An - Najah National University Faculty of Graduate Studies

Rheological Properties for Olive Oil in Palestine

By

Ahmad Mustafa Bahti

Supervisor

Prof. Issam Rashid Abdelraziq

Co - Supervisor

Dr. Sharif Mohammad Musameh

This Thesis is Submitted in Partial Fulfillment of the Requirements for the Degree of Master of Physics, Faculty of Graduate Studies, An-Najah National University - Nablus, Palestine.

Rheological Properties for Olive Oil in Palestine

By

Ahmad Mustafa Bahti

This thesis was successfully defended on 6/3/2014 and approved by:

Defense Committee Members

-Prof. Dr. Issam Rashid Abdelraziq (Supervisor)

-Dr. Sharif Mohammad Musameh (Co - Supervisor)

-Dr. Abedel - Rahman Abu - Lebda (External Examiner)

-Prof. Dr. Ghassan Saffarini (Internal Examiner)

Signature

Dedication

This thesis is dedicated to my parents, as well as, to my brothers and sisters. With respect and love.

Acknowledgement

I would like to express my sincere appreciation to my supervisors, Prof. Dr. Issam Rashid Abdelraziq for his helpful comments and continual encouragement, and Dr. Sharif Mohammed Musameh, for his cooperation which helped me in the completion of this research.

Special thanks to the members working in the laboratories of the Faculty of Science for their cooperation, whom contributed considerably to the completion of this work, Mr. Omir Nabulsi, Mr. Mohammad Bahjat, Mr. Sameeh Abed Al - Azeez, Mr. Ameed Amirah, Mr. Mohammad Al - Masri, Mr. Mohammad Al - Qareene, and Mr. Nafeth Zakrea.

الإقرار

أنا الموقع أدناه مقدم الرسالة التي تحمل العنوان:

Rheological Properties for Olive Oil in Palestine

اقر بأن ما اشتملت عليه هذه الرسالة إنما هو نتاج جهدي الخاص، باستثناء ما تمت الإشارة إليه حيثما ورد، وان هذه الرسالة ككل أو جزء منها لم يقدم من قبل لنيل أية درجة أو بحث علمي أو بحثي لدى أية مؤسسة تعليمية أو بحثية أخرى .

Declaration

The work provided in this thesis, unless otherwise referenced, is the researcher's own work, and has not been submitted elsewhere for any other degree or qualification.

Student's name:	اسم الطالب:
Signature:	التوقيع:
Date:	التاريخ:

Table of Contents

No.	Contents	Page
	Dedication	III
	Acknowledgement	IV
	Declaration	V
	Table of Contents	VI
	List of Tables	VIII
	List of Figures	X
	List of Abbreviations	XV
	Abstract	XVI
	Chapter One: Introduction	1
1.1	Olive Oil	1
1.2	Previous Studies	1
1.3	Objectives of the Study	5
1.4	Organization of the thesis	5
	Chapter Two: Theoretical Formulation	7
2.1	Rheology	7
2.2	Viscosity	7
2.2.1	Newtonian Systems	8
2.2.2	Non - Newtonian Systems	8
2.3	The Dependence on Temperature	9
2.3.1	The Andrade's Equation	10
2.3.2	Abramovic's Equation	10
2.3.3	Three Constant Andrade's Equation	11
2.4	Models Regarding Stress and Strain Rate	12
2.4.1	Newtonian Fluid	12
2.4.2	Non - Newtonian Fluid Models	13
2.5	Olive Oil Acidity	14
2.6	Refractive Index	15
2.7	Mass Density	15
	Chapter Three: Methodology	16
3.1	Measurement Equipment	17
3.1.1	Viscosity Apparatus	17
3.1.2	Temperature Apparatus	17
3.1.3	Density Apparatus	18
3.1.4	Refractive Index Apparatus	18
3.1.5	Acidity Measurement	19
	Chapter Four: Results and Analysis	21
4.1	Density Results	21
4.2	Refractive Index Results	23

4.3	Acidity Results	26
4.4	Refractive Index as a Function of Density of Olive Oil	31
4.5	Viscosity Results	33
4.5.1	Viscosity Results and Theoretical Predictions	35
4.5.2	Modifications of the Abramovic's and Andrade's	42
	equations	42
4.5.3	Proposed Equations	52
4.6	Shear Stress Versus Shear Rate	62
	Chapter Five: Discussion and Conclusion	66
	References	69
	Appendix	75
	الملخص	ب

VIII

List of Tables

No.	Caption	Page	
Table (2.1)	Constants of equations 2.5 and 2.6	11	
Table (2.2)	Constants of equation 2.7	12	
Table (2.3)	Classification of olive oil according to FFA%	15	
Table (4.1)	Measured density (gm/cm ³) of olive oil samples in	21	
	different regions and for different storage ages	<i>4</i> 1	
Table (4.2)	Measured refractive index of olive oil samples of	24	
	different regions and different storage ages		
Table (4.3)	Measured acidity in FFA% of olive oil samples in	27	
	different regions for different storage ages	-	
Table(4.4)	The acidity of olive oil samples from different	28	
	regions of different storage ages		
Table(4.5)	Results of kinematic viscosity of L_1 and L_8 samples	33	
	at different storage age.		
Table (4.6)	Results of dynamic viscosity of L_1 and L_8 samples at different storage egg	34	
	The measured and calculated dynamic viscosity of		
	olive oil versus temperature using Abramovic's		
Table (4.7)	onve on versus temperature using Abramovie's A	37	
	equation (formula 1) $[\log \eta = \frac{1}{T} - B]$ for L ₁ and L ₈		
	The measured and calculated dynamic viscosity of		
Table (4.8)	olive oil versus temperature using Abramovic's	39	
	equation (formula 2)[$\eta = A - B \log t$] for L ₁ and L ₈		
	The measured and calculated dynamic viscosity \mathcal{P}		
Table (4.9)	using Andrade's equation $[\ln \eta = A + \frac{B}{T} + CT]$	/1	
1 abic (4.7)	of olive oil samples collected from L_1 and L_8 versus	41	
	temperature		
	The measured and calculated dynamic viscosity		
	using modified Abramovic's equation (formula1)		
Table (4.10)	$\left[\log n = \frac{A}{2} - B\right]$ of olive oil samples collected from	44	
	L_1 and L_2		
	The measured and calculated dynamic viscosity		
Table (4.11)	using modified Abramovic's equation(formula2)		
	$[n = A - B \ loat]$ of olive oil samples from L ₁ and	47	
	L_8 versus temperature		
	The measured and calculated dynamic viscosity		
Table (4.12)	using modified Andrade's equation		
	$[\ln \eta = A + \frac{B}{2} + CT]$ of olive oil samples from L ₁	50	
	and L $_{\circ}$ versus temperatur		
	and L8 versus temperatur		

	IX	
Table (4.13)	The measured and calculated dynamic viscosity using the proposed equation (formula 1) $[\eta = A + Bt + Ct^2 + Dt^3]$ of olive oil samples collected from L ₁ and L ₈ versus temperature	54
Table (4.14)	The measured and calculated dynamic viscosity using proposed equation (formula 2) $[\eta = Aln(-Blnt)]$ of olive oil samples collected from L ₁ and L ₈ versus temperature	56
Table (4.15)	The measured and calculated dynamic viscosity using proposed equation (formula 3) $[\eta = A + Bt + Ct^2 + Dt^3 + Et^4 + Ft^5]$ of olive oil samples collected from L ₁ and L ₈ versus temperature	59
Table (4.16)	Values of AAD% obtained from fitting viscosity- temperature relationship using different equations.	61
Table (4.17)	The results of power law fit to experimental data form L_1 and L_8	63
Table (5.1)	The linear equations by linear fitting for density, refractive index and acidity versus storage age, and refractive index versus density for different regions	67
Table (A.1)	Measured density of olive oil samples in different regions and for different storage AGES	75
Table (B.1)	Measured refractive index of olive oil samples in different region and for different storage ages	81
Table (C.1)	Measured acidity of olive oil samples in different regions and for different storage ages	88

X List of Figures

No.	Caption			
Figure (2.1)	Flow curve of a Newtonian fluid at power law	12		
Figure (2.2)	Flow curve of a Non - Newtonian fluid at power law	14		
Figure (3.1)	NDJ - 1 Rotational Viscometer	17		
Figure (3.2)	Pycnometer	18		
Figure (3.3)	The way - 2s ABBE digital refractometer	19		
Figure (4.1)	Measured density versus storage age of olive oil samples of L ₁	22		
Figure (4.2)	Measured density versus storage age of olive oil samples of L_{0}			
Figure (4.3)	Measured refractive index versus storage age of olive oil samples of L_1	25		
Figure (4.4)	Measured refractive index versus storage age of olive oil samples of L_8	25		
Figure (4.5)	Measured acidity versus storage age of olive oil samples of L_1	30		
Figure (4.6)	Measured acidity versus storage age of olive oil samples of L_8	30		
Figure (4.7)	Measured refractive index versus density of olive oil samples of L_1	31		
Figure (4.8)	Measured refractive index versus density of olive oil samples of L_8	32		
Figure (4.9)	Measured viscosity versus storage age of olive oil samples of L_1 and L_8 at 25 °C	35		
Figure (4.10)	Measured and calculated values of dynamic viscosity using Abramovic's equation (formula 1) for four different samples	38		
Figure (4.11)	Measured and calculated values of dynamic viscosity using Abramovic's equation (formula 2) for four different samples from different regions	40		
Figure (4.12)	Measured and calculated values of dynamic viscosity using Andrade's equation for four different samples.	42		
Figure (4.13)	Measured and calculated values of dynamic viscosity of olive oil of L_1 using modified Abramovic's equation (formula 1)	45		
Figure (4.14)	Measured and calculated values of dynamic viscosity of olive oil of L_8 using modified Abramovic's equation (formula 1)	45		
Figure (4.15)	Measured and calculated values of dynamic viscosity of olive oil of L_1 using modified Abramovic's equation (formula 2)	48		

	711	
Figure (4.16)	Measured and calculated values of dynamic viscosity of olive oil of L_8 using modified Abramovic's equation (formula 2)	48
Figure (4.17)	Measured and calculated values of dynamic viscosity of olive oil of L_1 using modified Andrade's equation	51
Figure (4.18)	Measured and calculated values of dynamic viscosity of olive oil of L_8 using modified Andrade's equation	51
Figure (4.19)	Measured and calculated values of dynamic viscosity using proposed equation (formula1) fit for olive oil sample from L_1 and L_8	55
Figure (4.20)	Measured and calculated values of dynamic viscosity using proposed equation (formula 2) fit for olive oil sample from L_1 and L_8	57
Figure (4.21)	Measured and calculated values of dynamic viscosity using proposed equation (formula 3) fit for olive oil sample from L_1	60
Figure (4.22)	Measured and calculated values of dynamic viscosity using proposed equation (formula 3) fit for olive oil sample from L_8	60
Figure (4.23)	Relationship between shear stress and shear rate for olive oil sample from L_1 (2 years storage age) at different temperatures.	64
Figure (4.24)	Relationship between shear stress and shear rate for olive oil sample from L_1 (13 years storage age) at different temperature	64
Figure (A.1)	Measured density versus storage age of olive oil samples of L_2	76
Figure (A.2)	Measured density versus storage age of olive oil samples of L_3	76
Figure (A.3)	Measured density versus storage age of olive oil samples of L_4	77
Figure (A.4)	Measured density versus storage age of olive oil samples of L_5	77
Figure (A.5)	Measured density versus storage age of olive oil samples of L_6	78
Figure (A.6)	Measured density versus storage age of olive oil samples of L_7	78
Figure (A.7)	Measured density versus storage age of olive oil samples of L_9	79
Figure (A.8)	Measured density versus storage age of olive oil samples of L_{11}	79
Figure (A.9)	Measured density versus storage age of olive oil samples of L_{12}	80

XI

XII			
Figure (A.10)	Measured density versus storage age of olive oil samples of L ₁₂	80	
Figure (B.1)	Measured refractive index versus storage age of olive oil samples of L_2	82	
Figure (B.2)	Measured refractive index versus storage age of olive oil samples of L_3	82	
Figure (B.3)	Measured refractive index versus storage age of olive oil samples of L_4	83	
Figure (B.4)	Measured refractive index versus storage age of olive oil samples of L_5	83	
Figure (B.5)	Measured refractive index versus storage age of olive oil samples of L_6	84	
Figure (B.6)	Measured refractive index versus storage age of olive oil samples of L_7	84	
Figure (B.7)	Measured refractive index versus storage age of olive oil samples of L_9	85	
Figure (B.8)	Measured refractive index versus storage age of olive oil samples of L_{10}	85	
Figure (B.9)	Measured refractive index versus storage age of olive oil samples of L_{11}	86	
Figure (B.10)	Measured refractive index versus storage age of olive oil samples of L_{12}	86	
Figure (B.11)	Measured refractive index versus storage age of olive oil samples of L_{13}	87	
Figure (C.1)	Measured acidity versus storage age of olive oil samples of L_2	89	
Figure (C.2)	Measured acidity versus storage age of olive oil samples of L_3	89	
Figure (C.3)	Measured acidity versus storage age of olive oil samples of L_4	90	
Figure (C.4)	Measured acidity versus storage age of olive oil samples of L_5	90	
Figure (C.5)	Measured acidity versus storage age of olive oil samples of L_6	91	
Figure (C.6)	Measured acidity versus storage age of olive oil samples of L_7	91	
Figure (C.7)	Measured acidity versus storage age of olive oil samples of L_9	92	
Figure (C.8)	Measured acidity versus storage age of olive oil samples of L_{10}	92	
Figure (C.9)	Measured acidity versus storage age of olive oil samples of L_{11}	93	
Figure (C.10)	Measured acidity versus storage age of olive oil samples of L_{12}	93	

