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Synthetic methods for biologically relevant organofluorine compounds

Meher Perambuduru

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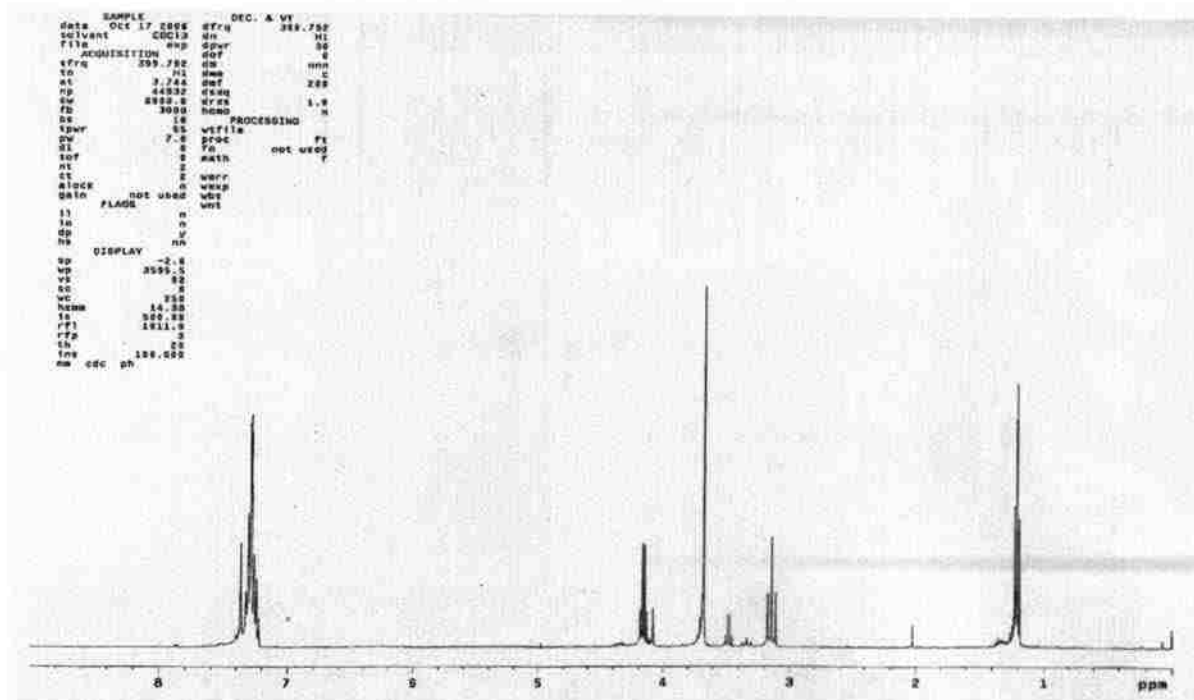
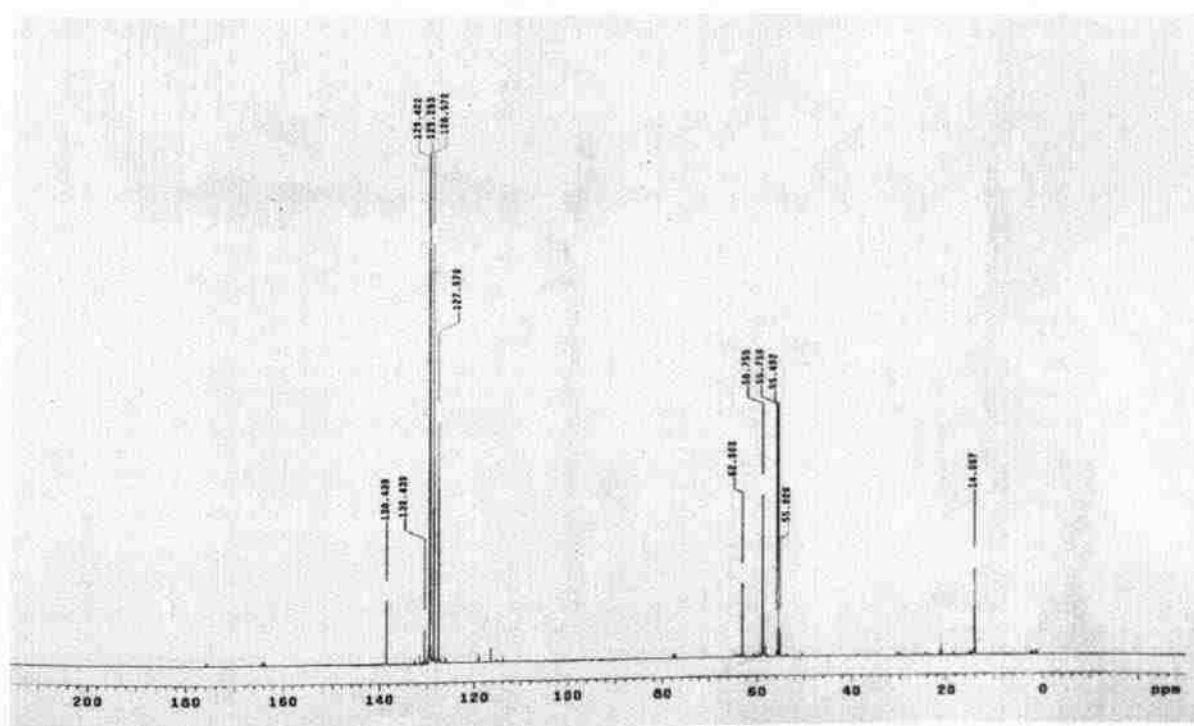
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1.4.3 NMR spectra of the products

Figure 1.33: ^1H NMR spectra of compound 1.80Figure 1.34: ^{13}C NMR spectra of compound 1.80

