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# Linear DNA-Linked Nanoparticle Building Blocks (nBLOCKs) for Modular Self-Assembly of Nanostructures

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Linear DNA-Linked Nanoparticle Building Blocks (nBLOCKs) for Modular Self-Assembly of  
Nanostructures

A thesis submitted in partial fulfillment  
of the requirements for the degree of  
Master of Science in Cell and Molecular Biology

by

Jakob Hockman  
University of Arkansas  
Bachelor of Science in Biomedical Engineering, 2016

August 2017  
University of Arkansas

This thesis is approved for recommendation to the Graduate Council.

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## Abstract

Controlling the shapes and sizes of nanomaterials often enables controlling their properties for certain applications. The most promising methods for controlling the shapes and sizes of nanostructures use base-pairing between complementary DNA strands to self-assemble nanostructures from DNA and nanoparticles. DNA Brick-based self-assembly is a particularly useful method for creating DNA nanostructures. It offers a large amount of control over the final shapes and sizes because it uses building blocks that are anisotropic and have predictable geometry. However, this control has not been extended to the self-assembly of nanostructures from nanoparticles. Applying DNA Brick based self-assembly to the self-assembly of nanostructures from nanoparticles would require DNA-linked nanoparticles that are anisotropic and have predictable geometry.

To this end, Solidworks models were used to study the interactions between DNA Bricks so that detailed information could be gained about their mode of self-assembly. This information was used to generate Solidworks models of DNA-linked nanoparticle building blocks (nBLOCKs) that can be used for DNA Brick-based self-assembly. These nBLOCKs could be created by attaching a single 43 base pair (bp) long DNA strand to gold nanoparticles using the anisotropic monofunctionalization technique. However, accomplishing this feat would require improving the efficiency of the anisotropic monofunctionalization method first.

Attempts to improve all three steps of the anisotropic monofunctionalization technique yielded mixed results. The efficiency of the first step, binding DNA to a solid support, was improved by implementing the photocleavable (PC) biotin – streptavidin interaction. UV-Vis absorbance spectroscopy revealed that the PC biotinylated DNA strands became bound to streptavidin-coated magnetic beads with nearly 100% efficiency. However, the second and third steps, binding gold nanoparticles (AuNPs) to DNA and cleaving DNA-linked AuNPs from the beads, still suffer from low yields. The efficiency of the second step was incrementally improved from 3% to 25% by tuning the reaction conditions. The third step was carried out at a maximum of 10% efficiency. The method was successfully used to generate nBLOCKs but the overall yield was less than 5%. Explanations of and possible solutions to the low-yield are suggested for future experiments.

## **Acknowledgements**

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Finally, I would like to thank my family and friends for all of their heartfelt support.

## **Dedication**

To Dad, Mom, Kasey and Luke.

## Table of Contents

1. Introduction .....	1
1.1 Structural DNA Nanotechnology .....	2
1.2 Self-Assembling Gold Nanostructures .....	9
1.3 Anisotropic Monofunctionalization “One-At-A-Time” Method .....	12
1.4 Thesis Objectives .....	18
2. Materials and Methods .....	19
2.1 Materials. ....	19
2.2 Computer Modelling. ....	19
2.3 DNA-Functionalization of AuNPs. ....	19
2.4 Characterization .....	20
3. Results and Discussion .....	21
3.1 Modelling the Self-Assembly of DNA Bricks and nBLOCKs .....	21
3.2 Improving the Efficiency of the Anisotropic Monofunctionalization Technique .....	34
4. Conclusion .....	46
5. Future Work .....	47
6. References .....	49
Appendix A: Strand Sequences for Small nBLOCK cube nanostructures .....	52

## List of Figures

Figure 1. Strategy of DNA Origami [16] .....	3
Figure 2. Nanostructures created using DNA Origami [16] .....	4
Figure 3. Strand displacement reaction leading to the opening of a DNA origami lockbox [19] .....	5
Figure 4. Periodic DNA building block in the shape of tensegrity triangle [22] .....	5
Figure 5. DNA Brick method of self-assembly [23].....	7
Figure 6. Nanostructures created using DNA Bricks [23].....	8
Figure 7. DNA Origami scaffold used to self-assemble nanostructures of AuNPs [38] .....	10
Figure 8. DNA Brick scaffold used to create nanostructures from AuNPs [39] .....	10
Figure 9. Isotropically functionalized AuNPs can be used to create nanostructures from the inside out (modified from [40]).....	11
Figure 10. Summary of Anisotropic Monofunctionalization Technique [32] .....	13
Figure 11. TEM characterization of nBLOCK geometries (modified from [32]) .....	14
Figure 12. Current limitations of nBLOCKs (modified from [32]).....	16
Figure 13. Adaptation of DNA Brick-based self-assembly for linear nBLOCKs (modified from [23]) .....	17
Figure 14. Formation of DNA Bricks from Linear ssDNA .....	22
Figure 15. Rigidity of dsDNA.....	23
Figure 16. Abstract model of DNA Bricks (adapted from [23]).....	24
Figure 17. Modified DNA Bricks with cavity for nanoparticles.....	26
Figure 18. Different possible geometries of nBLOCKs with one DNA strand attached .....	27
Figure 19. Creation of nanostructures from the inside-out using linear nBLOCKs with one DNA strand attached .....	29
Figure 20. nBLOCKs with two DNA strands attached are not as rigid as dsDNA.....	31
Figure 21. Creation of nanostructures from linear nBLOCKs with two DNA strands attached.....	32
Figure 22. The anisotropic monofunctionalization technique.....	35
Figure 23. DNA binding to streptavidin coated magnetic beads .....	37
Figure 24. TCEP reduction of protected oligonucleotides (modified from [43]) .....	38
Figure 25. Preparing AuNPs and binding DNA .....	39
Figure 26. Summary of attempts to increase efficiency of AuNP .....	40
Figure 27. Photocleavage of PC biotin using UV light (adapted from [45]) .....	42
Figure 28. Efficiency of Photocleavage in H <sub>2</sub> O .....	43
Figure 29. Efficiency of Photocleavage in Buffer .....	45

## 1. Introduction

There is an overarching goal within the field of nanotechnology to control the size and shape of nanostructures because many nanomaterials have size and shape dependent properties. Methods that allow researchers to control the overall size and shape of nanostructures also allow them to tune their material properties for specific applications.

For example, attempts to self-assemble structures from gold nanoparticles (AuNPs) are motivated in part by gold's size and shape dependent localized surface plasmon (LSP) [1-5]. LSP is the collective oscillation of electrons around a nanoparticle in response to incident light. The phenomenon known as plasmon coupling causes electrons within a certain distance of each other ( $< 5\text{nm}$ ) to synchronize their oscillations [6].

The LSP has a resonant frequency at which energy transmission is maximized. Changing the size and shape of the nanoparticle changes the number and relative positions of the electrons in range of plasmon coupling which in turn changes the LSP resonant frequency. LSP causes nanoparticles to vibrate which in turn generates heat and pressure waves.

One specialized application of these size and shape dependent properties is in cancer theranostics (therapy and diagnostics) [7-11]. By tuning the size and shape of AuNP nanostructures, their resonant LSP frequency can be tuned to near-infrared (NIR) light, to which human tissues are maximally transparent. Antibody-conjugated AuNPs injected into the bloodstream will seek out and bind to tumors through specific antibody-antigen interactions. NIR light can then be shined through human tissues to where the nanoparticles are bound on the tumor cells. The resulting mechanical vibrations generate pressure waves that can be detected using an ultrasound device. At high enough concentrations, the mechanical vibrations also generate enough heat to destroy the tumor cells.



## 1.1 Structural DNA Nanotechnology

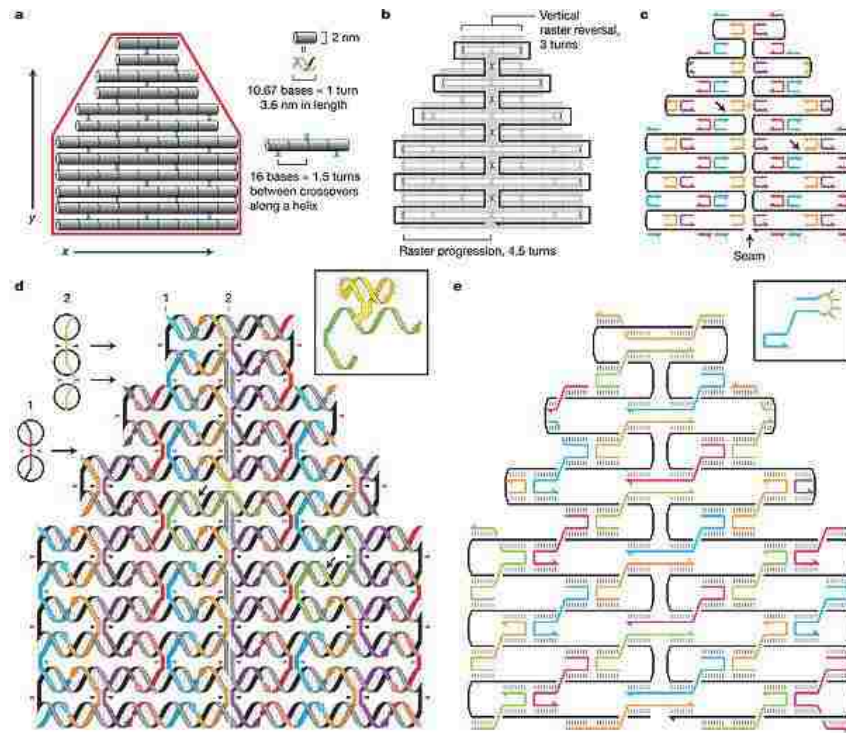
Among the many different methods for controlling the size and shape of nanostructures [3,12,13], none show as much versatility as Structural DNA Nanotechnology. This strategy takes advantage of the base-pairing between DNA strands to manipulate matter at the nanoscale.

There are various ways to create nanostructures within the field of Structural DNA Nanotechnology but they all exploit three important properties of DNA [14]. Single-stranded DNA (ssDNA) is flexible. It can be bent and woven into a variety of shapes and patterns. Double-stranded DNA (dsDNA), on the other hand, is much more rigid with a persistence length of 50nm [15]. One of the most unique features of DNA is its ability to recognize and bind to specific complementary sequences. A structure can thus be created by bending ssDNA strands into a certain pattern and binding complementary sequences in key locations to form dsDNA so that the desired structure is stabilized.

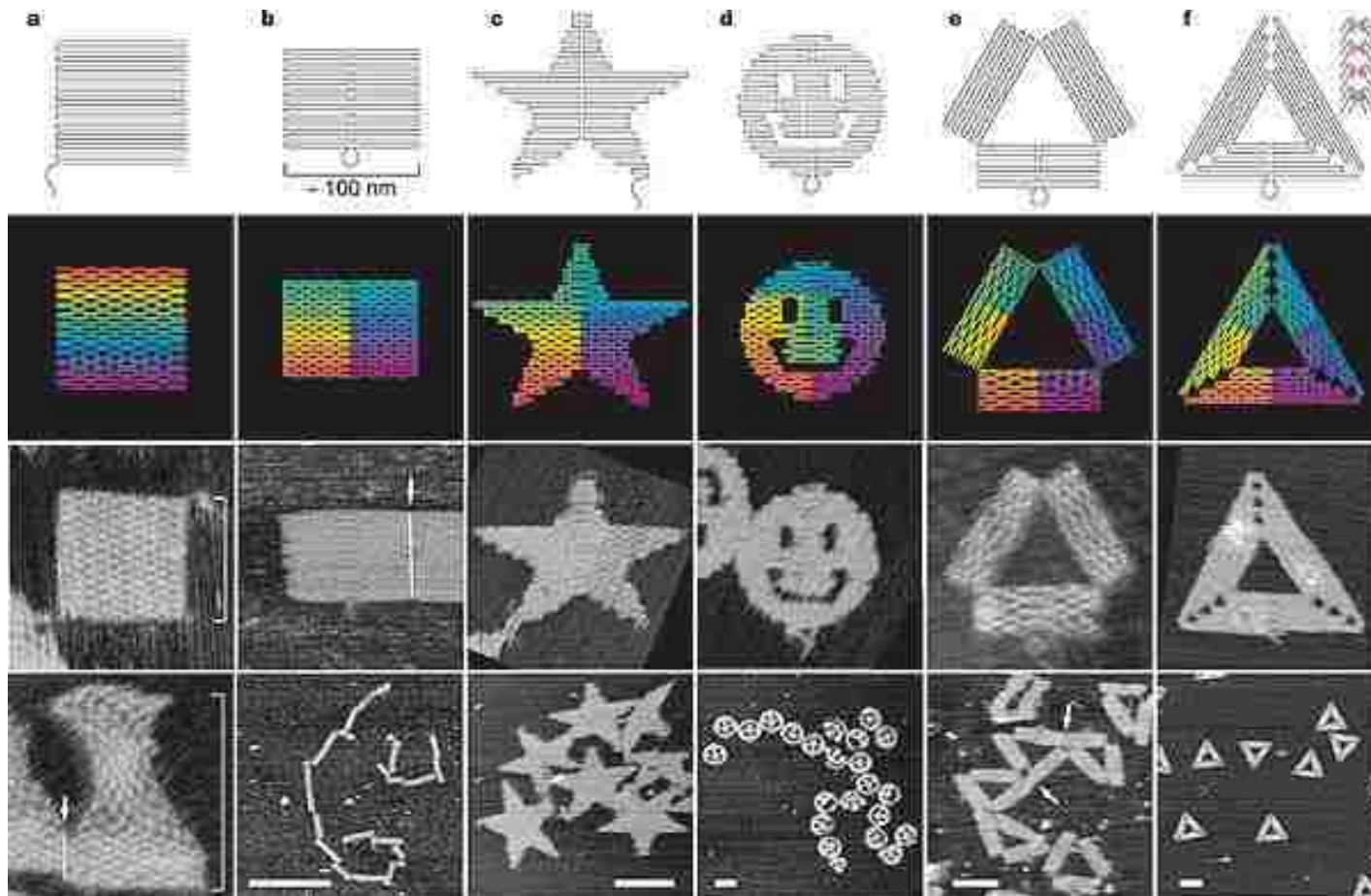
These properties have facilitated a variety of techniques for assembling nanostructures from DNA by itself. One common method of doing so, known as DNA origami [16], involves bending and folding a long DNA template strand into the desired structure, similar to how long, linear protein chains fold into 3-D structures. DNA origami structures are stabilized by dozens to hundreds of shorter staple strands that are complementary to binding domains dispersed throughout the template strand (Fig. 1). This technique has been used to create a wide variety of structures from smiley faces [16] (Fig 2.) to lockboxes (Fig. 3) [17-19].

The major advantage of this method is its versatility since it can be used to create nearly any shape. There are also readily available software packages, caDNAno and CanDO, that aid in designing and testing the stability of DNA origami structures before they are created experimentally [37].

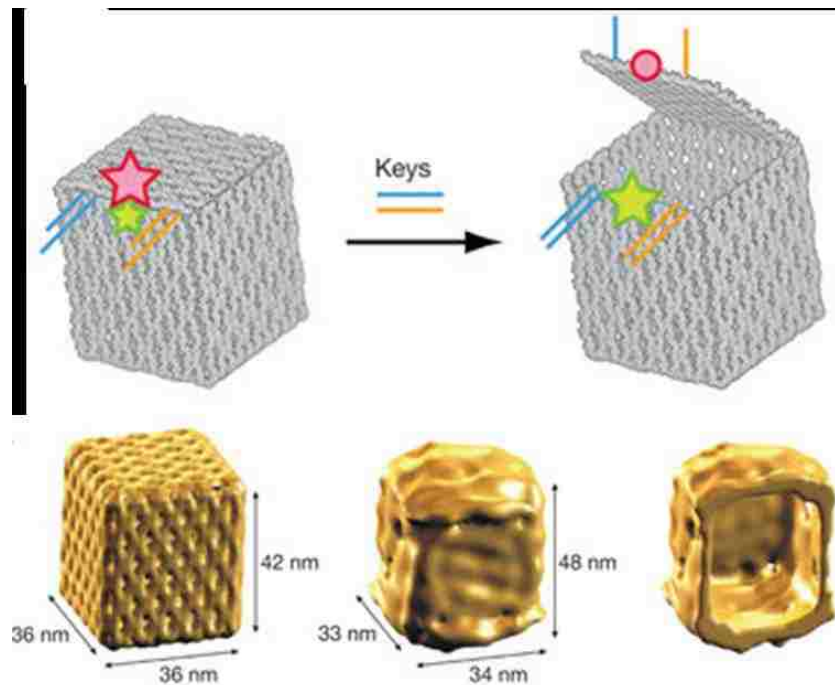
One piece of technology that complements DNA origami well is called strand displacement, which involves adding specific ssDNA strands to existing structures [21]. These additional strands are more complementary to the scaffold strand than some of the staple strands bound in key locations throughout the structure. The new strands displace these staple strands which causes dynamic structure changes, e.g. the opening of a DNA origami lockbox (Fig. 3) [19].



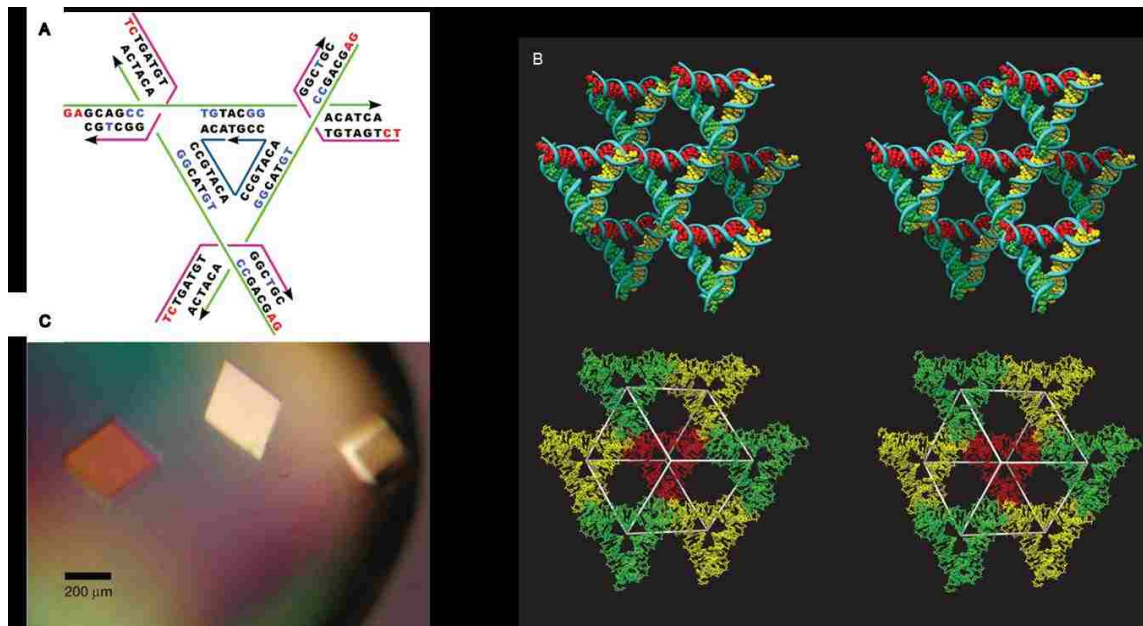
**Figure 1. Strategy of DNA Origami [16]** | DNA origami involves folding a long circular DNA molecule (template strand) into shapes using dozens to hundreds of staple strands. The staple strands bring together distant regions of the molecule and stabilize them in specific positions.



