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Control Oriented NOx and Soot Emission

Estimation for Diesel Engine

by

Fangfang Lin

A Thesis

Submitted to the Faculty of Graduate Studies

through the Department of Mechanical, Automotive, and Materials Engineering

in Partial Fulfillment of the Requirements for

the Degree of Master of Applied Science at the

University of Windsor

Windsor, Ontario, Canada

2011

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DECLARATION OF ORIGINALITY

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ABSTRACT

Comparing to conventional Electric Vehicles (EVs) and Hybrid Electric Vehicles (HEV), Plug-in Hybrid Electric Vehicles (PHEVs) have gained more attraction from researchers due to their advantages of extended all-electric range, decreased emission and being less dependent on recharging infrastructure. In order to develop power control and management for diesel engine-generator set in a series PHEV system, an estimation mechanism based on the NOx and soot emission with the impact of Exhaust Gas Recirculation (EGR) is developed in this thesis. In particular, the single zone combustion thermodynamic engine sub-model, NOx estimation sub-model (based on extended Zeldovich mechanism), soot estimation sub-model (based on Hiroyasu's two-step empirical model) and EGR sub-model are incorporated together. In order to illustrate the impact of EGR on the NOx and soot emission, the EGR sweep experiment has been carried out under certain working conditions. And, the simulation results have been validated with the empirical data. Finally, based on the validation of the emission estimation mechanism and the selected eight operating points working condition, the results EGR sweep results have been generated for the series PHEV engine-generator control design. In general, the present work targets on estimating real-time NOx and soot emission based on the inputs information, such as intake boost, fuel injection, injection timing etc, and the purpose of this thesis is to aid development of engine emission management with study about the effect of EGR on the reduction and the soot trade-off.

Key words: diesel emission estimation, PHEVs, thermodynamic engine model, EGR;

DEDICATION

To my beloved,

Adrian Wydra

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NOMENCLATURE

BSFC	Break Specific Fuel Consumption
CA	Crank Angle
EGR	Exhaust Gas Recirculation
EV	Electric Vehicle
FSN	Filter Smoke Number
HEV	Hybrid Electric Vehicle
HRR	Heat Release Rate
ID	Ignition Delay
IMEP	Indicated Mean Effective Pressure
L	Liter
NO	Nitric Oxide
RPM	Revolution per Minute
TDC	Top Dead Center
PHEV	Plug-in Hybrid Electrical Vehicle
Pinj	Injection Pressure
RPM	Revolution Per Minute
SOC	Start of Combustion
SOI	Start of Injection
Tw	Cylinder Wall Temperature

CHAPTER 1 INTRODUCTION

Due to the increasing oil price and stringent regulations on emissions around the globe, Electric Vehicles (EVs) and Hybrid Electric Vehicles (HEVs) [1] have received much research attention in the past few decades. Recently, many studied have been carried out in the field of Plug-in Hybrid Electric Vehicles (PHEVs) [3][4], which are considered as the new generation of HEV, because of rapidly improving battery technology. Gasoline engine (or other spark-ignition engine) equipped with an electric motor is still the most popular hybrid powertrains [5]. Comparing with conventional gasoline engine and HEVs, gasoline PHEVs reduce CO₂ emissions by 37%-67% and 19-54% [6]. But being encouraged by the Diesel engine (also known as compression-ignited engine) fuel economy benefits and high engine performance potential under low speed condition, researchers have increasing interests in the development of combination of diesel engines and electric motor in HEVs. Especially in Europe, automakers such as Volvo, Mercedes-Benz and Volkswagen are paying much attention to the investigation of diesel PHEVs.

1.1 Diesel Engine Emission Estimation

Emissions from internal combustion engine are released to the environment mainly into air and sometimes it is absorbed in water in marine application. The impact of air pollution is significant and few areas around the world are not impacted by its effects. Road transportation is considered a significant contributor to air pollution in cities. It's the main contributor for the emission of nitrogen dioxide. In recent decades, the epidemiological and toxicological evidence on the effects of transport-related air pollution on health has increased heavily [6]. The National Pollutant Inventory Guide list and detailed the four types of emission estimation techniques, which are sampling or direct measurement, mass balance, fuel analysis or other engineering calculations and emissions factors [10].

Since an early successful diesel engine was demonstrated by Rudolph Diesel at the 1900 Paris Exhibition [8], this technology has changed the world. High compression ratio and lean fuel running condition bring the high thermal efficiencies to Diesel engine. Comparing with gasoline engine, today's turbocharged diesel engine has 30% better fuel economy [8]. The most difficult challenges for diesel engine designers are the formation of soot particles and NOx. When the high velocity injected diesel fuel is directly sprayed into the air in the main combustion chamber, which is compressed to a high temperature and pressure, it vaporizes quickly and mixes with the air. Without spark plug applied, the mixture is ignited after undergoing a series of chemical reactions. Because the fuel distribution is non-uniform, NOx forms in the high flame temperature area, and the highest formation rate is in the close-to-stoichiometric regions [9]. The soot particles are generated from the incomplete combustion of hydrocarbon fuel because of the highly non-homogeneous environment. Over the past few years, due to the imposed emission regulations, extraordinary achievements of developing clean diesel engine have been made when new techniques are utilized such as common rail system, fuel injection control strategies, exhaust gas recirculation and exhaust gas after treatment etc [12].



Figure 1-1 Typical Direct Ignition (DI) Diesel Engine Heat Release Rate Diagram

To order to predict the engine performance and NOx and soot emission formation, it is necessary and important to understand the details of diesel engine combustion process [1]. For Diesel engine, the pollutant formation processes are mainly determined by fuel mixture distribution in the cylinder. Figure 1- 1 shows the typical Direct Ignition (DI) Diesel engine heat release rate diagram. After the ignition delay period, the fuel combustion takes place rapidly with the high heat-release rate during the premixed combustion phase.

To cope with the air pollution problem, stringent emission limits have been implemented to diesel engine. The following diagrams Figure 1- 2 show the European Emission Standards for HD Diesel Engine with respect to NOx and soot emission ranging from the 1992 to 2013. These two diagrams reveal the restriction of the emission standard over time. For example, the

concentrations in 2013 are expected to be roughly 25% of those in 2008.

An estimation of Diesel engine emission is essential for the Diesel engine-generator set control design. The real diesel engine works as a highly non-linear mechanical system [14], it is extremely hard to produce a model with truly predictive capabilities to achieve more precise estimation. With the rapid development of microprocessor technology, these engines models can be applied in the field of real-time state estimation, control and fault diagnostics [15][16]. Early efforts in diesel engine modeling were focus on specific closed part of the engine cycle, and the linear dynamic models were developed based on the usage of empirical data [17][2]. The great drawbacks of these quasi-linear models are their huge dependence on the empirical data, which need to be collected prior to the simulations, and poor performance to the transient response.