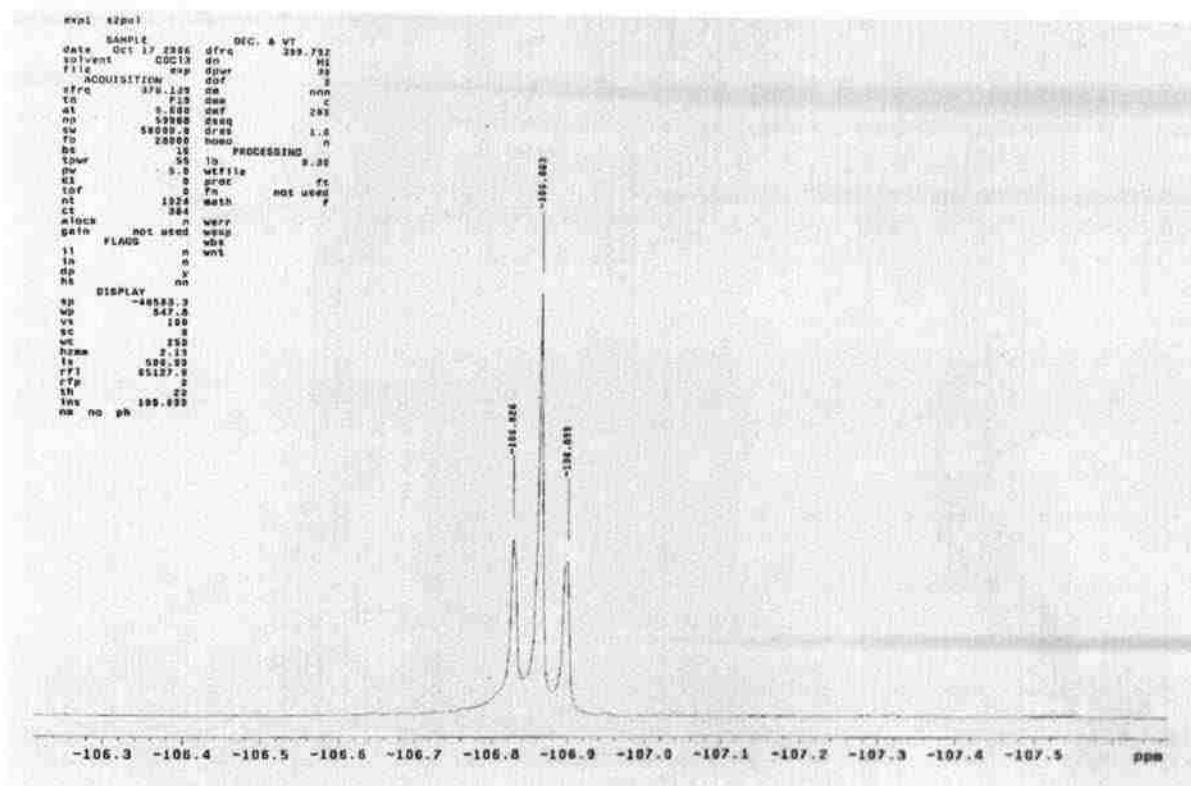


Figure 1.35: ^{19}F NMR spectra of compound 1.80

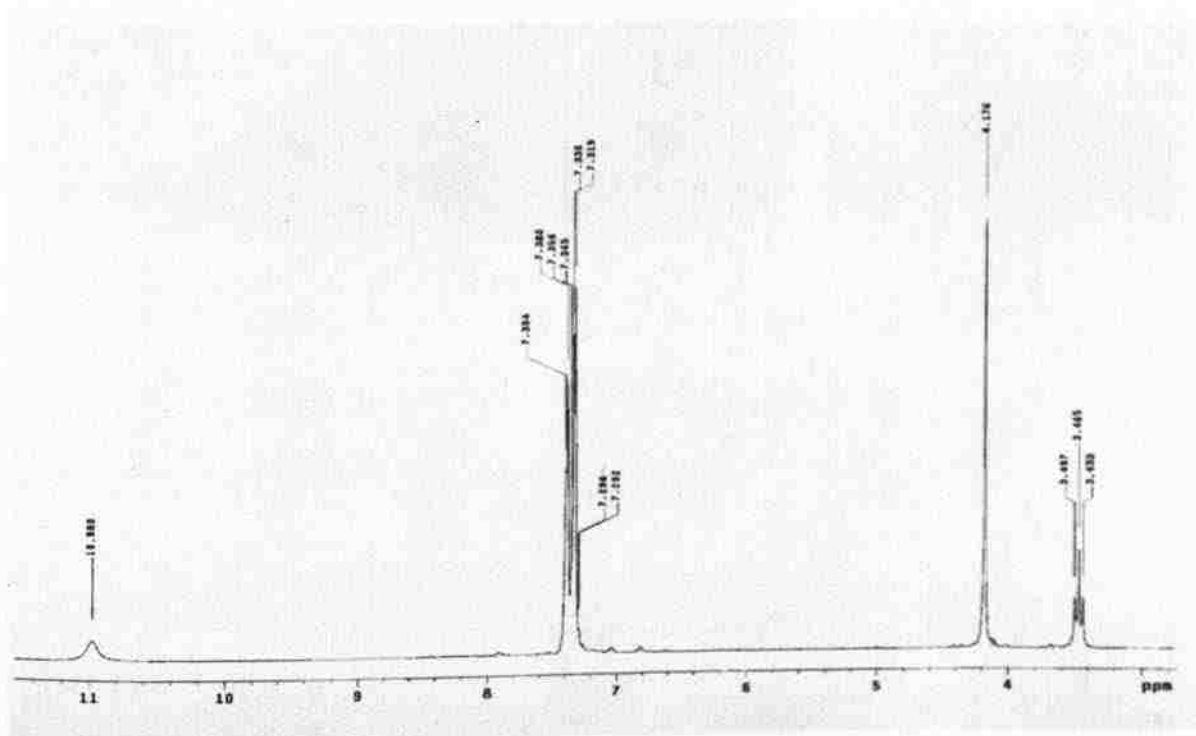


Figure 1.36: ^1H NMR spectra of compound 1.81