**Figure 2. Nanostructures created using DNA Origami [16]** | Different types of structures can be created by controlling the placement of staple strands throughout the template strand. The only limitation is the size of the final structure which is dependent on the length of the template strand (usually 7000 base pairs (bp's) long).



**Figure 3. Strand displacement reaction leading to the opening of a DNA origami lockbox [19]** | A box with a controllable lid was created using DNA origami and strand displacement reactions. DNA origami was used to create the geometry of the box itself. A strand displacement reaction involving ssDNA “keys” resulted in the displacement of the strands holding the box closed, which caused the lid to open.



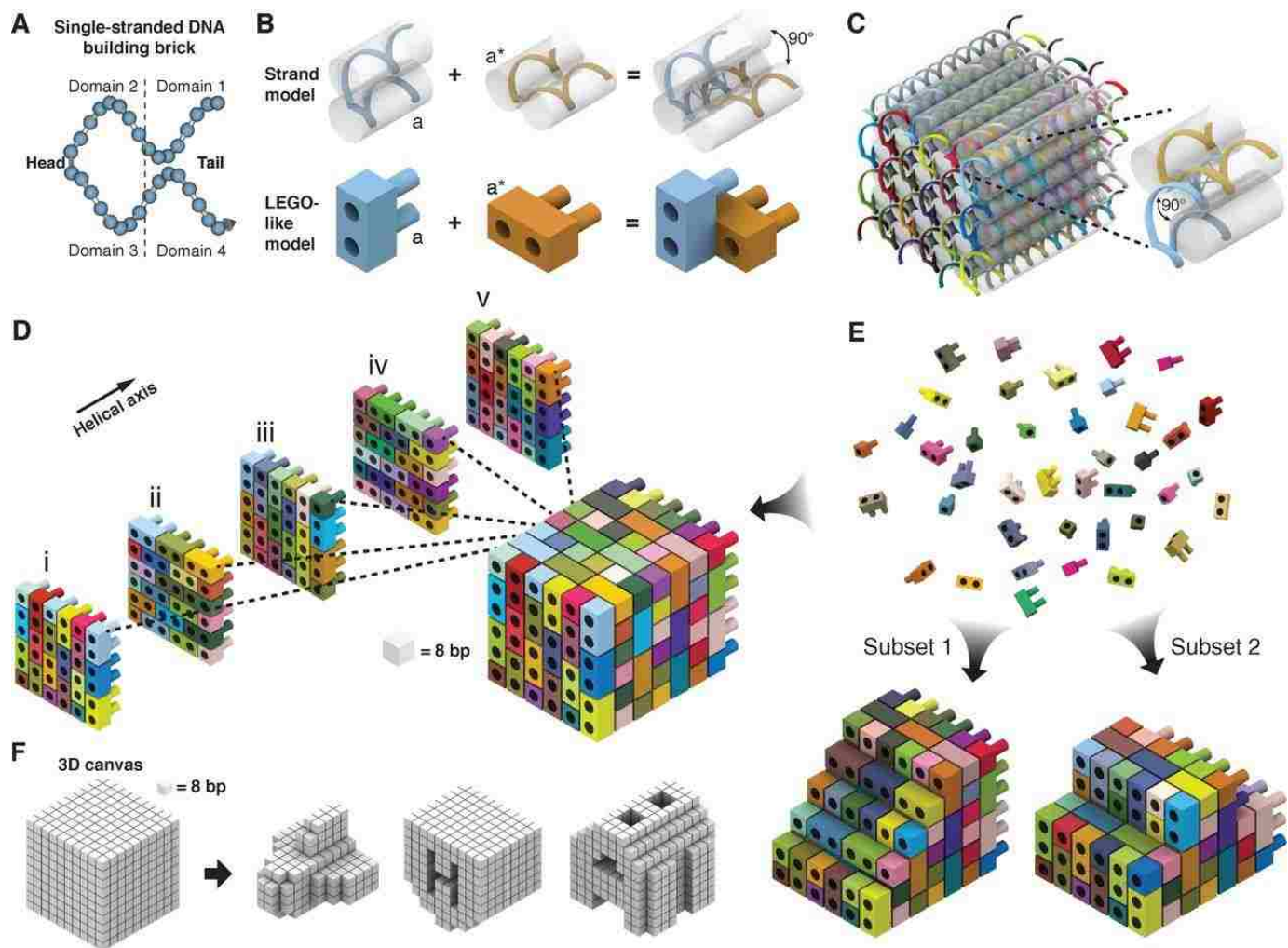
**Figure 4. Periodic DNA building block in the shape of tensegrity triangle [22]** | **A.** Schematic of the strand sequences used to form a tensegrity triangle. Sticky ends (red) interact with one another in a repetitive manner to build up a crystal lattice. **B.** 3-dimensional model of interactions between tensegrity triangles that lead to crystallization. **C.** Macroscopic DNA crystals made from these tensegrity triangles measuring .25 mm.

Another method of creating DNA nanostructures involves creating building blocks that stack on top of one other to form larger structures. This strategy can be further subdivided based on the interactions between building blocks. One strategy uses multiple identical DNA building blocks that interact periodically to create lattices. For example, a tensegrity triangle building block with self-complementary sticky ends was successfully used to create a 3-dimensional DNA crystal that could be seen with the naked eye (Fig. 4) [22]. The benefits of this strategy are its simplicity and scalability – only a few building blocks are needed to create structures that can be expanded to nearly any scale. The major limitation is that there is little control over the final shape of the resulting nanostructures since the building blocks stack uncontrollably.

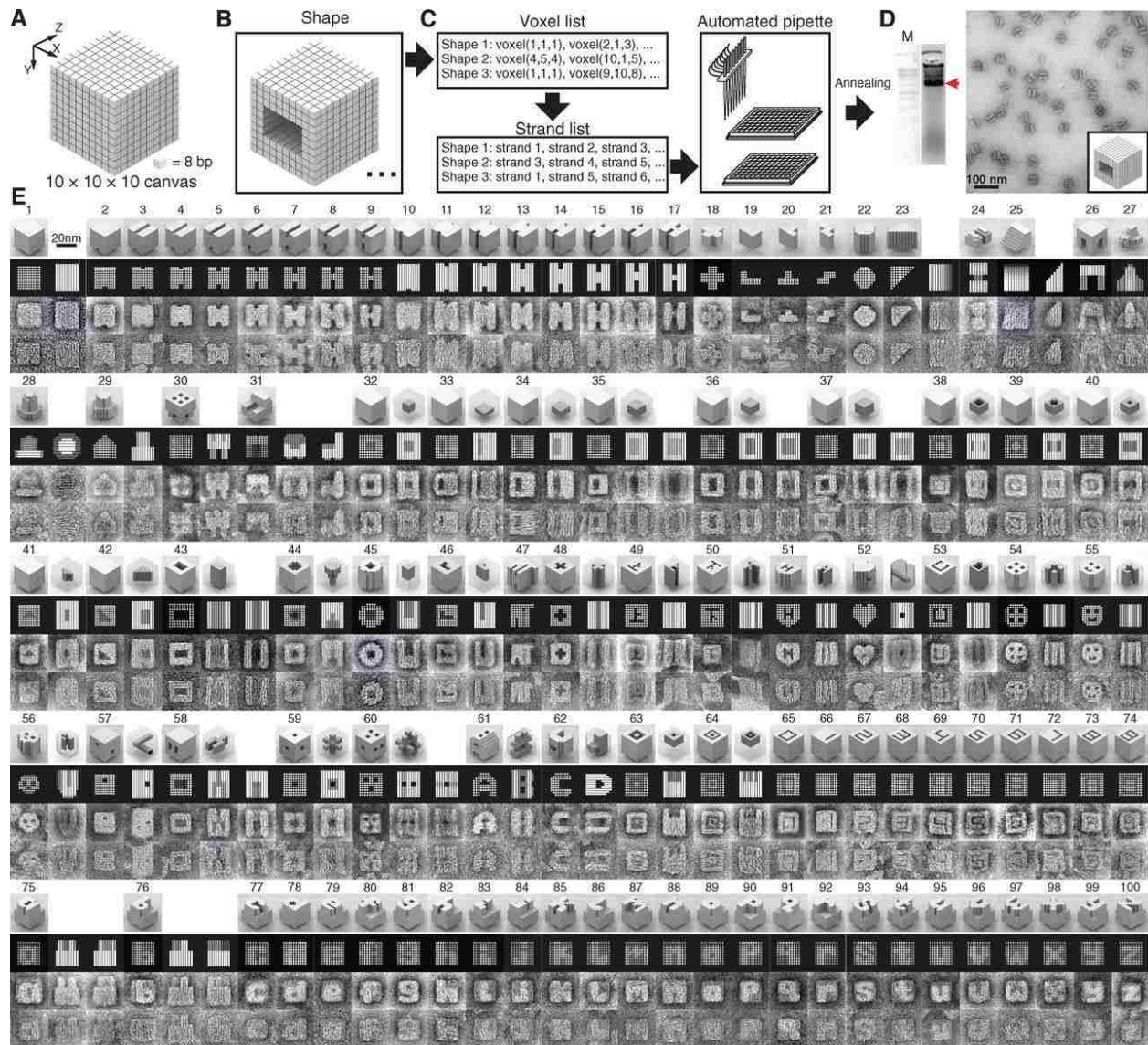
A different method of assembling nanostructures from DNA involves designing unique interactions for every DNA building block in an assembly. An excellent example of this “hard-coding” approach is the self-assembly of DNA Bricks [23,24]. Each DNA Brick is a linear 32 base pair (bp) long ssDNA strand that folds back on itself in between bp’s 16 and 17, creating two parallel helices (Fig. 5). These parallel helices interact with four others through four 8 bp binding domains. Each 8bp domain causes a  $270^\circ$  turn of the DNA helix. This helical turning causes every binding partner of the parallel helices to be oriented at  $270^\circ$  angles from each other. Each ssDNA strand is effectively a “brick” that stacks with other bricks in a specific orientation. When hundreds to thousands of these building blocks come together, they pack together very tightly and form nanostructures (Fig. 6)

The keys to this technology are anisotropy and predictable geometry. Each building block is anisotropic which means that they all have unique interactions with their neighbors. Anisotropy is extremely powerful in building nanostructures since only the desired interactions between building blocks take place, which creates a way to control the final shapes and sizes of the structures. Predictable geometry allows for consistent interactions to occur between building blocks so that they stack up on top of one another in a predictable manner. This predictability aids in designing structures since the relative orientations of all the building blocks are known. The only limitation to this method is the large number of unique building blocks required to create each structure. Many of the structures require hundreds to thousands of unique DNA strands. Designing their interactions can be complicated so there are software packages (e.g. 3DNA [43]) that automatically design the DNA Brick sequences to form the desired structures.





**Figure 5. DNA Brick method of self-assembly [23]** | **A.** DNA Bricks are formed from 32 bp ssDNA strands with a fold in between the 16<sup>th</sup> and 17<sup>th</sup> base pairs. **B.** Each Brick has 4 binding domains, each consisting of 8 bps. These domains can be abstractly represented as pegs and holes in “Bricks”. **C.** Each 8 bp binding domain causes a 270<sup>o</sup> turn of the helix. Each 270<sup>o</sup> binding domain orients its binding partners at a 90<sup>o</sup> angle from all of its other binding partners. **D.** This system allows the Bricks to pack together very tightly and self-assemble into predetermined shapes. **E.** Interactions can be customized so that each set of Bricks forms a unique structure. **F.** A cube can be created from a specific set of Bricks. By leaving out some of these bricks, new structures can be created without having to design new interactions or purchase new DNA strands.



**Figure 6. Nanostructures created using DNA Bricks [23]** | **A.** Researchers created a set of sequences that form a 10x10x10 cube structure. **B.** By leaving out key bricks, certain types of geometries could be formed. **C.** These strand sequences were selected using the help of a software program and added together in the lab using an automated robot. **D.** The correct structures were separated using gel electrophoresis and characterized using a TEM microscope. **E.** Various structures were created using this method, from letters to emojis.

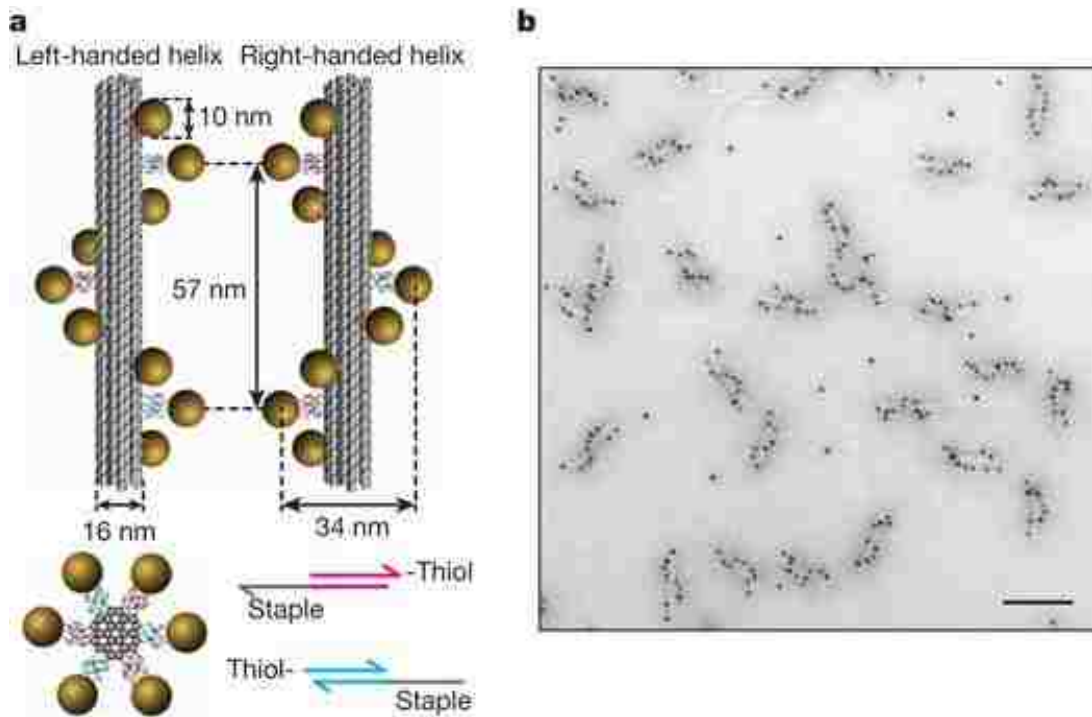
One final approach that attempts to merge these two methods is called algorithmic self-assembly. Algorithmic self-assembly involves creating building blocks that adhere to a certain set of rules [25,26] so that they stack up on top of each other in a predictable manner. These building blocks also require anisotropy and predictable geometry. However, one additional variable is the strength of binding between building blocks. Tuning this strength allows for the building blocks to be reused in very specific locations throughout the assembly as determined by an algorithm contained in the DNA building blocks and the temperature of the system. This method is extremely powerful in that it matches the versatility of the hard-coding approach with the smaller number of building blocks championed by the periodic approach. However, it is generally much more difficult to create these building blocks which is why they are more limited in use.

## **1.2 Self-Assembling Gold Nanostructures**

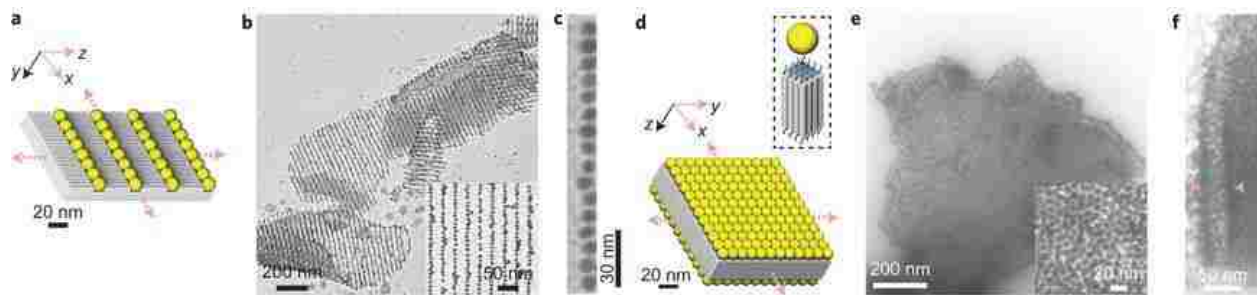
Any structures created using one of the above methods can be functionalized with sticky ends on which external DNA strands can bind. AuNPs can then be attached to these structures by attaching to them to complementary DNA strands and binding them to the sticky ends. These DNA-linked AuNPs can also serve as building blocks themselves if they can interact with one another in one of the manners described above. Thus, many methods have emerged to create different types of DNA-linked AuNPs depending on the desired applications.

One such method, known as isotropic functionalization, involves directly adding nanoparticles to a solution of free floating DNA strands so that many different copies of the same strand coats the surface of each nanoparticle. These DNA-linked nanoparticles can then be used to pattern the surface of DNA origami (Fig. 7) and DNA Brick (Fig. 8) structures [27-29]. The major advantage of creating AuNP nanostructures this way is that nearly any structure that can be built using DNA origami or DNA Bricks can be used to pattern AuNPs. However, the primary limitation of these strategies is that they can only be used to pattern nanoparticles on the surface of the structures. It would be difficult to create structures of AuNPs from the inside-out using either of these two methods.