Figure 1-2 European NOx and PM Emissions Standard for Heavy-Duty Diesel Engine

1.2 Literature Review

Nowadays, most of the phenomena related to combustion have been revealed by empirical study.

However, there is no experimental equipment or technique which can provide and clarify all detailed relevant chemical reactions. Based on the understanding of chemical processes, the simulation modeling and analysis investigation is necessary.

NOx emissions consists of NO and NO₂. The NO₂ formation is via the oxidation of NO. Therefore, the simulation of NOx formation is reduced to understanding of NO formation [21]. Based on the mechanism developed by Ahmed and Plee [17] that the majority of NO formation is via thermal path, Zeldovich mechanism and extended Zeldovich mechanism were postulated by Zeldovich and further developed by Lavoie [9]. The empirical heat transfer correlation was utilized in these phenomenological models. Miller et al [18] and Sundar R. Krishnan *et al* [19] improved the mechanism to super extended Zeldovich mechanism. Sundar R. Krishnan *et al* 's model accounts for 43 reactions and 20 species and Miller *et al*'s model includes 67 reactions and 13 species ,which provided more accurate prediction results, instead of the typical three reactions used in the extended Zeldovich mechanism. But then more reactions and species are included in the calculation results in higher computational complexity.

Although the soot formation phenomenon in the in-cylinder turbulent spray combustion is far from being fully understood, various models have been developed for soot predictions [21]. Khan and Greees first presented the model for the prediction of soot for diesel engines. Later on, detailed kinetic soot models were proposed [22] to describe soot dynamic formation. Complex gas phase chemistry reaction and particle growth and distribution were considered in detailed kinetic approach. Due to the complexity of turbulence, thermochemistry, and the appearance of Arrheius terms in detailed kinetic models, there have been concerns about using this approach to simulate diesel combustion condition. Hiroyasu et al. proposed a two-step phenomenological model to predict soot formation distribution [23] for combustion engines. Because Hiroyasu's model is easy to be implemented into CFD simulation code, most of the modern prediction of soot formation for multidimensional diesel engine calculation was based on this empirical approach. Comparing to the detailed kinetic soot model, the phenomenological model consists one Arrhenuis term in the soot formation step with two empirical constants and the particle growth and the dynamic of soot has not been taken into account. Niklas Winkler [25] incorporated a multi-zone diesel engine model for NOx prediction with a simplified soot formation estimation model, which were tuned with constant parameters during the entire engine operating range. The calculated engine performance which was produced from a 1-dimensional fluid dynamic code GT-Power for transient operation was validated with the experimental data.

1.3 Research Objectives

Figure 1-1 represents the clean diesel series plug-in hybrid electrical vehicle (PHEV) powertrain system. The author has been working on this project with other colleague at Clean Diesel Engine Laboratory. As shown in Figure 1-3, the development of the Diesel Engine model and the real time emission estimation model have been carried out by the author. The series HEVs proposed in this thesis has three power sources: the primary power sources are the high voltage battery and the super capacitors while the assistant power source is the diesel -generator set.



Figure 1-3 Series PHEV Powertrain System

The main goal of this thesis is developing an engine thermodynamic model and emission (NOx and Soot) estimation for a smart engine, incorporating of the EGR model into the engine model and the validation of the engine and emission estimation. The proposed single dimensional engine thermodynamic model provides the in-cylinder pressure, temperature, heat release rate (HRR), species fraction during the combustion process. Sub-models include:

- 1) Ignition delay model
- 2) Heat release model
- 3) Heat transfer model

These models are integrated together to predict engine performance and estimate emission formation. At the end of the thesis, the calculation results of the 8 operating points for the smart engine are also included.



Figure 1-4 NOx and Soot Emission Estimation Block

CHAPTER 2 BACKGROUND

PHEVs are a potentially important technology for reducing the fossil fuel consumption and CO2 emission because they can run on electricity for a certain distance after each recharge, depending on their battery's energy storage capacity, which is expected to be typically between 20km and 80km [1]. PHEVs offer great promise for petroleum displacement [29]. According to the 2009 National Household Travel Survey [28], in 2009, people travels 36.1 miles per day and average daily trips per person is 3.8. Although PHEVs will be more expensive than conventional and hybrid vehicles, the cost can be recovered by fuel saving. Meanwhile, people will benefit from emissions reduction.

The powertrain configurations include series, parallel and power split vehicle configurations. Comparing with other configurations, the series engine configuration is often considered to be closer to a pure electric vehicle. Since engine operation is directly related to fuel efficiency and gaseous emissions, it plays an important part in the hybrid vehicle control strategy [30][27].

Engine operating point is one of the important parameters affecting the vehicle control strategy. Engine operating point is directly related to fuel efficiency and emissions [31]. As mentioned before in Fig 1-2, in the case of series PHEVs configuration, the propulsion power for the vehicle is exclusively coming from the electrical energy source traction motor. As a result of that, engine speed is entirely decoupled from the wheel axles, and the engine operation point is independent of vehicle operations [31]. From the diesel engine perspective, as shown in Fig 2-1 the brake specific fuel consumption (BSFC) curve, with the characteristic series PHEVs system, diesel engine can work along the best efficiency cure while satisfying the power demand. Meanwhile, when the diesel engine works in the optimal engine operating points, NOx and PM emission increase dramatically. It bring another challenging constrain in the control strategy.

In the Argonne National Laboratory, Maxime research group conducted the project to test the effect of powertrain system control on diesel engine emissions and fuel efficiency, base on the conventional diesel engine and PHEVs. The trade off relation between fuel economy and emission from different configuration was studied, however, the NOx and PM emission was not considered as one of constrains in the control theory in simulation model. Aymeric Rousseau [29] tested the fuel efficiency of power split and series configuration midsize vehicles on more than real world driving conditions cycles. In Oak Ridge National Laboratory, Vehicle Technologies Program started the advanced PHEV Engine Systems and Emissions Control Modeling and Analysis program at 2005. One of the project objectives is to apply the simulation to test the fuel efficiency and emissions impact of advanced combustion (HCCI, PCCI) versus conventional SI and diesel engine.