	XIII	
Figure (C 11)	Measured acidity versus storage age of olive oil	
Figure (C.II)	samples of L ₁₃	74
Figure (D.1)	Measured refractive index versus density of olive	95
- igui (201)	oil samples of L_2	20
Figure (D.2)	Measured refractive index versus density of olive	95
	Oil samples of L_3	
Figure (D.3)	Measured refractive index versus density of onve	96
	On samples of L_4 Massured refractive index versus density of elive	
Figure (D.4)	oil samples of L.	96
	Measured refractive index versus density of olive	
Figure (D.5)	oil samples of L	97
	Measured refractive index versus density of olive	
Figure (D.6)	oil samples of L_7	97
	Measured refractive index versus density of olive	
Figure (D.7)	oil samples of L_9	98
F* (D 9)	Measured refractive index versus density of olive	09
Figure (D.8)	oil samples of L_{10}	98
Figure (D 0)	Measured refractive index versus density of olive	00
Figure (D.9)	oil samples of L_{11}	99
Figure (D 10)	Measured refractive index versus density of olive	99
Figure (D.10)	oil samples of L ₁₂	"
	Measured viscosity versus temperature of olive oil	100
Figure (E.1)	samples of L_2 (1year and 3 years storage age)	100
	Measured viscosity versus temperature of olive oil	100
Figure (E.2)	samples of L_2 (14 years and 15 years storage age)	
	Measured viscosity versus temperature of olive oil	101
Figure (E.3)	samples of L_3 (1 year and 2 years storage age)	
Figure (E.4)	Measured viscosity versus temperature of onve on $(14 \text{ years and } 15 \text{ years storage age})$	101
	samples of L_3 (14 years and 15 years storage age)	
Figure (E.5)	samples of L $(1 \text{ year and } 2 \text{ years storage age})$	102
	Measured viscosity versus temperature of olive oil	
Figure (E.6)	samples of L_4 (3 years and 4 years storage age)	102
	Massured viscosity versus temperature of olive oil	
Figure (F 7)	samples of L $_{c}$ (1 year 2 years and 3 years storage	103
riguit (E.7)	age)	105
Figure (E.8)	Measured viscosity versus temperature of olive oil	102
	samples of L_5 (6 years, 7 years and 8 years storage	105
Figure (E.9)	Measured viscosity versus temperature of olive oil	104
	samples of L_6 (5 years and 4 years storage age)	
Figure (E.10)	f interview viscosity versus temperature of onve of f is a samples of f is (12 years and 13 years storage age)	104
1	samples of $L_6(12)$ years and 13 years storage age)	

	XIV			
Figure (E.11)	Measured viscosity versus temperature of olive oil samples of L_6 (14 years, 15 years and 16 years storage age)	105		
Figure (E.12)	Measured viscosity versus temperature of olive oil samples of L_7 (3 years and 4 years storage age)	105		
Figure (E.13)	Measured viscosity versus temperature of olive oil samples of L_7 (5 years, 6 years and 19 years storage age)	106		
Figure (E.14)	Measured viscosity versus temperature of olive oil samples of L_9 (2 years and 3 years storage age)	106		
Figure (E.15)	Measured viscosity versus temperature of olive oil samples of L_9 (5 years and 7 years storage age)	107		
Figure (E.16)	Measured viscosity versus temperature of olive oil samples of L_{10} (1 year and 14 years storage age)	107		
Figure (E.17)	Measured viscosity versus temperature of olive oil samples of L_{10} (14 years and 15 years storage age)	108		

AAD%	Percentage of Average Absolute Deviation
ANOVA	Analysis of Variance
CGS	Centimeter-Gram-Second
cP	Centipoise
cSt	Centistokes
Ea	Activation Energy
Eq.	Equation
EVOO	Extra Virgin Olive Oil
VOO	Virgin Olive Oil
FFA	Free Fatty Acids
Fig.	Figure
IOOC	International Olive Oil Council
N	Newton
Р	poise
Pa	Pascal
P-value	Probability
R	Gas Constant
RPM	Revolution Per Minute
SD	Standard Deviation
SI	Système International d' Unités
SMC	Spindle Multiplier Constant
SP	Spindle
T	Temperature in Kelvin
L	Saida Region
L_2	Allar Region
L_3	Beit Lid Region
L_4	NazlatIssa Region
L ₅	Meithaloon Region
L ₆	Jeet Region
L ₇	Ti'innik Region
L_8	Yasid Region
L ₉	Borqa Region
L_10	Arraba Region
L_11	Asira Region
L ₁₂	Beta Region
L ₁₃	Jenin Region
ρ	Density
η	Dynamic Viscosity
ν	Kinematic Viscosity
η_{cal}	Calculated Dynamic Viscosity
η_{exp}	Measured Dynamic Viscosity

XV List of Abbreviations

XVI Rheological Properties for Olive Oil in Palestine By Ahmad Mustafa Bahti Supervisor Prof. Issam Rashid Abdelraziq Co - Supervisor Dr. Sharif Mohammad Musameh

Abstract

In this study, olive oil samples of different storage ages and regions in Palestine were studied. The density, refractive index, acidity and viscosity of the samples were measured. The refractive index of the olive oil samples were studied against storage ages and results showed that the refractive index decreases as a function of storage age. The acidity of olive oil samples from different regions and different crops showed that acidity increases as a function of storage age. Most of olive oil samples (storage age ≤ 12 years) acidity did not exceed the international quality standards (< 3.3%). It is not noting that olive oil can be stored until 12 years without exceeding the international quality standards of acidity in proper conditions.

The viscosity of olive oil samples of 2012 crop from different regions was studied, and the results showed that most of the olive oil samples are classified to be extra - virgin.

Two and three constant equations were proposed to obtain more suitable prediction of temperature dependence of dynamic viscosity of olive oil. The experimental results of viscosity were compared with the power law equation, and the behavior of olive oil was found to be Newtonian.

Chapter One Introduction

1.1 Olive Oil

Olive oil is a fat obtained from the olive fruit by mechanical or chemical means. Olive oil is commonly used in cooking, cosmetics, pharmaceuticals, soaps, and as a fuel for traditional oil lamps. Olive oil is used throughout the world, but especially in the Mediterranean countries and, in particular, in Greece, which has the highest consumption per person (NAOOA, 2013).

Olives are very important for the Palestinian, not only because they are the biggest crop in what remains a largely agricultural economy, but also for their deep cultural significance as a symbol of traditional society and ties to the land. It is estimated that olive trees account for nearly 45 percent of cultivated land in Palestine and in good years can contribute as much as 15 - 19 percent of agriculture output. Given that agriculture accounts for nearly 25 percent of gross domestic product, olives are an important element of the Palestinian economy and estimates suggest that about 100,000 families depend to some extent upon the olive harvest for their livelihoods (The World Bank, 2012).

1.2 Previous Studies

Vegetable oils have become increasingly important for nutritional purposes and in a wide range of industrial applications which include fuels, skin care products, high pressure lubricants and alkyd resins for paint. These applications require extensive studies on the physic - chemical properties of oils in order to ascertain their suitability as raw materials. Such properties include viscosity and acidity which are an important parameters in the design of process equipment for oils (Eromosele, and Paschal,2003). (Nierat *et al.*, 2012).

In 1988, a model was used by Patil and his group to describe the liquid liquid thermal hypothesis of vegetable oils. Extensive data on hydrolysis equilibrium and rate have been obtained (Patil *et al.*, 1988).

Noureddin and his group have presented the range of temperatures in which the viscosity and temperature of vegetable oils are correlated (Noureddini *et al.*, 1992).Van Wazer and his group discussed the sensitivity of the eye in judging viscosity of Newtonian liquids (James, 1996).

Studies by Bayrak on vegetable oils determined the relationship between viscosity and average molecular weight (Bayrak *et al.*, 1997). Hsieh predicted viscosity of vegetable oils from density data (Hsieh *et al.*, 1999). The palm oil was proved to be a good diesel - generator fuel by Almeidaa and his group, they found that the performance of diesel generator is increased by increasing the palm oil temperature (Almeidaa *et al.*, 2002).

In a given range of temperatures, Farhoosh found that the natural logarithm of the kinetic rate of five different vegetable oils varies linearly with respect to temperature (Farhoosh *et al.*, 2008).

In his study, Ahmad evaluated the viscosity changes of vegetable oils, and fitted the viscosity with well - known rheological equations (Ahmad *et al.*, 2009). He identified model limitation through graphical and numerical

observations. Vegetable oils were subjected to viscometer measurements of viscosity at shear rate (3 - 100 RPM) and temperature (40 - 100°C).

Stanciu proposed four relationships of dynamic viscosity temperature dependence for vegetable oils. In his studies he found a polynomial or exponential dependence between temperature and dynamic viscosity of vegetable oil using the Andrade's equation changes (Stanciu, 2012).

Adnan and his group studied the characterization of different oils and their rheological properties. Eight different natural oils, namely olive, coconut, almond, castor, sesame, cotton seed, sunflower, and paraffin oils. All the oils investigated were found to possess Non - Newtonian behavior (Adnan *et al.*, 2009).

Effect of fatty acid composition on dynamic and steady shear rheology of oils was studied by Hasan Yalcin and his group (Yalcin *et al.*, 2012).

Studies in New Zealand by Sims indicated that vegetable oils, particularly rapeseed oil, could be used as a replacement for diesel fuel (Sims *et al.*, 1981).

Reid and his group evaluated the chemical and physical properties of 14 vegetable oils. These injection studies pointed out that the oils behave very differently from petroleum - based fuels (Reid *et al.*, 1989).

Goering and his group studied the characteristic properties of eleven vegetable oils to determine which oil would be the best suited for use as an alternative fuel source (Goering *et al.*, 1981).

Bruwer and his group studied the use of sunflower seed oil as a renewable energy source. When operating tractors with 100% sunflower oil instead of diesel fuel, an 8% power loss occurred after 1000 hours of operation (Bruwer *et al.*, 1981).

Viscosity of oil samples used by Fasina and Colley was shown to decrease with temperature, in the same study they found that the specific heat capacity increases with increasing temperature (Fasina and Colley, 2008).

In their work, Toscano and his group investigated the effects of two analytical chemical parameters, their results confirmed the incidence of molecular characteristics of triglycerides of an oil with respect to its viscosity (Toscano *et al.*, 2012). The concentration of polyunsaturated fatty acids (PUFAs) was found to be a predominant parameter that influences the low - temperature properties of vegetable oil - based lubricants (Quinchia *et al.*, 2012).

Many researches concerned on olive oil as a vegetable oil, Lupi and his group prepared different samples of olive oil based organogels by using cocoa butter, dynamic temperature ramp tests, carried out at 5 °C/min, allowed the determination of rheological characteristics (Lupi *et al.*, 2012). To study the influence of operative conditions adopted during the malaxation of pastes on the quality of resulting oils, Angerosa and his coworkers found that low temperatures and times, ranging between 30 and 45 min, according to the rheology of the olive pastes, were the optimal operative conditions for the malaxation (Angerosa *et al.*, 2001).

The relationship between density, viscosity, oil/water interfacial tension and structure of vegetable oils after heating at frying temperatures were studied by Adolfo. He aimed to explore the possibility of reusing waste vegetable oils as solid agglomerants for different purposes. Commercial olive and sunflower oils were heated at 150 and 225 0 C in the time interval of 1 – 15 days to achieve a wide range of alteration degrees. Structural changes in the oils were monitored, of the two vegetable oils studied, sunflower oil was found to be more sensitive to thermal treatment, undergoing greater changes in its properties, especially in viscosity, which may show a marked increase (Adolfo *et al.*, 2006).

1.3 Objectives of the Study

- The goal of this work is to check whether olive oil in Palestine shows Newtonian or Non Newtonian behavior.
- The physical properties (viscosity, acidity, density, refractive index) of olive oil in Palestine will be measured and compared with standard values.
- The viscosity of olive will be measure as a function of temperature.
- Equation of the behavior will be suggested to describe the rheoligical effect.
- The experimental data will be fitted by using SPSS and Excel programs to get the relationship of dynamic viscosity as a function of temperature.

1.4 Organization of the Thesis:

Given below is a brief outline of the topics discussed in this thesis:

 Introduction: the characteristics of olive oil are presented here, previous studies concerning the problems into account, and the objectives of the study.

- 2- Theoretical formulation: the theory of rheology, viscosity, stress and strain rate, acidity of oils, refractive index and mass density.
- 3- Methodology: the samples used in the research and the function of the equipments used in measurements.
- 4- Results and analysis of data obtained.
- 5- Discussion about different results obtained by analysis of measured data.

Chapter Two

Theoretical Formulation

2.1 Rheology

Rheology is defined as the branch of physics that studies the deformation and flow of matter (Larson, 1999).

Rheology applies to substances which have a complex microstructure, such as muds, suspensions, polymers and other glass formers, it also applies to many foods and additives, bodily fluids and other biological materials or other materials which belong to the class of soft matter (Themelis, 1995).

The role of rheology is important in the field of cosmetic science, especially in the field of emulsions and lotions (Martin *et al.*, 2004).

Since the different creams have different consistencies and they are used for long terms, the effects of different rheological parameters of oils are studied for understanding the performance of the system (Remington, 2006).

2.2 Viscosity

Viscosity is a measure of the resistance to flow or shear. Viscosity can also be termed as a drag force and is a measurement of the frictional properties of the fluid. It can be expressed in two distinct forms:

- a. Dynamic viscosity (η)
- b. Kinematic viscosity (v)

Dynamic viscosity is defined as the ratio of shear stress (force over cross section area) to the rate of deformation (the difference of velocity over a sheared distance), and it is presented as:

$$\eta = \frac{\tau}{\frac{\partial u}{\partial x}}$$
(2.1)

Where, η is the dynamic viscosity in Pascal-second (Pa.s); τ is shear stress (N/m²); and, $\frac{\partial u}{\partial x} = \gamma$ is rate of deformation or velocity gradient or better

known as shear rate (1/s) (Dutt N. et al, 2007).

The Kinematic viscosity requires knowledge of mass density of the liquid (ρ) at that temperature and pressure. It is defined as:

$$v = \frac{\eta}{\rho} \tag{2.2}$$

Where, v is kinematic viscosity in centistokes (cSt), ρ is in g/cm³ (Dutt N. *et al*, 2007).