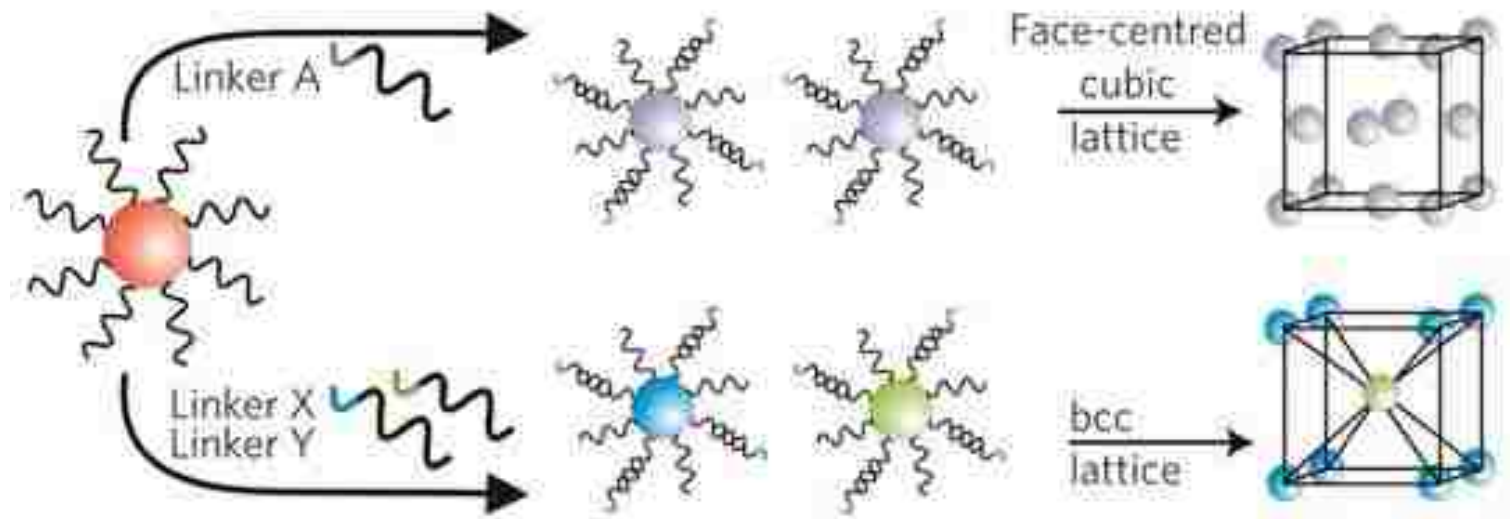




**Figure 7. DNA Origami scaffold used to self-assemble nanostructures of AuNPs [38] | A.** Illustration showing how the DNA scaffold interacts with the DNA-bound AuNPs through complementary sticky ends. **B.** TEM images of nanostructures with precise placement of the nanoparticles on the surface of DNA scaffolds.



**Figure 8. DNA Brick scaffold used to create nanostructures from AuNPs [39] |** Illustration showing how DNA Brick scaffolds can be used to pattern AuNPs. It is possible to precisely place AuNPs on the surface of these structures.



**Figure 9. Isotropically functionalized AuNPs can be used to create nanostructures from the inside out (modified from [40] | Sets of nanoparticles with different complementary DNA strands or self-complementary DNA strands can be used to self-assemble nanostructures with different lattice structures.**

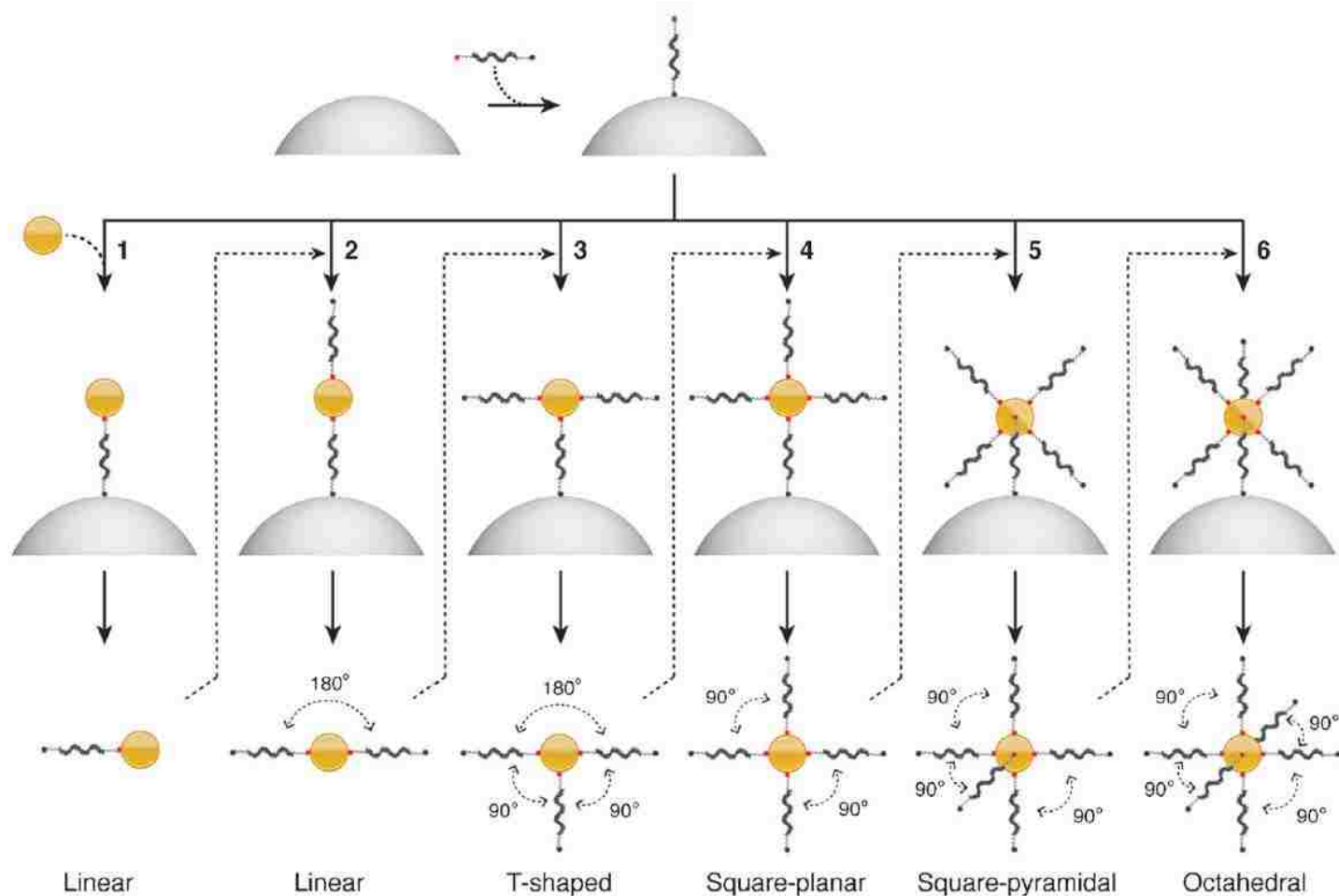
To create structure made entirely of DNA-linked AuNPs, many researchers have begun using isotropically functionalized AuNPs as periodic DNA-linked building blocks (Fig. 9). To date, face-centered cubic and body-centered cubic lattices are the only structures to have been created using this method [30-31]. The major limitation of these building blocks is that it is difficult to control their interactions so that they self-assemble into arbitrary shapes – there is no way to control the overall size and shape of the nanostructures that they form.

### **1.3 Anisotropic Monofunctionalization “One-At-A-Time” Method**

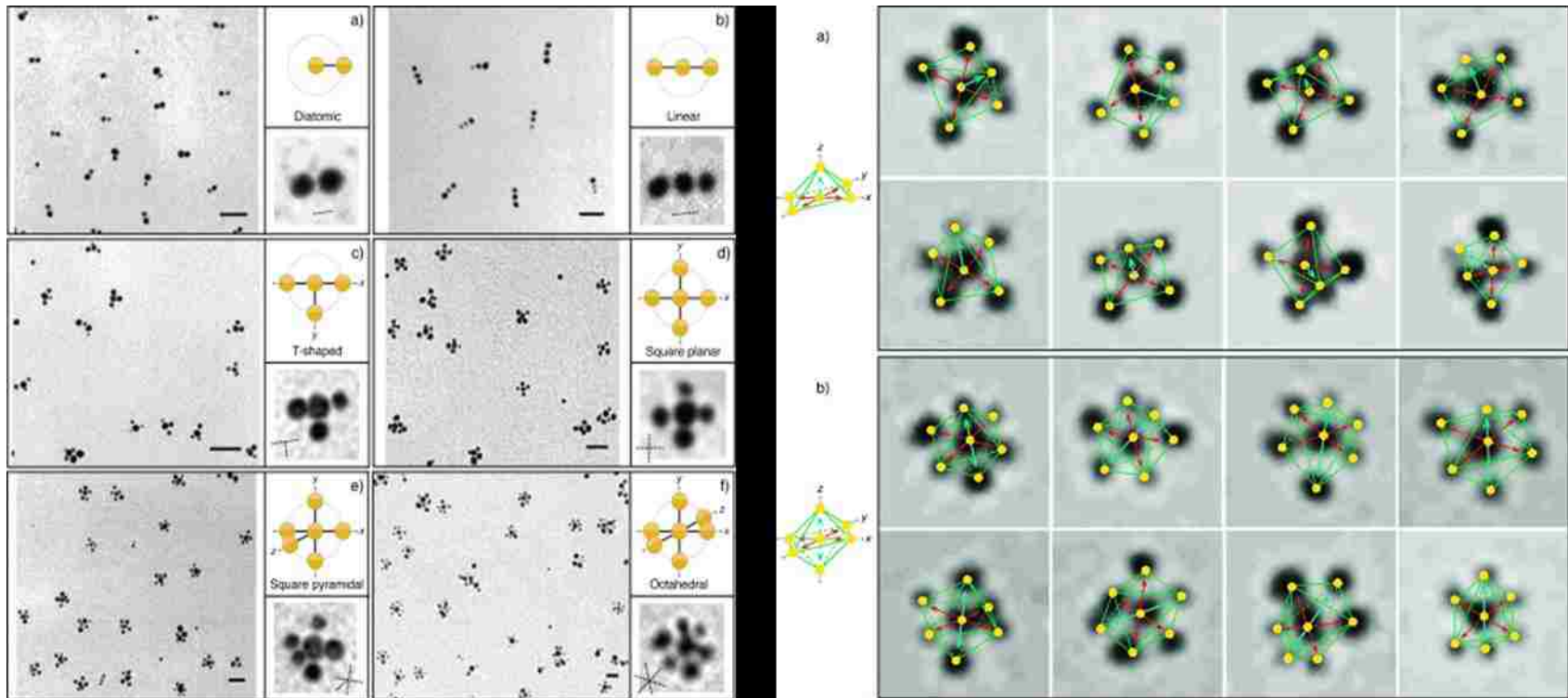
To overcome these limitations, it would be necessary to create building blocks with the anisotropy and predictable geometry needed for the hard-coded self-assembly method. Creating these building blocks requires a different method of functionalization.

A more powerful method of attaching DNA to AuNPs is the anisotropic monofunctionalization “one-at-a-time” technique [32,41] (Fig. 10). This method involves controlling the number, composition, and orientation of each DNA strand that binds to the surface of an AuNP. These building blocks have DNA strands that are oriented symmetrically in one (linear geometry), two (square planar geometry), or three dimensions (octahedral geometry). Each DNA strand is also unique since they are attached one at a time. Thus, the 3-dimensional nBLOCKs appear to have both predictable geometry and anisotropy which would make them ideal for creating hard-coding nanostructures. However, to date, these building blocks have not been used to create 3-dimensional structures due to the related problems of complexity and inefficiency.

The overall efficiency of the anisotropic monofunctionalization method is determined by the combined efficiencies of three cyclic steps. The first step is to bind DNA strands to a solid-support (e.g. magnetic beads). The solid support has specific locations on which the DNA molecules can bind so that the strands are spaced apart. The second step is to bind the AuNPs to the DNA. Because the DNA strands are spaced apart on the surface of the solid support, each AuNP binds to only one DNA strand at a time. The third step is to cleave the DNA-linked AuNPs from the beads and collect them.



**Figure 10. Summary of Anisotropic Monofunctionalization Technique [32]** | DNA is spaced apart on the surface of a solid support. AuNPs are added and the space between the DNA strands prevents more than one of them from binding to each AuNP. The DNA-linked AuNPs are cleaved from the beads and collected. This process can be repeated up to six times to create nBLOCKs with different geometries.



**Figure 11. TEM characterization of nBLOCK geometries (modified from [32])** | (Left) TEM images of the nBLOCK geometries. (Right) 3-dimensional analysis showing the ability to create nBLOCKs with specific geometries.

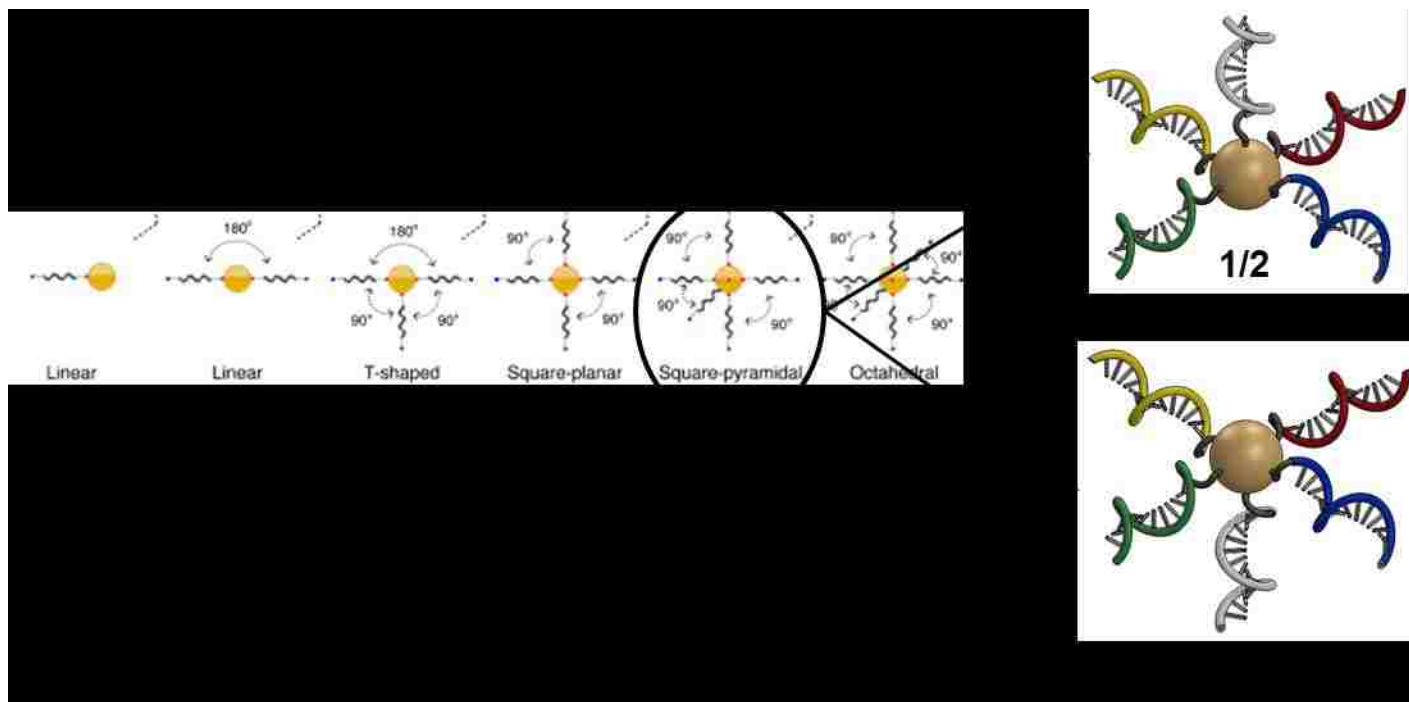
Repeating the monofunctionalization steps a second time binds a second DNA strand to the particle at a different location. Electrostatic repulsion and steric hindrance between the two negatively charged DNA molecules causes the new strand to bind as far away from the first as possible, i.e. on the opposite face of the particle (Fig. 11). An nBLOCK with two DNA strands attached thus has linear geometry. The third DNA strand will attach in between the previous two at  $90^\circ$  angles to both. The fourth DNA strand will bind at  $180^\circ$  from the third strand and at  $90^\circ$  angles from the other two, creating a square planar geometry.

The 5<sup>th</sup> DNA strand can attach on either side of the plane created by the four other DNA strands. Because of this, there is a 50% chance of the fifth and sixth strands attaching in an undesired orientation on the AuNP, which cuts the yield for creating these 3-dimensional nBLOCKs in half (Fig. 12). Creating structures using these nBLOCKs would require isolating the correctly formed nBLOCKs, which would be difficult.