Figure 2-1 Engine Brake Specific Fuel Consumption Contour

Chemical reaction inside cylinder during the combustion process is very difficult to model due to its transient and heterogeneous character, primarily controlled by the turbulent mixing of injected fuel and the fresh intake air [32]. So it is hard to predict the NOx and Soot emission. Researchers attempted to develop two-zone combustion model, multi-zone combustion models driven by the desire to predict accurate exhaust emissions. Hongusk Kim [33] investigated the effect of Exhaust Gas Recirculation on heavy-duty diesel engine performance and developed and modified KIVA-3V code to test the NO and Soot emissions. The multidimensional engine model was used in investigation. Pia Kilpinen [13] improved a sub-model for NO emission prediction for medium-speed, four-stroke, direct-injection marine engine and the trend of increasing NO emission with increasing load was correctly predicted.

Comparing with lower zones diesel combustion models, the advantage of multi-dimensional models is available for the detailed spatial information and interactions of phenomena. However, the accurate prediction results from multi zone models are based on the relative inadequacy of turbulence, combustion chemistry sub-models. And multi-dimensional models require longer computing time and storages capacity [9]. Therefore, in our case, the single-zone model is a reasonable choice for our control purpose.

EGR was considered as an effective means for reducing flame temperature and NOx emission [35]. The control inputs to the diesel engine include the fuel injection, injection time, boost and EGR, while the outputs are the in-cylinder pressure, in-cylinder temperature and heat release rate etcetera. Figure 2- 1 shows a flow chart of the diesel thermodynamic model and NOx and soot emission model. As mentioned before the thermodynamic model receives the input signals of fuel injection, injection timing, boost and EGR. From this, in-cylinder pressure, heat release rate and in-cylinder temperature can be calculated. The instantaneous in-cylinder pressure, temperature and heat release rate can be used as inputs of the emission estimation models.

After comparing the experimental data from Clean Diesel Engine Lab to the estimation results, the model was tuned in order to complete the model validation. In the following sections the detailed information about the diesel thermodynamic model and sub-models and the emission estimation model is shown.



Figure 2-2 Thermodynamic Engine Model and Emission Estimation Model

2.1 Combustion Model

According to the first law of the thermodynamics, the change of the internal energy of a system

during a process is the sum of the changes in its heat transfer and work transfer. Figure 2- 3 indicates the thermodynamic engine model. The first law of thermodynamics is implemented with the ideal -gas equation of state [34].

$$dQ = dE + pdV \tag{2.1}$$

$$[OH] = 2.129 \times 10^{2} \times T^{(-0.57)} \times e^{(-4595/T)} \times [O]^{0.05} \times [H_2O]^{1/2}$$
(2.2)

Where: *R* is the universal gas constant (R= $8.31447 \ kJ/kmol \cdot K$);

P (pa) is the absolute pressure;

 $T(\mathbf{K})$ is the absolute temperature;

 $V(m^3)$ is the instantaneous cylinder volume;

The instantaneous cylinder volume consists of a clearance volume V_c and the instantaneous displacement volume. The instantaneous displacement volume is a function of crank angle, cylinder bore, crank radius, compression ratio and the length of connecting rod.

$$r_{c} = \frac{Maximum \ Cylinder \ Volume}{Minimum \ Cylinder \ Volume} = \frac{V_{d} + V_{c}}{V_{c}}$$
(2.3)

$$V = \frac{\pi D^2}{4} \left\{ \frac{S}{r_c - 1} + \frac{S}{2} \left[\left(1 + \frac{1}{R} \right) - \cos(\frac{\pi}{180}\theta) - \frac{1}{R} \sqrt{1 - R^2 \sin^2(\frac{\pi}{180}\theta)} \right] \right\}$$
(2.4)

Where V_c is the clearance volume;

 V_d is the displaced volume;

 r_c is the compression ratio;



Figure 2-3 Thermodynamic Engine Model

For the air surrounding the injected fuel, the first law of thermodynamic can be written as following equation:

$$\frac{dU}{d\theta} = p \frac{dV}{d\theta} - \frac{dQ_b}{d\theta} - \frac{dQ_w}{d\theta}$$

$$(2.5)$$

$$(M_{air} + X_{(i)} \times M_{fuel}) C_p \left(T_{(i)} - T_{(i-1)}\right) = LHV \times M_v \times \left(\theta_{(i)} - \theta_{(i-1)}\right) - p_{(i-1)} \left(V_{(i)} - V_{(i-1)}\right) - Q_{w(i)}$$

$$(2.6)$$

The equation for the mean mixture temperature is derived as follows:

$$T_{(i)} = T_{(i-1)} + \frac{LHV \times M_v \times (\theta_{(i)} - \theta_{(i-1)}) - p_{(i-1)} (V_{(i)} - V_{(i-1)}) - Q_{w(i)}}{M_{air} + X_{(i)} \times M_{fuel}}$$
(2.7)

Where: *LHV* is the lower heating value; (Assuming *LHV*=42.5 MJ/kg)

 M_{fuel} (kg) is the injected fuel;

 C_p is the specific heat capacity;

 M_{air} (kg) is the mass of in-cylinder air;

 $M_{\nu}(\text{kg})$ is the evaporated fuel at each step size;

 Q_w is the heat transfer from cylinder wall, cylinder head and piston;

X is the burn fraction of the injected fuel;

T is the in-cylinder temperature;

The compression process is assumed as a polytrophic compression process;

$$P_0 V_0^{\gamma} = P V^{\gamma} \tag{2.8}$$

Where: P_0 (pa) is the in-cylinder pressure when the crank angle starts from 180°;

 V_0 (m³) is the in-cylinder volume when the crank angle stats from 180°;

P (pa) is the in-cylinder pressure when the crank angle is after 180° ;

 $V(m^3)$ is the in-cylinder volume when the crank angle is after 180°;

The equations for the in-cylinder pressure and in-cylinder temperature are as follows during the compression process.

$$P_{(i+1)} = P_{(i)} \left(\frac{V_{(i)}}{V_{(i+1)}}\right)^{\gamma}$$
(2.9)

$$T_{(i+1)} = \frac{P_{(i+1)}V_{(i+1)}}{P_{(i)}V_{(i)}}T_{(i)}$$
(2.10)

2.2 Heat Release Model

The in-cylinder combustion is a very complex process, and the extensive modeling would be helpful. The analytical functions computing the burn rate is used in this thesis. The prevailing of these functions is the Wiebe function, which can be used to predict the burn fraction and burn rate for different engine systems [43][44].

