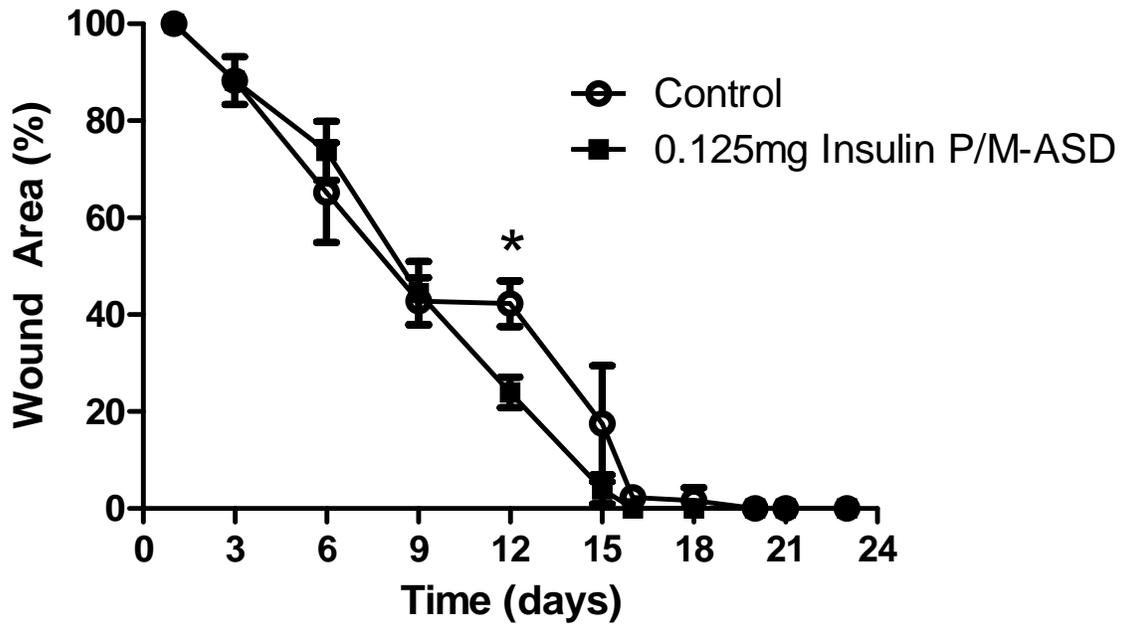


Figure 58. Normalized wound areas measurements for control group rats treated with P/M-ASD and treatment rats receiving 0.125mg P/M-ASD with 3 animals in each treatment group. The asterisk denotes significance of $p < 0.05$.