The flow characteristics of liquids are mainly dependent on viscosity and are broadly divided into two categories:

- 1- Newtonian systems.
- 2- Non Newtonian systems.

2.2.1 Newtonian Systems

These fluids have the same viscosity at different shear rates (different revolution per minute) (rpm). These fluids are called Newtonian over the shear rate range they are measured. Water is an example of these fluids (James F., 1996).

2.2.2 Non - Newtonian Systems

These fluids have different viscosity at different shear rates. They are classified into two groups:

a) Time Independent

Time independent means that the viscosity behavior does not change as a function of time when it is measuring at a specific shear rate. Pseudoplastic materials such as lava, ketchup, whipped cream, and blood are examples of such fluids which display decrease in viscosity with an increase in shear rate. This type of fluid is known as "shear thinning".

b) Time Dependent

Time Dependent means that the viscosity behavior changes as a function of time when measuring at a specific shear rate (the duration for which the fluid has been subjected to shearing as well as their previous kinematic history). A thixotropic material is an example of that fluid which has decreasing viscosity under constant shear rate. Many gels are classified to be thixotropic material (James, 1996).

In this study, the viscosity of different olive oil samples will be measured as a function of shear rates over a given shear rate range and at different temperatures.

The nature of liquids is complex, so there has been no comprehensive theory explaining the relationship between liquid viscosity and other properties, so empirical methods are used in addition to mathematical expressions to get the best fit of the experimental data.

2.3 The Dependence on Temperature

Clements and his group was the first who fitted the dependence of viscosity on temperature using the Arrhenius - type relationship which is given by:

$$\eta = \eta_{\infty,T} e^{\frac{E_a}{RT}} \tag{2.3}$$

Where η is the dynamic viscosity in Pa.s, $\eta_{\infty,T}$ is the viscosity at infinite temperature in Pa.s, E_a is the exponential constant that is known as activation energy (J/mol); R is the gas constant (J/mol.K) and T is the absolute temperature in Kelvin (Ahmad, 2009; Clements *et al.*, 2006).

When applied to real phenomena, equation (2.3) failed to provide good representation, so new models are needed.

2.3.1 The Andrade's Equation

The Andrade's equation is the simplest form of representation of liquid dynamic viscosity as a function of temperature, it takes the form:

$$\eta = A e^{\frac{B}{T}} \tag{2.4}$$

Where η is the dynamic viscosity in cP, T is the temperature in Kelvin, A is a constant in cP, and B is a constant Celsius, A and B are characteristics of each substance (De Guzman, 1913; Andrade, 1930).

The constants of Andrade's equation were evaluated experimentally for a number of substances by several researchers such as Duhne, Dutt and his group, Visvwanath and Natarjan (Duhne, 1979; Dutt *et al.*, 2007; Natarajan *et al.*, 1989)

2.3.2 Abramovic's Equations

Abramovic proposed the following equations to describe the dependence of dynamic viscosity on temperature:

Abramovic's equation (formula 1):

$$\log \eta = \frac{A}{T} - B \tag{2.5}$$

Abramovic's equation (formula 2):

$$\eta = A - Blog t \tag{2.6}$$

Where η is the dynamic viscosity in cP, T is the temperature in Kelvin, t is the temperature in degrees Celsius. A and B are constants.

Table 2.1 shows the values of the constants of the above two equations for olive oil (Abramovic *et al.*, 1998)

	$\log \eta = \frac{A}{T} - B$			$\eta = A - Blog t$		
Oil	A (K)	В	η (cP) at	A (cP)	В	η (cP) at
			298.15 K		(cP)	25.15 °C
Olive oil	1558.2	3.433	62.12	235.4	124.1	61.59
Refined corn						
oil	1464.1	3.207	50.54	186.6	97.4	50.19
Refined						
sunflower oil	1443.3	3.157	48.29	177.2	92.3	47.93

 Table (2.1): Constants of equations 2.5 and 2.6

2.3.3 Three Constant Andrade's Equation

Three constants equation was proposed by Andrade which is used by Abramovic. The equation has the following form :

$$\ln \eta = A + \frac{B}{T} + CT \tag{2.7}$$

Where η is the dynamic viscosity in cP, T is the temperature in Kelvin. A,

B and C are constants (Vogel, 1921; Andrade, 1930; Abramovic, 1998).

Table (2.2) shows the values of the constants A, B, and C of equation 2.7(Andrade, 1930; Abramovic *et al.*, 1998).

	$ln \eta = A + \frac{B}{T} + CT$			
Oil	А	B (K)	$C(K^{-1})$	η (cP) at
				298.15 K
Olive oil	-32.72	7462.27	0.04	69.03
Refined corn oil	-27.89	6572.41	0.03	22.16
Refined sunflower oil	-28.09	6575.60	0.03	18.34

Table (2.2): Constants of equations 2.7

2.4 Models Regarding Stress and Strain Rate

2.4.1 Newtonian Fluid

Newtonian fluid means that when shear stress is plotted against shear rate at a given temperature, the plot shows a straight line with a constant slope that is independent of shear rate. (Fig.2.1)



Fig. (2.1): Flow curve of a Newtonian fluid at power law

The simplest constitutive equation is Newton's law of viscosity:

$$\tau = \eta \dot{\gamma} \tag{2.8}$$

where η is the Newtonian viscosity and $\dot{\gamma}$ is the shear rate or the rate of strain. The Newtonian fluid is the basis for classical fluid mechanics. Gases, for example, exhibit characteristics of Newtonian viscosity.

2.4.2 Non - Newtonian Fluid Models

One of the most widely used forms of the general non - Newtonian constitutive relation is a power law model, which can be described as (Middleman, 1968; Munson *et al.*, 1998; Bird *et al.*, 1987):

$$\tau = m\dot{\gamma}^n \tag{2.9}$$

Where τ is stress and $\dot{\gamma}$ is strain rate, m and n are power-law model constants. The constant, m is a measure of the consistency of the fluid with dimensions of cP.(s)ⁿ⁻¹, the higher the *m* is, the more viscous the fluid is. *n* is a measure of the degree of non - Newtonian behavior. The greater the departure from the unity, the more pronounced the non - Newtonian properties of the fluid are.

The viscosity for the power - law fluid can be expressed as (Middleman, 1968; Munson *et al.*, 1998; Bird *et al.*, 1987):

$$\eta = m \dot{\gamma}^{n-1} \tag{2.10}$$

Where η is non - Newtonian apparent viscosity, if n = 1, a Newtonian fluid is obtained. If n deviates from 1, a non - Newtonian fluid is obtained. (Fig.2.2).



Fig.(2.2): Flow curve of a Non - Newtonian fluid at power law

Herschel - Bulkley described the behavior of fluids by the following equation:

$$\tau = m\gamma^n + \tau_\gamma$$
, where $\tau \ge \tau_\gamma$
 $\tau = 0$, where $\tau \le \tau_\gamma$ 2.11)

Where τ is stress and $\dot{\gamma}$ is strain rate, m and n are model constants, τ_{γ} is a constant that is interpreted as yield stress.

The model shows both yield stress and shear - thinning non - Newtonian viscosity, and is used to describe the rheological behavior of food products and biological liquids (James F., 1996).

2.5 Olive Oil Acidity

The acidity of olive oil is effected by different parameters such as degree of ripeness, industrial processes employed for oil extraction, altitude, the cultivator, climate and other factors.

Olive oil is classified qualitatively according to its acidity into many classes as given in Table (2.3) (IOOC, 2000).

Category	FFA%
Extra virgin olive oil	≤ 0.8
Virgin olive oil	≤ 2.0
Ordinary virgin olive oil	≤ 3.3
Lampante oil	> 3.3

 Table (2.3): Classification of olive oil according to FFA%

2.6 Refractive Index

Refractive index (n) of a medium is defined as the ratio of the speed of light in a vacuum to the speed of light traveling through this medium, and mathematically it is written as:

$$n = \frac{c}{v} \tag{2.12}$$

Where c is the speed of light in vacuum and v is the speed of light in the substance. The refractive index for olive oil extends from 1.4677 to 1.4707 at 20 °C (IOOC, 2000).

2.7 Mass Density

Mass density is defined as the ratio of mass of the material in grams and the volume in cm³. Robert has measured the density of olive oil to be 0.918 gm/cm³ at 15 °C and at atmospheric pressure (Robert *et al.*, 1979).

Chapter Three Methodology

Olive oil samples were collected from different region in Palestine, they were all produced by Palestinian industrial olive oil mills, from the crop of 1994 until the crop of 2012 at least four samples were collected from each region representing different olive oil ages.

The samples were collected from different regions, these are:

 L_1 , L_2 , L_3 , L_4 , L_5 , L_6 , L_7 , L_8 , L_9 , L_{10} , L_{11} , L_{12} , and L_{13} . The samples were kept in closed glass bottles in dark place at 25 °C.

The viscosity of each olive oil sample was measured in wide range of temperatures extend from 8 °C to 73 °C. Each time acidity, refractive index, and mass density were measured.

The viscosity was measured using the ND - 1 rotational viscometer. The refractive index was measured using the refractometer. Chemical titration was used to measure the acidity, while temperature of the samples was measured by using Digital Prima Long Thermometer.

Measured data were analyzed and relationships between different parameters were studied. In addition the relationship between density and refractive index, refractive index and age, acidity and age, viscosity and age, viscosity and temperature, and density and age were plotted. The curves representing the relationship between viscosity and temperature were fitted using previously used equations. A comparison with experimental data was done. Moreover, new equations were suggested to fit the experimental data of viscosity versus temperature. Finally, the relationship between viscosity and shear rate was studied to determine whether the olive oil samples under study are Newtonian or Non Newtonian.

3.1 Measurement Equipment

3.1.1 Viscosity Apparatus

The viscosity was measured by using NDJ - 1 Rotational Viscometer. It has four spindles (RV SPINDLE SET) and accuracy of 5%. The rotational speeds of the spindles are: 6, 12, 30, 60 RPM, the ranges of viscosities measured by the spindles are 0.1 to 100000 cP.



Fig.(3.1): NDJ - 1 Rotational Viscometer

The viscosity was measured using a appropriate spindle for each rotational speed: 6, 12, 30, 60 RPM at different temperatures. The temperature ranges from 8 $^{\circ}$ C to 73 $^{\circ}$ C.

3.1.2 Temperature Apparatus

Digital Prima Long Thermometer was used to measure the temperature of olive oil samples. The accuracy of this apparatus is $\pm 1\%$. It measures temperature ranges from -20 °C to +100 °C.

The temperature of the olive oil samples was incremented using the Fried Electric Model WB - 23.

3.1.3 Density Apparatus

The density of olive oil samples will be measured using a 2ml Pycnometer. The Pycnometer was first weighted empty and then weighted full of olive oil then the difference was divided by 2 ml to get the density.



Fig.(3.2): Pycnometer

The error in measuring the density is calculated using the following equation:

$$\Delta \rho = \pm \rho \left[\frac{\Delta m}{m} + \frac{\Delta V}{V} \right] \tag{3.1}$$

The analytical balance HR - 200 with accuracy ± 0.00005 was used to measure the mass.

3.1.4 Refractive Index Apparatus

The index of refraction of the olive oil samples was measured using the way - 2s ABBE digital refractometer.



Fig.(3.3): The way - 2s ABBE digital refractometer

The measurement range of the device extends from 1.3000 - 1.7000 with accuracy equals to ± 0.0002 .

3.1.5 Acidity Measurement

The acid value of olive oil was determined by the titrimetric method used in (AOAC 1997). The acid value of olive oil is equal to the mass of KOH in mg required to neutralize 1 g of olive oil dissolved in ethanol - ether mixture, and titrated with standard KOH solution.

Three main steps were followed to measure the acid value of olive oil:

<u>Firstly</u>: a 0.1 M of ethanolic KOH solution is prepared and standardized as follows:

- A 0.56 g of solid KOH is transferred into a 100 mL volumetric flask and dissolved in absolute ethanol (to get roughly 0.01 moles with 0.1 M of KOH ethanol solution).
- A 0.204 g of dry primary standard KHP (Molar mass = 204.23 g/mol) was weighted into a 250 mL conical flask and was dissolved in 50 mL of distilled water (to get accurately 0.01 moles with 0.2 M of KHP solution).
- 3 drops of phenolphthalein are added (in order not to increase volume and correspondingly changing values of molarity) and

titrated dropwise in the vicinity of the end point with KOH until a pink color is obtained and persisted for 30 seconds.

- The last two steps were repeated three times.
- The average molar concentration of KOH solution was calculated.

$$KHP + KOH \longrightarrow K_2P + H_2O$$

Where the exact molarity of KOH is
$$\frac{weight of KHP(g) \times 100\%}{204.23 \times mL of KOH}$$

Secondly: Ethanol - ether mixture was prepared:

50 mL of absolute ethanol and 50 mL of ether were mixed in a conical flask, and 3 drops of phenolphthalein solution then ehanolic KOH were added to faint pink color.

<u>Thirdly:</u> The acid value of olive oil was determined:

- 5 10 mg oil was weighted into 250 mL conical flask, then 50 mL of ethanol ether mixture and 3 drops phenolphthalein solution are added.
- The resulted solution is titrated the standard ethanolic KOH solution until permanent faint pink appears and persists for 30 seconds.

Then the acid value was calculated according to the following equation:

Acid value =
$$\frac{\text{mL KOH standardsolution} \times \text{molarity of KOH standard solution} \times 56.1}{wt \text{ of sample (g)}}$$

The acid value may be expressed in terms of % free fatty acids as flows:

$$FFA\% = \frac{Acid \ value}{1.99}$$
21

Chapter Four

Results and Data Analysis

4.1Density Results

The densities in (gm/cm³) of the collected olive oil samples were measured. The overall density results are given in Table

4.1. The density results are tabulated according to the region of the olive oil sample and the sample storage age in years.