Equation (2.11) and (2.12) are used to calculate the mass burning fraction and absolute value of heat release [42].

$$X_{b} = \left(1 - \alpha_{wall}\right) \left\{ 1 - \exp\left(-a\left(\frac{\theta - \theta_{0}}{\Delta \theta}\right)^{m+1}\right) \right\} + \alpha_{wall} \left\{ 1 - \exp\left(-a\left(\frac{\theta - \theta_{0}}{K_{wall}\Delta \theta}\right)^{m+1}\right) \right\}$$
(2.11)

$$\frac{dQ_B}{d\theta} = M_{fuel} \times LHV \times \frac{dX_b}{d\theta}$$
(2.12)

Where : X_b is the mass fraction burned;

 θ is the crank angle;

 θ_0 is the crank angle when the combustion starts (SOC);

- $\Delta\theta$ is the total combustion duration;
- *M* is the adjustable parameter;

A is the adjustable constant relating to the combustion duration;

 α_{wall} is the fraction of the mixture that burns in the slow combustion region;

 K_{wall} is the ratio of slow burn duration to the standard burn duration;

$$a = a_{10-90} = \left(\left(\ln(\frac{1}{1-0.9}) \right)^{1/(m+1)} - \left(\ln(\frac{1}{1-0.1}) \right)^{1/(m+1)} \right)^{m+1}$$
(2.13)

For diesel engines, the in-cylinder pressure, temperature and swirl ratio directly affect the ignition delay (ID). Many corrections have been made to predict the ignition delay for diesel engines [45][46]. The most widely accepted correlation was given by an Arrhenius expression [45], which is a function of ambient gas pressure and temperature.

$$ID = Fp^{-N} \exp(\frac{E}{R_{\mu}T})$$
(2.14)

Where: p and T are the cylinder pressure and temperature, E is the activation energy, R_u is the universal gas constant, N and F are experimental constants.

For diesel fuel, Wolfer (1950) [23] achieved a reasonable estimate from the following equation.

$$ID = 3.45 \exp(\frac{2100}{T}) p^{-1.02}$$
 (2.15)

2.3 Heat Transfer Model

Convective mode of heat transfer contributes to 80% of the total engine heat transfer [9].For typical diesel engines, the heat transfer process starts from the in-cylinder mixture within the cylinder via the combustion chamber wall to the coolant. Both the convective heat transfer and radiation heat transfer contribute to the heat flux through the wall. Based on the experiment results, Annand [9] developed the convective heat-transfer correlation for cylinder head. In this thesis, Woschni's approach is used.

$$\frac{dQ_w}{d\theta} = \sum_{i=1}^3 \frac{dQ_w}{d\theta} = \frac{1}{w} \times \sum_{i=1}^3 h_c \times A_i \times (T - T_w)$$
(2.16)

Where: *T* is the in-cylinder temperature (k).

 T_w is the temperature of cylinder wall, cylinder head and piston head.

 h_c is the coefficient of convective heat transfer;

i=1: Cylinder head; *i*=2: Piston; *i*=3: Cylinder wall;

A is the heat transfer area (m^3) ;

According to Woschni's correlation, the equation for the coefficient of heat transfer is summarized as:

$$h_c = 3.26 \times D^{-0.2} \times p^{0.8} \times T^{-0.55} \times w^{0.8}$$
(2.17)

Where: *D* is the cylinder bore (m);

p is the in-cylinder pressure (kPa);

T is the cylinder temperature (K);

W is the average cylinder gas velocity (m/s);

The average cylinder gas velocity for a four stroke, direct injection compression ignition engine was expressed as follows:

$$w = C_1 S_p + C_2 \frac{V_d T_r}{p_r V_r} (p - p_m)$$
(2.18)

Where: V_d is the displaced volume; p is the instantaneous in-cylinder pressure; P_r, T_r, V_r are the in-cylinder pressure, temperature and volume when the crank angle is 180° ; p_m is the motored cylinder pressure at the same crank angle as p; $C_1 = 2.28; C_2 = 3.24 \times 10^{-3}$;

2.4 EGR System

EGR systems work by recirculating different amounts of an engine's exhaust gas back to the engine cylinder. Combining the exhaust gas recirculation technology is recognized as the only possible way to achieve future emission restriction target by internal measures [54]. EGR is defined as the ratio of intake manifold CO_2 concentration to exhaust manifold CO_2 concentration [53]. EGR system interferes with the diesel combustion system through thermal, chemical and dilution effects. When an EGR system is incorporated into the diesel engine, the amount of NOx emissions are reduced due to lower in-cylinder temperatures during the combustion process. The application of EGR is regarded as one of the essential means of reducing oxides of nitrogen (NO_X) emission from diesel engines. However, for intermediate and low engine loads, EGR increases the particulate emissions and specific fuel consumption [36]. Equation (2.19) describes the complete combustion process without EGR while the equation (2.20) describes the combustion process when diesel engine incorporates EGR system. In the diesel engine model calculations for this work, the second equation was used. In order to validate the simulation results, the calculation was conducted by using a typical diesel fuel formulation. Ultimately, the

formula $C_1H_{1.7}$ was used in the simulation.

The complete combustion equation (without EGR) when Lambda > 1

$$C_{\alpha}H_{\beta} + \lambda\left(\alpha + \frac{\beta}{4}\right)\left(O_{2} + 3.76N_{2}\right) \rightarrow \alpha CO_{2} + \frac{\beta}{2}H_{2}O + 3.76\lambda\left(\alpha + \frac{\beta}{4}\right)N_{2} + \left(\lambda - 1\right)\left(\alpha + \frac{\beta}{4}\right)O_{2}$$

$$(2.19)$$

The complete combustion equation (with EGR)

$$C_{\alpha}H_{\beta} + a\left(O_{2} + 3.76N_{2}\right) + \left(\frac{\alpha \times EGR}{1 - EGR}\right)CO_{2} + \left(\frac{\beta}{2}EGR\right)H_{2}O + 3.76 \times a \times \frac{EGR}{1 - EGR}N_{2} + \frac{(a - \left(\alpha + \frac{\beta}{4}\right))EGR}{1 - EGR}O_{2} \rightarrow \left(\frac{a}{1 - EGR}\right)CO_{2} + \left(\frac{\beta}{2}\right)H_{2}O + \left(\frac{3.76 \times a}{1 - EGR}\right)N_{2} + \left(\frac{a - (\alpha + \frac{\beta}{4})}{1 - EGR}\right)O_{2}$$

$$(2.20)$$

Where
$$EGR = \frac{N_{CO_2,intake}}{N_{CO_2,exhaust}} a = \lambda \times (\alpha + \frac{\beta}{4})$$

The mass fraction of the CO₂ from exhaust:

$$MASS_CON_CO_{2} = \frac{\frac{\alpha \times EGR}{1 - EGR} \times 44}{\frac{\beta}{1 - EGR} \times 44 + \frac{\beta}{2} EGR} \times 18 + 3.76 \times a \times \frac{EGR}{1 - EGR} \times 28 + \frac{(a - (\alpha + \frac{\beta}{4}))EGR}{1 - EGR} \times 32}$$