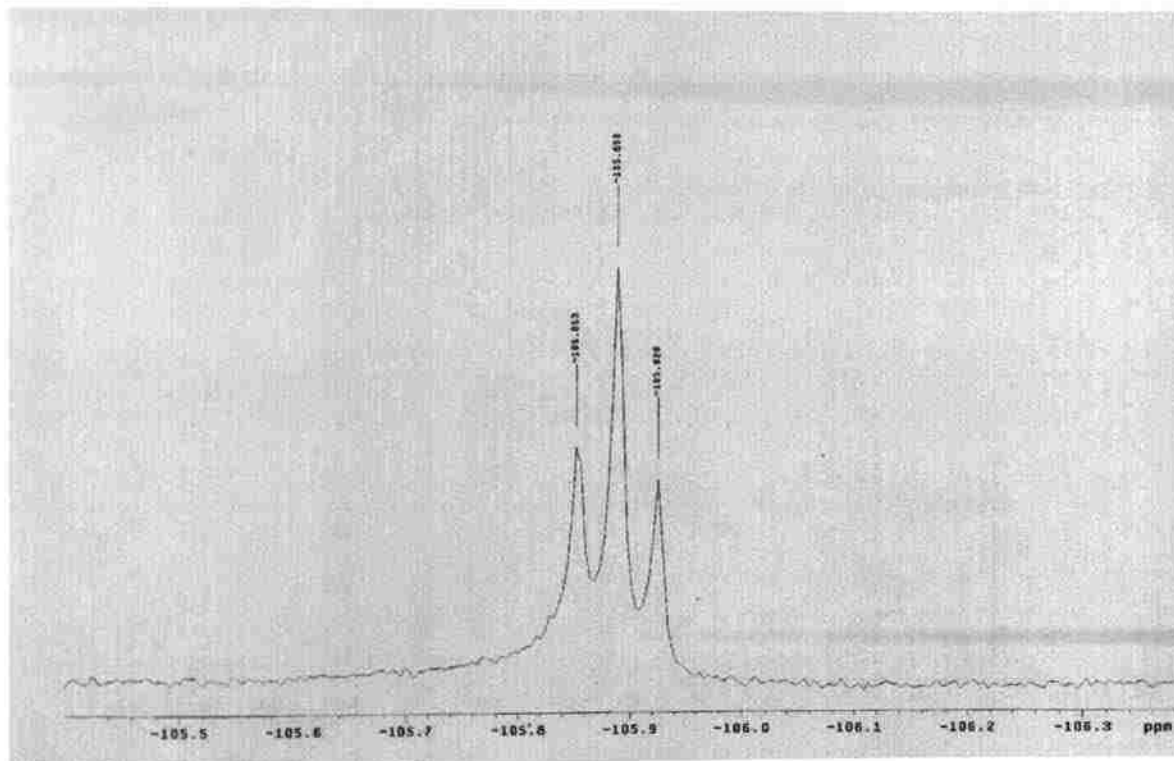


Figure 1.37: ^{19}F NMR spectra of compound 1.81

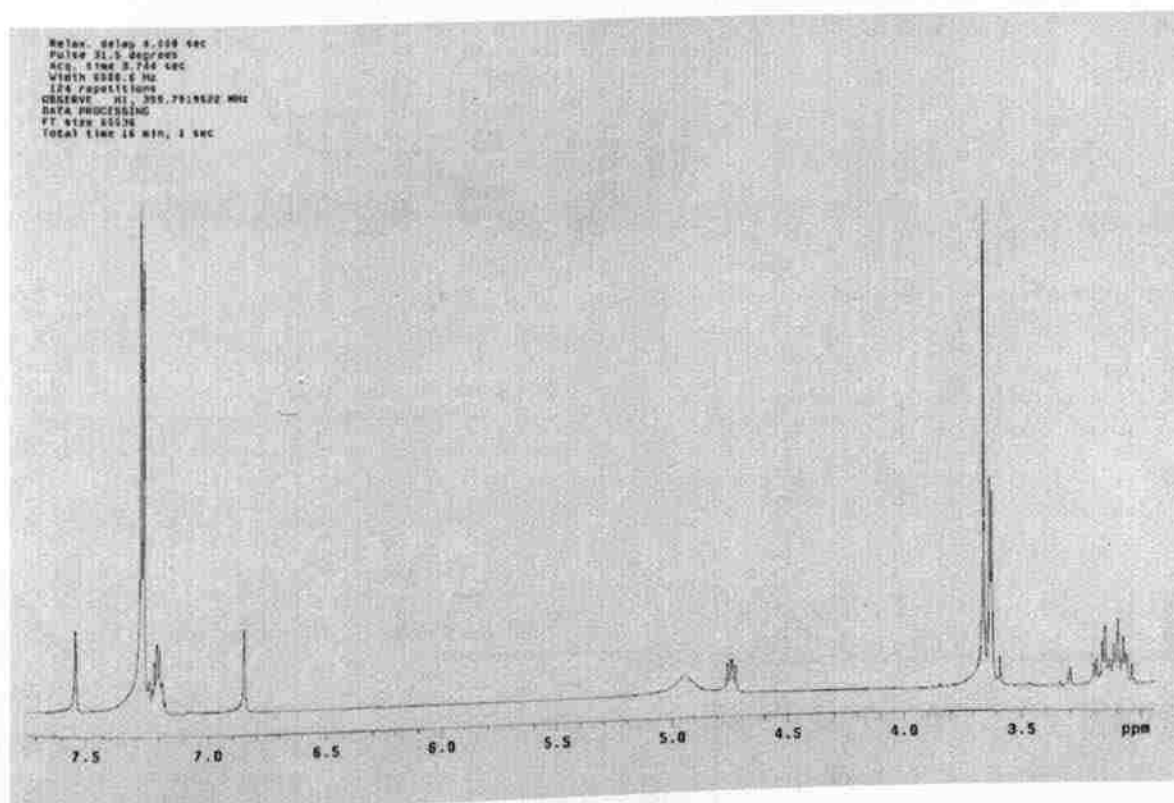


Figure 1.38: ^1H NMR spectra of compound 1.83

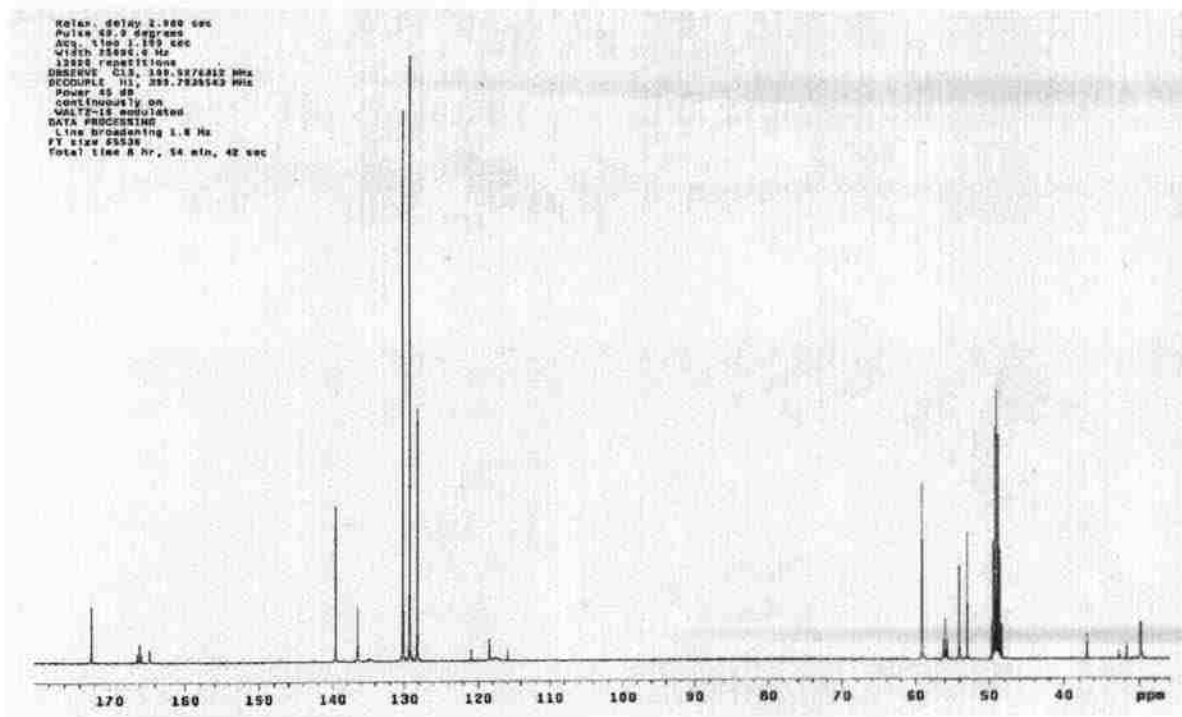