Day 0

Day 3

Day 6

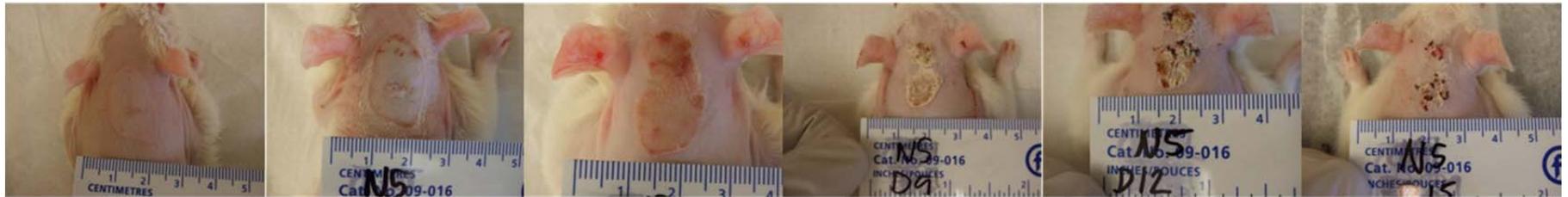
Day 9

Day 12

Day 15



Blank Control



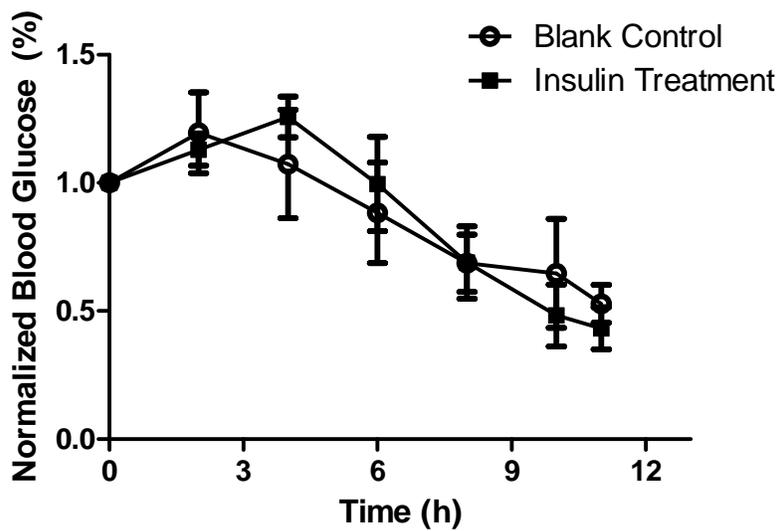
Insulin Treated

Figure 59. Photographs of the burn wounds in both the blank control and insulin treated groups. Note the difference on day 15, where the insulin treated wound is nearly completely healed, while the control wound remains over a significant surface area.

5.4.2 Blood Glucose

Blood glucose was recorded and showed no difference between treatment groups within the first 12h as seen in Figure 60A-B. Normalized blood glucose levels increased within 3 hours post dressing application, followed by a steady rate of decrease after 12h. The following day, blood glucose levels leveled in all test groups over a period of 12 days, without significant deviation. Insulin loaded P/M-ASD therefore, had no systemic effects on lowering blood glucose levels compared to the control.

A



B

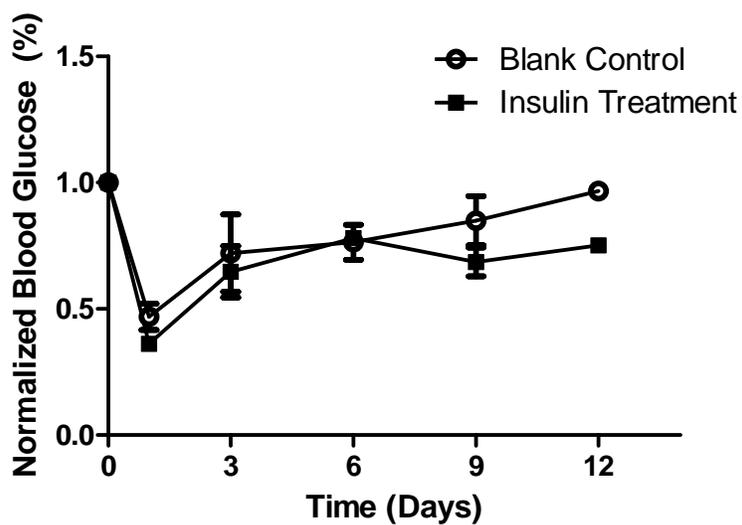
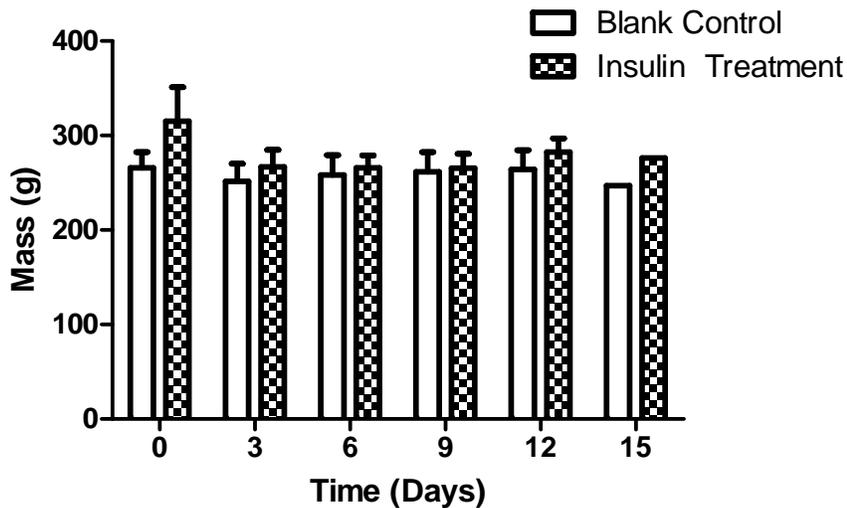


Figure 60. Graph (A) shows the normalized blood glucose levels for a period of 12h after the application of the control and insulin loaded P/M-ASD. Graph (B) shows normalized blood glucose levels for the same groups over a 12 day period.

5.4.3 Animal Weight

Animal weight was recorded over a period of 15 days as an indicator for changes in metabolism, or health as a result of insulin application. Figure 61 shows no significant differences in animal weight over the 15 day period, indicating that topical insulin delivery had no effect on systemic metabolism. Weight in both groups remained approximately 250g throughout the 2 week period, which is different from the daily weight gain of 2-3% for female Sprague Dawley rats.[350] According to Burke *et al.*, (1979), burn injuries can place a substantial burden on growth and development. As a result, normal weight gain over the study period was most likely decreased by the burn injury itself, and not from insulin delivery.