Storage Age (years)	L ₁	L ₂	L ₃	L_4	L_5	L ₆	L ₇	L ₈	L ₉	L ₁₀ L ₁₁	L ₁₂	L ₁₃	Average
1	0.9190	0.9189	0.9183	0.9184	0.9189			0.9191					0.9187
2		0.9189	0.9178	0.9184				0.9191	0.9185		0.9185		0.9185
3	0.9188	0.9185		0.9183	0.9188	0.9183	0.9188	0.9189	0.9182	0.9185	0.9185	0.9188	0.9185
4	0.9188			0.9182	0.9186	0.9182	0.9186					0.9185	0.9184
5	0.9186						0.9185		0.9177				0.9182
6					0.9186		0.9181	0.9177			0.9180		0.9181
7					0.9182				0.9175				0.9178
8								0.9177		0.9180			0.9178
12						0.9177						0.9179	0.9178
13	0.9180					0.9176							0.9178
14	0.9177	0.9160	0.9176			0.9176				0.9179		0.9179	0.9174
15	0.9176	0.9155	0.9175			0.9175				0.9174		0.9179	0.9172
16	0.9179					0.9175				0.9165		0.9175	0.9173
19							0.9174						0.9174
Average	0.9183	0.9176	0.9178	0.9183	0.9186	0.9178	0.9183	0.9185	0.9180	0.9172 0.9180	0.9183	0.9181	0.9180

	4 1	3.6 1	1 • 4	1 1 5		1	1	•	1100 4	•	1 0	1100	4
Table (4	• Measured	density	$(\sigma m/cm^{\circ})$) At (nive ni	l cami	Meg in	different	regions (and for	different	storage ages
	T • I)	• Micasul cu	uchsity	(gm/cm	<i>)</i> DI (i sam	nes m	unititu	licgions	anu ivi	unititut	sionage ages

The overall average density of olive oil samples is 0.9180 gm/cm³.

The highest values of density recorded were the 1 year storage age and 2 years storage age of L_8 samples (0.9190 gm/cm³). While the lowest value was found for 15 years storage age of L_2 samples (0.9155 gm/cm³). The maximum value of the average density was found for L_5 samples (0.9186 gm/cm³) while the minimum average value of density was found for L_{10} samples (0.9172 gm/cm³).

The relationship between density of olive oil samples of L_1 and L_2 and storage age is given in Figs. (4.1, 4.2), respectively.



Fig.(4.1): Measured density versus storage age of olive oil samples of L₁



Fig.(4.2): Measured density versus storage age of olive oil samples of L₈

The density shows a linear proportional relationship with sample storage age. The relationship between density and storage age for the rest of regions are shown in Appendix A.

4.2 Refractive Index Results

The measured refractive indexes of olive oil for all samples, from all regions are given in Table 4.2.

	,					- ····			- 0				0	
Storage Age (years)	L ₁	L ₂	L ₃	L_4	L_5	L_6	L ₇	L_8	L ₉	L ₁₀	L ₁₁	L ₁₂	L ₁₃	Average
1	1.4717	1.4712	1.4710	1.4717	1.4716			1.4718						1.4715
2			1.4710	1.4716				1.4713	1.4713			1.4713		1.4713
3	1.4714	1.4711		1.4711	1.4714	1.4711	1.4715	1.4711	1.4713		1.4713	1.4712	1.4711	1.4712
4	1.4714			1.4707	1.4710		1.4714						1.4705	1.4710
5	1.4710						1.4712		1.4712					1.4711
6					1.4708		1.4712	1.4706				1.4709		1.4709
7					1.4707				1.4709					1.4708
8					1.4707			1.4705			1.4711			1.4708
12						1.4705							1.4703	1.4704
13	1.4708					1.4702								1.4705
14	1.4706	1.4708	1.4709			1.4701				1.4708			1.4701	1.4706
15	1.4705	1.4706	1.4708			1.4704				1.4698	1.4710		1.4702	1.4705
16	1.4702					1.4700				1.4690			1.4700	1.4698
19							1.4711							1.4711
Average	1.4710	1.4709	1.4709	1.4713	1.4710	1.4704	1.4713	1.4711	1.4712	1.4699	1.4711	1.4711	1.4704	1.4708

 Table (4.2): Measured refractive index of olive oil samples of different regions and different storage ages

The average value of refractive index of all olive oil samples is 1.4708. The range of refractive index of all samples extends from 1.4690 (16 years storage age L_{10} sample) to 1.4718 (1 year storage age L_8 sample). The relationship between refractive index and storage age for samples collected from L_1 and L_8 are shown in Figs 4.3 and 4.4.



Fig.(4.3): Measured refractive index versus storage age of olive oil samples of L₁



Fig.(4.4): Measured refractive index versus storage age of olive oil samples of L_8

One can notice from Figs 4.3 and 4.4 that the refractive index decreases as the storage age of the olive oil sample increases. The relationship between refractive index and storage age for the rest of regions are showed in Appendix B.

4.3Acidity Results

The results of olive oil samples acidity for samples collected from all regions and different storage ages are given in Table 4.3.

Storage	_	_	_	_	_		_	_	_	_	_	_	_
Age	L_1	L_2	L_3	L_4	L_5	L_6	L_7	L_8	L ₉	L_{10}	L ₁₁	L ₁₂	L ₁₃
(years)													
1	1.18	1.72	0.94	0.51	0.45			0.44					
2			0.70	0.53				0.50	0.67			0.77	
3	1.55	2.53		0.56	1.49	1.03	0.80	1.96	0.96		0.96	1.88	0.56
4	2.64			1.96	2.07	1.80	1.12						1.03
5	3.97						1.56		1.18				
6					2.40		2.50	4.77				2.00	
7					2.88				1.20				
8					2.94			5.25			3.92		
12						3.18							2.31
13	8.79									4.40			
14	9.09	5.09	5.98							5.04			
15	11.13	5.77	5.72			3.81					7.07		3.95
16	13.01					4.49				9.49			4.60
19						5.02	4.00						5.22

 Table 4.3: Measured acidity in FFA% of olive oil samples in different regions for different storage ages

The range of acidity extends from 0.44% for the 1 year storage age (L_8) to 13.01% for the 16 years storage (L_1).

Table 4.4 shows the acidity of olive oil samples from different regions as a function of storage age. The Classification of olive oil is given according to Table 2.3.

Region	Storage Age(years)	Acidity	Olive Oil Classification
L ₃	2	0.70	Extra virgin olive oil
L ₄	1	0.51	Extra virgin olive oil
L ₄	2	0.53	Extra virgin olive oil
L ₄	3	0.56	Extra virgin olive oil
L ₅	1	0.45	Extra virgin olive oil
L ₈	1	0.44	Extra virgin olive oil
L ₈	2	0.50	Extra virgin olive oil
L ₉	2	0.67	Extra virgin olive oil
L ₁₂	2	0.77	Extra virgin olive oil
L ₁₃	3	0.56	Extra virgin olive oil
L ₁	1	1.18	Virgin olive oil
L ₁	3	1.55	Virgin olive oil
L ₂	1	1.72	Virgin olive oil
L ₃	1	0.94	Virgin olive oil
L ₄	4	1.96	Virgin olive oil
L ₅	3	1.49	Virgin olive oil
L ₆	3	1.03	Virgin olive oil
L ₆	4	1.80	Virgin olive oil
L ₇	3	0.80	Virgin olive oil
L ₇	4	1.12	Virgin olive oil
L ₇	5	1.56	Virgin olive oil
L ₈	3	1.96	Virgin olive oil
L ₉	3	0.96	Virgin olive oil
L ₉	5	1.18	Virgin olive oil
L ₉	7	1.20	Virgin olive oil
L ₁₁	3	0.96	Virgin olive oil
L ₁₂	3	1.88	Virgin olive oil
L ₁₂	6	2.00	Virgin olive oil
L ₁₃	4	1.03	Virgin olive oil

Table 4.4: The acidity of olive oil samples from different regions ofdifferent storage ages

		2)	
L ₁	4	2.64	Ordinary virgin olive oil
L ₂	3	2.53	Ordinary virgin olive oil
L ₅	4	2.07	Ordinary virgin olive oil
L ₅	6	2.40	Ordinary virgin olive oil
L ₅	7	2.88	Ordinary virgin olive oil
L ₅	8	2.94	Ordinary virgin olive oil
L ₆	12	3.18	Ordinary virgin olive oil
L ₇	6	2.50	Ordinary virgin olive oil
L ₁₃	12	2.31	Ordinary virgin olive oil
L ₁	5	3.97	Lampante oil
L ₁	13	8.79	Lampante oil
L ₁	14	9.09	Lampante oil
L ₁	15	11.13	Lampante oil
L ₁	16	13.01	Lampante oil
L ₂	14	5.09	Lampante oil
L_2	15	5.77	Lampante oil
L_3	14	5.98	Lampante oil
L_3	15	5.72	Lampante oil
L_6	14	3.81	Lampante oil
L_6	15	4.49	Lampante oil
L ₆	16	5.02	Lampante oil
L_7	19	4.00	Lampante oil
L ₈	6	4.77	Lampante oil
L ₈	8	5.25	Lampante oil
L_{10}	13	4.40	Lampante oil
L ₁₀	14	5.04	Lampante oil
L_{10}	16	9.49	Lampante oil
L ₁₁	8	3.92	Lampante oil
L ₁₁	15	7.07	Lampante oil
L ₁₃	14	3.95	Lampante oil
L ₁₃	15	4.60	Lampante oil
L ₁₃	16	5.22	Lampante oil

Three samples of crop 2013 (L_4 , L_5 , and L_8) are extra virgin, while the other three samples of the same crop are virgin (L_1 , L_2 , and L_3). The five samples of crop 2012 are all extra virgin. In general, one can notice from Table (4.4) that the olive oil samples are good for human consumption up to 10 years storage age.

29

The relationship between acidity and storage age for the collected samples from L_1 and L_8 are shown in Figs 4.5 and 4.6, respectively.



Fig.(4.5): Measured acidity versus storage age of olive oil samples of L₁



Fig.(4.6): Measured acidity versus storage age of olive oil samples of L₈The olive oil acidity increases as the storage age of the sample increases.The relationship between olive oil acidity and storage age is shown to be

linear. The acidity versus storage age for the rest of the samples of different regions are shown in the Appendix C.

4.4 Refractive Index as a Function of Density of Olive Oil

The refractive index of olive oil samples were plotted against density mass and the results of the L_1 and L_8 are shown in Figs.4.7 and 4.8.The refractive index versus density for the rest of the samples are shown in the Appendix D.



Fig.(4.7): Measured refractive index versus density mass of olive oil samples of L₁



Fig.(4.8): Measured refractive index versus density mass of olive oil samples of L_8

The relationship between refractive index and density mass is linear.

The reason for the increasing of acidity with storage age is the definition of acidity. A concentration of free fatty acids, oils and fats are made up of triglycerides and chains of free fatty acids, when chains of free fatty acids are liberated, they can increase the acidity free fatty acids which leads to an increase in acidity. The reason for the breakup of the of fatty chains is an lipase enzyme that gives the association between fatty chains and triglycerides then transformed into free fatty chains. The presence of water with the presence of holes in the olive grain or high oil temperature will lead to increase in the activity of the lipase enzyme and thus higher acidity. As the temperature of the oil increases the spaces between particles

increases then their motion became easier which leads to a decrease in viscosity because the viscosity is the resistance of flow of the liquid. The decrease of density storage with storage age is due to the increase of free fatty acids (i.e., the breakup of the liquid molecules) and thus less density between molecules, leading to a decrease in mass density.

4.5 Viscosity Results

The Kinematic viscosity of olive oil samples was calculated at room temperature (25 $^{\circ}$ C). Table 4.5 gives the results of L₁ and L₈ samples at different storage age.

Table 4.5: Results of kinematic viscosity of L_1 and L_8 samples at different storage age

		(v) in (cSt)												
$t (^{0}C)$		L_1		L ₈										
	Storage	Storage	Storage	Storage	Storage	Storage								
(())	age:	age:	age:	age:	age:	age:								
	2 years	5 years	13 years	3 years	6 years	10 years								
25	74.00	74.02	69.71	81.61	80.63	77.36								

The dynamic viscosity of olive oil samples was measured at different temperatures. Table 4.6 shows the results of L_1 and L_8 samples at different storage age.

	0	0											
		(η_{exp}) in cP											
		L_1			L_8								
$t (^{0}C)$	Storage	Storage	Storage	Storage	Storage	Storage							
(\mathbf{C})	age:	age:	age:	age:	age:	age:							
	2 years	5 years	13 years	3 years	6 years	10 years							
23	75	72	70	80	79	78							
25	68	68	64	75	74	71							
28	62	67	63	72	71	69							
30	58	61	59	68	66	67							
33	58	55	53	63	63	62							
37	56	49	49	59	61	59							
40	52	47	45	55	57	57							
42	52	43	44	52	57	54							
44	45	39	42	45	50	52							
47	38	33	42	40	47	50							
50	32	31	36	33	46	46							
52	31	25	32	28	43	42							
55	29	25	28	26	41	36							
58	22	21	24	23	38	32							
60	17	18	22	20	37	30							
63	17	17	22	19	33	25							

Table 4.6: Results of dynamic viscosity of L_1 and L_8 samples at different storage age

The dynamic viscosity decreases with storage age sample as a function of temperature. It also decreases with temperature at a given sample age.

The relationship between dynamic viscosity and storage age of the samples from L_1 and L_8 are shown in Fig.4.9.



Fig.(4.9): Measured viscosity versus storage age of olive oil samples of L_1 and L_8 at 25 °C

Figure 4.9 shows that the viscosity decreases as the storage age of the olive oil sample increases.

4.5.1 Viscosity Results and Theoretical Predictions

The experimental results of dynamic viscosity versus temperature were compared with equations obtained by Abramovic and Andrade. The percentage of absolute deviations and standard deviation between the measured and theoretical data were calculated.