Figure 1.39: ^{13}C NMR spectra of compound 1.83

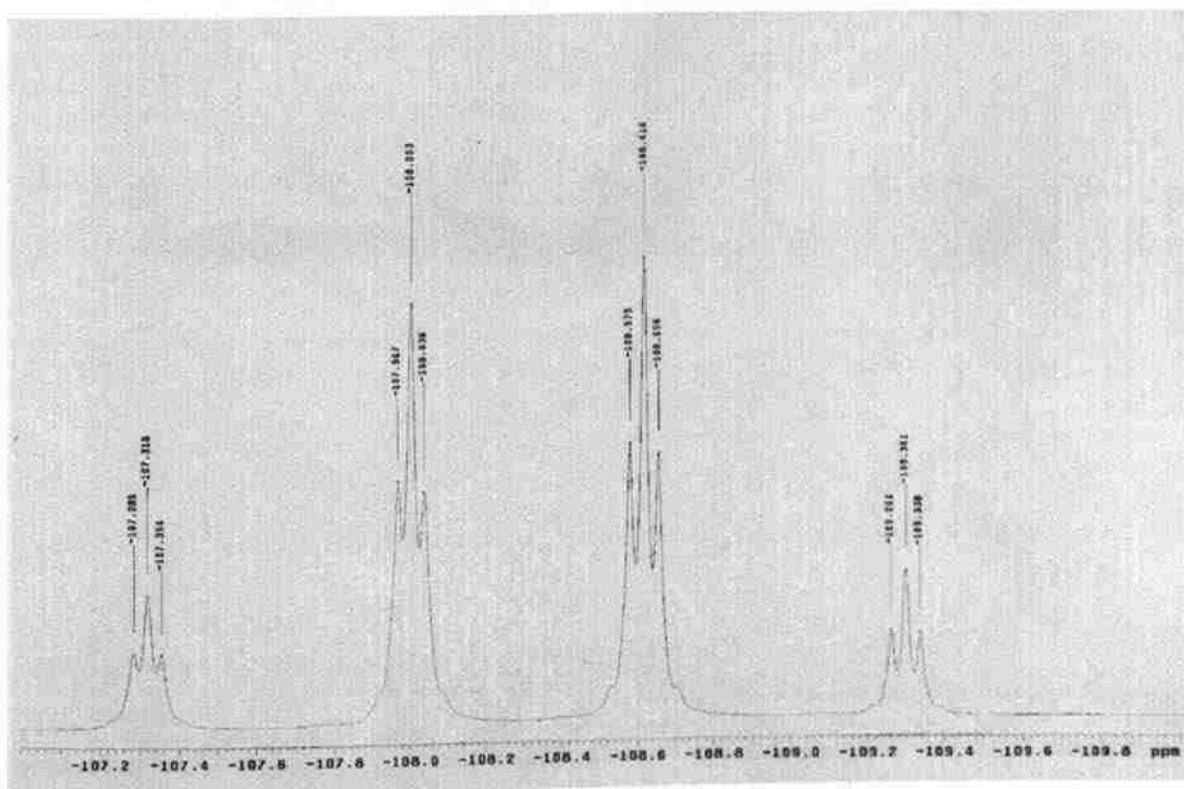


Figure 1.40: ^{19}F NMR spectra of compound 1.83

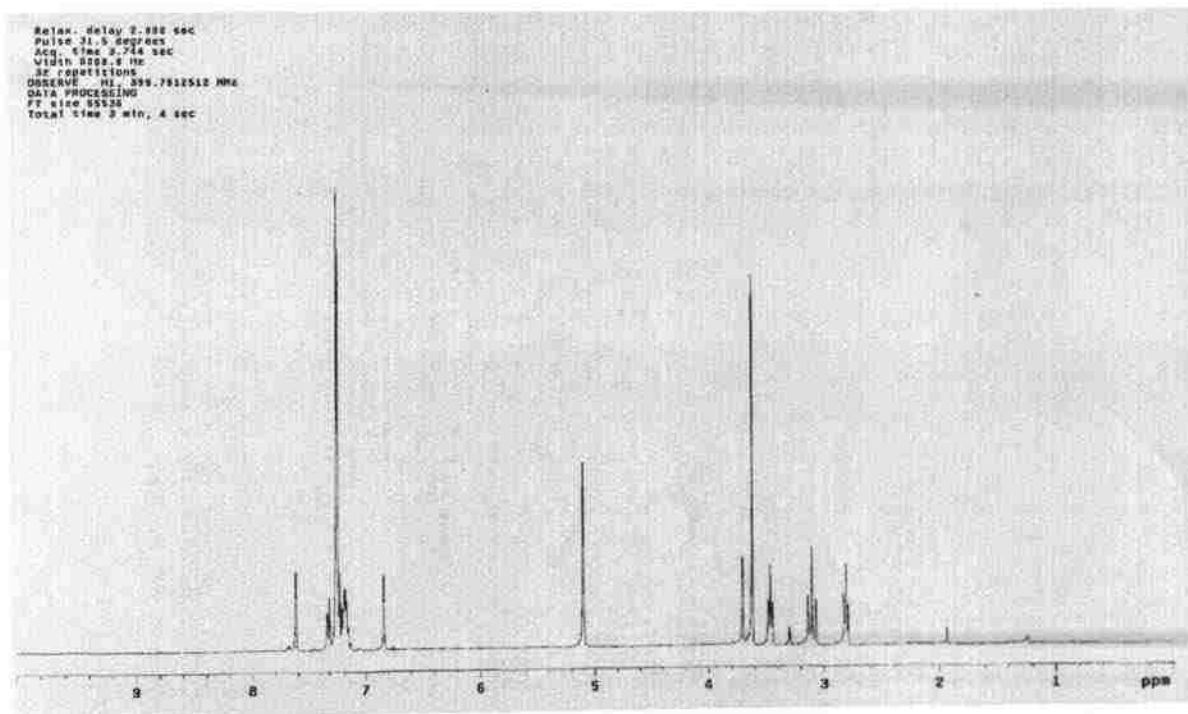