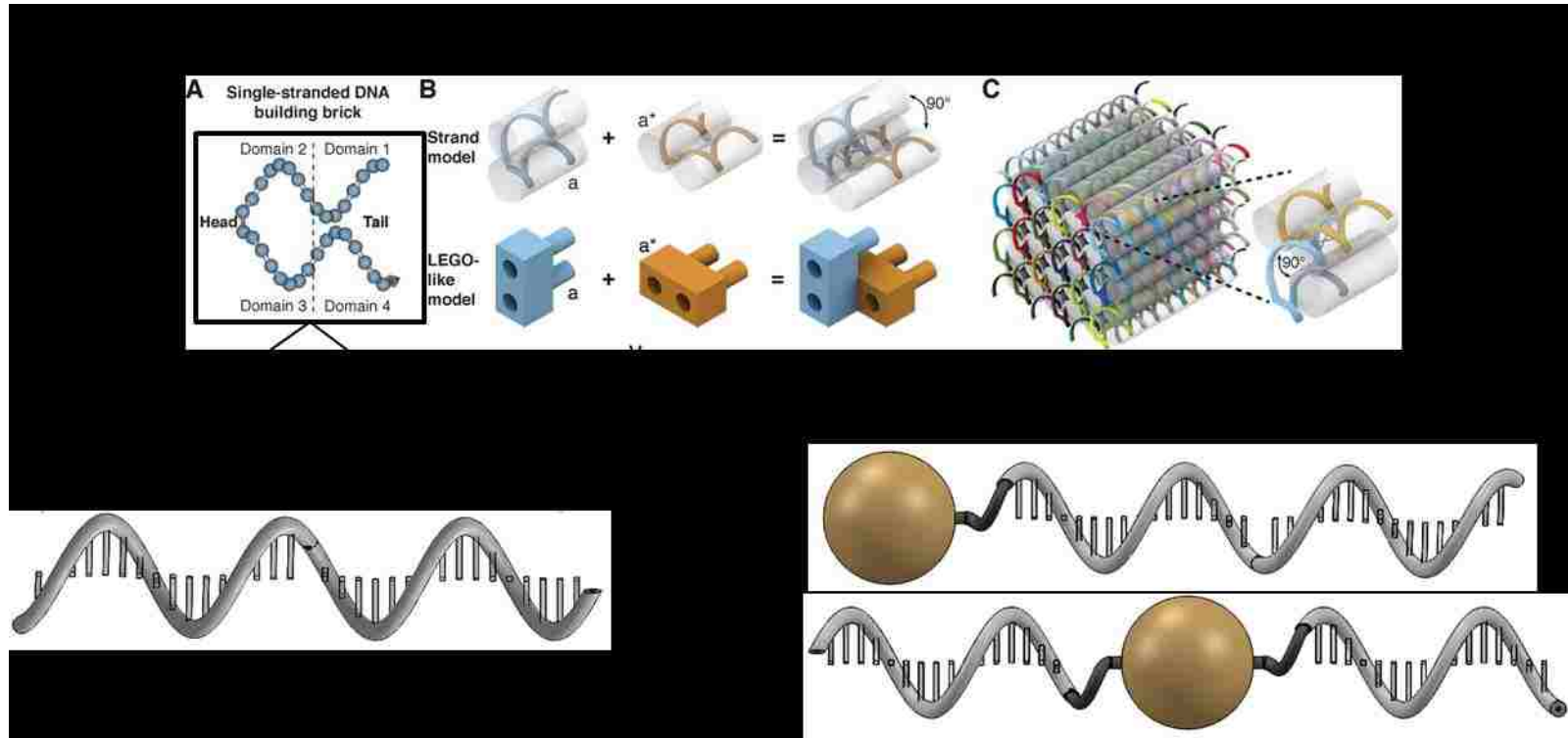
The overall yield of the method also drops with each repetition of the cycle (Fig. 12). Assuming there is a high yield for attaching AuNPs to the DNA (e.g. 95%), each repetition of the steps causes a further drop in yield ( $.95^5$  for 5 steps, or 77%). Thus, creating structures using more complex building blocks results in lower yields.

To summarize, the two primary limitations of using 3-dimensional nBLOCKs to build modular nanostructures are: (1) the yield decreases with each repetition of the monofunctionalization steps and (2) the chirality of 3-dimensionally symmetric nBLOCKs makes it more difficult to create structures without separating out the desired nBLOCKs. To overcome these issues, it would be useful to create structures using one of the simpler building blocks (i.e. linear nBLOCKs).

Accomplishing this task will require a new strategy for designing the interactions between nBLOCKs. DNA Brick-based self-assembly offers a strategy for creating complex nanostructures using linear building blocks (Fig 13). By applying DNA Brick-based self-assembly to nBLOCKs, it would be theoretically possible to create modular nanostructures from the inside out using DNA-linked AuNPs.



**Figure 12. Current limitations of nBLOCKS (modified from [32])** | (Left) The yield decreases with each repetition of the monofunctionalization cycle. Assuming 95% yield for each monofunctionalization step, the overall yield drops by  $.95^N$  for  $N$  steps (e.g.  $.95^5 = 77\%$  for the fifth step). (Right) The fifth step creates an additional drop in yield due to the chirality of the nBLOCKS. The fifth DNA strand can bind on either side of the particle which creates two different possible geometries. Creating structures using these building blocks would require separating out the correct nBLOCKS.



**Figure 13. Adaptation of DNA Brick-based self-assembly for linear nBLOCKs (modified from [23])** | Because of the similarities between the two linear building blocks, it is theoretically possible to adapt DNA Brick-based self-assembly to the self-assembly of linear nBLOCKs. This would allow for the creation of AuNP nanostructures from the inside out with precise control over their final size and shape.



## **1.4 Thesis Objectives**

The goal of this thesis is to reduce the complexity and improve the efficiency creating modular nanostructures from nBLOCKs. In order to accomplish this goal, it will be necessary to complete two objectives.

The first objective is to use computer models to study the self-assembly of DNA Bricks and nBLOCKs. DNA Brick-based self-assembly will be studied in some detail so that its unique features can be applied to the self-assembly of nanostructures from linear nBLOCKs. The information gained will be used to design interactions between linear nBLOCKs so that they self-assemble into nanostructures in a similar manner. The second objective is to improve the three steps of the anisotropic monofunctionalization method so that the linear nBLOCKs required for structure building can be reliably produced.

## **2. Materials and Methods**

### **2.1 Materials.**

Powdered agarose, methanol (MeOH), and hydrochloric acid (HCl) were purchased from EMD Millipore (Billerica, MA, USA). Tetraoctylammonium bromide (TOAB) was purchased from Acros Organics (Geel, Belgium), Tris HCl Buffer from Bio-Rad (Hercules, CA, USA), Tris-Acetate-EDTA (TAE) Buffer from G-Biosciences (St. Louis, MO, USA), and 4-dimethylaminopyridine (DMAP) from GFS Chemicals (Powell, OH, USA). Sodium hydroxide (NaOH), sulfuric acid (H<sub>2</sub>SO<sub>4</sub>), dichloromethane (DCM) and Tween 20 (Tween) were purchased from VWR (Radnor, PA, USA). Ethylenediaminetetraacetic acid (EDTA), sodium borohydride (NaBH<sub>4</sub>), sodium 2-mercaptoethanesulfonate (MESA) and sodium chloride (NaCl) were purchased from Sigma Aldrich (St. Louis, MO, USA). Streptavidin coated magnetic beads (M-280 Dynabeads), SYBR Gold nucleic acid gel stain, and tris(2-carboxyethyl)phosphine-hydrochloric acid (TCEP) were purchased from Thermo Fisher (Waltham, MA, USA). All oligonucleotides were purchased from Integrated DNA Technologies (Coralville, IA, USA.) The following strand sequences were used: Strand 1 – 5' AACGAACCTTCTAGAGTATG – 3'; Strand 1C – 5' CATACTCTAGAAGGTTTCGTT – 3'. These strands were either non-functionalized or functionalized with either a photocleavable biotin or biotin linker on their 5' end and/or a protected thiol group on their 3' end.

### **2.2 Computer Modelling.**

Computer models of DNA strands and nBLOCKs were created in Solidworks. DNA parameters were adapted from the parameters used in caDNAo to design DNA origami structures [18,44]: 10.67 base pairs per 360° helical turn, 3.4 nm horizontal translation per turn, 2 nm diameter. 2.8 nm DMAP-MESA AuNPs were modelled as solid spheres.

### **2.3 DNA-Functionalization of AuNPs.**

DMAP-capped AuNPs were synthesized per established protocols [34]. The MESA concentration in one mg/ml DMAP-capped AuNP solution was raised to 105 µM MESA.

To bind DNA to the solid support, 10  $\mu$ l of streptavidin-coated magnetic beads (.01 mg) were washed twice with 1.5 ml of DNA Binding Buffer containing Tween (5 mM Tris HCl, 1 M NaCl, .05% Tween, pH 7.4) and once with 1.5 ml DNA Binding Buffer (5 mM Tris HCl, 1 M NaCl, pH 7.4) with five minutes of gentle mixing during each washing step. Two hundred  $\mu$ l of 500 nM oligonucleotide in DNA Binding Buffer was added to the beads and left to incubate for 1 hr.

To bind AuNPs, the resulting DNA-conjugated beads were washed twice with 1.5 ml of 1x TE Buffer containing Tween (10 mM Tris HCl, 1 mM EDTA, .05% Tween, pH 7.4) and once with 1x TE Buffer (10 mM Tris HCl, 1 mM EDTA, pH 7.4) under nitrogen. One ml of 3.5  $\mu$ M TCEP in 1x TE Buffer was added and the mixture was incubated at 37°C. After 1 hr of incubation, the TCEP solution was replaced with fresh 3.5  $\mu$ M TCEP solution under nitrogen and allowed to incubate for an additional 1 hr at the same temperature. After TCEP reduction, the beads were washed twice with 1.5 ml ddH<sub>2</sub>O containing Tween (.05% Tween) and twice with ddH<sub>2</sub>O under nitrogen. 450  $\mu$ l of .2  $\mu$ M DMAP-MESA AuNPs in ddH<sub>2</sub>O were added and the solution was left to incubate overnight.

To cleave the DNA-bound AuNPs, the AuNP-bound-beads were washed 2x more in 1.5 ml of ddH<sub>2</sub>O containing Tween, 2x in 1.5 ml of ddH<sub>2</sub>O and finally resuspended in 200  $\mu$ l of ddH<sub>2</sub>O. The sample was irradiated with UV light (Luxo UV-Inspection Lamp) for 15 minutes. After magnetic separation, the supernatants of the samples were collected and characterized.

## 2.4 Characterization

UV-Visible absorbance spectra were collected using a Thermo Fisher Nanodrop 2000C. Extinction coefficients obtained from Integrated DNA Technologies were used to calculate DNA concentrations using the  $A_{260}$  value. The diameter of the AuNPs was calculated using the  $A_{peak}/A_{450}$  ratio [42]. This value was used to calculate the extinction coefficient using the logarithmic relationship between diameter and  $\epsilon$  [35]. For data analysis, the concentration drops in the supernatants after DNA binding or AuNP binding were compared against control samples (non-functionalized DNA for DNA Binding and non-TCEP-activated DNA strands for AuNP binding).

### 3. Results and Discussion

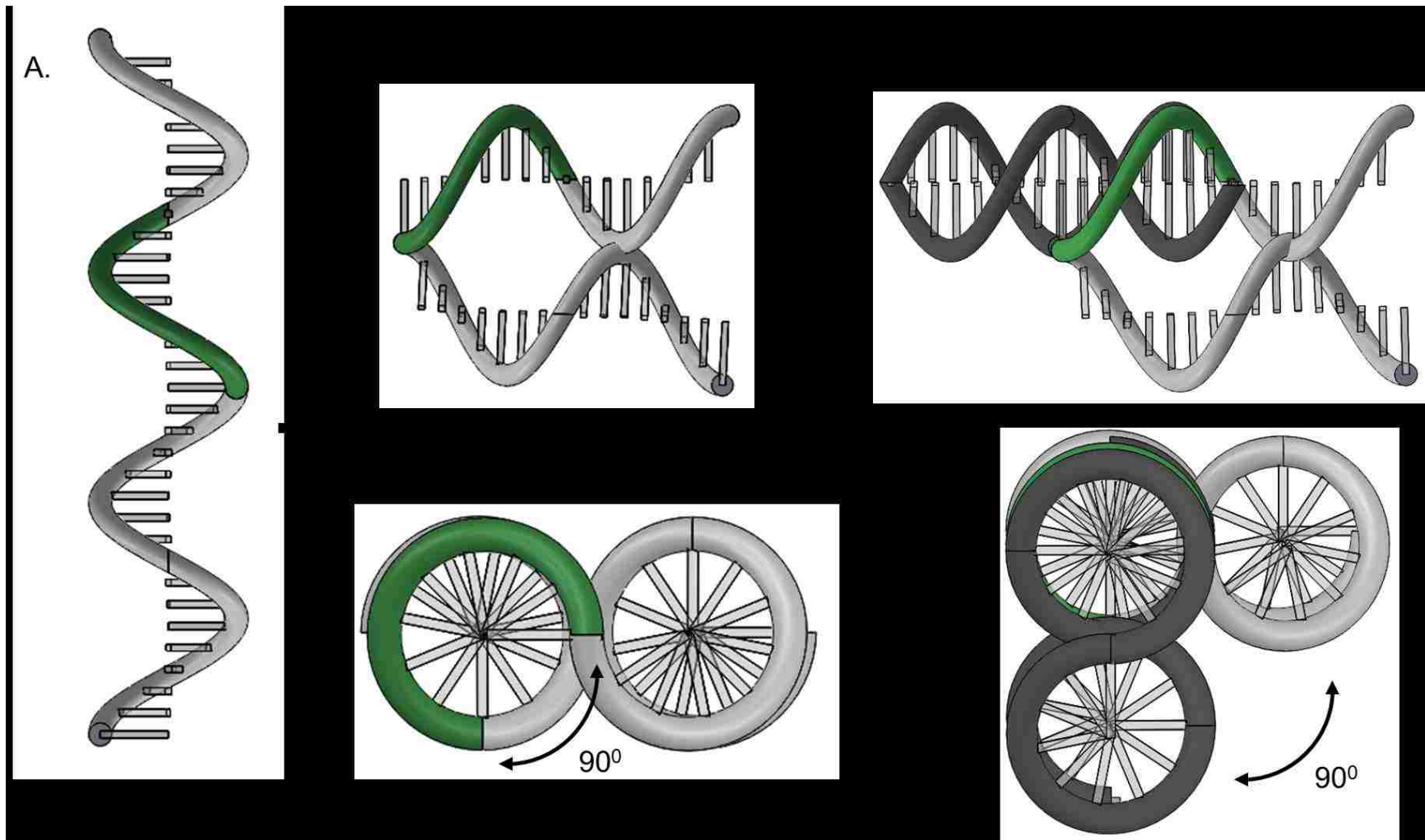
The results and discussion are divided into two sections. The first section deals with modelling DNA Bricks and nBLOCKs and their modes of self-assembly. The third section details an attempt to improve the yield of the anisotropic monofunctionalization method so that the nBLOCKs required to build structures could be produced.

#### 3.1 Modelling the Self-Assembly of DNA Bricks and nBLOCKs

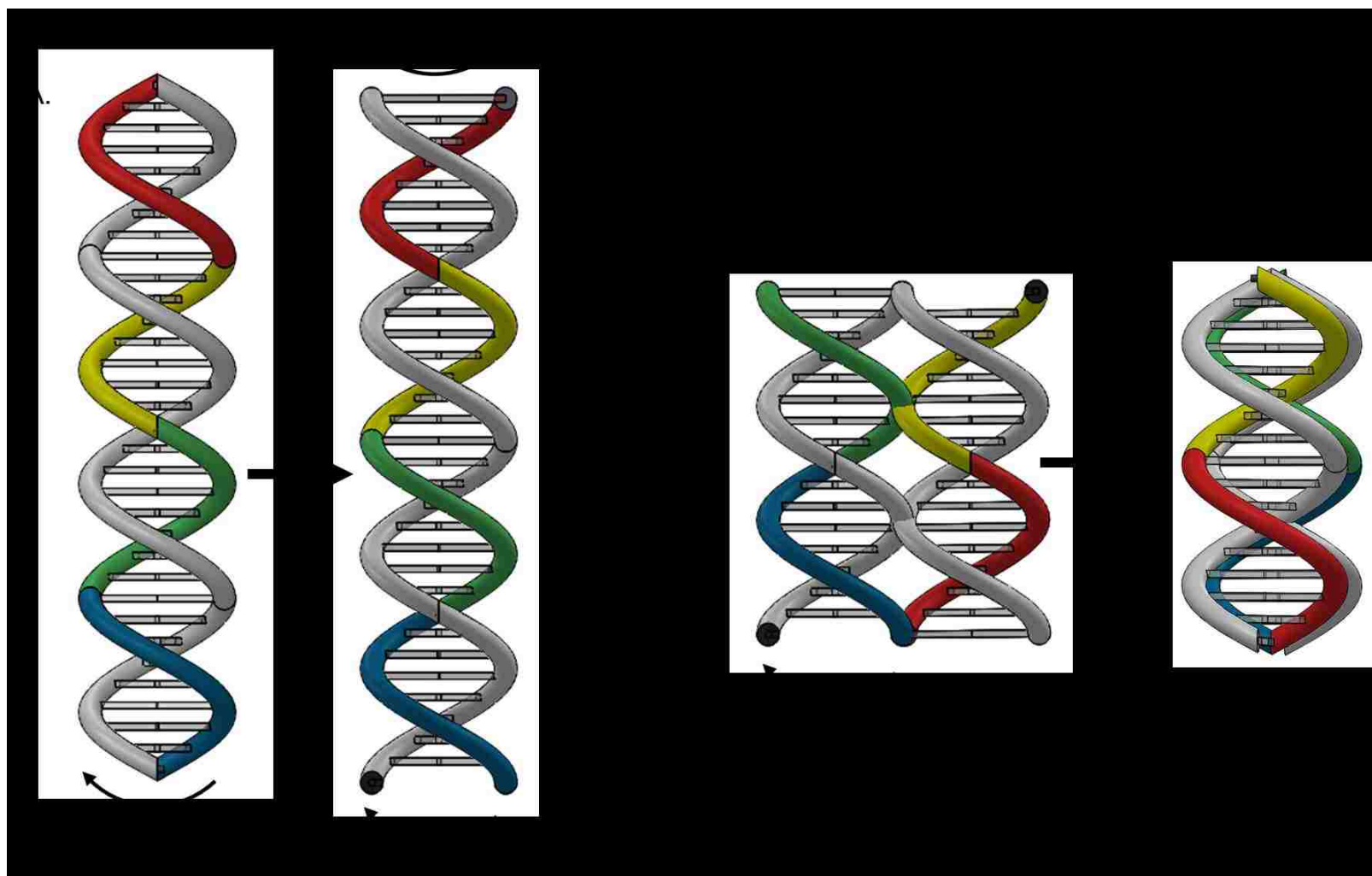
As mentioned previously, there are two properties of DNA Bricks that make their method of self-assembly powerful: anisotropy and predictable geometry. Anisotropy means that every DNA Brick has four unique binding domains and thus has a specific position and orientation in a structure. The size and shape of nanostructures can thus be controlled by adding or removing DNA Bricks in specific locations.

Creating structures also requires knowing how the DNA Bricks interact through each of their 4 binding partners. DNA Bricks have predictable geometry due to the combination of DNA's helical turning and the rigidity created by base pairing interactions. As mentioned earlier, each DNA Brick is formed by bending a 32bp long ssDNA strand in between the middle 16<sup>th</sup> and 17<sup>th</sup> bps (Fig. 14). This fold creates a parallel helix geometry. Each DNA Brick also has four, 8 bp long binding domains. Each 8 bp binding domain causes an exact 270° of the helix (Fig. 14B). When a DNA Brick binds to one of its binding partners, the relative orientations of their parallel helices are offset by 90° (Fig. 14C). This parallel helix geometry is stable because the dsDNA regions of overlapping binding partners are rigid. Take, for example, a linear DNA Brick bound to four 8bp long binding partners (Fig. 15). When the Brick rotates, all of its binding partners rotate with it.