Three equations were used to fit the experimental data of this work, these are: Abramovic's equation (formula 1)

$$\log \eta = \frac{A}{T} - B \tag{4.1}$$

and Abramovic's equation (formula 2)

$$\eta = A - B \log t \tag{4.2}$$

and Andrade's equation

$$36$$
$$\ln \eta = A + \frac{B}{T} + CT \tag{4.3}$$

a. Two Constants Abramovic's Equation (Formula 1)

The dynamic viscosity versus temperature using the two constants Abramovic's equation (formula 1) has been calculated and gives: $\log \eta = \frac{A}{T} - B$ with A = 1558.2 (K) and B = 3.433 (Abramovic *et al.*, 1998). A comparison between the measured values of dynamic viscosity and the calculated values using Abramovic's equation (formula 1) of four different olive oil samples. The values of AAD% and SD are also given in Table 4.7.

 Table 4.7: The measured and calculated dynamic viscosity of olive oil versus temperature using Abramovic's equation (formula 1)

	ncal (cP)			η_{exp}	(cP)		
	lical (CF)		L_1			L_8	
$t(^{0}C)$	Abramovic's			Storage	Storage		Storage
(())	equation	Storage	Storage	age:	age:	Storage	age:
	(formula 1)	age:	age:	13	3 years	age:	10
	(1011111111111)	2 years	5 years	years		6 years	years
23	67	75	72	70	80	79	78
25	62	68	68	64	75	74	71
28	55	62	67	63	72	71	69
30	51	58	61	59	68	66	67
33	45	58	55	53	63	63	62
37	39	56	49	49	59	61	59
40	35	52	47	45	55	57	57
42	32	52	43	44	52	57	54
44	30	45	39	42	45	50	52
47	27	38	33	42	40	47	50
50	24	32	31	36	33	46	46
52	23	31	25	32	28	43	42
55	21	29	25	28	26	41	36
58	19	22	21	24	23	38	32
60	18	17	18	22	20	37	30
63	16	17	17	22	19	33	25
AAD%		13.2%	12.0%	15.3%	15.8%	24.1%	22.1%
SD		1.72	1.24	1.42	2.17	3.03	2.76

 $[\log \eta = \frac{A}{T} - B]$ for L₁ and L₈

The values of AAD% are high for all samples. Their ranges are from 12.0% to 24.1%, which means that the constants of Abramovic's equation (formula1) used by Abramovic do not represent a good fit for our results. Fig 4.10 shows the relationship between the dynamic viscosity of olive oil versus temperature for L_1 and L_8 samples.



Fig.(4.10): Measured and calculated values of dynamic viscosity using Abramovic's equation (formula 1) for four different samples

b. Two Constants Abramovic's Equation (Formula 2)

The viscosities of olive oil for 2, 5 and 13 years storage age of L₁ sample and 3, 6, and 10 years storage age of L₈ sample were calculated using Abramovic's two constants formula 2 which is: $\eta = A - Blog t$ with A = 235.4 cP and B = 124.1cP (Abramovic *et al.*, 1998). The results of calculated and measured values of viscosity are given in Table 4.8.

Table 4.8: The measured and calculated dynamic viscosity of olive oilversus temperature using Abramovic's equation (formula 2)

	_			η_{exp}	(cP)		
	usi		L_1			L_8	
t (°C)	η _{cal (cP)} ng Abramovic's equation (formula 2)	Storage age: 2 years	Storage age: 5 years	Storage age: 13 years	Storage age: 3 years	Storage age: 6 years	Storage age: 10 years
23	66	75	72	70	80	79	78
25	62	68	68	64	75	74	71
28	56	62	67	63	72	71	69
30	52	58	61	59	68	66	67
33	47	58	55	53	63	63	62
37	41	56	49	49	59	61	59
40	37	52	47	45	55	57	57
42	34	52	43	44	52	57	54
44	31	45	39	42	45	50	52
47	28	38	33	42	40	47	50
50	25	32	31	36	33	46	46
52	22	31	25	32	28	43	42
55	19	29	25	28	26	41	36
58	17	22	21	24	23	38	32
60	15	17	18	22	20	37	30
63	25	17	17	22	19	33	25
AAD%		14.3%	18.2%	21.3%	17.1%	24.6%	22.8%
SD		1.63	1.31	1.51	2.09	3.04	2.73

 $[\eta = A - Blog t]$ for L₁ and L₈

AAD% values are very high and extend from 14.3% for 2 years storage age of L_1 sample to 24.6% for 6 years storage age sample. This implies a failure fit constants for abramovic's equation (formula 2).

Fig. 4.11 shows the relationship between dynamic viscosity versus temperature for two samples from L_1 and L_8 .



Fig.(4.11): Measured and calculated values of dynamic viscosity using Abramovic's equation (formula 2) for four different samples from different regions

c. Andrade's Equation

Andrade's equation $\ln \eta = A + \frac{B}{T} + CT$ is used to calculate the dynamic viscosity of olive oil versus temperature. The values of A, B, and C are found to be equal -32.72, 7462.27 K, and 0.04 (K⁻¹), respectively (Andrade, 1930; Abramovic *et al.*, 1998).

The calculated and experimental values of viscosity versus temperature are given in Table 4.9 for two samples from L_1 and L_8 .

Table 4.9: The measured and calculated dynamic viscosity using Andrade's equation $[\ln \eta = A + \frac{B}{T} + CT]$ of olive oil samples collected from L₁and L₈ versus temperature

		$\eta_{\exp}(cP)$										
	usii		L ₁			L_8						
t (°C)	η _{cal (α} Ρ) ng Andrade's equation	Storage age: 2 years	Storage age: 5 years	Storage age: 13 years	Storage age: 3 years	Storage age: 6 years	Storage age: 10 years					
23	75	75	72	75	80	79	78					
25	69	68	68	69	75	74	71					
28	61	62	67	61	72	71	69					
30	56	58	61	56	68	66	67					
33	49	58	55	49	63	63	62					
37	42	56	49	42	59	61	59					
40	38	52	47	38	55	57	57					
42	35	52	43	35	52	57	54					
44	33	45	39	33	45	50	52					
47	30	38	33	30	40	47	50					
50	27	32	31	27	33	46	46					
52	25	31	25	25	28	43	42					
55	23	29	25	23	26	41	36					
58	21	22	21	21	23	38	32					
60	20	17	18	20	20	37	30					
63	19	17	17	19	19	33	25					
AAD%		9.6%	6.4%	9.5%	10.4%	19.7%	17.4%					
SD		1.30	0.77	1.02	1.63	2.51	2.25					

High AAD% values were calculated, 9.6%, 6.4% and 9.5% for the 2 years, 5 years, and 13 years of L_1 , and 10.4%, 19.7%, and 17.4% for the 3 years, 6 years, and 10 years of L_8 . These indicate that Andrade's equation with the given constants is not suitable to describe the experimental results.

Fig. 4.12 shows the measured and calculated dynamic viscosity using Andrade's equation of olive oil samples collected from L_1 and L_8 versus temperature.



Fig.(4.12): Measured and calculated values of dynamic viscosity using Andrade's equation for four different samples

4.5.2 Modifications of the Abramovic's and Andrade's equations

In section 4.5.1, Abramovic's first and second formulas and Andrade's equation with their constants failed to describe the relationship between dynamic viscosity and temperature of olive oil samples. Modifications were proposed to these equations by fitting the experimental data with different constants to get good prediction with the least error. In order to achieve that, experimental data was fitted using these equations and new constants are introduced and the values of AAD% and SD were calculated.

a. Modification to Abramovic's Equation (Formula 1)

Measured values of dynamic viscosity versus temperature of two samples from L₁ and L₈ were fitted using Abramovic's equation (formula 1) $\log \eta = \frac{A}{T} - B$. The values of A, B for each samples of different storage age were introduced, and values of AAD% and SD were calculated. The results of this fit are given in Table 4.10.

			L	41			L_8						
$t(^{0}\mathbf{C})$	Storag	e age:	Stora	ge age:									
(())	2 ye	ears	5 ye	ears	13 y	ears	3 ye	ears	6 уе	ears	10	years	
	$\eta_{exp}(cP)$	$\eta_{cal}(cP)$											
23	75	78	72	77	70	73	80	86	79	81	78	81	
25	68	72	68	71	64	68	75	80	74	77	71	77	
28	62	65	67	64	63	62	72	72	71	72	69	71	
30	58	61	61	59	59	58	68	67	66	68	67	67	
33	58	55	55	53	53	53	63	60	63	63	62	62	
37	56	49	49	46	49	47	59	52	61	58	59	56	
40	52	44	47	42	45	43	55	47	57	54	57	52	
42	52	42	43	39	44	41	52	44	57	51	54	49	
44	45	39	39	37	42	38	45	41	50	49	52	47	
47	38	36	33	33	42	35	40	38	47	46	50	44	
50	32	33	31	30	36	33	33	34	46	43	46	41	
52	31	31	25	28	32	31	28	32	43	41	42	39	
55	29	28	25	26	28	29	26	29	41	39	36	36	
58	22	26	21	24	24	26	23	27	38	37	32	34	
60	17	25	18	22	22	25	20	25	37	35	30	32	
63	17	23	17	20	22	23	19	23	33	33	25	30	
A(K)	1334	4.35	143	9.45	1234	4.13	1420	0.86	970	.83	106	53.17	
В	2.615 2.975		75	2.3	06	2.864		1.369		1.681			
AAD%	7.7	7%	5.7	7%	7.3%		6.8%		2.4%		4.6%		
SD	0.9 0.7		.7	0.	5	1.2		1.	.8	1.6			

Table 4.10: The measured and calculated dynamic viscosity using modified Abramovic's equation (formula 1) $[\log \eta = \frac{A}{T} - B]$ of olive oil samples collected from L₁ and L₈ versus temperature

Figs 4.13 and 4.14 show the relationship between viscosity and temperature for two samples of L_1 and L_8 respectively. In addition, the fitted line using the modified Abramovic's first formula is given.



Fig.(4.13): Measured and calculated values of dynamic viscosity of olive oil of L_1 using modified Abramovic's equation (formula 1)



Fig.(4.14): Measured and calculated values of dynamic viscosity of olive oil of L_8 using modified Abramovic's equation (formula 1)

The values of AAD% given in Table 4.9 are still high. They are 7.7%, 5.7%, and 7.3%, for the 2 years, 5 years, and 13 years of L_1 , and 6.8%, 2.4%, and 4.6% for the 3 years, 6 years, and 10 years of L_8 . Therefore one can conclude that modified Abramovic's equation (formula 1) was not suitable for describing the relationship between dynamic viscosity and temperature.

b. Modification to Abramovic's Equation (Formula 2)

Modification was made on Abramovic's equation (formula 2)

 $[\eta = A - Blog t]$ of olive oil behavior of samples from L₁ and L₈, the results of measured and calculated values of dynamic viscosity are given in Table 4.11.

			Ι	-1			L ₈						
$t(^{0}C)$	Storage age:		Storage age:		Storag	ge age:	Storage age:		Storage age:		Storage age:		
(\mathbf{C})	2 years		5 years		13 years		3 years		6 years		10 years		
	$\eta_{exp}(cP)$	$\eta_{cal}(cP)$											
23	75	77	72	76	70	71	80	85	79	79	78	80	
25	68	72	68	71	64	67	75	79	74	76	71	76	
28	62	66	67	64	63	62	72	72	71	71	69	70	
30	58	62	61	60	59	58	68	68	66	68	67	67	
33	58	57	55	55	53	54	63	61	63	64	62	62	
37	56	50	49	48	49	48	59	54	61	58	59	57	
40	52	46	47	43	45	45	55	49	57	55	57	53	
42	52	43	43	41	44	42	52	46	57	53	54	51	
44	45	41	39	38	42	40	45	43	50	51	52	48	
47	38	37	33	34	42	37	40	39	47	48	50	45	
50	32	33	31	30	36	34	33	35	46	45	46	42	
52	31	31	25	28	32	32	28	32	43	44	42	40	
55	29	28	25	25	28	29	26	28	41	41	36	38	
58	22	25	21	22	24	27	23	25	38	39	32	35	
60	17	23	18	20	22	25	20	23	37	37	30	34	
63	17	21	17	17	22	23	19	20	33	35	25	31	
A(cP)) 251.06		257.41		221	51	286.70		217.24		230.80		
B(cP)	128.	058	133	133.583		.397	148.	372	101.243		110.926		
AAD%	6.3	8%	3.	7%	3.0)%	4.4	4.4%		3.5%		4.3%	
SD	0.8		0	.6	0	.3	1.2		1.7		1.4		

Table 4.11: The measured and calculated dynamic viscosity using modified Abramovic's equation (formula 2) $[\eta = A - Blog t]$ of olive oil samples from L₁ and L₈ versus temperature

The measured and fitted viscosity - temperature relationship using the modified Abramovic's equation (formula 2) are shown in Figs. 4.15 and 4.16.



Fig.(4.15): Measured and calculated values of dynamic viscosity of olive oil of L_1 using modified Abramovic's equation (formula 2)



Fig.(4.16): Measured and calculated values of dynamic viscosity of olive oil of L₈ using modified Abramovic's equation (formula 2)

The calculated values of AAD% extend from 3.0% to 6.3%. Which are relatively high values. This implies that modified Abramovic's equation (formula 2) is not proper to fit the relationship between viscosity and temperature.

c. Modification to Andrade's Equation

Measured and calculated values of dynamic viscosity versus temperature of two samples from L₁ and L₈ were fitted using Andrade's equation $[\ln \eta = A + \frac{B}{T} + CT]$. The values of A, B, and C for each sample were introduced, and values of AAD% and SD were calculated. The results of this fit are given in Table 4.12.