[349]

Figure 61. Weight recordings from both control group and insulin treated animals over a 15 day period. Animal weight in both groups remained close to 250g and did not fluctuate.

5.4.4 Histology

Hematoxylin and eosin (H&E) staining was performed by Prof. Martins-Green, on tissue samples obtained from Dr. Yan Liu and Michael Hrynyk at the conclusion of the study period. Figure 62 shows the position of the epidermal migration tongue which is migrating into the wound. Rats treated with 0.125mg insulin, applied every 3 days over a 15 day period, resulted in a longer and thicker migration tongue than control animals. According to Liu *et al.*, (2009) insulin stimulates keratinocyte migration, resulting in enhanced re-epithelialization. The exact mechanism is still unknown, however insulin has been shown to stimulate DNA replication and protein synthesis within keratinocytes.[264][266] Topical insulin delivery resulted in enhanced re-epithelialization through faster keratinocyte migration as compared to blank control dressings, as seen in H&E tissue sections.

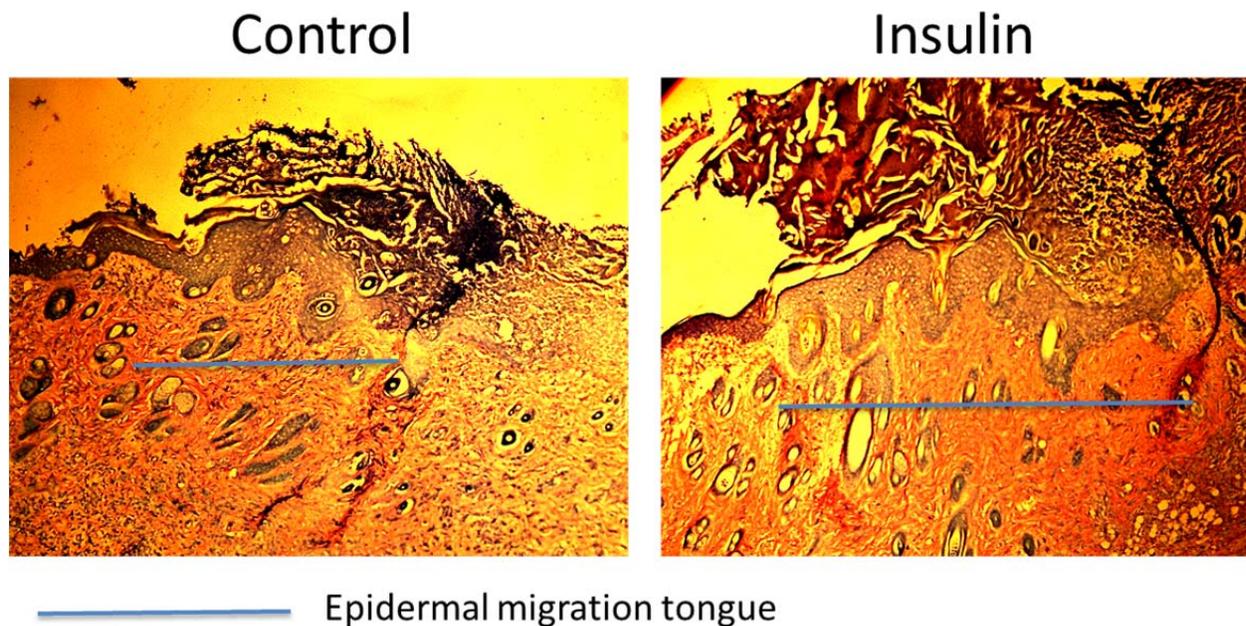


Figure 62. H&E stained wound tissue from rats treated with a blank control and insulin loaded P/M-ASD. The blue bar indicates the length of the epidermal migration tongue.

Wound tissues obtained at the end of the study from both control and insulin treated groups were stained with Mason's trichrome stain for collagen as seen in Figure 63 by Dr. Yan Liu and analyzed by Prof. Martins-Green. In the insulin treated group, collagen fibers stained blue, appear thicker, denser and more oriented than compared to the control. In the control group, collagen fibers appear thinner and not as well organized. According to Bashey *et al.*, (1972) insulin is known to increase protein synthesis, either by enhancing the incorporation of amino acids through the cell membrane, or by an action which is independent of amino acid transport. Bashey showed that insulin enhanced collagen biosynthesis *in vitro*, through the stimulation of proline and lysine hydroxylases, which help with the biosynthesis of collagen.[351] Topical insulin delivery through the P/M-ASD was able to promote greater biosynthesis of collagen, potentially promoting keratinocyte migration through the provision of a dense collagen network.