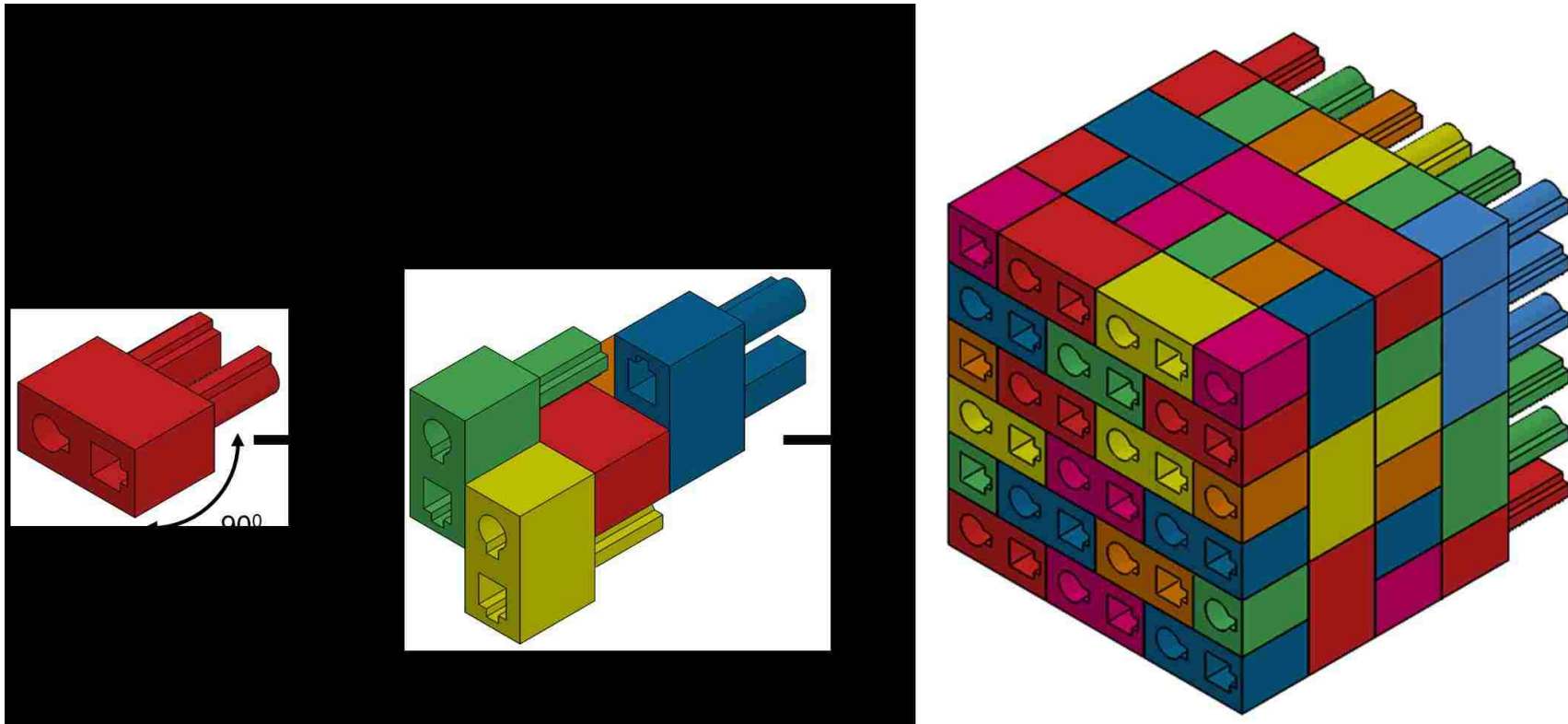
These two properties together allowed researchers to create an abstract model to show how DNA Bricks form structures (Fig. 16A) [23]. Each abstract DNA Brick has four binding domains. The front binding domains are offset from the back ones by a 90° angle (Fig. 16B). When the front of one DNA Brick binds into the back of another, the two Bricks become offset by 90°. When hundreds of DNA Bricks come together, they do so according to this simple abstract model (Fig. 16C).



**Figure 14. Formation of DNA Bricks from Linear ssDNA | A.** DNA Bricks are 32 bp long ssDNA strands with 4 unique 8 bp binding domains (one of four shown in green). **B.** The ssDNA backbone folds back on itself when it forms bonds with other DNA Bricks. Each 8 bp binding domain is equivalent to a  $270^\circ$  turn of the helix. The Brick viewed from the top (Top) and back (Bottom). **C.** The  $270^\circ$  turns orient each of the four binding partners (one shown in dark grey) at  $90^\circ$  angles from the original. This helps create predictable geometry, which is a requirement for DNA Brick-based self-assembly.



**Figure 15. Rigidity of dsDNA | A.** A single 32 ssDNA strand bound to four other 8bp strands. Rotation on one end of the molecule causes rotation at the other. **B.** The dsDNA molecule formed when four Bricks come together is also rigid. Rotational movement of one Brick gets transferred to all four of its binding partners. This rigidity helps orient all of the DNA Bricks when they come together in an assembly.



**Figure 16. Abstract model of DNA Bricks (adapted from [23])** | **A.** Each DNA Brick can be abstractly modelled as a brick with four binding domains (2 pegs and 2 holes). Each peg is oriented at  $90^\circ$  angles from their complementary holes on the other side. **B.** Self-assembly of a small number of DNA Bricks. Each abstract Brick is oriented at  $90^\circ$  from all of its binding partners. **C.** DNA Bricks can be stacked on top of one another in a predictable manner to create well-defined structures.

Designing structures using the abstract model is relatively simple. The only source of complexity comes from designing the base pairing sequences of the hundreds to thousands of DNA Bricks required to make a structure. To help in this task, there are computer aided design tools that can be used to create the necessary strand sequences (e.g. 3DNA [43]).

To summarize, DNA Brick-based self-assembly requires building blocks that are anisotropic and have predictable interactions. If these two conditions are met, designing and building structures is made simple by software programs that design the sequences of all the necessary DNA strands.

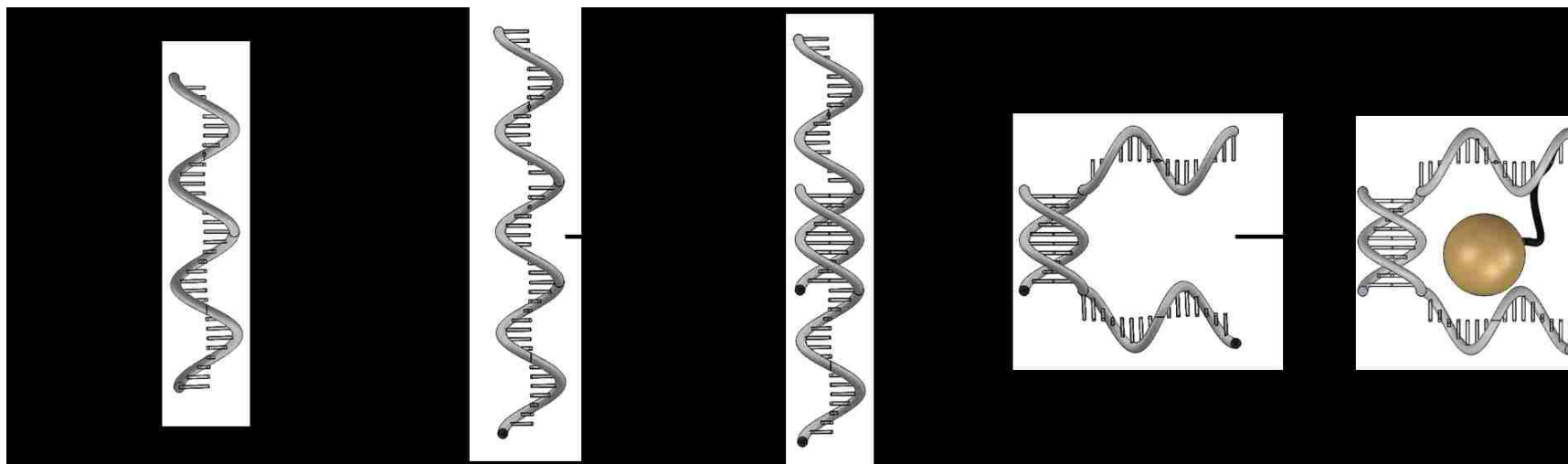
Creating structures from nanoparticles would require DNA-linked nanoparticles that are anisotropic and interact predictably. An nBLOCK can be created with multiple unique DNA strands attached at specific locations on the surface of the AuNP using the anisotropic monofunctionalization technique. Thus, the condition for anisotropy is met due to the capabilities of the method used to create nBLOCKs.

Linear nBLOCKs are also geometrically similar to DNA Bricks. Because of this, it was hypothesized that 32 bp long ssDNA strands attached to AuNPs would be capable of DNA Brick-based self-assembly. However, this hypothesis was proven wrong since DNA Brick structures are packed together too tightly to incorporate nanoparticles (Fig. 16).

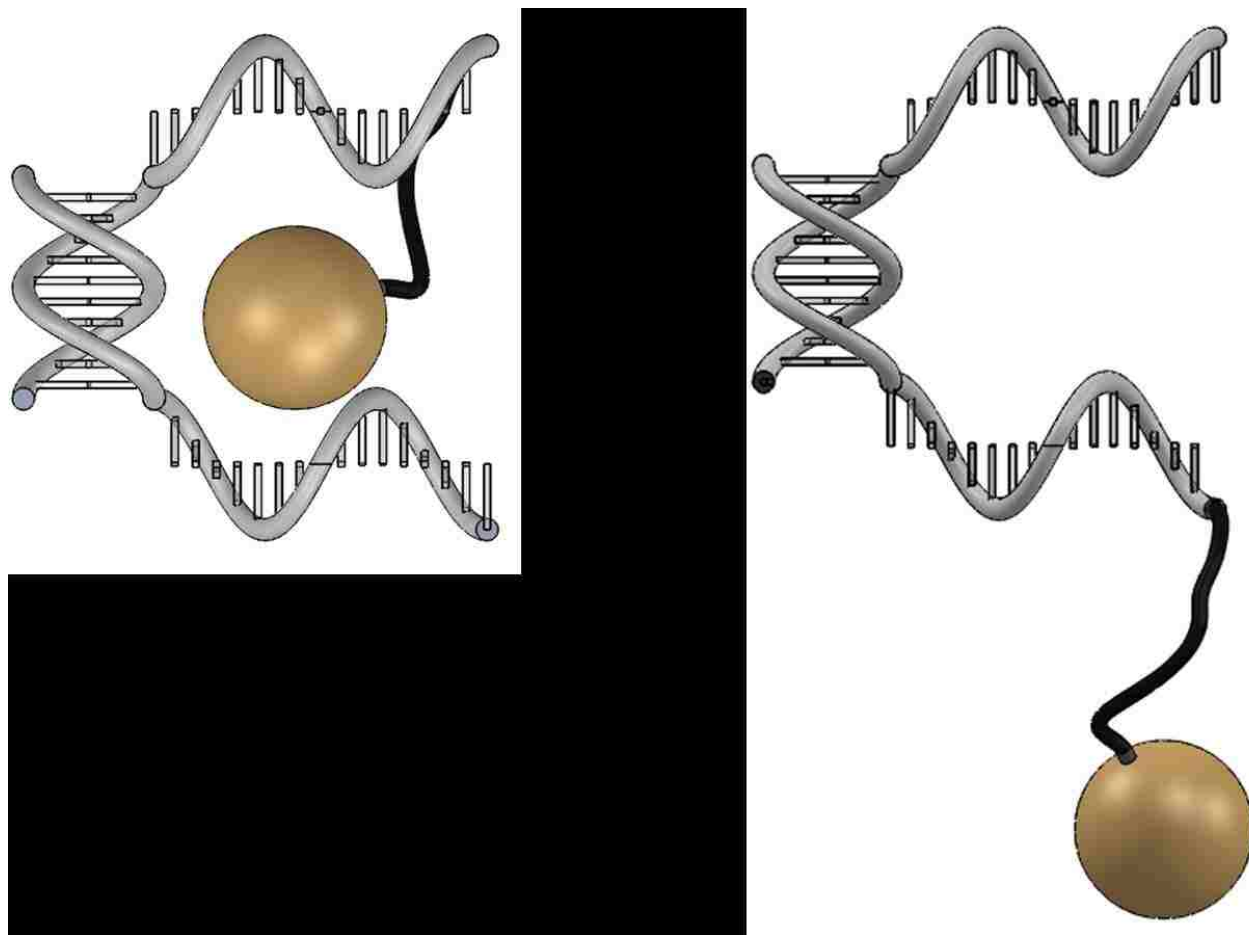
However, a slight modification to the DNA Brick geometry allows for the incorporation of nanoparticles (Fig. 17). By extending the 32 bp ssDNA strand to 43 bp and creating a rigid 11bp dsDNA region in the middle of the molecule, it is possible to extend the distance between the parallel helices by ~3.5 nm. The nanoparticles would fit in the created cavities since they are only 2.8 nm in diameter. This 11bp dsDNA region causes a ~360° turn of the helix (371° exactly) and thus would not influence the parallel helix geometry. Because the parallel helix geometry is maintained, these building blocks would still interact like DNA Bricks.

However, there is a flexible carbon linker that attaches the AuNPs to the DNA strands. This carbon linker would allow free movement of the AuNPs in the area surrounding its point of attachment. This extra movement could negatively influence the stability of the structure. It might also make it more difficult to accurately predict the shape that the AuNP nanostructures would form.





**Figure 17. Modified DNA Bricks with cavity for nanoparticles** | The modified Bricks are created by extending a 32 bp long DNA Brick by 11 bp. The middle 11bp becomes double-stranded and gives some rigidity to the created cavity. The 11bp region causes a  $\sim 360^\circ$  turn of the helix so that when the ends of the DNA strand fold, they create a parallel helix geometry nearly identical to that of unmodified DNA Bricks. These modified DNA Bricks create enough space for a nanoparticle to reside in the center.

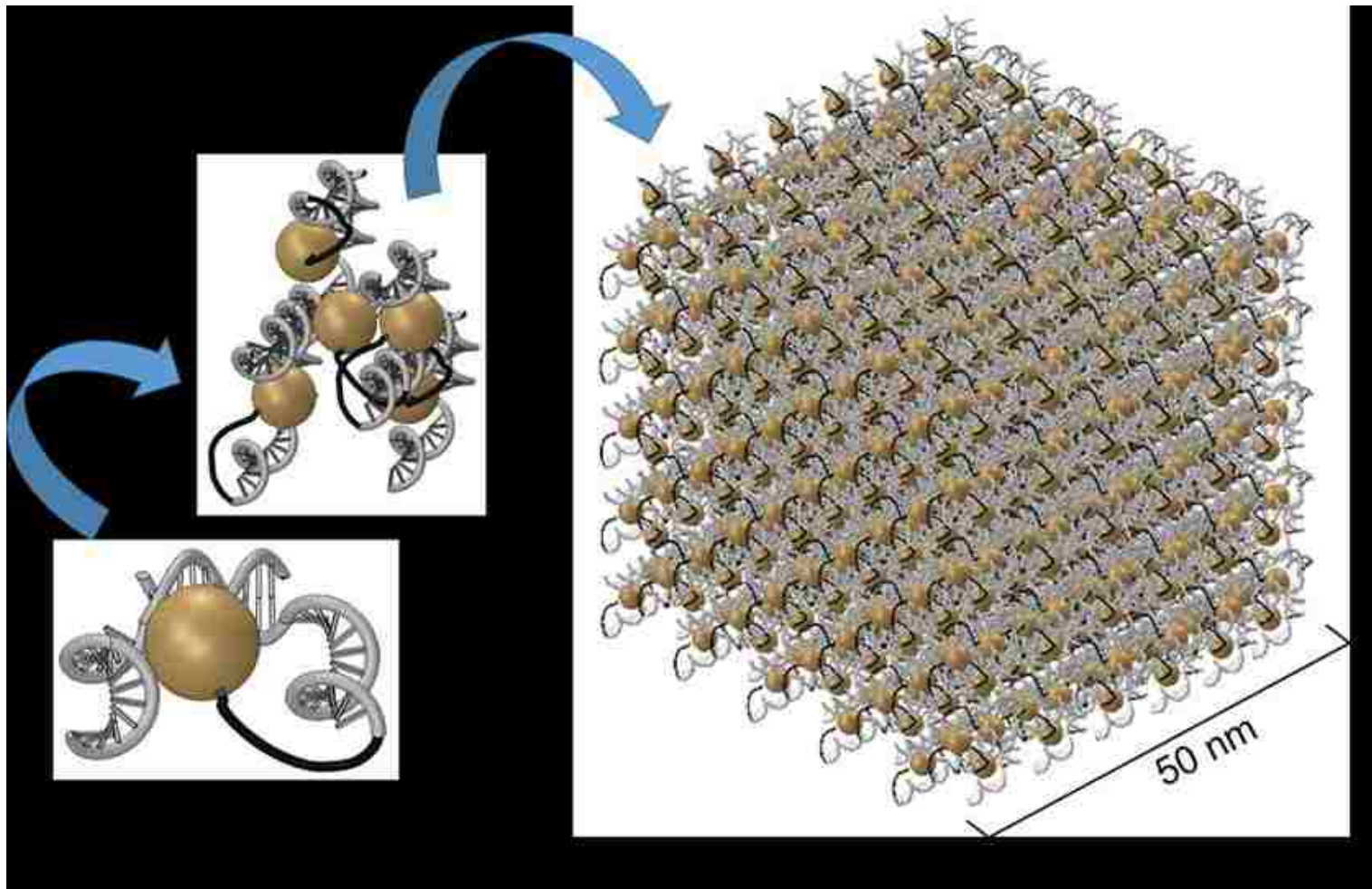


**Figure 18. Different possible geometries of nBLOCKs with one DNA strand attached** | The flexible carbon linker binding the AuNP to the DNA strand allows free movement. Tuning the length of this linker would help stabilize certain geometries.

Nevertheless, according to these models, nBLOCKs with 43bp DNA strands attached could be used to create nanostructures (Fig. 19). There are three assumptions on which this conclusion is based. The first is that the  $371^\circ$  turn created by the 11bp dsDNA region does not change the relative orientations of the parallel helices. According to the caDNAo software used to design DNA origami structures, there is  $\pm 12^\circ$  of angle forgiveness in helical turns [18,44]. This means that any helical turns between the angles of  $348^\circ$  and  $372^\circ$  will have properties nearly identical to those of an exact  $360^\circ$  turn. The  $371^\circ$  turn used in this model is within this zone of forgiveness and thus will create a geometry similar to that of a DNA Brick without the extra 11bp backbone.