Where: $a = \alpha + \frac{\beta}{4}$

Amount of O₂ (mole) consumed each step size (0.1 Crank Angle):

$$n_{rac} O_{2} = n_{fuel} \times (\alpha + \beta/4)$$
(2.22)

(2.21)

Where: *n_fuel* is the amount of fuel consumed each step size (0.1 Crank Angle);

Amount of CO₂, H₂O produced each step size (0.1 Crank Angle):

$$n_rac_CO_2 = n_fuel \times (\frac{a}{1-EGR} - \frac{\alpha \times EGR}{1-EGR})$$
(2.23)

$$n_rac_H_2O = n_fuel \times \left(\frac{\frac{\beta}{2}}{1 - EGR} - \frac{\frac{\beta}{2}EGR}{1 - EGR}\right)$$
(2.24)

$$n_{O_2}(i) = n_{O_2}(i-1) + n_{rac} O_2$$
(2.25)

$$n_{CO_{2}}(i) = n_{CO_{2}}(i-1) + n_{rac} CO_{2}$$
(2.26)

$$n_{H_2}O(i) = n_{H_2}O(i-1) + n_{rac} - H_2O$$
(2.27)

$$n_{total}(i) = n_{CO_{2}}(i) + n_{H_{2}}O(i) + n_{O_{2}}(i) + n_{N_{2}}(i)$$
(2.28)

The mole fraction of O_2 , N_2 , CO_2 , H_2O are as follows:

$$O_2 _ mole _ frac(i) = \frac{n _ O_2(i)}{n _ total(i)}$$
(2.29)

$$N_2 _ mole _ frac(i) = \frac{n _ N_2(i)}{n _ total(i)}$$
(2.30)

$$CO_2 _ mole _ frac(i) = \frac{n_CO_2(i)}{n_total(i)}$$
(2.31)

$$H_2O_mole_frac(i) = \frac{n_H_2O(i)}{n_total(i)}$$
(2.32)

CHAPTER 3 CONTROL ORIENTED NOx AND SOOT ESTIMATION

3.1 NOx Emission Estimation

In-cylinder NOx formation occurs from three fundamentally different reactions. In year 1946, Zeldovich [9] was the first to suggest the importance of reactions equation (3.1) and (3.2). For internal combustion engine, the primary source is the thermal NO forming from the thermal dissociation equation (3.1) and the subsequent reaction of nitrogen and oxygen molecules from the combustion air [10].

The governing equations for the thermal NOx from molecular nitrogen are as follows:

$$O + N_2 \rightleftharpoons N + NO \tag{3.1}$$

$$N + O_2 \rightleftharpoons O + NO$$
 (3.2)

The mechanism of NO formation from Zeldovich has been well accepted and understood extensively. Later on, Lavoie [9] added the reaction (3.3) to the Zeldovich Mechanism and contributed significantly to the study. Due to the temperature sensitivity, the mechanism including equation (3.1), (3.2) and (3.3) is referred to as the thermal mechanism.

$$N + OH \rightleftharpoons H + NO \tag{3.3}$$

Applying the chemical kinetic rate law for the equations presented above, the formation rate of NO is given by:

$$\frac{d[NO]}{dt} = k_{f,1}[O][N_2] + k_{f,2}[N][O_2] + k_{f,3}[N][OH]$$
$$-k_{r,1}[NO][N] - k_{r,2}[NO][O] - k_{r,3}[NO][H]$$

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Note: All concentrations have a unit of mol/m³

The rate constants measured from experimental studies have been evaluated by Hanson and Salimian[49]and the expressions are shown below.

$$k_{f,1} = 1.8 \times 10^8 e^{-38370/T} \tag{3.5}$$

$$k_{f,2} = 1.8 \times 10^4 T e^{-4680/T} \tag{3.6}$$

$$k_{f,3} = 7.1 \times 10^7 e^{-450/T} \tag{3.7}$$

$$k_{r,1} = 3.8 \times 10^7 \, e^{-425/T} \tag{3.8}$$

$$k_{r,2} = 3.81 \times 10^3 T e^{-20820/T} \tag{3.9}$$

$$k_{r,3} = 1.7 \times 10^8 e^{-24560/T} \tag{3.10}$$

Where $k_{f,1}$, $k_{f,2}$, $k_{f,3}$ are the rate constants for the forward reactions. $k_{r,1}$, $k_{r,2}$, $k_{r,3}$ are the rate constants for the corresponding reverse rate constants.

According to Westenberg[50], the equilibrium O-atom concentration can be derived as follows:

$$[O] = k_p \times [O_2]^{1/2}$$
(3.11)

$$[O] = 36.64 \times T^{-1/2} \times [O_2]^{1/2} \times e^{-27123/T}$$
(3.12)

The partial equilibrium approach was chosen to determine the OH radical concentration.

$$[OH] = 2.129 \times 10^{2} \times T^{-0.57} \times e^{-4595/T} \times [O]^{1/2} \times [H_{2}O]^{1/2}$$
(3.13)

Finally, the expression for the rate of change NO concentration is shown as follows

$$\frac{d[NO]}{dt} = 2k_{f,1}[O][N_2] \left(\frac{1 - \frac{k_{r,1}k_{r,2}[NO]^2}{k_{f,1}[N_2]k_{f,2}[O_2]}}{1 + \frac{k_{f,1}[NO]}{k_{f,2}[O_2] + k_{f,3}[OH]}} \right)$$
(3. 14)

As mentioned previously, the rate of formation of NO is significant at high temperatures, which results from the large amount of energy required to break the strong N_2 triple bond. This effect is illustrated by the high activation energy of reaction equation (3.1). Since the activation energy for nitrogen-atom oxidation is small, with sufficient oxygen provided, the free nitrogen atoms consumption rate equals to it formation rate and therefore a quasi-steady state can be achieved [50]. Except for extremely fuel-rich combustion conditions, this assumption can be applied for most combustion cases. The indicated forward and reverse reaction rates introduce the uncertainty to the prediction with actual in-cylinder pressure and temperature.

3.2 Soot Emission Estimation

The most popular semi-empirical model is the two-step Hiroyasu model [51]. According to his model, the soot formation process considered involves two reaction steps: the formation step and the oxidation step. When the formation reaction begins, at the same time, the oxidation step is carried on. Therefore the amount of soot produced is equal to the soot production of the formation reaction minus the oxidation reaction. In the formation step, shown in equation (3.16), soot is directly related to fuel vapor molecules whereas during the oxidation step, shown in the equation (3.17), soot particles decrease due to the presence of molecular oxygen. The pressure of O2 can be derived from the engine combustion model.