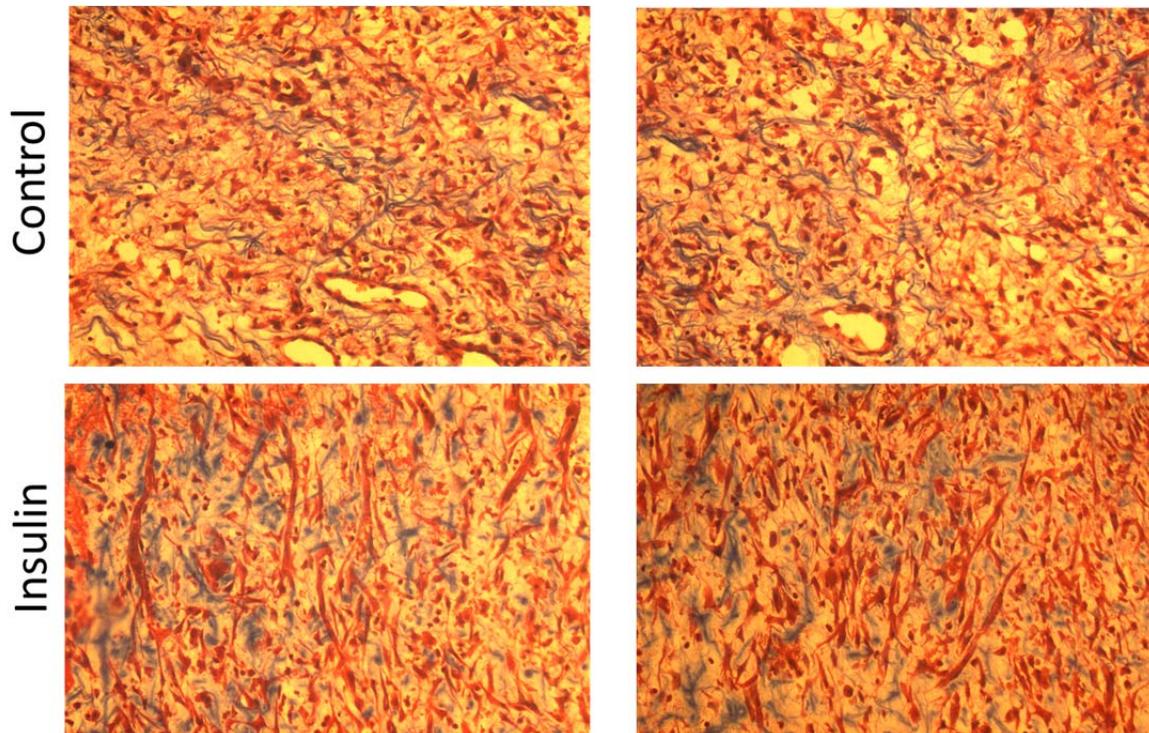


Figure 63. Mason's trichrome staining performed on tissue samples obtained from control and insulin treated animals. Collagen has been stained blue, and appears in dense, thick and oriented fibers in tissues exposed to insulin, as compared to controls.

5.5 ASD with Low Insulin Dose (0.0625mg)

The present study was performed by Dr. Yan Liu and Sandeep Dhall, under the supervision of Prof. Martins-Green with samples made by Michael Hrynyk. A half dose of insulin (0.0625mg) was administered by cutting a P/M-ASD loaded with 0.125mg insulin in half. Dressings were applied to the same partial-thickness wounds and separated into 2 control rats and 2 insulin treated rats. Dressings were changed every 3 days for 15 days. Wound surface area measurements were recorded and plotted in Figure 64. Between day 0 and 3, animals dosed with insulin showed nearly 10% more epithelialization as compared to control animals. By day 6, both study groups remained at approximately 75% non-epithelialized. On days 6 to 15, re-epithelialization was re-established in insulin treated animals, achieving a 30% reduction in

wound surface area as compared to the blank control group. A large difference was observed between control and insulin treatment groups, but was not statistically significant. A possible reason for this could have been attributed to an inconsistent delivery of insulin. Since the microparticles are heterogeneously distributed within the dressing, each half would have an unequal amount of insulin.