Figure 1.41: ^1H NMR spectra of compound 1.85

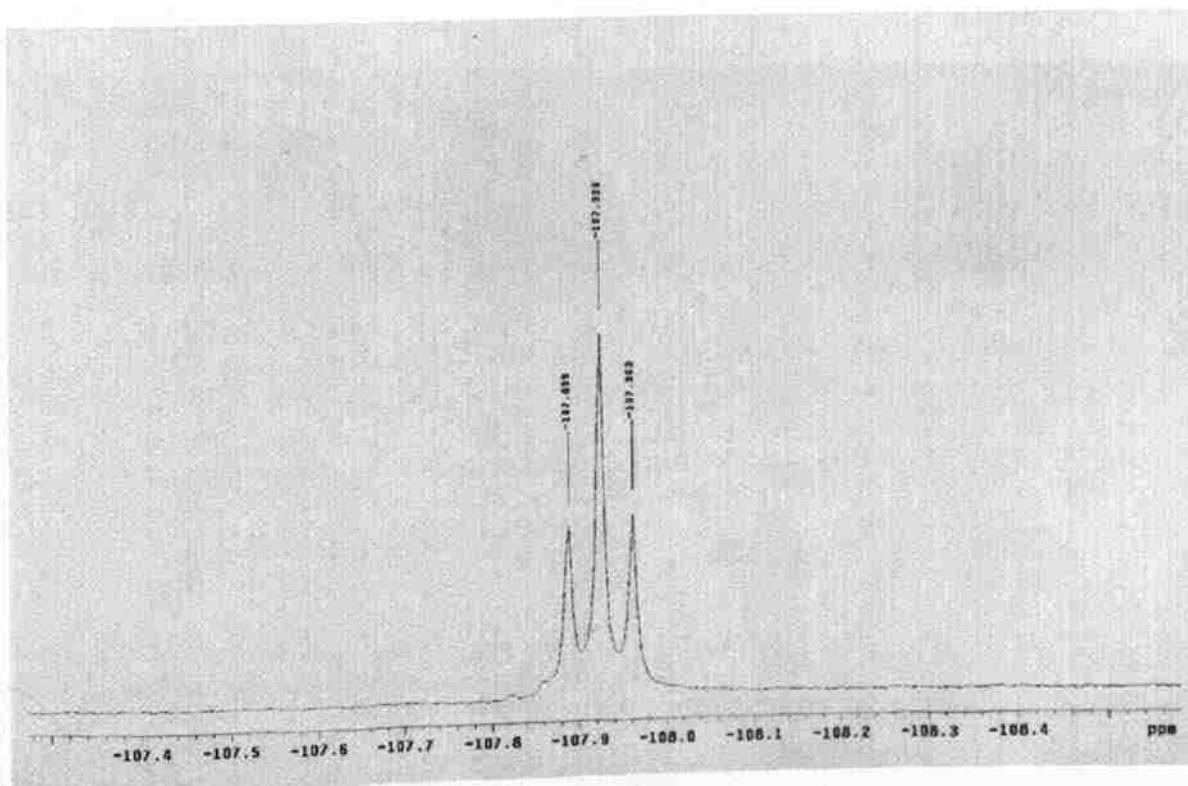


Figure 1.42: ^{19}F NMR spectra of compound 1.85

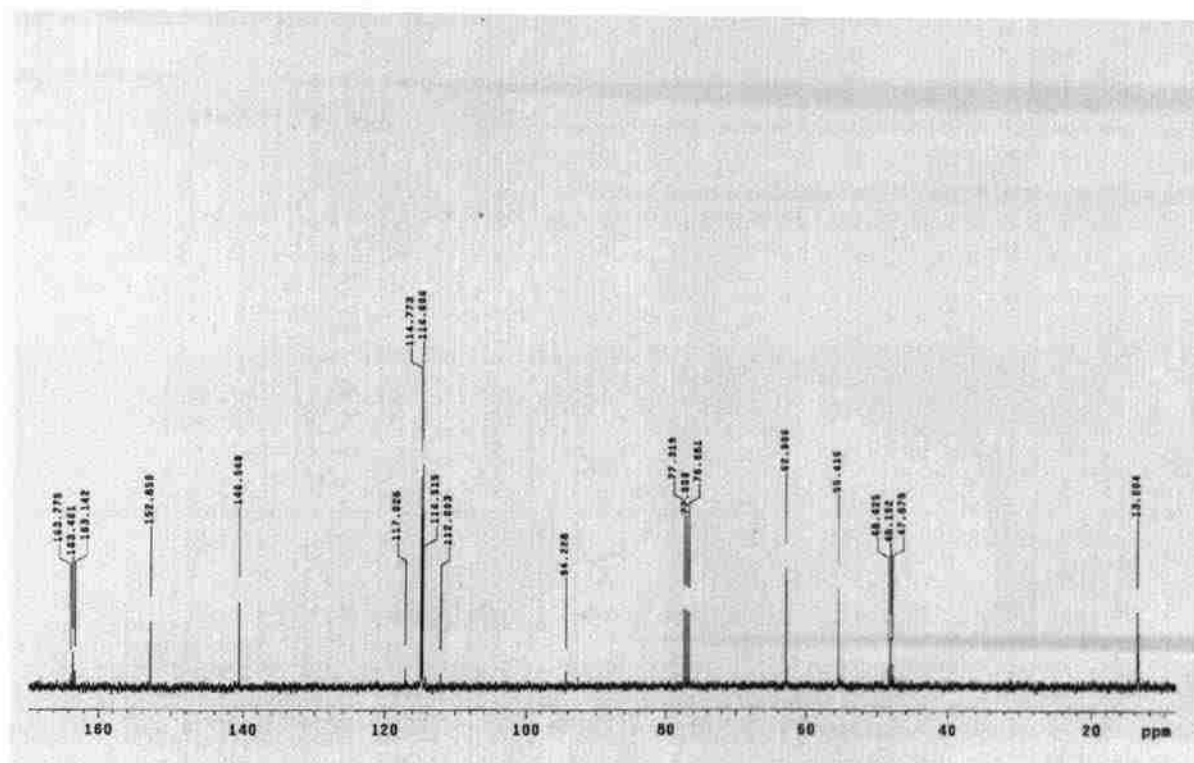