		_	L ₁	L ₈									
$t(^{O}C)$	Storage age:		Storage	e age:	Storag	Storage age:		Storage age:		Storage age:		Storage age:	
(())	2 years		5 ye	ars	13 y	ears	3 ye	ears	6 ус	6 years		10 years	
	$\eta_{exp}(cP)$	$\eta_{cal}(cP)$											
23	75	70	72	72	70	68	80	78	79	78	78	74	
25	68	68	68	69	64	66	75	76	74	75	71	73	
28	62	66	67	65	63	62	72	73	71	71	69	70	
30	58	64	61	62	59	59	68	70	66	68	67	68	
33	58	60	55	57	53	55	63	65	63	64	62	65	
37	56	54	49	50	49	50	59	58	61	59	59	60	
40	52	50	47	45	45	46	55	53	57	56	57	56	
42	52	47	43	42	44	44	52	49	57	54	54	54	
44	45	43	39	39	42	41	45	45	50	52	52	51	
47	38	39	33	34	42	38	40	40	47	49	50	47	
50	32	34	31	30	36	34	33	34	46	46	46	43	
52	31	31	25	27	32	32	28	31	43	44	42	41	
55	29	27	25	24	28	29	26	26	41	41	36	37	
58	22	23	21	20	24	26	23	22	38	38	32	34	
60	17	21	18	18	22	24	20	20	37	37	30	32	
63	17	17	17	16	22	21	19	16	33	34	25	28	
А	143.30		109.15		69.71		155	155.66		21.16		76.62	
B(K)	-20277.051		-14732	1.423	-8948	3.565	-2199	6.926	-1694.496		-10264.864		
C(K ⁻¹)	-0.238		-0.186		-0.119		-0.260		-0.037		-0.127		
AAD%	3.0	5%	1.8	%	2.1%		1.8	1.8%		1.2%		1.9%	
SD	0.7		0.	6	0.3		1.1		1.6		1.4		

Table 4.12: The measured and calculated dynamic viscosity using modified Andrade's equation $[\ln \eta = A + \frac{B}{T} + CT]$ of olive oil samples from L₁ and L₈ versus temperature

Figs. 4.17 and 4.18 show the relationship between viscosity and temperature using the modified Andrade's equation for the 5 years storage age of L_1 and 6 years storage age of L_8 sample.



Fig.(4.17): Measured and calculated values of dynamic viscosity of olive oil of L₁ using modified Andrade's equation



Fig.(4.18): Measured and calculated values of dynamic viscosity of olive oil of L_8 using modified Andrade's equation

The values of AAD% when new constants are substituted into Andrade's equation are smaller than that of Abramovic's equations as given in Table 4.12. Taking 13 years storage age of L₁ as an example, one can notice from Tables (4.7 - 4.12) that AAD% is 15.3% for Abramovic's equation (formula1) $[\log \eta = \frac{A}{T} - B]$ where it is 7.3% for modified Abramovic's equation (formula 1), 21.3% for Abramovic's equation (formula 2) $[\eta = A - Blog t]$ and 3.0% for modified Abramovic's equation (formula 2), and 9.5% for Andrade's equation $[\ln \eta = A + \frac{B}{T} + CT]$ and 2.1% for modified Andrade's equation.

The Abramovic's and Andrade's equations with modified constants were better in describing the viscosity – temperature relationship since low values of AAD% (relative to the same equations with the original constants) were obtained. However, the modification of these equations by imposing new constants did not produce an acceptable description to the experimental data because still they have high values of AAD%.

The using of Abramovic's and Andrade's equations failed to describe the viscosity temperature relationship. New proposed equations should be imposed to describe the experimental data.

In the following section, three new equations are proposed to describe the dynamic viscosity relationship of olive oil versus temperature.

4.5.3 Proposed Equations

a. Proposed equation (formula 1)

Three order polynomial of equation was proposed as:

$$\eta = A + Bt + Ct^2 + Dt^3 \tag{4.4}$$

to fit the experimental dynamic viscosity - temperature relation of three samples from L_1 and three samples from L_8 . Table 4.13 shows the results of fitting.

	L_							L ₈						
$t (^{0}C)$	Storage	e age:	Storag	e age:	Stora	ge age:	Storag	ge age:	Storage age:		Storage age:			
ι(C)	2 years		5 years		13 years		3 years		6 years		10 years			
	$\eta_{exp}(cP)$	$\eta_{cal}(cP)$												
23	75	70	72	72	70	68	80	78	79	79	78	75		
25	68	68	68	69	64	66	75	76	74	75	71	73		
28	62	66	67	65	63	62	72	72	71	70	69	70		
30	58	64	61	61	59	59	68	70	66	67	67	68		
33	58	60	55	57	53	55	63	65	63	63	62	65		
37	56	54	49	50	49	50	59	58	61	59	59	60		
40	52	50	47	45	45	46	55	53	57	56	57	56		
42	52	47	43	42	44	44	52	49	57	54	54	54		
44	45	43	39	39	42	41	45	45	50	52	52	51		
47	38	39	33	34	42	38	40	40	47	49	50	47		
50	32	34	31	30	36	34	33	35	46	47	46	44		
52	31	31	25	27	32	32	28	31	43	45	42	41		
55	29	27	25	24	28	29	26	27	41	42	36	37		
58	22	23	21	21	24	26	23	23	38	39	32	33		
60	17	21	18	19	22	24	20	21	37	37	30	31		
63	17	18	17	17	22	21	19	18	33	33	25	27		
A(cP)	51.05		88.06		96.19		64.98		152.52		84.09			
B(cP/°C)	2.616		0.2	0.257		-1.077		543	-4.812		0.150			
C(cP/°C ²)	-0.094		-0.0	53	-0.	008	-0.	105	0.083		-0.029			
D(cP/°C ³)	0.00071		0.00	049	0.00010		0.00083		-0.00059		0.00019			
AAD%	0.2	.%	0.3	%	0.	2%	0.1%		0.2%		0.1%			
SD	0.7		0.	7	0.1		0.1		0.4		0.5			

Table 4.13: The measured and calculated dynamic viscosity using the proposed equation (formula 1) $[\eta = A + Bt + Ct^2 + Dt^3]$ of olive oil samples collected from L₁ and L₈ versus temperature

AAD% are relatively very small for all samples, they range from 0.1% to 0.3%. This result implies that the proposed formula 1 is suitable to describe the viscosity - temperature relationship.

Fig. 4.19 shows the experimental and calculated values of the dynamic viscosity versus temperature for two olive oil samples.



Fig.(4.19): Measured and calculated values of dynamic viscosity using proposed equation (formula 1) fit for olive oil sample from L_1 and L_8

One can observe from Fig 4.19 that the fitted lines coincide with most of the experimental data.

b. Proposed equation (formula 2)

The measured dynamic viscosity versus temperature was fitted using the proposed formula 2:

$$\eta = Aln(-Bln t) \tag{4.5}$$

Results of this fitting for L_1 and L_8 samples are given in Table 4.14.

			L_1				L ₈						
$t(^{0}\mathbf{C})$	Storage age:		Storage	age:	Storag	e age:	Storage age:		Storage age:		Storage age:		
(\mathbf{C})	2 years		5 years		13 years		3 years		6 years		10 years		
	$\eta_{exp}(cP)$	$\eta_{cal}(cP)$											
23	75	78	72	77	70	72	80	86	79	80	78	81	
25	68	72	68	71	64	68	75	80	74	76	71	76	
28	62	66	67	64	63	62	72	72	71	71	69	70	
30	58	61	61	60	59	58	68	67	66	67	67	67	
33	58	56	55	54	53	53	63	61	63	63	62	62	
37	56	49	49	47	49	48	59	53	61	58	59	56	
40	52	45	47	43	45	44	55	48	57	54	57	52	
42	52	43	43	40	44	42	52	45	57	52	54	50	
44	45	40	39	37	42	40	45	42	50	50	52	48	
47	38	37	33	34	42	37	40	38	47	48	50	45	
50	32	33	31	30	36	34	33	34	46	45	46	42	
52	31	31	25	28	32	32	28	32	43	44	42	40	
55	29	29	25	25	28	30	26	29	41	41	36	38	
58	22	26	21	22	24	27	23	26	38	39	32	36	
60	17	24	18	21	22	26	20	24	37	38	30	34	
63	17	22	17	18	22	24	19	21	33	36	25	32	
A(cP)	-200.60		-210.	07	-173	3.51	-232	-232.78		-159.29		-173.75	
В	-0.216		-0.221		-0.210		-0.220		-0.193		-0.200		
AAD%	0.6	5%	0.79	%	0.5	5%	0.5	0.5%		0.8%		0.4%	
SD	0.2		0.7	7	0.	4	0.4		1.7		0.5		

Table 4.14: The measured and calculated dynamic viscosity using proposed equation (formula 2) [$\eta = Aln(-Bln t)$] of olive oil samples collected from L₁ and L₈ versus temperature
Good prediction was obtained as given by Table 4.14, where very low values of AAD% are obtained. They are 0.6%, 0.7% and 0.5% for the 2 years, 5 years, and 13 years of L_1 , and 0.5%, 0.8% and 0.4% for the 3 years, 6 years, and 10 years of L_8 . The proposed equation (formula 2) is suitable in describing the viscosity versus temperature relationship.

Experimental and calculated data using the proposed equation (formula 2) of viscosity versus temperature for samples from L_1 and L_8 are shown in Fig.4.20.



Fig.(4.20): Measured and calculated values of dynamic viscosity using proposed equation (formula 2) fit for olive oil sample from L_1 and L_8

c. Proposed equation (formula 3)

The experimental results for dynamic viscosity versus temperature for four olive oil samples were fitted using the following proposed formula 3:

$$\eta = A + Bt + Ct^2 + Dt^3 + Et^4 + Ft^5$$
(4.6)

The results of this fitting in addition to experimental values, constants A, B, C, D, E, F, AAD%, and SD are given in Table 4.15.

			L	1			L_8							
t (°C)	Storag	e age:	Storag	ge age:	Storag	e age:								
	2 ye	ears	5 years		13 years		3 years		6 y	ears	10 years			
	$\eta_{exp}(cP)$	$\eta_{cal}(cP)$												
23	75	75	72	72	70	69	80	80	79	79	78	77		
25	68	67	68	69	64	67	75	75	74	74	71	73		
28	62	61	67	65	63	62	72	70	71	69	69	68		
30	58	60	61	61	59	58	68	68	66	67	67	66		
33	58	58	55	57	53	54	63	65	63	64	62	63		
37	56	56	49	50	49	49	59	59	61	60	59	59		
40	52	52	47	45	45	46	55	54	57	57	57	57		
42	52	49	43	42	44	44	52	50	57	54	54	55		
44	45	46	39	39	42	43	45	46	50	52	52	53		
47	38	40	33	34	42	40	40	40	47	48	50	49		
50	32	34	31	30	36	36	33	33	46	45	46	45		
52	31	30	25	27	32	33	28	30	43	43	42	42		
55	29	26	25	24	28	28	26	25	41	41	36	37		
58	22	22	21	20	24	24	23	22	38	38	32	32		
60	17	20	18	19	22	22	20	21	37	37	30	29		
63	17	16	17	17	22	22	19	19	33	33	25	25		
A(cP)	125	5.24	64.	70	-411	l.16	724	.97	665	5.72	205	.43		
B(cp/K)	-144	.179	3.4	64	70.2	235	-79.	770	-72.	.015	-9.5	27		
C(cP/K ²)	6.8	04	-0.2	226	-3.8	361	3.8	43	3.4	169	0.1	68		
D(cP/K ³)	-0.1	156	0.0	05	0.1	.00	-0.0)90	-0.0	083	0.0	02		
E(cP/K ⁴)	0.002	1734	-0.00	0059	-0.00	1250	0.002	1019	0.00	0961	-0.00	0082		
F(cP/K ⁵)	-0.0000	07485	0.0000	00297	0.0000	06041	-0.0000	04412	-0.000004357		0.0000	00617		
AAD%	0.4	1%	0.2	2%	0.1	L%	0.1	L%	0.1%		0.1%			
SD	0.	7	0.	3	0.	.2	0.	1	0	.2	0.	3		

Table 4.15: The measured and calculated dynamic viscosity using proposed equation (formula 3) $[\eta = A + Bt + Ct^2 + Dt^3 + Et^4 + Ft^5]$ of olive oil samples collected from L₁ and L₈ versus temperature

Figs. 4.21 and 4.22 show the experimental and calculated dynamic viscosity versus temperature of olive oil samples of L_1 and L_8 .



Fig.(4.21): Measured and calculated values of dynamic viscosity using proposed equation (formula 3) fit for olive oil sample from L_1



Fig.(4.22): Measured and calculated values of dynamic viscosity using proposed equation (formula 3) fit for olive oil sample from L_8

The proposed formula 3 is suitable for viscosity - temperature relationship.

The values of AAD% are very small, they are 0.4%, 0.2% and 0.1% for the 2 years, 5 years, and 13 years storage ages of L_1 , and 0.1%, 0.1%, and 0.1% for the 3 years, 6 years and 10 years storage ages of L_8 . This indicates a good theoretical prediction.

The validity of good prediction is emphasized by Fig. 4.21 and 4.22 that show that both the experimental and calculated results are rather identical.

The results of fitting the viscosity temperature relationship using different equation are given in Table 4.16.

Table 4.16: Values of AAD% obtained from fitting viscosity -temperature relationship using different equations

		≻ A					
		L_1			verage AAD%		
Equations	Storage age: 2 years	Storage age: 5 years	Storage age: 13 years	Storage age : 3 years	Storage age: 6 years	Storage age: 10 years	
Abramovic's formula 1	13.2%	12.0%	15.3%	15.8%	24.1%	22.1%	17.08%
Abramovic's formula 2	14.3%	18.2%	21.3%	17.1%	24.6%	22.8%	19.72%
Andrade's formula	9.6%	6.4%	9.5%	10.4%	19.7%	17.4%	12.17%
Modified Abramovic's formula 1	7.7%	5.7%	7.3%	6.8%	2.4%	4.6%	5.75%
Modified Abramovic's formula 2	6.3%	3.7%	3.0%	4.4%	3.5%	4.3%	4.20%
Modified Andrade's formula	3.6%	1.8%	2.1%	1.8%	1.2%	1.9%	2.07%
Our proposed formula 1	0.2%	0.3%	0.2%	0.1%	0.2%	0.1%	0.18%
Our proposed formula 2	0.6%	0.7%	0.5%	0.5%	0.8%	0.4%	0.58%
Our proposed formula 3	0.4%	0.2%	0.1%	0.1%	0.1%	0.1%	0.17%

The values of AAD% is very high when Abramovic's and Andrade's equations are used, the average values of AAD% are 17.08%, 19.72%, and 12.17% for Abramovic's 1, Abramovic's 2, and Andrade's equations respectively, these high values of AAD% indicate that these equation are not suitable to describe the viscosity - temperature relationship.