Soot formation rate equals to the amount of formed soot minus the oxided soot [37].

$$\frac{dm_{soot}}{dt} = \left(\frac{dm_{soot}}{dt}\right)_{form} - \left(\frac{dm_{soot}}{dt}\right)_{oxid}$$
(3.15)

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$$\frac{dm_{soot_form}}{dt} = A_1 \times \exp\left(-\frac{T_{A1}}{T}\right) \times M_{fuel_v} \times \left(\frac{P_{gas}}{P_{gas_ref}}\right)^{n1}$$
(3.16)

$$\frac{dm_{soot_oxid}}{dt} = A_2 \times \exp(-\frac{T_{A2}}{T}) \times (m_{soot})^{n^2} \times \left(\frac{p_{O_2}}{p_{O_2_ref}}\right)^{n^3}$$
(3.17)

Where: A_1 is the constant for soot formation;

 A_2 is the constant for soot oxidation;

 T_{A1} is the activation temperature for soot formation reaction [6313K];

 T_{A2} is the activation temperature for soot oxidation reaction [7070K];

 M_{fuel-v} is the currently vapor fuel mass (kg);

 p_{gas} is the in-cylinder pressure (bar);

 p_{gas_ref} is the gas reference pressure (bar);

 p_{O2} is the partial pressure of O₂ (bar);

 p_{O2-ref} is the O₂ reference partial pressure (bar);

$$n_1 = 1.8; n_2 = 1.0; n_3 = 1.0;$$

Due to its simplicity, this model has been widely utilized [37][38][39]. It can be incorporated into the single-zone, two-zone and multi-zone engine combustion models [40][26].

In order to calculate the soot formation rate, in-cylinder current vapor fuel mass is required. C.O. Schmalzing [41] proposed a holistic injection model to optimize fuel injection in the Diesel engine system. Comparing to measured liquid and vapor phase penetration lengths for different operating conditions, the calculated results were validated. According to the simulation results from [41], the relation of vaporizing percentage and crank angle was generated and utilized for the current vapor fuel mass calculation, which is shown in Figure 3-1.



Figure 3-1 Vaporizing Percentage VS. Crank Angle (Injection Pressure Pinj=80 MPa)

CHAPTER 4 EXPERIMENT SETUP AND ESTIMATION RESULT VALIDATION

4.1 Experiment Setup

The experimental investigation was conducted on a single cylinder, four-stroke, common rail diesel engine of 0.76 lit displacement volume. The experiment engine bench setup and engine geometrical specifications are shown in Figure 4- 1 and Table 4- 1.



Figure 4-1 Single Cylinder Research Engine Test Platform

The experimental investigations in this thesis have been carried out under independently controlled levels of engine boost, exhaust backpressure and EGR valve opening. In order to

measure the in-cylinder pressure to obtain combustion information, an AVL G13P pressure piezoelectric transducer was mounted and connected through a Kistler 5010B charge amplifier. Based on the computer code developed in Labview control software, the control of high speed acquisition system had been achieved. A dual-bank exhaust analyzer system (for NOx, HC, CO_2 and O_2) has been installed for the exhaust emissions and intake gas concentration for exhaust analysis. A CAI 6000 Series chemiluminescence detector and an AVL smoke meter are used to measure NOx and Filter Smoke Number. The following equation was used to convert FSN to soot mass concentration in the unit of $[mg/m^3]$.

$$Soot[mg/m^{3}] = 1.8419 \times [FSN]^{3} + 2.9545 \times [FSN]^{2} + 12.815 \times [FSN] - 0.0305$$
(4.1)

4.2 Estimation Result Validation

In order to affect the comparison results, it was necessary to carry out the parametric analysis of the sub-model constants to tune the model.

- 1) Determining the values for adjustable constants of Weibe heat release model in order to make sure that the calculated in-cylinder pressures match the experimental pressures measured from the cylinder pressure transducer. When *m* value equals to 1, the in-cylinder pressure and heat release rate curve match with the empirical results with $k_{wall} = 0.34$ and $\alpha = 0.36$. The temperature of the cylinder walls was chosen equal to 500K.
- 2) Determining the values for the reaction constants of the NOx formation so that the calculated results of NOx formation will match the experimental values. According to the NOx estimation mechanism, the rate constants for the forward reactions and corresponding reverse rate constants were set as follows:

$$k_{f,1} = 2.8 \times 10^8 e^{-42000/T} \tag{4.2}$$

$$k_{f,2} = 1.8 \times 10^4 T e^{-4680/T} \tag{4.3}$$

$$k_{f,3} = 7.1 \times 10^7 e^{-450/T} \tag{4.4}$$

$$k_{r,1} = 3.8 \times 10^7 e^{-425/T} \tag{4.5}$$

$$k_{r,2} = 3.81 \times 10^3 T e^{-20820/T} \tag{4.6}$$

$$k_{r,3} = 1.7 \times 10^8 e^{-24560/T} \tag{4.7}$$

The constants for the equilibrium O-atom and OH radial concentration were chosen as follows:

$$[O] = 36.64 \times T^{1.24} \times [O_2]^{1.4} \times e^{-28400/T}$$
(4.8)

The partial equilibrium approach was chosen to determine the OH radical concentration.

$$[OH] = 2.129 \times 10^{2} \times T^{(-0.57)} \times e^{(-4595/T)} \times [O]^{0.05} \times [H_{2}O]^{1/2}$$
(4.9)

3) Determining the values for the reaction constants of the soot formation so that the soot formation will match the experimental values. For soot estimation mechanism, the constants for soot formation and oxidation were set as $A_1 = 1.6 \times 10^{-4}$ and $A_2 = 3200$. And the exponent of pressure in the soot formation and oxidation equations $n_1=3$ and $n_3=0.9$ are chosen to fit the experimental results. The activation temperatures are found in [37], $T_{A1}=3200k$ and $T_{A2}=7070k$.

Before utilizing the thermodynamic model and emission estimation for the 8 operating points, it is necessary to validate their ability to provide the engine performance information and emission profile. In this chapter, the calculated and measured results of in-cylinder pressure, temperature, cumulative heat release, NO formation concentration and soot formation concentration as a function of crank angle will be presented. The measured experimental data and calculated results are depicted in the following figures in a comparative way. Table 4-1 presents the specification of single cylinder research engine used for the engine performance and emission estimation validation.