The second assumption is that the AuNPs will occupy the spaces left by the 11bp backbones without any problems. The carbon linker between the AuNPs and DNA strands is flexible which would allow the AuNPs to move freely in the vicinity of their attached DNA strand. This freedom of movement would be emphasized at the boundaries where there are no DNA backbones to provide specific cavities. One way to overcome this problem at the boundaries is to replace the nBLOCKs with 43 bp ssDNA strands without any AuNPs attached. The length of the carbon linker could be also changed to give the AuNPs more or less freedom of movement throughout the rest of the structure. The size and shape of the AuNPs could also be modified to tune the freedom of movement. The structures' tolerances to movement of the AuNPs will be determined by applications of interest.

One final assumption is that the electrostatic repulsion between the AuNPs' negatively charged capping ligands and the phosphate backbones of the DNA strands will not prevent structure building. Electrostatic repulsion could make the structures less stable than DNA Brick structures. To overcome this, it would be possible to use stronger bonds between the DNA strands by increasing the number of overlapping base pairs. For example, 19 bp binding domains could be used instead of 8 bp binding domains since they both cause  $\sim 270^\circ$  turns of the helix. Additionally, lowering the concentration of capping ligand on the surface of the particles could decrease the electrostatic repulsion between the AuNPs and DNA strands if it became a problem. Tuning these variables and creating the optimal conditions for creating structures using nBLOCKs with one DNA strand attached will be the subject of future experiments.



**Figure 19. Creation of nanostructures from the inside-out using linear nBLOCKs with one DNA strand attached |** By creating strand sequences that self-assemble in a manner identical to DNA Bricks, it would be possible to create the desired structures from linear nBLOCKs.

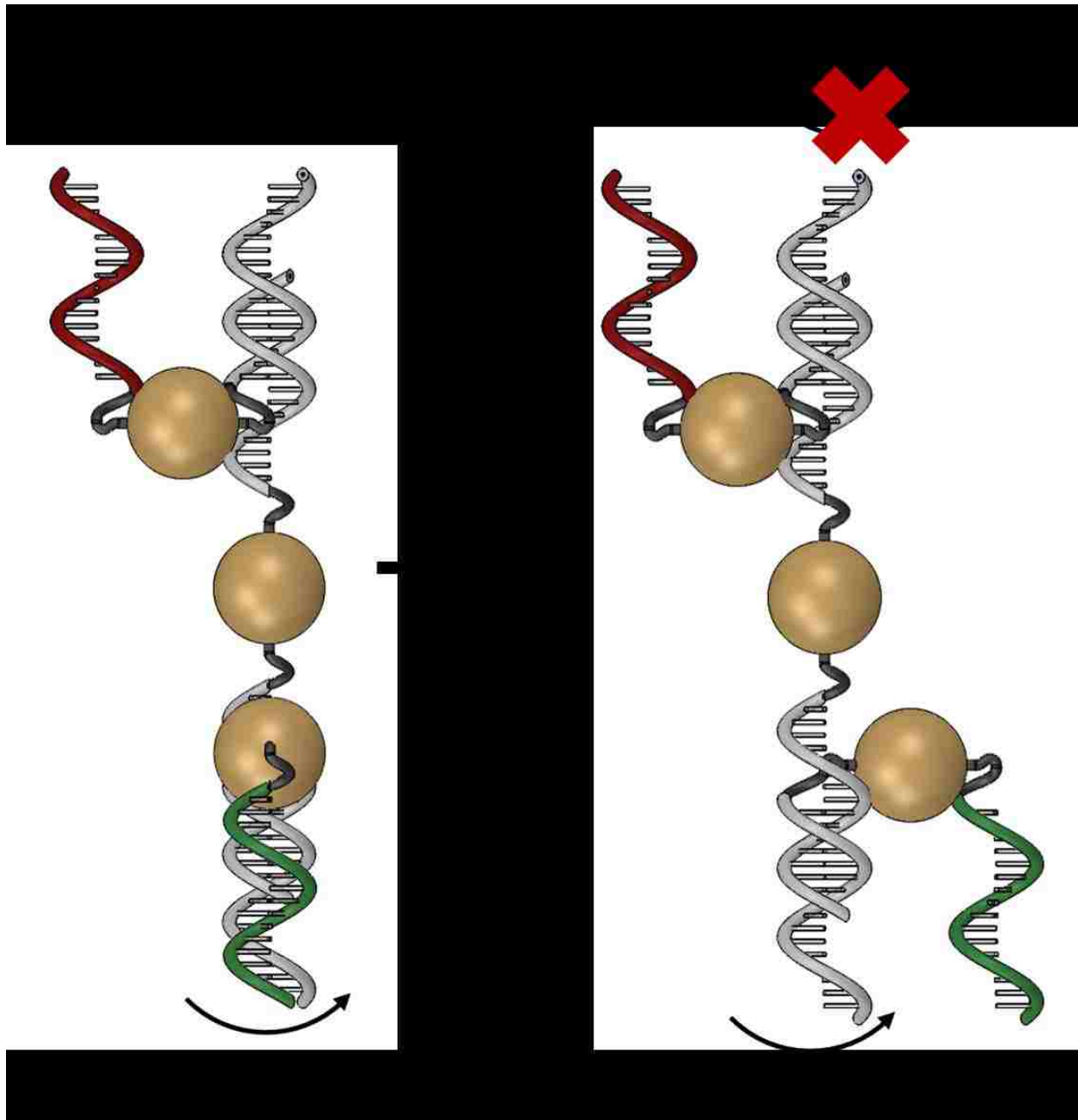
There is another nBLOCK geometry that could theoretically be used for DNA Brick-based self-assembly. It is possible to attach two 16 bp ssDNA strands on either side of an AuNP using the anisotropic monofunctionalization technique. Because the AuNPs are in the middle of two halves of a DNA Brick, it would be unnecessary to change the DNA Brick geometry to create space for the nanoparticles. The positions of the nanoparticles would also be better defined since there are two points of attachment to the DNA structure rather than one. Thus, this geometry of nBLOCK could provide several key benefits for creating structures compared to nBLOCKs with only one DNA strand attached.

However, having two points of attachment to the DNA Brick also creates a problem. As mentioned previously, predictable geometry is a key requirement for DNA Brick-based self-assembly. The rigidity of dsDNA ensured that all of the DNA Bricks became oriented and positioned in specific locations when they came together. This was possible because rotational movement of one DNA Brick became transferred to all of its binding partners.

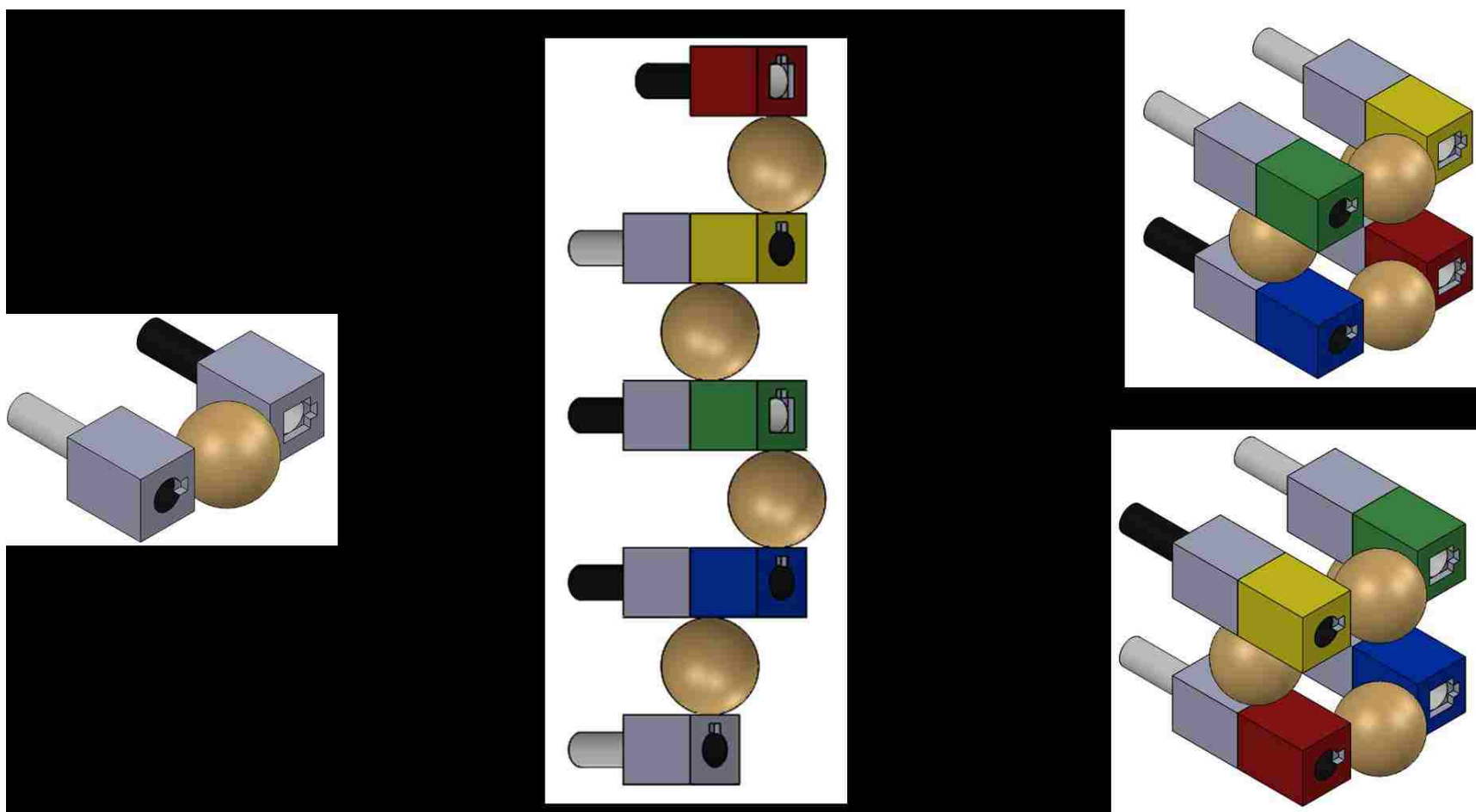
The same cannot be said for nBLOCKs with two DNA strands attached (Fig. 20). The AuNPs are attached in the middle of the two halves of a DNA Brick through carbon linkers. These linkers are made up of  $\sigma$  bonds that all allow free, independent rotation of the DNA molecules. Because of this, nBLOCKs and their binding partners are also free to rotate independently.

An abstract model of these nBLOCKs were created to reflect their differences with DNA Bricks (Fig. 21). The abstract nBLOCKs have four binding domains like DNA Bricks. However, these binding domains are smooth rather than rigidly shaped. Additionally, there is no  $90^\circ$  offset between the binding domains since the added flexibility makes it impossible to control their relative orientations.

A structure consisting of four nBLOCKs was created using this model. Because of the added flexibility, there is no way to accurately predict the structure that a group of these nBLOCKs will form (Fig. 21C-D). There would be mixtures of structure-types, each slightly different from the rest. It might be possible to separate out the correct structures from incorrect ones. However, this separation step would add an additional level of complexity to building structures.



**Figure 20. nBLOCKs with two DNA strands attached are not as rigid as dsDNA |** When a rotational force is applied to one end of the structure (green), that rotational movement is not transferred to the nBLOCKs on the other side (red). This makes it difficult to predict the relative geometries of the nBLOCKs and their binding partners.



**Figure 21. Creation of nanostructures from linear nBLOCKs with two DNA strands attached** | **A.** Abstract model of nBLOCK with two DNA strands attached. Each of the four binding domains are smooth rather than rigid. There is no control over the relative orientation of the building blocks. **B.** Linear chain of nBLOCKs. The front of the top red binding domain (black peg) is complementary to the back of the bottom grey binding domain (black hole). When these binding domains interact, they can drive the formation of two different structures. **C.** The order of the binding domains starting from red (R) is RYGB in the counter clockwise direction. **D.** The same building blocks forming a different structure with the binding domains ordered RYGB in the clockwise direction.

Because of these complications, it would be worthwhile to focus on using nBLOCKs with one DNA strand attached to create structures. These nBLOCKs form structures more reliably than those with two DNA strands attached. In fact, these nBLOCKs are extremely promising candidates for self-assembling AuNP nanostructures from the inside out with precise control over their final size and shape. Nevertheless, it will be necessary to test the key assumptions using a more detailed modelling software in the future. These tests will be discussed in the Future Works section.



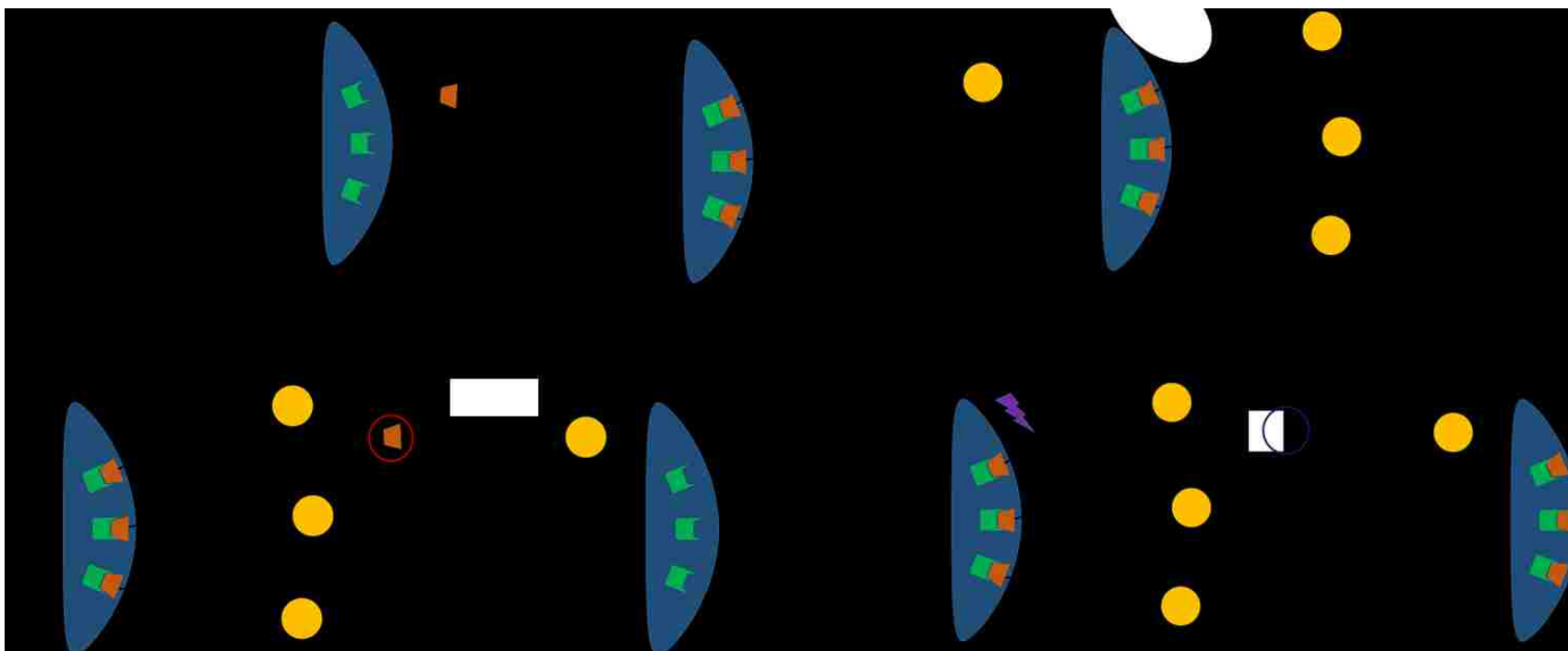
### 3.2 Improving the Efficiency of the Anisotropic Monofunctionalization Technique

The anisotropic monofunctionalization technique [32] is capable of producing nBLOCKs that are capable of DNA Brick-based self-assembly. However, recent experiments have shown that the method suffers from low yield. Thus, improving the method's efficiency is a key step towards realizing the goal of building structures from nBLOCKs.

The anisotropic monofunctionalization strategy involves three steps (Fig. 22). The first step is to bind DNA to a solid support. The solid support spaces out the DNA strands so that when AuNPs bind, only one DNA strand can bind to each particle. The second step is to bind AuNPs to the thiol groups on the DNA strands. The third step is to cleave the linkage between the DNA strands and solid support and collect the nBLOCKs.

The original protocol used an electrostatic interaction between a positively charged amine group bound to the oligonucleotides and negatively charged carboxyl groups bound to the solid support to carry out the first step [32]. This strategy resulted in a large amount of nonspecific binding between the DNA strands and solid support which lowered the yield of the method. Additionally, the DNA strands on the nanoparticles were still attached to the amine group after the cleavage step (Fig. 22B). There are two limitations to leaving the amine group attached. The first is that a positively charged molecule could interfere with the close-packing of DNA strands during DNA Brick-based self-assembly. The second limitation is that certain downstream molecular biology applications, e.g. polymerase chain reaction or DNA ligation, require an unmodified DNA end.

Overcoming the limitations of the original protocol would require a new linkage reaction. The photocleavable (PC) biotin-streptavidin interaction is one potential alternative. Biotin and streptavidin have an extremely high affinity for one another with a dissociation constant of  $4 \times 10^{-14}$  M [36]. It is one of the most specific and reliable reactions currently known. A photocleavable nitrobenzene linker ensures that the DNA strands are left with an unmodified end after cleavage (Fig. 22C). Based on this information, it was hypothesized that the PC biotin-streptavidin interaction could be used to overcome the limitations of the anisotropic monofunctionalization method.



**Figure 22. The anisotropic monofunctionalization technique | A.** The first step of the method involves binding DNA strands to a solid support through a reversible reaction. The second step is to bind AuNPs to the beads through a thiol linker on the DNA strands. The third step is to reverse the binding reaction and collect the DNA-linked AuNPs. **B.** The original protocol left an amine group on attached to the DNA strands. **C.** A photocleavage reaction would leave an unmodified end for downstream molecular biology applications.