Consequently, new P/M-ASD were made with 0.0625mg insulin, relieving the need to adjust the dose by cutting the dressing in half. Wound surface area measurements were recorded and plotted in Figure 65, showing the wound area for 4 separate control and 4 insulin treated rats. Both control and insulin loaded P/M-ASD animal groups had the same rate of recovery throughout the 22 day period, of approximately 4.5%/day. No significant difference in wound healing was observed between treated groups, suggesting that the insulin dose was below the therapeutic threshold to stimulate measurable re-epithelialization. It was decided that full area dressings would therefore need at least 0.125mg, or 0.04mg/cm² insulin to induce faster re-epithelialization when compared to a blank P/M-ASD control.

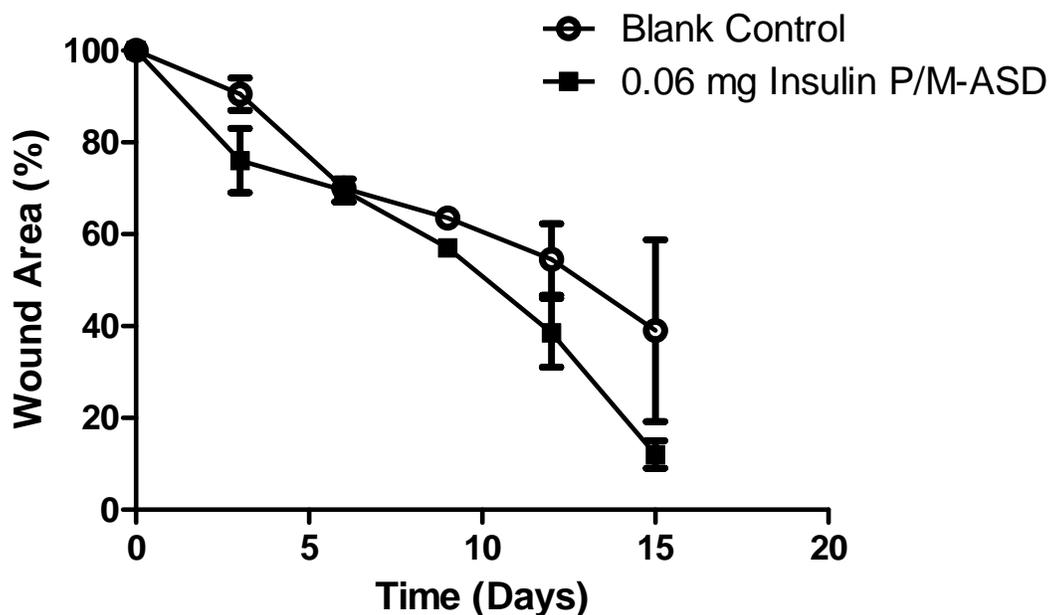


Figure 64. A plot of the wound healing response over a 15 day period with 2 control animals treated with blank and 2 control animals treated with insulin-loaded P/M-ASD.

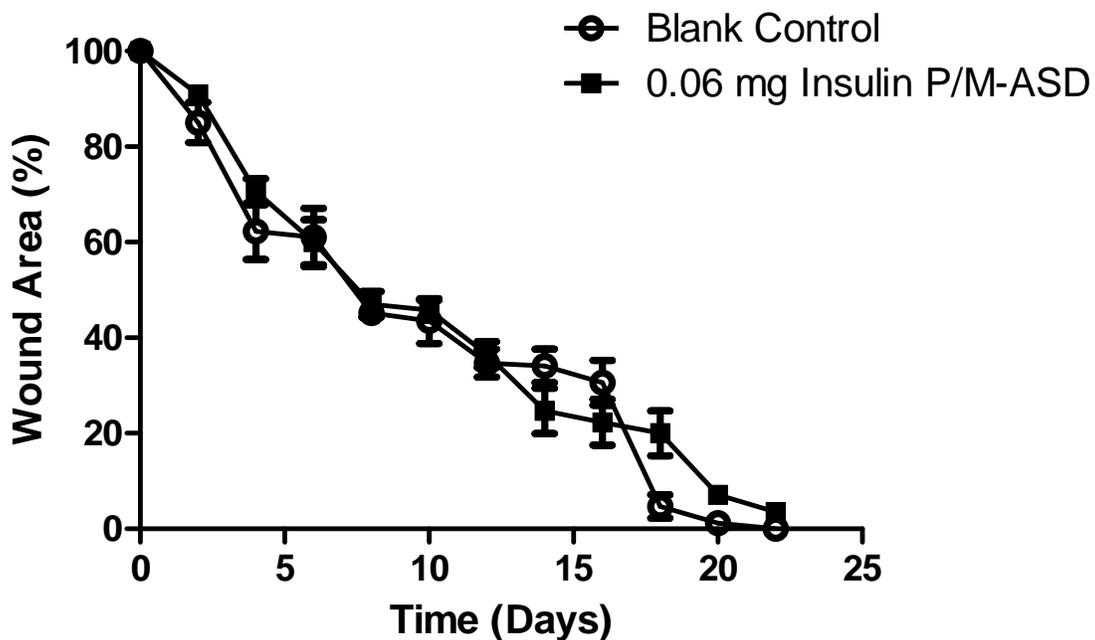


Figure 65. The wound healing response over a 22 day period of animals treated with full P/M-ASD loaded with 0.0625mg insulin, and blank controls. None of the P/M-ASD were cut. Animal study groups contained 4 rats each.