Bore (mm)	96
Stroke (mm)	105
Connecting Rod (mm)	176
Compression Ratio	14.3
Combustion System	Direction Injection
Injection System	Common rail

Table 4-1 Single Cylinder Research Engine Geometrical Specification

SINGLE CYLINDER I-TEC ENGINE SPECIFICATIONS

The experimental data shown here was collected from the Clean Diesel Engine Laboratory (University of Windsor). The operating conditions for the EGR Sweep are as follows: Engine speed = 1200 rpm, IMEP = 8 bar, Boost = 1.4 bar, Injection pressure = 1200 bar and CA 50 = 368 CA. The range of the EGR sweep is from 8% to 43% depending on the engine operating condition examined.

The detailed information of governing equations for heat release rate calculation was presented in Chapter 2. Figure 4-2 illustrates the heat release fraction with different m values between 0.1 and 3 for the combustion duration of 8 crank angle degrees. With higher m value, the combustion is retarded. The burning rate is shown in Figure 4-3 when the combustion duration was fixed at 40 crank angle degrees with different m values.

[CHAPTER 4]



Figure 4- 2 Heat release rate with different *m* value



Figure 4-3 Burn rates for constant combustion duration and different values of m

In the internal combustion engine analysis, the in-cylinder pressure has been considered to be an important experimental diagnostic in the development of automotive engine research [43]. The in-cylinder pressure profiles directly reflect the effects of in-cylinder heat release, heat transfer to the cylinder wall and head surface and work transfer. Figure 4- 4 and 4-7 present the predicted and measured in-cylinder pressure traces diagram when the EGR is 21% and 34% at speed of 1200 rpm. The computation time interval is 0.1 CA. It can be seen from the figure that the experimental data and simulation results appear to agree well with one another. After the first step of the analysis, it was verified that the model can effectively predict the in-cylinder pressure.



Figure 4- 4 Comparison of the calculated and experimental in-cylinder pressure (EGR=21%)

The good coincidence between the measured and calculated in-cylinder pressure was also achieved in other different EGR operating conditions. Appendix A shows the detail information about the comparisons.



Figure 4- 5 Comparison of heat release rate from experiment and simulation (EGR=21%)

[CHAPTER 4]



Figure 4- 6 Average in-cylinder temperature (EGR=21%)

The Figure 4-7 and Figure 4-8 illustrate the comparison of the computed and empirical results of the in-cylinder pressure and heat release rate when EGR equals to 34%. And they both follow the same trend as when EGR is 21%. Comparing the temperature results shown in Figure 4-6 (when EGR=21%), the mean cylinder temperature (shown in Figure 4-9) decreases due to the increase of EGR rate.



Figure 4-7 Comparison of the calculated and measured in-cylinder pressure (EGR=34%)



Figure 4-8 Comparison of heat release rate from experiment and simulation (EGR=34%)



Figure 4- 9 Average in-cylinder temperature (EGR=34%)

Figure 4-10 and Figure 4-11 show the species mole fraction of O_2 , CO_2 and H_2O during the combustion process with EGR=21% and EGR=34%.



Figure 4- 10 Species (O₂, CO₂ and H₂O) mole fraction (EGR=34%)



Figure 4- 11 Species (O₂, CO₂, and H₂O) mole fraction (EGR=21%)

Figure 4- 12 shows the comparison of the calculated nitric oxide concentration history for EGR=21% and EGR=34%. It presents that when higher EGR is applied, value of NO formation decreases.

Figure 4- 13 shows the soot density history for EGR=21% and 34%. And when the higher EGR is applied, soot emission increases.



Figure 4-12 Calculated and measured in-cylinder Nitric Oxide (NO) concentration



Figure 4-13 Calculated and measured in-cylinder soot concentration

EGR is one of the most effective ways to reduce NOx emission for diesel engines. From Figure 4- 14, it can be observed that the NOx decreases when higher EGR rate is applied under fixed operating condition. It is hard to simulate the diesel engine combustion process and precisely predict the emission. But for the control purpose, as it can be observed, this model presents high reliability to predict the trend and relative change in NO emissions within different operating conditions.



Figure 4-14 Calculated and measured in-cylinder Nitric Oxide (NO) formation

Figure 4-15 illustrates the comparison of the simulated and measured soot density when EGR rate increases from 15% to 45%.Soot formation is primarily determined by the engine load [9]. When the load increases, larger amount of fuel is injected into cylinders, which results in the increased temperature during the mixing controlled combustion [47].



Figure 4-15 Calculated and measured in-cylinder Nitric Oxide (NO) formation

CHAPTER 5 APPLICATION OF ESTIMATION RESULTS TO SMART DIESEL ENGINE

After the calculation results of thermodynamic model and NO and soot emission estimation have been validated, the computation results of the 8 operating points will be shown in this chapter for the development of the engine-generator set controller design. The good agreement achieved in Chapter 4 between the empirical and calculated results under certain engine operating condition confirmed that the estimation will provide valuable information for the engine-generator control design. Table 5-1 presents the specification of the smart engine used for the engine-generator control design.

Table 5-1 Smart Engine Specification

Engine Type	3 in line,4 Stroke Diesel Engine			
Bore (mm)	65.5			
Stroke (mm)	79			
Compression Ratio	18			
Cylinder Displacement(L)	0.8			
Rated Torque (lb-ft)	95			
Fuel Injection System	Common Rail High Pressure			



Figure 5-1 Smart Car Cdi Engine



Figure 5-2 Photo of engine-generator set (Courtesy Ricardo)

Table 5-2 shows the detailed information about the 8 operating points for the engine controller design. The speed ranges from 1600 rpm to 2800 rpm and the torque, output power and fuel flow all increase as well.

Operating	Engine	Torque	Power	Fuel Flow	Efficiency	BMEP	CA 50
Point	Speed (rpm)	(Nm)	(kW)	(g/s)	(%)	(bar)	
#1	1600	42	7	0.51	0.33	6.6	365
#2	1950	51	10	0.72	0.345	8.0	365
#3	2150	58	13	0.89	0.35	9.1	365
#4	2250	67	16	1.07	0.352	10.5	365
#5	2450	75	19	1.31	0.351	11.8	365
#6	2570	82	22	1.52	0.345	12.9	365
#7	2630	90	25	1.75	0.337	14.1	365
#8	2800	95	28	2.05	0.324	14.9	365

 Table 5-2
 Detailed Information about 8 Operating Points

The following figures show the benefit of EGR in reducing of NOx emission for the eight operating points. It can be seen that with the increase of EGR rate, the NO emission decreases under all of the eight different operating conditions. In general, there are two reasons which can explain relationship between NOx formation and EGR rate. The first reason for NOx emission reduction with higher EGR is because that the higher the values of the EGR rate that is used, the lower the in-cylinder temperature during combustion process. The second reason is the reduction of oxygen concentration which restrains NOx generation. As mentioned previously, NOx emission is very sensitive to temperature and oxygen concentration.