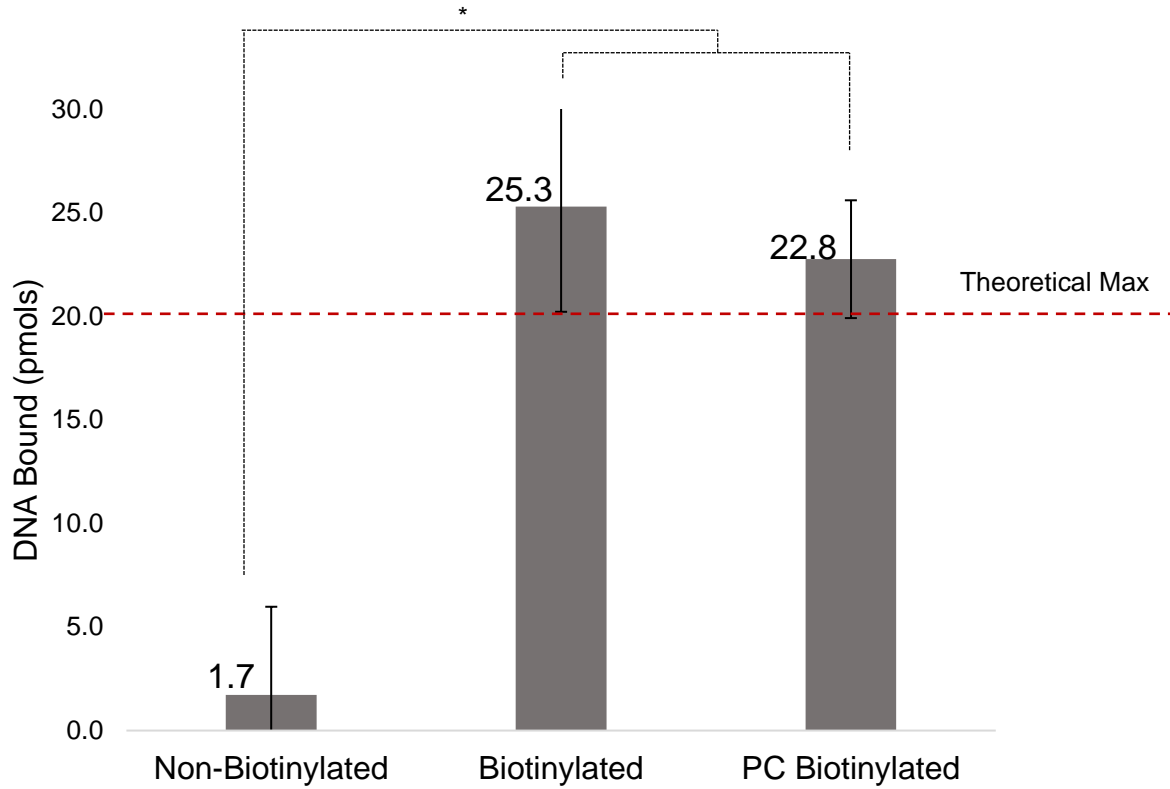
To test this hypothesis, DNA strands with a PC biotin modification were added to streptavidin coated magnetic beads. The absorbance of the DNA solutions at 260nm were measured before and after binding and the extinction coefficient provided by the manufacturer was used to calculate the amount of DNA bound to the beads (Fig. 23). The 25.3 and 22.0 pmols of DNA binding for biotinylated and PC biotinylated oligonucleotides respectively were slightly higher than the amount predicted by the manufacturer. However, both values were within a single standard deviation. On the other hand, only 1.7 pmols of the non-biotinylated DNA strands bound to the beads. These results suggested that the PC biotin-streptavidin interaction is much more specific and reliable than the electrostatic interaction.

The second step of the anisotropic monofunctionalization method is to bind AuNPs to the DNA strands on the beads. The DNA strands were purchased with a dithiol protecting group on their 3' ends (Fig. 24). This protecting group had to be cleaved using a reducing agent, e.g. TCEP, before the DNA could bind to the AuNPs. It is important to note here that oxygen can convert the free thiols back to dithiols if it is present (Fig. 24).

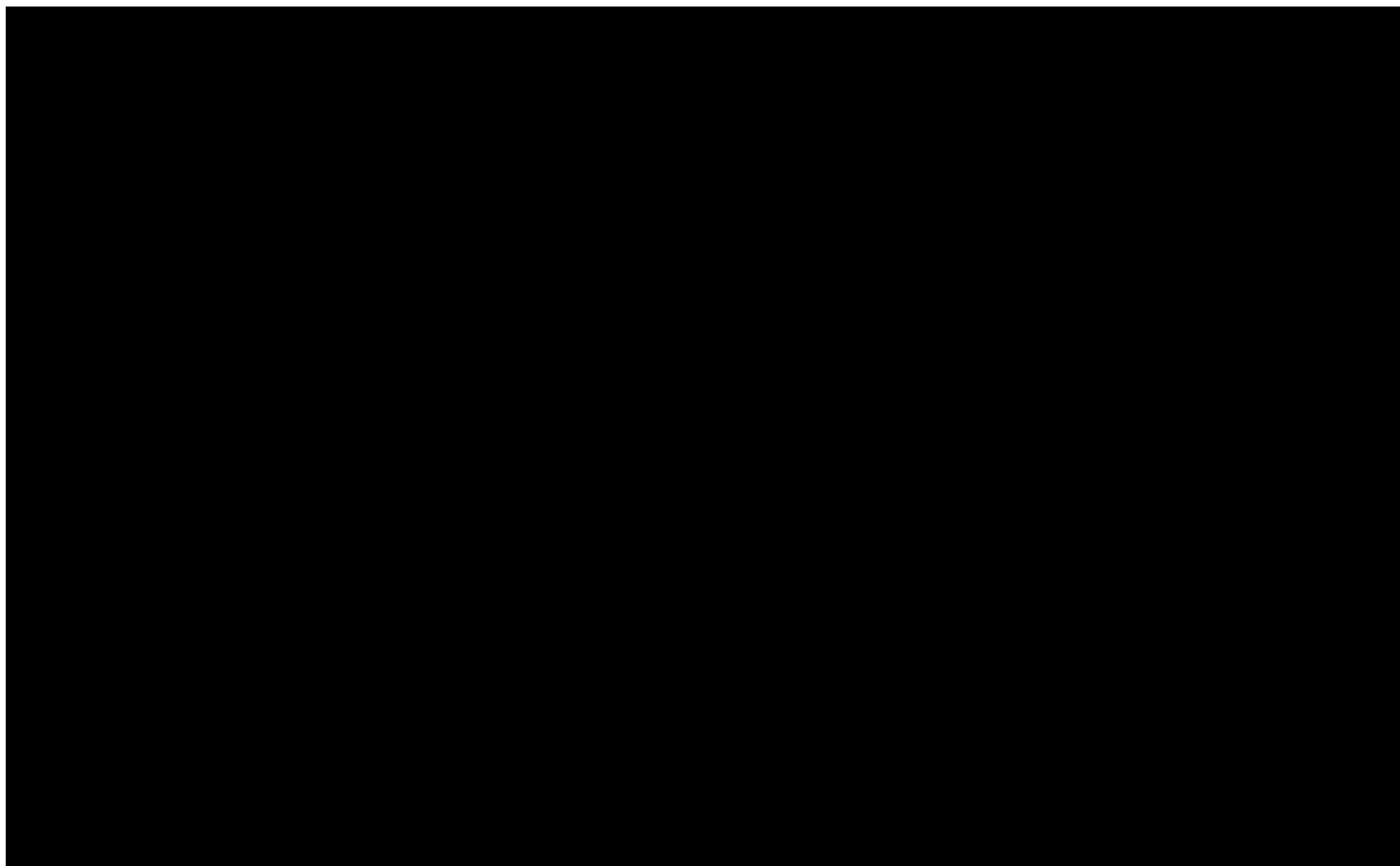
After deprotecting the DNA strands, AuNPs were added and the mixture was incubated overnight. After 12 hours of incubation, only 3% of the DNA strands became attached to AuNPs (Fig. 26). Two primary factors were identified as likely causes of this low efficiency. The first was electrostatic repulsion between the DNA strands and AuNPs. The AuNPs have a negatively charged capping ligand (MESA) and the DNA strands have negatively charged phosphate backbones. Electrostatic repulsion between the like-charges could have prevented the AuNPs from getting close enough to the DNA strands to attach (Fig. 25). The second factor was that the thiol linkages could have reformed into disulfides after reduction due to the presence of oxygen (Fig. 24)

Numerous experiments were carried out to test these various factors and their influence on the efficiency of ligand exchange (Fig. 26). It is important to note here that there were not enough repetitions of each experiment to prove statistical significance. Conditions that improved the yield were accepted and implemented but the experiments themselves were not repeated so that ligand exchange could be improved further.

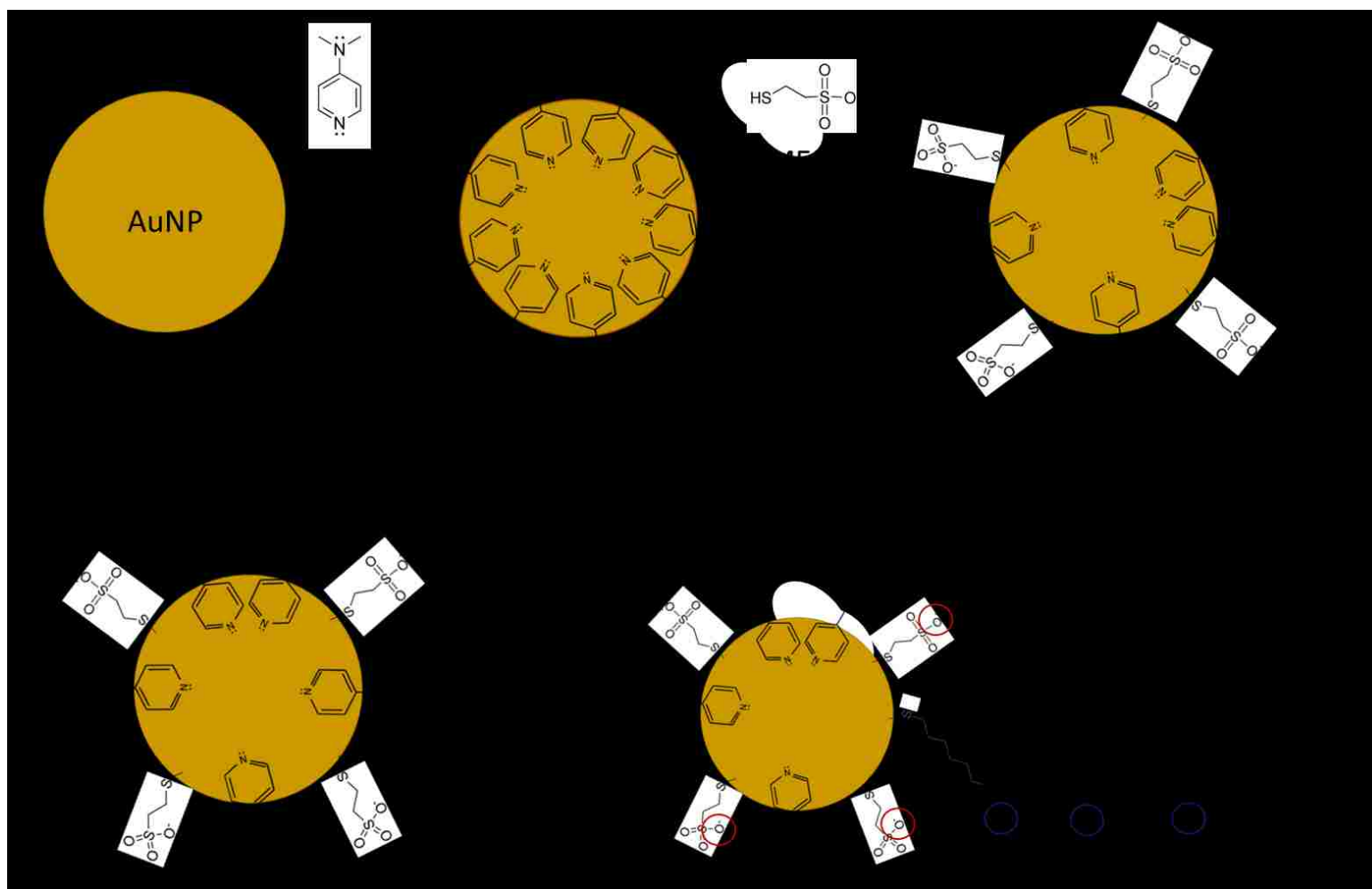
# DNA Binding to Beads



**Figure 23. DNA binding to streptavidin coated magnetic beads** | Concentration decreases were calculated using the absorbance value at 260 nm and an extinction coefficient provided by the manufacturer. Theoretical maximum was ~ 20 pmols as provided by the manufacturer of streptavidin-coated magnetic beads. Biotinylated and PC-Biotinylated DNA strands became bound within one standard deviation of the theoretical max. Only 1.7 pmols of non-biotinylated DNA strands became bound. There was a statistically significant difference between biotinylated and non-biotinylated DNA strands as well as between PC biotinylated and non-biotinylated DNA strands (indicated by \*), but not between biotinylated and PC biotinylated DNA strands.

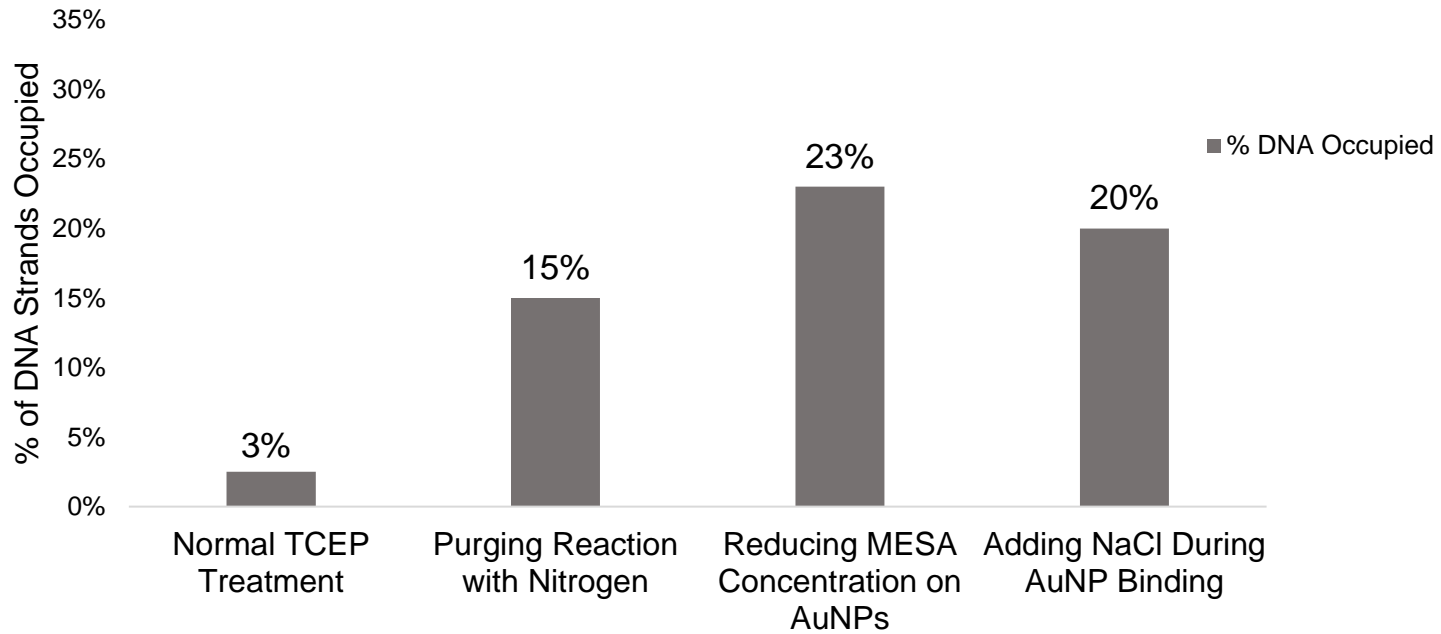


**Figure 24. TCEP reduction of protected oligonucleotides (modified from [43])** | Oligonucleotides were purchased with a disulfide protecting group. This group must be removed in order to bind the oligonucleotides to the AuNPs. TCEP, a reducing agent, was used to carry out this reaction. The disulfide bond can reform if oxygen is present.



**Figure 25. Preparing AuNPs and binding DNA** | AuNPs are prepared with a DMAP capping ligand (Top). MESA molecules bind to the AuNPs through covalent thiol bonds and thus replace the weaker DMAP molecules. The DMAP-MESA AuNPs are added to the DNA-bound beads. They attach through a similar ligand exchange reaction involving DMAP molecules and covalent thiol bonds. Negative charges from the MESA capping ligands (red circles) and phosphate backbone (blue circles) could potentially repel each other (bottom right).