5.6 ASD Repeat Study with 0.125mg Insulin Dose

The present study was performed by Dr. Yan Liu, under the supervision of Prof. Martins-Green with samples made by Michael Hrynyk. P/M-ASD insulin loaded with 0.125 mg insulin were tested as full wound area dressings. Dressing changes were performed every three days for all test groups, at which point wound measurements were also determined. As seen in Figure 66 between days 0 to 3, wound re-epithelialization was significantly greater by 15% in the control group, as compared to the insulin treated group. From days 3 to 9, wound re-epithelialization was comparable between both groups, with no significant difference. After day 9, re-epithelialization in the insulin treated group was comparable to that of the control group. At the end of the study on day 15, 20% less wound area was measured in the insulin treated group as compared to the control group. This was a significant difference, however complete wound healing was not achieved in any of the test groups because the trial was terminated. The results are comparable to those in section 5.4 supporting the conclusion that a dose of 0.125 mg, or 0.04 mg/cm² is sufficient to stimulate significant differences in wound recovery over a 2 week period.

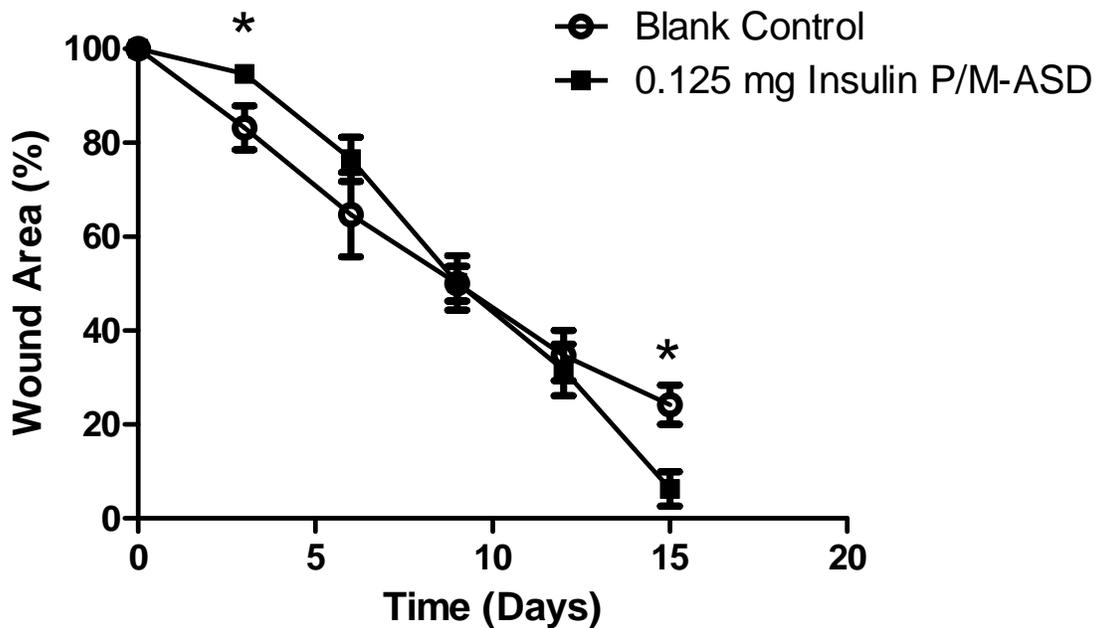


Figure 66. Wound area plotted over a 15 day period, comparing P/M-ASD loaded with 0.125 mg insulin and blank controls. Dressings were changed every 3 days. Asterisks indicate significance with a $p < 0.05$.