The modifications are done on these three equations, the values of AAD% are 5.57%, 4.20%, and 2.07% for modified Abramovic's 1, modified Abramovic's 2, and modified Andrade's equations respectively. These values are still high and these equations did not present a good fit for the viscosity - temperature relationship.

Our proposed equations presented the best fit with the lowest values of AAD%. These values are 0.18%, 0.58%, and 0.17% for our proposed equations.

4.6 Shear Stress Versus Shear Rate

The power law model is used to check whether the behavior of the olive oil samples is Newtonian or Non - Newtonian. A linear fit was done to the experimental results by the power law equation ($\tau = \eta \dot{\gamma}^n$) to find the value of the exponent *n*. The results of two samples from L₁ and L₈ are given in Table 4.17.

				L_1			L_8							
$+ (^{0}C)$	Storage age:		Storage age:		Stora	ge age:	Stora	Storage age:		ge age:	Storage age:			
(\mathbf{C})	2 year		5 years		13 years		3 years		6 years		10 years			
	n	Error	n	Error	n	Error	n	Error	n	Error	n	Error		
8	1.00	0.006	1.01	0.008	1.03	0.021	1.01	0.008	1.01	0.010	1.01	0.004		
18	1.00	0.006	1.01	0.008	1.00	0.006	1.00	0.006	1.00	0.008	1.00	0.006		
28	0.99	0.016	1.00	0.000	0.99	0.010	1.00	0.008	1.00	0.008	1.00	0.000		
40	1.01	0.014	1.01	0.006	0.99	0.006	1.02	0.002	1.02	0.002	1.01	0.006		
47	0.98	0.012	1.01	0.006	0.98	0.011	1.01	0.010	1.02	0.009	1.00	0.000		
58	1.02	0.014	1.05	0.004	1.00	0.003	1.01	0.006	1.04	0.002	1.00	0.004		
70	1.02	0.006	1.00	0.000	0.98	0.006	1.05	0.004	1.00	0.004	1.02	0.008		

Table 4.17: The results of power law fit to experimental data from L_1 and L_8

The relationships between shear stress and shear rate for L_1 (2 and 13 years storage age) for different temperatures are shown

in Figs. 4.23 and 4.24, respectively.



Fig.(4.23): Relationship between shear stress and shear rate for olive oil sample from L_1 (2 year storage age) at different temperatures



Fig.(4.24): Relationship between shear stress and shear rate for olive oil sample from L_1 (13 years storage age) at different temperatures

Table 4.17 shows the value of n is closed to one within an accepted error bars. This means that the olive oil samples are Newtonian fluid.

Figs. 4.23 and 4.24 show that the relationship between shear stress and shear rate for olive oil is always linear at all temperatures. This is another indication of the Newtonian behavior of the olive oil samples as indicated by the simplest equation of Newton's law of viscosity.

Chapter Five

Discussion and conclusion

Three samples from L_1 and three samples from L_8 are selected to be analyzed. The reasons of choosing these two regions are:

Firstly: they are far enough from each other.

Secondly: the altitude are different, it is 350 m for L₁, and 890 m for L₈.

Thirdly: the quantities of rain are different for both regions, since we have different crops.

5.1 Density

The olive oil density is shown to be decreasing linearly as a function of age. The average value of the measured result of density of olive oil samples is found to be 0.9180 gm/cm^3 . This result is in a good agreement with Robert and his group value. They found the density of olive oil to be 0.918 gm/cm^3 (Robert *et al.*, 1979).

5.2 Refractive index

The average value of refractive index is measured to be 1.4708 while the standard value is 1.4677 - 1.4705 (International Olive Council, 2011).

Our values of refractive index are in good agreement with the standard ones. A linear fit showed that the refractive index is decreasing as a function of storage age.

5.3 Acidity

The experimental results of acidity showed that the quality of olive oil samples varies from Extra virgin to Lampante olive oil according to Table (2.3). The samples showed that the acidity increase with storage age. For

example, the acidity was measured to be 0.44% (Extra virgin) for crop 2012 of L_8 and, 1.56% (Virgin) for crop 2008 of L_7 , 3.18% (Ordinary virgin) for crop 2001 of L_6 , and 9.94% (Lampante) for crop 1997 of L_{10} .

Falque found that the value of the acidity of the extra - virgin olive oils is 0.39%. (Falque *et al.*, 2007). These values are close to the extra - virgin olive oils value of acidity of crop 2012 of L_8 (0.44%).

The equations from linear fit of density versus storage age, refractive index versus storage age, acidity versus storage age, and refractive index versus density are given in Table 5.1 for different regions.

Table 5.1: The linear equations by linear fitting for density, refractive index and acidity versus storage age, and refractive index versus density for L_1 and L_8

Region		Equation
L ₁	Density as a function of	$\rho = -0.00009 t + 0.91908$
L_8	storage age	$\rho = -0.00024 t + 0.91946$
L ₁	Refractive index as a	n = -0.00008 t + 1.47166
L_8	function of storage age	n = -0.00017 t + 1.47175
L ₁	FFA% as a function of	FFA% = 0.73817 t - 0.13134
L_8	storage age	FFA% = 0.77305 t - 0.50782
L ₁	Refractive index as a	$n = 0.75966 \ \rho + 0.77342$
L_8	function of density	$n = 0.65741 \ \rho + 0.86723$
L ₁	Viscosity as a function of	$\eta = -0.55263 t + 76.63158$
L ₈	storage age	$\eta = -0.55465 t + 69.81658$

where ρ is the density in (gm/cm³), t is the storage age in years, FFA% is the free fatty acid composition, and *n* is the refractive index.

The coefficients of the linear equations relating density and storage age, refractive index and storage age, FFA% and storage age, and refractive index and density differ from one region to another as seen from Table 5.1.

5.4 Viscosity

Many theoretical predictions were suggested to describe the relationship between dynamic viscosity and temperature of olive oil. Three models were tested but failed to describe the relationship because storage age was not taken into consideration. Modifications were done on these equations to improve the theoretical prediction, but still the equations are not proper.

Three new equations were proposed to describe the viscosity - temperature relationship. Our proposed equations presented the best fit for the experimental results.

The results of viscosity measurements showed that the behavior of olive oil is Newtonian since the value of the flow indices (n) is very close to one.

Our work is not consistent with the work of Adnan and his group who found the flow index of olive oil to be 0.84, and so olive oil was considered Non - Newtonian (Adnan *et al.*,2009).

References

- Abramovic H., klofutar G., "The temperature dependence of dynamic viscosity for some vegetables oils", Acta Chim. Slov. 45(1), 69-77 (1998).
- Adolfo, V. F. and Ana, G B., "Study of the evolution of the hysicochemical and structural characteristics of olive and sunflower oil after heating at frying temperatures", Food Chemistry 98, 214-219 (2006).
- Ahmad M., Amran A., Giap S.G.E. and Nik W. M., "The Assessment of Rheological Model Reliability In Lubricating Behavior of Vegetable Oils", Engineering e-Transaction, 4(2), 81-89 (2009).
- Adnan Q., Ahmad M., Akhtar N., Farzana K. and Mehmood A.,
 "Rheological Studies and Characterization of Different Oils", J.
 Chem. Soc. Pak., 31(2), 201 206 (2009).
- Almeidaa S., Belchiora C., Nascimentob M., Vieirab L. and Fleuryb G., *"Performance of a Diesel Generator Fuelled With Palm Oil"*, FUEL 81 (16), p.p 2097–2102(2002).
- Andrade E. N. C., **Nature**, **125**, 309-318(1930).
- Angerosa F., Mostallino R., Basti C., Vito R., "Influence of malaxation temperature and time on the quality of virgin olive oils", Food Chemistry, 72 19-28 (2001).
- Bayrak Y., Iscan M., and Topallar H., "Kinetics of Autoxidative Polymerization of Sun Flower Seed Oil", Tr. J. of chemistry, 21, 118-125 (1997).

- Bird R.B., Armstrong R.C., and Hassager O., "Dynamics of Polymeric Liquids", Vol. 1, Wiley, New York, (1987).
- Bruwer J. J., BoshoffB. D., HugoF. J. CDuPlessis., L. M., FulsJHawkins., C., Vander Walt A. N., and Engelbert A., "The Utilization of Sunflower Seed Oil as Renewable Fuel Diesel Engines. In Agricultural Energy", ASAE, 2,4 -81(1981).
- Chhabra R.P., and Richardson J.F., "Non Newtonian Flow in the Process Industries", Butterworth Heinemann Publishers; (1999).
- Clements C., Craig-Schmidt M., Fasina O. O. and Hallman H., *"Predicting temperature - dependence viscosity of vegetable oils from fatty acid composition"*, Journal of the American Oil Chemists' Society, 83(10), 899-903 (2006).
- De Guzman J., "Relation between fluidity and heat of fusion",
 Anales Soc. Espan. Fis. Y. Quim. 11, 353-362 (1913).
- Duhne C. R., "Viscosity temperature correlations for liquids", Chem. Eng., 86 (15), 83 (1979).
- Dutt N., Ghosh T. K., Prasad D. H. L., Rani K. Y., and Viswanath D.
 S., "Viscosity of Liquids Theory, Estimation, Experiment, and Data", 444-553 (2007).
- Eromosele C., and Paschal N., "Characterization and Viscosity Parameters of Seed Oils from Wild Plants", Bioresource Technology 86 203–205(2003).
- Falque E. and Mendez A. I., "Effect of storage time and container type on the quality of extra-virgin olive oil", Since Direct Food Control, 18, 521–529, (2007).

- Farhoosh R., Niazmand R., Rezaei M. and Sarabi M., "Kinetic Parameter Determination of Vegetable Oil Oxidation Under Rancimat Test Conditions", Eur. J. Lipid Sci. Technol., 110, 587– 592(2008).
- Fasina O., Colley Z., "Viscosity and Specific Heat of Vegetable Oils as a Function of Temperature: 35°C to 180°C", Int. J. Food Prop., 11(4): 738-746. (2008).
- Goering C. E., A. W. Schwab, M. J. Daugherty, E. H. Pryde, and A. J. Heakin, "Fuel Properties of Eleven Vegetable Oils", ASAE, 81-3579(1981).
- Hsieh F.H., Rodenbush C.M. and Viswanath D.S., "Density and Viscosity of Vegetable Oils", JAOCS., 76, 1415–1419 (1999).
- <u>http://www.aoac.org/iMIS15_Prod/AOAC</u>, AOAC, Association of

```
Official Agricultural Chemists, 1997.
```

- <u>http://www.naooa.org/</u>, NAOOA , North American Olive Oil Association, 2013.
- IOOC, International Olive oil Council "Trade standards applying to olive oil and olive pomace oil", E. /Conv. /Doc. no.1 /16 November 2000. International Olive Oil Council, Madrid, Spain.
- James F., "Rheological Methods in Food Process Engineering", 2nd
 Ed., 1-35 (1996).
- Larson R. G., "The Structure and Rheology of Complex Fluids", Oxford University Press, Oxford and New York, (1999).
- Lupi F.R., Gabriele D., Facciolo D., Baldino N., Seta L., de Cindio
 B., "Effect of organogelator and fat source on rheological

properties of olive oil-based organogels" Food Research International; **46**(1), 177–184 (2012).

- Martin A. and BustamanteP., "Physical Pharmacy", 4th Ed, Lippincot, USA, P. 453 (2004)
- Middleman S., "The Flow of High Polymers", Wiley Interscience, New York, (1968).
- Munson B.R., Young D.F., and Okiishi T.H., "Fundamentals of Fluid Mechanics", Wiley, New York, (1998).
- Natarajan G. and Viswanath D. S., "Data book on viscosity of liquids", Hemisphere, New York (1989).
- Nierat T., Musameh S., and Abdelraziq I., "Temperature and Storage Age (Yearly Basis)-Dependence of Olive Oil Viscosity in Different Locations in Palestine", J. Mater. Environ. Sci. 5 (1): 245-254 (2014).
- Nierat T., Musameh S., and Abdelraziq I., "Temperature and Storage Age (Weekly Basis)-Dependence of Olive Oil Viscosity in Different Locations in Palestine", MSAIJ 9(11): 445-451 (2013).
- Nierat T., "Temperature and Storage Age-Dependence of Olive Oil Viscosity in Different Locations in Palestine ", master thesis, An-Najah National University, (2012).
- Noureddini H., Teoh B.C. and Davis Clements L., "Viseosities of Vegetable Oils and Fatty Acids", JAOCS, 69,12 (1992).
- Patil T., Butala D., Raghunathan T., and Shankar H., "Thermal Hydrolysis of Vegetable Oils and Fats. 1. Reaction Kinetics", Ind. Eng. Chem. Res., 27 (5), p 727–735(1988).