To study the EGR effect on the engine performance and emission, for each operating point, certain parameters will be fixed, such as engine speed, injected fuel quality, intake pressure and temperature.

Figure 5-3 shows the typical effect of EGR on the NOx formation for operating point 1, 2, 3 and 4. For operating point 1, NOx formation is reduced drastically with 12% EGR. For operating point 2, it is reduced by approximately 70% within 15% EGR.



Figure 5-3 Relationship between EGR and NOx formation (for Operating Point 1-4)

From the simulation results, we can see that from operating point 5 to 8, the NO emissions are relatively high. Thus we require a higher EGR. For example, with the increase of the amount of fuel injection, for operating point 5, when EGR is 20%, the NO emission is around 1200 ppm, but for operating point 3, when EGR is increased to 20%, there is almost no NOx present.

Figure 5- 4 shows the relationship between NO formation and EGR rate of operating point #5, #6 and #7 when boost is increased from 1.6 bar to 2 bar. Comparing with Figure 5- 3, in the case of operating point #5, 6 and 7, the engine is running in a higher load and speed. Therefore higher EGR rate and boost are required.



Figure 5-4 Relationship between EGR and NOx formation (for Operating Point 5-7)



Figure 5- 5 Relationship between EGR and NOx formation (for Operating Point 8)

Figure 5- 6 and Figure 5- 7 show the calculation results for soot emissions which all follow the same trend with various EGR rates. With the increase of EGR, soot emissions increase. This is because when the amount of fuel injected into cylinder increases, more fresh air is required to complete the combustion process. As it is shown in Figure 5- 6, for operating point 1, the fuel injection is around 38.1 mg/cycle, when EGR is 17% soot emission is around 2 mg/m³. But when the fuel injection increases to 64 mg/cycle, when EGR is increased to 19%, the soot emission is already over 6 mg/m³.



Figure 5- 6 Relationship between EGR and soot density (for Operating Point 1-4)



Figure 5-7 Relationship between EGR and soot density (for Operating Point 5-7)

For operating point #8, the engine is working at 2800 rpm and the fuel injection quantity is increased to 87 mg/cycle. Figure 5- 8 and Figure 5-5 present the variation of soot and NOx formation with various EGR rates. When higher boost is applied to engine, higher NOx and lower soot density are achieved. The reason for the increase on the NOx formation and the reduction of soot formation is owing to excess air.



Figure 5-8 Relationship between EGR and soot density (for Operating Point 8)

Figure 5- 9 and Figure 5-10 illustrate the in-cylinder pressure trace calculated under the condition of operating point 3 and 8 with different EGR rates. It shows the peak of in-cylinder pressure has decreased with the increase of EGR rates and the combustion timing has been retarded in these two cases.



Figure 5-9 In-cylinder pressure traces with different EGR rates (Operating Point 3)



Figure 5-10 In-cylinder pressure traces with different EGR rates (Operating Point 8)



Figure 5-11 Heat release rate with different EGR rates (Operating Point 3)

Figure 5- 12 illustrates the effect of EGR on ignition delay. The ignition delay increases with increasing EGR rate. The reason is that higher EGR results in lower oxygen concentration in the intake mixture, which will slow down the oxidation processes and increase ignition delay.



Figure 5-12 Effect of boost on ignition delay

Figure 5- 13 illustrates the effect of the intake oxygen concentration and injection timing on the in-cylinder peak pressure for operating point 1. With the same injection timing, the in-cylinder peak pressure increases when intake oxygen concentration increases. For the in-cylinder combustion, higher oxygen concentration is equivalent to lower CO₂ concentration, which results in the decrease of specific heat capacity and the increase of in-cylinder temperatures. Therefore the in-cylinder peak pressures increase. With the same intake oxygen concentration, the in-cylinder peak pressure decreases when the injection timing is retarded. For example, when the injecting timing is at -9 CA ATDC and the oxygen concentration is below 18%, combustion did not occur. But when the injecting timing is at -20 CA ATDC, the low limit of oxygen concentration is 17%.



Figure 5-13 Effect of intake Oxygen concentration on the in-cylinder peak pressure

CHAPTER 6 CONCLUSION AND FUTURE WORK

6.1 Conclusion

The objective models for emission estimation for a series PEHV diesel engine-generator set control design, which combine the thermodynamic engine model, NOx model, soot model and EGR model described in Chapter One, have been developed. The heat release model was tuned to match the empirical data until the satisfactory results have been achieved, and the calculated in-cylinder pressure agrees well with the experimental data. Furthermore, the in-cylinder temperature, heat release rate, species mole fraction were calculated and provided as the inputs of the NOx and soot emission estimation models. Comparing to the previous work, the emission estimation model has been modified to take EGR system into account. Additionally, the NOx and soot emission estimation was validated and satisfactory for the Diesel engine-generator set control design. According to the implemented Model Predictive Control (MPC) design, eight operating points were selected to represent the diesel engine output power ranging from 7kW to 28kW. And the NO and soot emission results with EGR sweep for each operating point were provided for the diesel engine-generator controller design.

6.2 Future Work

In the validation part, more EGR sweep experiments shall be further conducted with various boundary operating conditions. Measured experimental data from Clean Diesel Engine Lab was satisfactory for certain limited operating range. More empirical data collected will bring improvement over the previous validation work.

The next stage of the research shall target on a two-zone thermodynamic engine model. NOx emission is very sensitive to the in-cylinder temperature and oxygen concentration. For the engine control application, the single zone dynamic engine model had been efficient to accomplish the estimation work. However, when the two-zone engine model is applied to the emission estimation model, it will provide more accurate prediction.

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APPENDIX A. ESTIMATION RESULT VALIDATION

The comparison of experimental and simulation results of in-cylinder pressure with EGR=21% and EGR=34% was shown in Chapter 4. More validation results with various EGR rates will be illustrated in this section.



Figure A-1 Comparison of the calculated and measured in-cylinder pressure (EGR=6.5%)

[APPENDIX]



Figure A- 2 Comparison of the calculated and measured in-cylinder pressure (EGR=12%)



Figure A- 3 Comparison of the calculated and measured in-cylinder pressure (EGR=15%)

[APPENDIX]





Figure A- 4 Comparison of the calculated and measured in-cylinder pressure (EGR=25%)

Figure A- 5 Comparison of the calculated and measured in-cylinder pressure (EGR=30%)

[APPENDIX]



Figure A- 6 Comparison of the calculated and measured in-cylinder pressure (EGR=35%)

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