5.7 Verification of ASD Dose and Application Protocol

The present study was performed by Dr. Yan Liu, under the supervision of Prof. Martins-Green with samples made by Michael Hrynyk. P/M-ASD loaded with 0.125 mg insulin dose was tested on a group of 4 control and 4 insulin treated animals to verify the effectiveness of the dose. Wound dressings were changed every 3 days, along with wound area measurements plotted in Figure 67. Wound re-epithelialization was similar between both control and insulin treated groups for the first 6 days, with an overall recovery of 45-55% across both groups. On days 9-12, significant improvements of nearly 10% in wound healing were observed in animals treated with insulin, as compared to blank P/M-ASD. By the end of the study on Day 22, only 4% of the wound area remained, as compared to 17% in the control with no significant difference observed. Overall, the insulin treated animal achieved 13% more re-epithelialization than the

control animals, a recovery consistent with previous trials using a dose of 0.125 mg, or 0.04mg/cm².

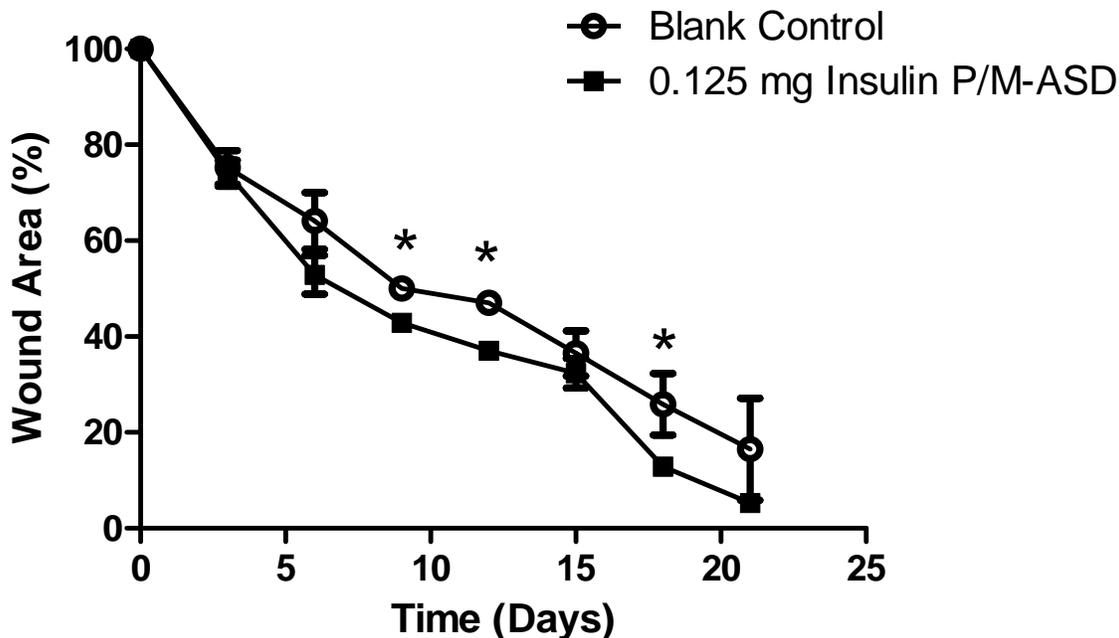


Figure 67. Wound area plotted over a 21 day period, comparing P/M-ASD loaded with 0.125 mg insulin and blank controls. Dressings were changed every 3 days. Asterisks indicate significance with a $p < 0.05$.

5.8 Summary

Animal experimentation, particularly involving wounding and burn injuries, is a technically and ethically challenging endeavour that is difficult to avoid in drug delivery research. The study operated under severe constraints imposed by the very necessary animal care protocols, limited access to animals, and extensive resources required. This study could not have been carried out without the support, supervision and direct involvement of Prof. Martins-Green and her lab staff.

In this study, insulin-loaded PLGA microparticles and alginate/PEG sponge dressings were tested on rats to determine the biocompatibility of the dressing materials, the method of application and the insulin dose required to stimulate healing. Histology, blood glucose and animal weight measurements provided information about the healing response on a molecular level and information about the animal's ability to cope with exogenous insulin delivery. Overall, a dose of 0.125 mg, or 0.04mg/cm² was able to reduce healing times by approximately 25% when applied every 3 days to partial-thickness scald burn wounds.

Burn wound injury itself, is a destructive injury that can causes pain that must be managed on a constant basis. As a result, the study was limited by the number of study animals available, thus each experiment and test animal was very carefully considered and planned. Without having access to large study groups, we have still been able to show that P/M-ASD containing insulin-loaded PLGA microparticles are capable of improving wound healing as compared to a blank control dressing. The current P/M-ASD has therefore, demonstrated the potential for sustained, controlled release of therapeutic peptides, as well as other growth factors, antibiotics, and pain killers which will be the subject of future studies.