- Quinchia L., Delgado M., Franco J., Spikes H. and Gallegos C., "Low
 Temperature Flow Behaviour of Vegetable Oil-Based Lubricants", Industrial Crops and Products, 37 p 383– 388(2012).
- Reid J. F., HansenA. C., and C. E. Goering, "Quantifying Diesel Injector Coking with Computer Vision"., ASAE,32(5), 1503-1506(1989).
- Remington, "The Science and Practice of Pharmacy", Vol. 1, 21st
 Ed, Lippincot, USA, P. 338 (2006).
- Robert C., "CRC Handbook of Chemistry and Physics", David R.
 Lide; (1979).
- Sims R.E., Raine R., and McLeod R., "Rapeseed Oil as a Fuel for Diesel Engines". SAE-Australia. Paper presented at the National Conference on Fuels from Crops of the Society of Automotive Engineers – Australia (1981).
- Skelland A.H.P., "Non Newtonian Flow and Heat Transfer", John Wiley and Sons Inc.; (1967).
- Stanciu I., "A New Viscosity Temperature Relationship for Vegetable Oil", J. Petroleum Technology and Alternative Fuels, 3(2), pp. 19-23, (2012).
- The World Bank 2012, West Bank and Gaza Program," Brief
 Overview of the Olive and the Olive Oil Sector in the
 Palestinian Territories", http://go.worldbank.org/MBK9GU1TD0
- Themelis, N. J., "Transport and chemical rate phenomena", Basel : Gordon and Breach 1995

- Toscano G., Riva G., FoppaPedretti E. and Duca D., "Vegetable Oil and Fat Viscosity Forecast Models Based on Iodine Number and Aponification Number", biomass and bioenergy, 46p.p 511-516(2012).
- Vogel H., "Das temperaturabhängigkeitsgesetz der viskosität von flüssigkeiten", Physics. 22, 645-646 (1921).
- Yalcin H, Toker O.S., and Dogan M., "Effect of Oil Type and Fatty Acid Composition on Dynamic and Steady Shear Rheology Of Vegetable Oils", J Oleo Sci., 61(4):181-7(2012).

Appendix A

Density Results

The densities in (gm/cm³) of the olive oils samples for all regions and for different storage ages are given in Table (A.1).

Storage Age (years)	L ₂	L_3	L_4	L ₅	L ₆	L ₇	L ₉	L_{10}	L_{11}	L ₁₂	L ₁₃	Average
1	0.9189	0.9183	0.9184	0.9189								0.9187
2	0.9189	0.9178	0.9184				0.9185			0.9185		0.9185
3	0.9185		0.9183	0.9188	0.9183	0.9188	0.9182		0.9185	0.9185	0.9188	0.9185
4			0.9182	0.9186	0.9182	0.9186					0.9185	0.9184
5						0.9185	0.9177					0.9182
6				0.9186		0.9181				0.9180		0.9181
7				0.9182			0.9175					0.9178
8									0.9180			0.9178
12					0.9177						0.9179	0.9178
13					0.9176							0.9178
14	0.9160	0.9176			0.9176			0.9179			0.9179	0.9174
15	0.9155	0.9175			0.9175				0.9174		0.9179	0.9172
16					0.9175			0.9165			0.9175	0.9173
19						0.9174						0.9174
Average	0.9176	0.9178	0.9183	0.9186	0.9178	0.9183	0.9180	0.9172	0.9180	0.9183	0.9181	0.9180

 Table (A.1): Measured density of olive oil samples in different regions and for different storage ages

Table (A.1) shows the average density in each region, for each storage age in years.

The relationship between density of olive oil samples and storage age are shown in figures (A.1 - A.11).



Fig. A.1: Measured density versus storage age of olive oil samples of L₂



Fig. A.2: Measured density versus storage age of olive oil samples of L₃



Fig. A.3: Measured density versus storage age of olive oil samples of L₄



Fig. A.4: Measured density versus storage age of olive oil samples of L₅



Fig. A.5: Measured density versus storage age of olive oil samples of L₆



Fig. A.6: Measured density versus storage age of olive oil samples of L_7



Fig. A.7: Measured density versus storage age of olive oil samples of L₉



Fig. A.8: Measured density versus storage age of olive oil samples of L_{11}



Fig. A.9: Measured density versus storage age of olive oil samples of L_{12}



Fig. A.10: Measured density versus storage age of olive oil samples of L₁₃

Appendix B

Refractive Index Results

The refractive index of the olive oils samples for all regions and for different storage ages are given in Table (B.1).

Table	(B.1):	Measured	refractive	e index o	of olive o	oil sam	ples in	different	region a	nd for	different	storage	ages
	· /								0			0	0

Storage Age (years)	L ₂	L ₃	L ₄	L_5	L ₆	L_7	L9	L ₁₀	L_{11}	L ₁₂	L ₁₃	Average
1	1.4712	1.4710	1.4717	1.4716								1.4715
2		1.4710	1.4716				1.4713			1.4713		1.4713
3	1.4711		1.4711	1.4714	1.4711	1.4715	1.4713		1.4713	1.4712	1.4711	1.4712
4			1.4707	1.4710		1.4714					1.4705	1.4710
5						1.4712	1.4712					1.4711
6				1.4708		1.4712				1.4709		1.4709
7				1.4707			1.4709					1.4708
8				1.4707					1.4711			1.4708
12					1.4705						1.4703	1.4704
13					1.4702							1.4705
14	1.4708	1.4709			1.4701			1.4708			1.4701	1.4706
15	1.4706	1.4708			1.4704			1.4698	1.4710		1.4702	1.4705
16					1.4700			1.4690			1.4700	1.4698
19						1.4711						1.4711
Average	1.4709	1.4709	1.4713	1.4710	1.4704	1.4713	1.4712	1.4699	1.4711	1.4711	1.4704	1.4708

Table (B.1) shows the average refractive index in all regions, for each storage age in years.

The relationship between refractive index of olive oil samples and storage age is shown in figures (B.1 - B.11).



Fig. B.1: Measured refractive index versus storage age of olive oil samples of L_2



Fig. B.2: Measured refractive index versus storage age of olive oil samples of L₃



Fig. B.3: Measured refractive index versus storage age of olive oil samples of L₄



Fig. B.4: Measured refractive index versus storage age of olive oil samples of L₅



Fig. B.5: Measured refractive index versus storage age of olive oil samples of L₆



Fig. B.6: Measured refractive index versus storage age of olive oil samples of L_7



Fig. B.7: Measured refractive index versus storage age of olive oil samples of L₉



Fig. B.8: Measured refractive index versus storage age of olive oil samples of L_{10}



Fig. B.9: Measured refractive index versus storage age of olive oil samples of L_{11}



Fig. B.10: Measured refractive index versus storage age of olive oil samples of L₁₂



Fig. B.11: Measured refractive index versus storage age of olive oil samples of L_{13}

Appendix C

Acidity Results

The Acidity of the olive oils samples for all regions and for different storage ages are given in Table (C.1).

Table	(C.1):	Measured	Acidity	of olive	oil sa	amples in	different	regions
I GOIC	$(\mathbf{\nabla} \mathbf{I} \mathbf{I})$	i i i cu cu	includy			mpres m		10510110

Storage Age (years)	L ₂	L ₃	L_4	L ₅	L ₆	L ₇	L ₉	L ₁₀	L ₁₁	L ₁₂	L ₁₃
1	1.72	0.94	0.51	0.45							
2		0.70	0.53				0.67			0.77	
3	2.53		0.56	1.49	1.03	0.80	0.96		0.96	1.88	0.56
4			1.96	2.07	1.80	1.12					1.03
5						1.56	1.18				
6				2.40		2.50				2.00	
7				2.88			1.20				
8				2.94					3.92		
12					3.18						2.31
13								4.40			
14	5.09	5.98						5.04			
15	5.77	5.72			3.81				7.07		3.95
16					4.49			9.49			4.60
19					5.02	4.00					5.22

and for different storage ages

The relationship between Acidity of olive oil samples and storage age is shown in figures (C.1 - C.11).



Fig. C.1: Measured acidity versus storage age of olive oil samples of L₂



Fig. C.2: Measured acidity versus storage age of olive oil samples of L₃



Fig. C.3: Measured acidity versus storage age of olive oil samples of L₄



Fig. C.4: Measured acidity versus storage age of olive oil samples of L₅



Fig. C.5: Measured acidity versus storage age of olive oil samples of L₆



Fig. C.6: Measured acidity versus storage age of olive oil samples of L_7



Fig. C.7: Measured acidity versus storage age of olive oil samples of L₉



Fig. C.8: Measured acidity versus storage age of olive oil samples of L_{10}


Fig. C.9: Measured acidity versus storage age of olive oil samples of L_{11}



Fig. C.10: Measured acidity versus storage age of olive oil samples of L_{12}



Fig. C.11: Measured acidity versus storage age of olive oil samples of L₁₃

Appendix D

Refractive Index as a Function of Density

Figs. (D.1 - D.11) show the relationship between measured refractive index and density for all regions and all storage ages.



Fig. D.1: Measured refractive index versus density of olive oil samples of L₂



Fig. D.2: Measured refractive index versus density of olive oil samples of L₃



Fig. D.3: Measured refractive index versus density of olive oil samples of L₄



Fig. D.4: Measured refractive index versus density of olive oil samples of L₅



Fig. D.5: Measured refractive index versus density of olive oil samples of L₆



Fig. D.6: Measured refractive index versus density of olive oil samples of L7



Fig. D.7: Measured refractive index versus density of olive oil samples of L9



Fig. D.8: Measured refractive index versus density of olive oil samples of L_{10}



Fig. D.9: Measured refractive index versus density of olive oil samples of L₁₁



Fig. D.10: Measured refractive index versus density of olive oil samples of L_{12}

Appendix E

Viscosity Results

Figs. (E.1 - E.22) show the relationship between measured viscosity and temperature for all regions and all storage ages.



Fig. E.1: Measured viscosity versus temperature of olive oil samples of L_2 (1year and 3 years storage age)



Fig. E.2: Measured viscosity versus temperature of olive oil samples of L_2 (14 years and 15 years storage age)



Fig. E.3: Measured viscosity versus temperature of olive oil samples of L_3 (1 year and 2 years storage age)



Fig. E.4: Measured viscosity versus temperature of olive oil samples of L_3 (14 years and 15 years storage age)



Fig. E.5: Measured viscosity versus temperature of olive oil samples of L_4 (1 year and 2 years storage age)



Fig. E.6: Measured viscosity versus temperature of olive oil samples of L_4 (3 years and 4 years storage age)



Fig. E.7: Measured viscosity versus temperature of olive oil samples of L_5 (1 year, 2 years and 3 years storage age)



Fig. E.8: Measured viscosity versus temperature of olive oil samples of L_5 (6 years, 7 years and 8 years storage age)



Fig. E.9: Measured viscosity versus temperature of olive oil samples of L_6 (3 years and 4 years storage age)



Fig. E.10: Measured viscosity versus temperature of olive oil samples of L_6 (12 years and 13 years storage age)



Fig. E.11: Measured viscosity versus temperature of olive oil samples of L_6 (14 years, 15 years and 16 years storage age)



Fig. E.12: Measured viscosity versus temperature of olive oil samples of L₇ (3 years and 4 years storage age)



Fig. E.13: Measured viscosity versus temperature of olive oil samples of L_7 (5 years, 6 years and 19 years storage age)



Fig. E.14: Measured viscosity versus temperature of olive oil samples of L₉ (2 years and 3 years storage age)



Fig. E.15: Measured viscosity versus temperature of olive oil samples of L₉ (5 years and 7 years storage age)



Fig. E.16: Measured viscosity versus temperature of olive oil samples of L_{10} (1 year and 14 years storage age)



Fig. E.17: Measured viscosity versus temperature of olive oil samples of L_{10} (14 years and 15 years storage age)

جامعة النجاح الوطنية كلية الدراسات العليا

الخصائص الديناميكية الفيزيائية لزيت الزيتون في فلسطين

إعداد أحمد مصطفى بهتي

إشراف أ. د. عصام راشد عبد الرازق د. شريف محمد مسامح

قدمت هذه الأطروحة استكمالاً لمتطلبات الحصول على درجة الماجستير في الفيزياء بكلية الدراسات العليا في جامعة النجاح الوطنية في نابلس، فلسطين.

الخصائص الديناميكية الفيزيائية لزيت الزيتون في فلسطين إعداد أحمد مصطفى بهتي إشراف أ. د. عصام راشد عبد الرازق د. شريف محمد مسامح

الملخص

تم دراسة عينات من زيت الزيتون في فلسطين ذات أعمار تخزين مختلفة ومن مناطق مختلفة. حيث تم قياس الكثافة ومعامل الانكسار و الحموضة و اللزوجة للعينات. تم دراسة معامل الانكسار لعينات زيت الزيتون كعلاقة مع عمر التخزين و أظهرت النتائج أن معامل الانكسار يقل بزيادة عمر التخزين. أظهرت الدراسة أن حموضة عينات زيت الزيتون من مختلف المحاصيل وفي مناطق مختلفة تزداد بزيادة عمر التخزين. اظهرت النتائج ان معظم عينات زيت الزيتون (عمر مناطق مختلفة من عامر التخزين و أظهرت النتائج أن معامل الانكسار يقل بزيادة عمر التخزين. أظهرت الدراسة أن حموضة عينات زيت الزيتون من مختلف المحاصيل وفي مناطق مختلفة تزداد بزيادة عمر التخزين. اظهرت النتائج ان معظم عينات زيت الزيتون (عمر مناطق مختلفة تزداد بزيادة عمر التخزين. اظهرت النتائج ان معظم عينات زيت الزيتون هما الانكريون المناطق مناطق مختلفة من ما التخزين. الموضة لها معايير الجودة الدولية (< 3.3 ٪). تجدر الإشارة هنا إلى أن زيت الزيتون يمكن تخزينه حتى 12 عاما دون ان تتجاوز الحموضة معايير الجودة الدولية في الطروف المناسبة.

وقد تم دراسة اللزوجة لعينات زيت الزيتون من محصول عام 2012 من مناطق مختلفة، و أظهرت النتائج أن تصنيف معظم عينات زيت الزيتون هو زيت الزيتون البكر.

تم اقتراح معادلات تتضمن ثابتين واخرى تتضمن ثلاث ثوابت للحصول على افضل علاقة مناسبة لوصف تغير اللزوجة الديناميكية لزيت الزيتون مع درجة الحرارة.

تمت مقارنة النتائج التجريبية للزوجة مع معادلة قانون القوة الخاصة باللزوجة، وقد تبين أن سلوك جميع عينات الزيت هو سلوك نيوتوني.