Figure 1.43: ^{13}C NMR spectra of compound 1.87

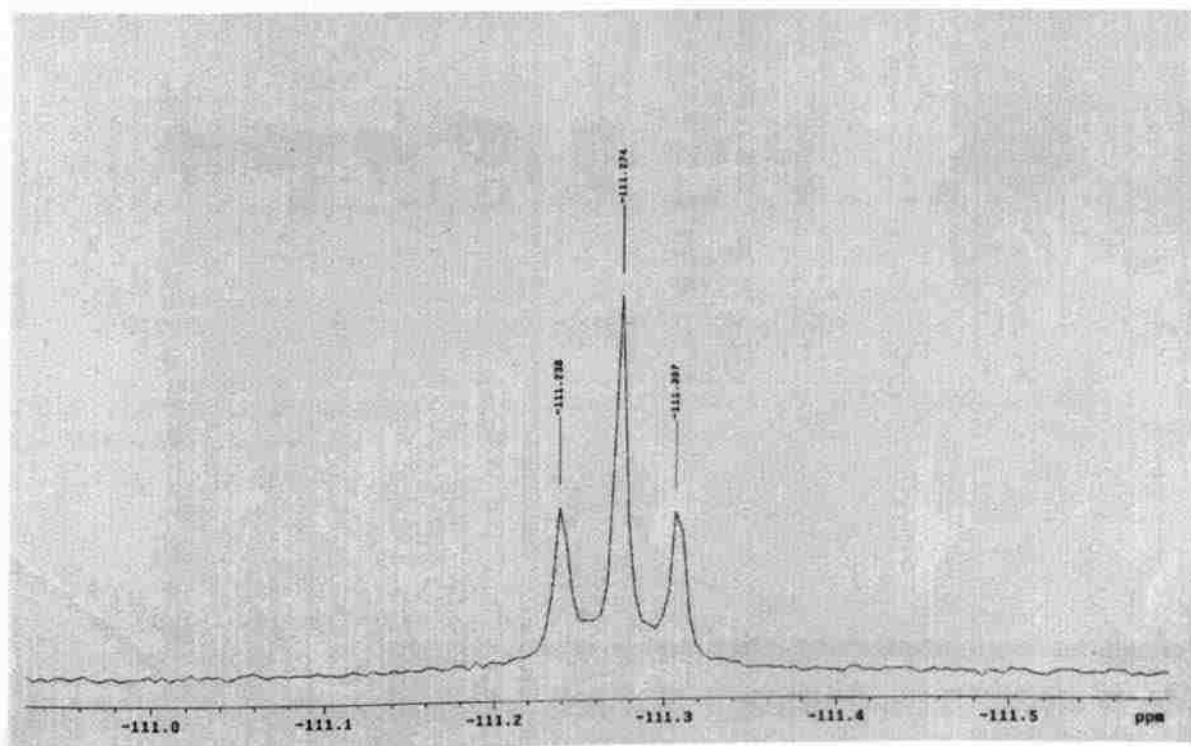


Figure 1.44: ^{19}F NMR spectra of compound 1.87

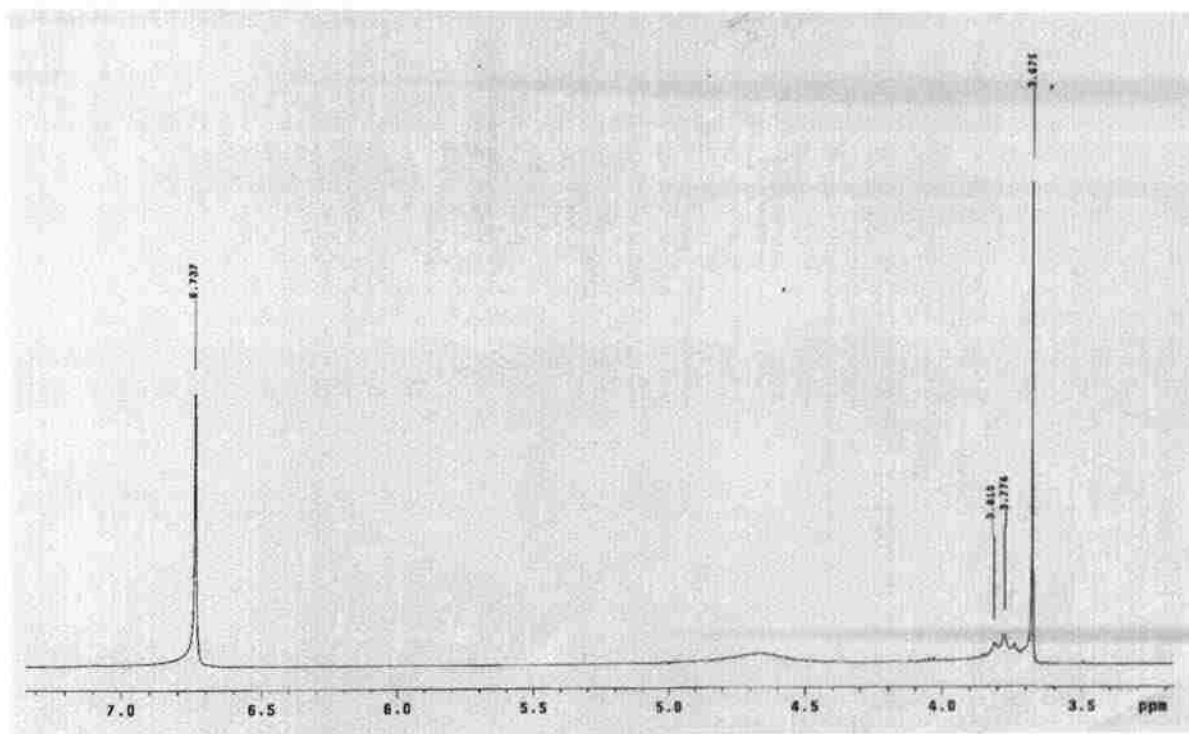