# Efficiency of AuNP Binding



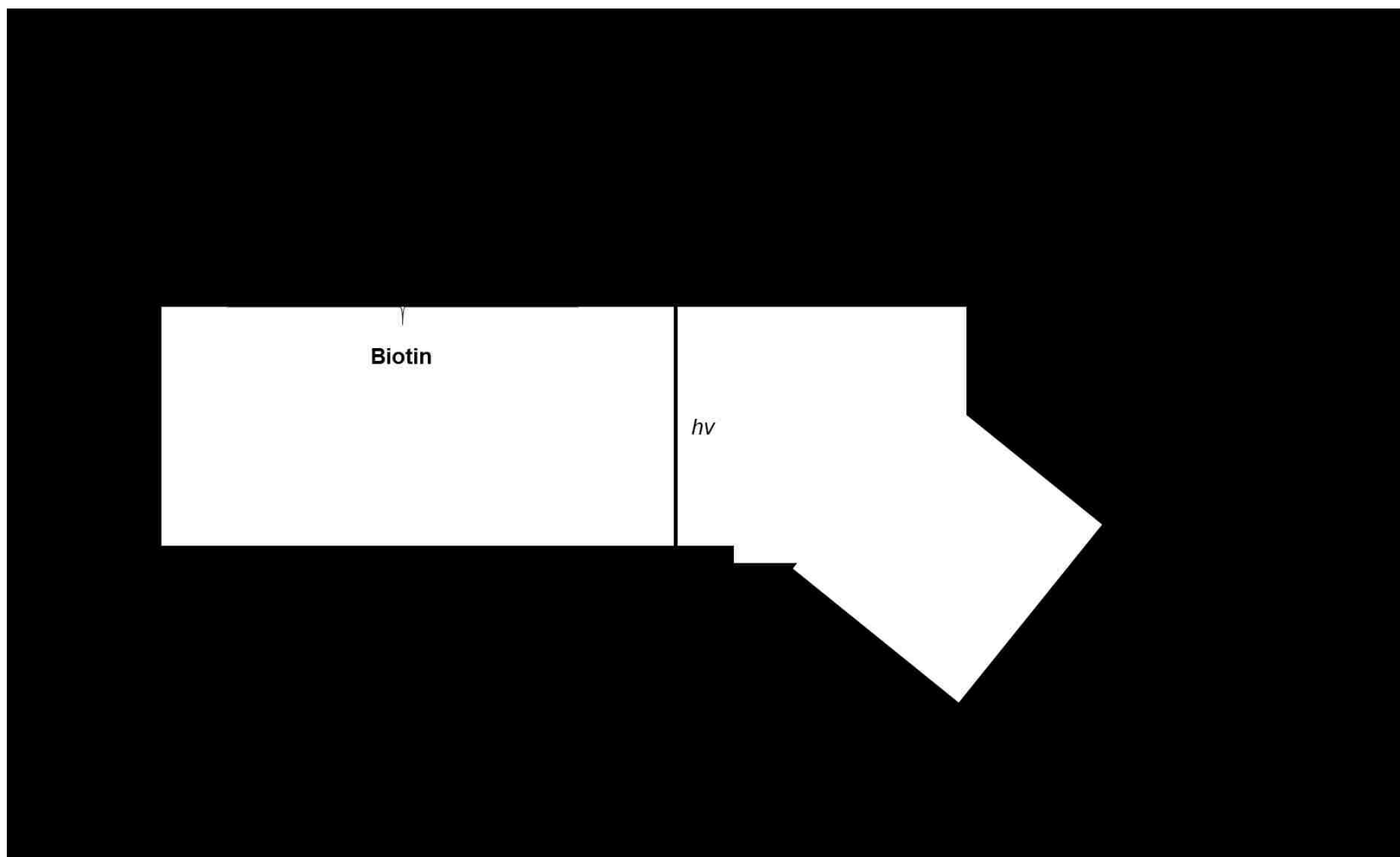
**Figure 26. Summary of attempts to increase efficiency of AuNP** | The concentration drop in the AuNP sample was measured using UV-Visible spectroscopy. The number of AuNPs that became bound was divided by the number of DNA strands that became bound in the first step of the protocol (~20 pmols). The original protocol yielded 3% AuNP binding. Nitrogen purging was added to prevent disulfide reformation. Under this condition, the efficiency was increased to 15%. The MESA concentration was lowered to reduce electrostatic repulsion between the AuNPs and DNA strands. The efficiency was increased to 23% under this condition. NaCl was added to shield the charges on the DNA strands and AuNPs. This condition did not influence the yield.

The first step was to purge the buffers with N<sub>2</sub> to reduce the amount of O<sub>2</sub> present and thus reduce the amount of disulfide reformation. This condition improved the efficiency to 15% so N<sub>2</sub> purging was implemented during follow-up experiments. The next step was to reduce the concentration of the negatively charged MESA capping ligand to reduce electrostatic repulsion between the AuNPs and DNA strands. The MESA concentration was reduced until a minimum concentration was reached (3x lower than original protocol), below which the AuNPs became unstable due to uncontrolled aggregation. This condition improved the efficiency of AuNP binding to 23%. It was hypothesized that the counter ions in NaCl solution could be used to shield the electrostatic repulsion further. However, adding NaCl did not improve the efficiency of AuNP binding.

These results together suggested that both electrostatic repulsion and disulfide reformation could have affected the efficiency of AuNP binding. However, the maximum yield of 25% suggested that there are other variables that have not been considered. Since lowering the MESA concentration slightly improved the yield while adding NaCl to shield the charge repulsion did not, it is possible that the increased yield was caused by something other than reduced electrostatic repulsion. MESA binds to the surface of the AuNPs through a thiol interaction which replaces some of the DMAP molecules (Fig. 25). This reaction is identical to the ligand exchange reaction that takes place when DNA strands bind to the AuNPs. Thus, it is possible that the MESA molecules themselves prevented the DNA strands from binding to the AuNPs via steric hindrance. To test this, NaCl was added to reactions involving AuNPs with the original amount of MESA (3x more). The efficiency of AuNP binding was 12% (data not shown). This result suggested that MESA itself is the cause of low ligand exchange efficiency and not electrostatic repulsion. Testing this hypothesis further will be the subject of future experiments.

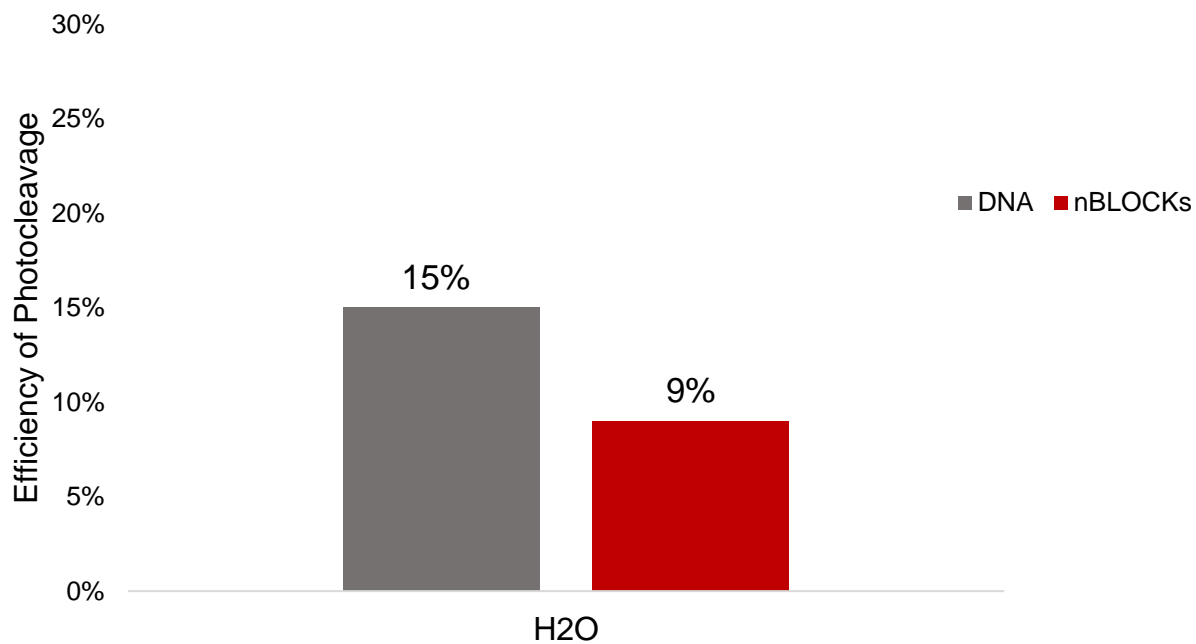
The final step in the anisotropic monofunctionalization method is to cleave the DNA-bound AuNPs from the beads. The PC biotin linker was cleaved using UV light (Fig. 27). UV-Vis data showed comparable yields of 9% and 15% photocleavage for nBLOCKs and DNA only respectively (Fig. 28). It is important to note here that many samples showed no photocleavage at all and that these results represent the maximum yields.





**Figure 27. Photocleavage of PC biotin using UV light (adapted from [45])** | UV light cleaves the bond between the oligonucleotide and biotin molecule. This reaction leaves the biotin molecules attached to the beads and the DNA strands with unmodified ends.

# Efficiency of Photocleavage



**Figure 28. Efficiency of Photocleavage in H<sub>2</sub>O** | Photocleavage was carried out on DNA only and on DNA-bound AuNPs. The efficiency was determined by calculating the concentration of DNA or nBLOCKs released into solution and dividing them by the number of DNA strands or AuNPs that became bound during previous steps. The overall yields of both were similar at around ~10-15% efficiency.

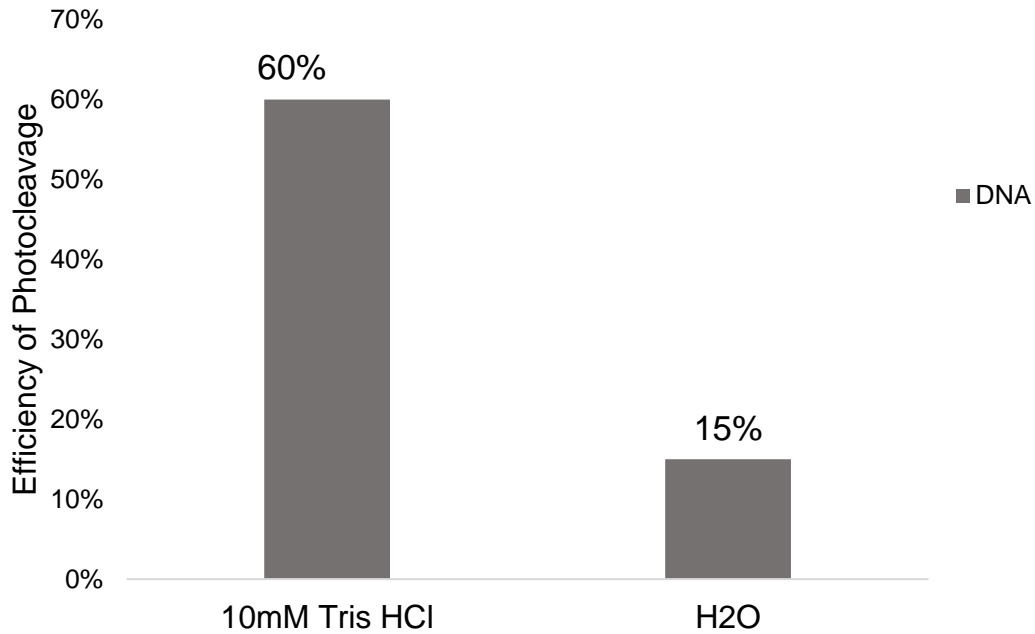
It was hypothesized that the efficiency of photocleavage was influenced by the buffer conditions of the photocleavage reaction. In buffer, ssDNA takes on a more linear form whereas in water, it is much more agglomerated. If the DNA strands agglomerated around the photocleavable linkers in water, it is possible that they prevented the UV light from reaching the nitrobenzene ring. To test this hypothesis, photocleavage was carried out in 10 mM Tris HCl buffer. The efficiency of photocleavage improved to 60% photocleavage for DNA only (Fig. 29). However, this improvement has not yet been replicated for nBLOCKs. Tuning the buffer condition will be the subject of future experiments.

To summarize, the PC biotin – streptavidin interaction can be used to overcome the limitations of the previous nBLOCK protocol. The first step of the reaction was carried out with maximum efficiency and specificity. Additionally, the nBLOCKs created using this method had unmodified DNA strands attached. This unmodified DNA strand is key for downstream molecular biology techniques and is also theoretically important for the close packing of nBLOCKs during DNA Brick-based self-assembly. Thus, some of the limitations of the previous nBLOCK protocol have been successfully overcome.

Nevertheless, the overall efficiency of the reaction is still low ( $25\%$  [Step 2]  $\times$   $9\%$  [Step 3]  $\approx 2\%$  total yield). Several attempts were made to improve the efficiency of the second step but the maximum efficiency achieved was around 25%. The efficiency of the photocleavage step peaked at around 10% yield for nBLOCKs, though the yield was improved to 60% for DNA only. Further improving the yields of these two steps will be the focus of future experiments.

One way to improve the yield of the second step would be to change the capping ligand on the AuNPs to something weaker than MESA. Bis(p-sulfonatophenyl)phenylphosphine dihydrate dipotassium (BSPP) has been used to cap AuNPs for similar ligand exchange reactions. These AuNPs bound to DNA strands with much higher efficiency than DMAP-MESA AuNPs [36]. It might also be possible to use a disulfide functional group to bind the DNA strands to the AuNPs since it forms a stronger bond than the thiol linkage. The combination of these two changes would likely significantly improve the yield of AuNP binding. The efficiency of the third step could be improved by tuning the buffer conditions during photocleavage.

# Efficiency of Photocleavage



**Figure 29. Efficiency of Photocleavage in Buffer** | It was hypothesized that the efficiency of photocleavage could be improved by changing the reaction solution from water to buffer. Doing so raised the efficiency of photocleavage to 60% for DNA only. Tuning the buffer during photocleavage of nBLOCKs will be the subject of future experiments.

#### 4. Conclusion

The goal of this project was to create nanostructures from the inside out using linear nBLOCKs. While this goal was not fully accomplished, several steps were taken towards realizing it.

Computer modelling revealed that anisotropy and predictable geometry were key components of DNA Brick-based self-assembly. This information was used to design linear nBLOCKs that are capable of DNA Brick-based self-assembly. These nBLOCKs consist of an AuNP attached to a single 43 bp long DNA Brick. The DNA Brick geometry was modified by adding an additional 11bp backbone region to create space for AuNPs. One key assumption for these building blocks was that the 371° turn created by the 11bp region would not significantly influence the parallel helix geometry of the DNA strands. This assumption was based on previous experimental validations of modelled DNA origami structures [18,44]. Additionally, it was assumed that the freely floating nanoparticles would populate specific cavities created by the dsDNA backbones. However, the AuNPs are free to move because they are attached to the DNA structure through flexible carbon linkers. There are several variables that can be tuned to increase or decrease the freedom of movement depending on the desired applications. Nevertheless, the modelling results showed that it would be possible to create nanostructures from the inside out using linear nBLOCKs.

Additionally, the modelling results revealed that it would be difficult to create structures using nBLOCKs with two DNA strands attached. The free, independent rotation of the DNA strands on either side of the particle make it difficult to create structures with specific geometries. Future efforts should be focused on using nBLOCKs with a single DNA strand attached to build structures.

The experimental work itself showed that the photocleavable biotin – streptavidin interaction can be used to improve the capabilities of the anisotropic monofunctionalization method. This new method was successfully used to create and recover nBLOCKs. However, the maximum overall yield was only 2%. The major limitation at this stage is the inefficiencies of AuNP binding and photocleavage. The yields have not yet reached the level necessary for structure building. Overcoming these roadblocks will eventually allow for the creation of nanostructures from linear nBLOCKs.

## 5. Future Work

Solidworks models are inherently limited because they can only simulate rigid objects. In reality, ssDNA is very flexible. Additionally, the strength of binding between DNA strands is dependent on such variables as temperature and salt concentration. Neither of these variables can be incorporated into the Solidworks models.

The Solidworks models showed that nBLOCKs with a single DNA strand are capable of DNA Brick-based self-assembly. Nevertheless, it is possible that the addition of an 11bp dsDNA backbone and the free floating AuNPs would negatively influence the structures' stability in a way that the Solidworks models could not predict. The structures' stability will also be influenced by the size and shape of the AuNPs, the length of the carbon linker attaching the AuNPs to the DNA strands, the concentration of the AuNPs' capping ligand and the strength of binding between building blocks (e.g. the length of overlapping complementary regions).

Simulating these variables will require a more powerful modelling program. The OxDNA molecular dynamics simulator is capable of simulating both DNA and nBLOCKs [33]. The program was recently limited by the time required to simulate large numbers of nBLOCKs but this limitation has since been overcome. Thus, it is likely that this model would help fill in knowledge gaps left by the Solidworks models. A set of strand sequences that can be used to study the self-assembly of nBLOCKs in OxDNA is given in Appendix A.

The Solidworks models themselves could also be used to design dynamic structure changes involving strand displacement reactions. This has not yet been tried for either DNA Brick or AuNP-based structures but would likely improve their versatility for specific applications.

Successfully implementing the modelling results will require improving the efficiency of the anisotropic monofunctionalization method as well. Based on the results of this thesis, there are three options to further improve the efficiency of the protocol. The first is to change the AuNP capping ligand. One capping ligand that binds more weakly than MESA and has been successfully used for a similar ligand exchange reaction is bis(p-sulfonatophenyl)phenylphosphine dihydrate dipotassium (BSPP) [36]. The

second option is to use a disulfide modified oligonucleotide instead of a thiol modified DNA strand. The dithiol moiety would allow the DNA strands to bind more tightly to the particles and push the equilibrium of the reaction further towards AuNP binding. The third option is to optimize the buffer conditions and maximize the efficiency of photocleavage. The combination of these three conditions will likely provide the best results.

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## Appendix A: Strand Sequences for Small nBLOCK cube nanostructures

<u>ID:</u>	<u>Sequence (3'-5'):</u>
ssDNA1	gtagagagta
n1	tttttttttctcagtg
n2	ttttaacgtgcttttttttactctctaacttttttctgaaagg
n3	ttttttttttccacactg
n4	tttttacttggttttttactctctaacttttttagctata
n5	tttttaatctgctttttttactctctaacttttttagctata
n6	ttttggcctctatttttttactctctaacttttttccttacac
n7	ttttgaatggttttttttactctctaacttttttaaactg
n8	ttttacctagcgtttttttactctctaacttttttaacaaac
n9	ttttgttcataattttttt
n10	tttttagtgatttttttt
n11	tttttaatgagggctactgagtactcttaacagcacggttgaggact
n12	tttttccttatcaggtgggtactcttaaccaagtaatacactgg
n13	ttttgcagattactacggat
n14	ttttacggcttctctcagtactcttaacagaggccgacatagg
n15	ttttggaatggtataagct
n16	ttttctataaacctggtggctactcttaacaaacattccacgtatg
n17	ttttgaaagcgtgtgtaagggtactcttaaccgtaggttcgtagag
n18	tttttatgaaccgggaaac
n19	ttttccagttgccaagttatactcttaacaatcacctctcccta
n20	ttttggacagttggtttt
n21	ttttcaggcgagatccgtagtactcttaacccctattctatgaa
n22	ttttctgggtgcagctca
n23	ttttcagagctccctatgtctactcttaacataaggaagatcgatc
n24	ttttatactcaaccagtga
n25	ttttcccgaagtcatacgtctactcttaacaaagccgttgtaacta
n26	ttttctcaaggcctctacgatactcttaacaccattccggaatacc
n27	tttttacgtcctgtttcccgtactcttaacgttataggagtacgt
n28	ttttaattgctctaagggagtactcttaacacgcttcttcggtg
n29	ttttgcaactgggacacaga
n30	ttttaactgtccaaggcca
n31	tttttttttttcataag
n32	ttttttttttgatcgatctactcttaacgcacccagtttttt
n33	ttttttgagtattttttt
n34	tttttttttttagttacatactcttaacctcgctgtttttt
n35	ttttttttttggattcctactcttaacgagctctgtttttt
n36	ttttttttttacgtactc
n37	ttttttttttcaccgaaatactcttaacactcgggtttttt
n38	ttttgccttgagtttttt
n39	ttttttttttctgtgtactcttaacaggacgtatttttt
n40	ttttttttttggcctgtactcttaacgagcaattttttt