Chapter 6

CONCLUSIONS, CONTRIBUTIONS TO KNOWLEDGE AND FUTURE DIRECTIONS

Skin is a dynamic and complex organ that relies on the interaction of different cell types, biomacromolecules and signaling molecules. When the skin becomes injured, a cascade of events is triggered to quickly restore the integrity of the skin. Depending on the size and severity of the wound, a dressing is typically used to provide a temporary barrier to protect the wound from dehydration, microorganisms and debris. Modern dressings are passive devices comprised of natural or synthetic fibers. Commonly, pharmaceuticals such as antibiotics and pain killers are used in conjunction to help with the recovery process. Unfortunately, modern dressings do not have the ability to promote faster recovery beyond the natural rate of healing. Consequently, for the first time, a PEG-alginate sponge wound dressing incorporating an insulin delivery device has been developed to address this issue. Although insulin has been documented to promote wound healing for the past century, no one has developed a successful delivery system for insulin to wound administration. As a result, the polymeric microparticle system developed in this study, has enabled the delivery of bioactive insulin for up to a 1 month period. The polymeric microparticles have been integrated into a composite wound dressing that supports moist wound healing, and enables insulin diffusion from the microparticles to the wound. Partial-thickness rodent burn models have shown that P/M-ASD reduced healing time by about 25%, thereby enabling a faster recovery and restoration of the skin's integrity.

This study has achieved the four main research goals originally outlined. Using PLGA microparticles, bioactive insulin delivery was sustained for a period of 25 days and demonstrated to have the ability to stimulate human keratinocyte migration *in vitro*. A dressing made from PEG and alginate was formulated incorporating the insulin-loaded PLGA microparticles.

Bioactive insulin release was achieved for nearly 3 weeks, along with favourable water handling and physical properties conducive for wound healing. Finally, animal testing confirmed that a constant dose of insulin from P/M-ASD loaded with 0.125mg, or 0.04mg/cm² insulin, with dressing changes every 3 days was sufficient to significantly improve wound healing as compared to an ASD without insulin.

The impact of this technology is significant in several areas of wound care. Firstly, P/M-ASD was demonstrated to be an effective drug delivery platform from which bioactive insulin can be released for an extended period of time, an accomplishment that has previously not been shown. The ability to stabilize bioactive insulin in a harsh wound environment for an extended period of time was achieved through the use of a highly stable crystalline form of the drug, which resists degradation during the liberation of acidic by-products during the polymer erosion process. Subsequently, a major hurdle in microencapsulating peptides in PLGA has been overcome through the development of this method. The significance of this discovery is that P/M-ASD can be adapted to deliver not only insulin, but other growth factors, analgesics and antimicrobials as well. No wound dressing on the market today is capable of providing this ability. Furthermore, the combination of alginate and PEG have enabled the dressing to be applied and removed without disturbing de novo tissue formed during the healing process. This is a substantial improvement over gauze or hydrocolloid dressings that often disturb the wound bed during dressing changes and contribute to increased scar tissue and pain for the patient. Therefore, this technology has advanced wound care in two major areas, both in modernizing the materials from which dressings are made, but also in the delivery of biologicals to ameliorate the healing process.

In future studies, three major areas of research focus will have to be examined to further refine this technology. The first area of research is examining the dissolution mechanism by which crystalline insulin dissolves and becomes released from the polymer. Our current understanding of the insulin release kinetics considers slow influx of water during polymer erosion results and a gradual dissolution of the crystals over extended period. To understand the impact of this on the release kinetics, a future study using soluble forms of insulin could help to elucidate the exact mechanism by which insulin is released, and how it could be further attenuated. The second area of focus is examining the biochemical events that take place during the wound healing process during insulin chemotherapy. The current focus has been on the role of keratinocytes and does not examine the role that fibroblasts have on the wound healing process. By examining the temporal release of insulin and the biochemical events that stimulate keratinocyte and fibroblast migration, a better determination of the therapeutic window can be established. This refinement in dose level and timing can help determine the optimal conditions under which insulin is released from P/M-ASD to be effective. Finally, the introduction of other growth factors such as EGF and PDG, along with antimicrobials would be the next step in advancing the current P/M-ASD formulation. The incorporation of additional growth factors would enable amplification of insulin's effects by stimulating different arms of the migration and proliferation pathways in keratinocytes and fibroblasts. Likewise, the incorporation of antimicrobials such as peptoids, antibiotics and silver sulfadiazine would help to maintain a microorganism-free environment and decrease the risk of developing a chronic wound. In addition, a slow release of pain killers such as lidocaine would reduce pain and improve patient well-being. As such, the P/M-ASDs are an excellent topical drug delivery platform that offers the potential to incorporate other molecules to accelerate wound healing.

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