Figure 1.45: ^1H NMR spectra of compound 1.88

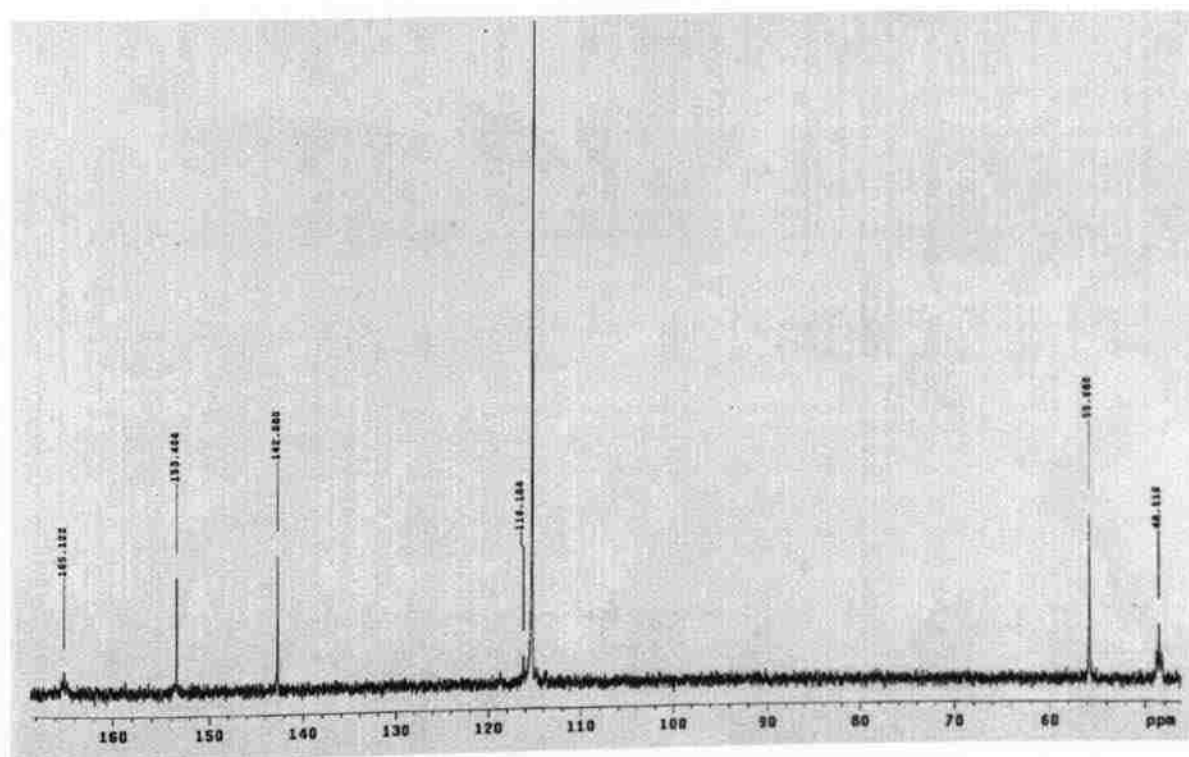


Figure 1.46: ^{13}C NMR spectra of compound 1.88

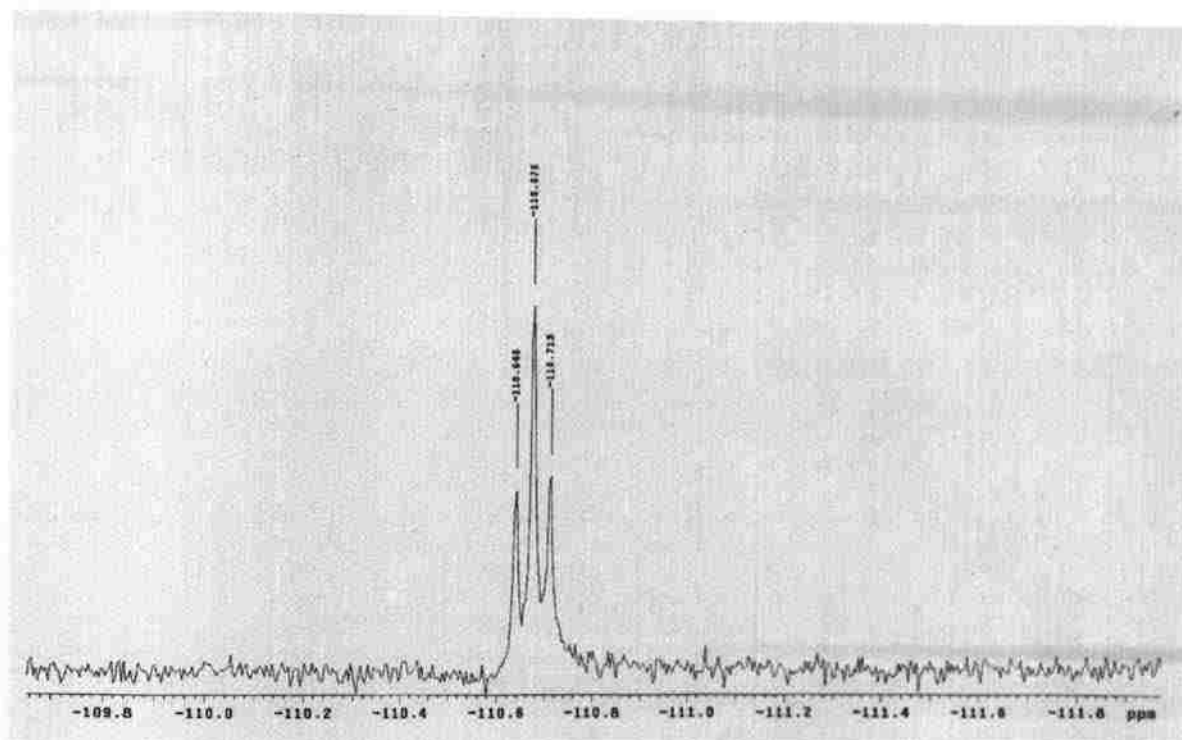


Figure 1.47: ^{19}F NMR spectra of compound 1.88

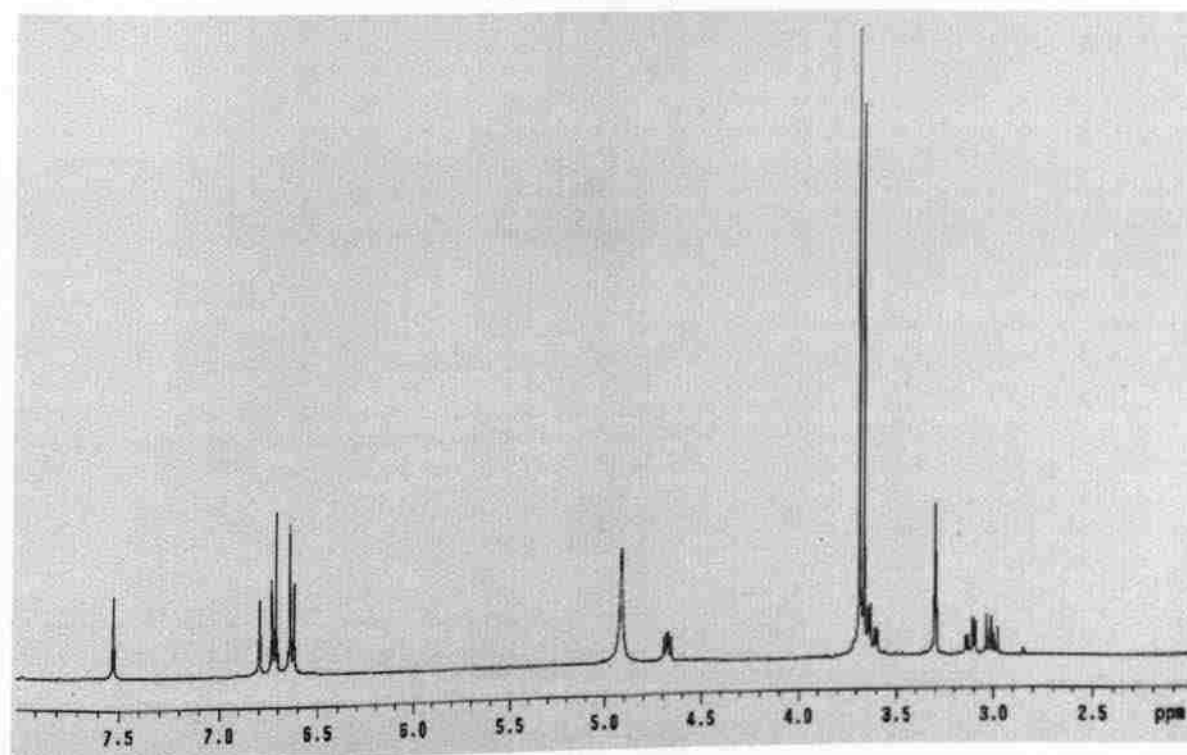


Figure 1.48: ^1H NMR spectra of compound 1.89

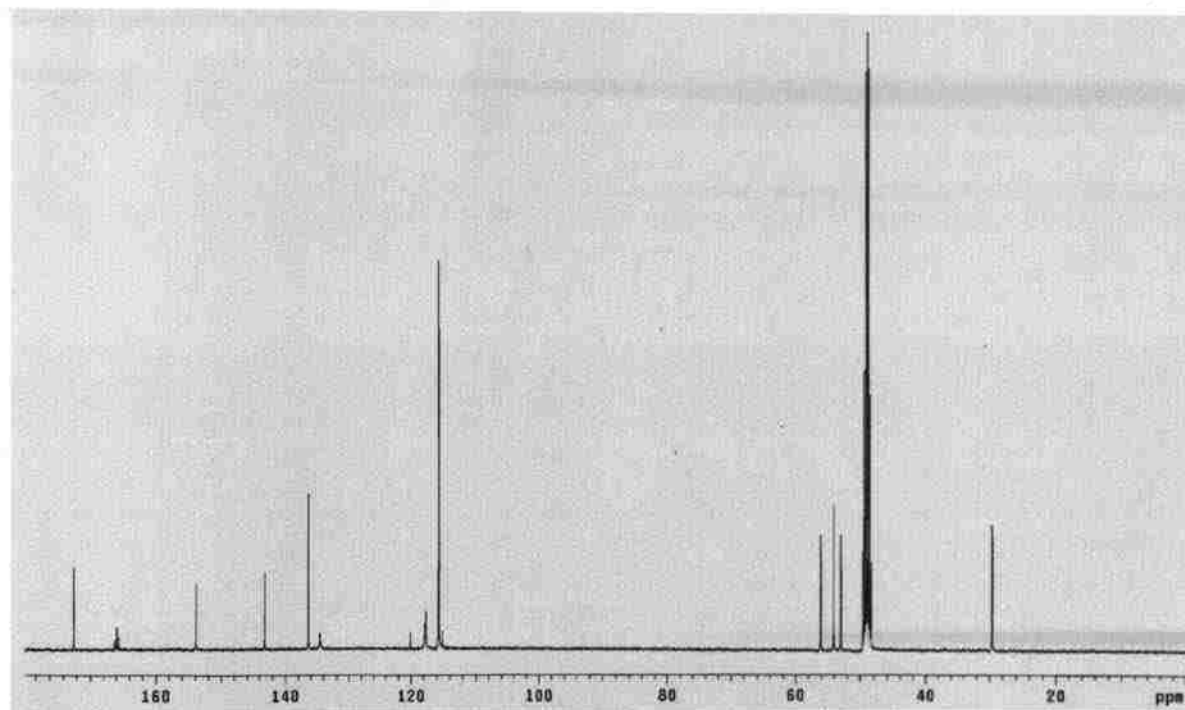


Figure 1.49: ^1H spectra of compound 1.89

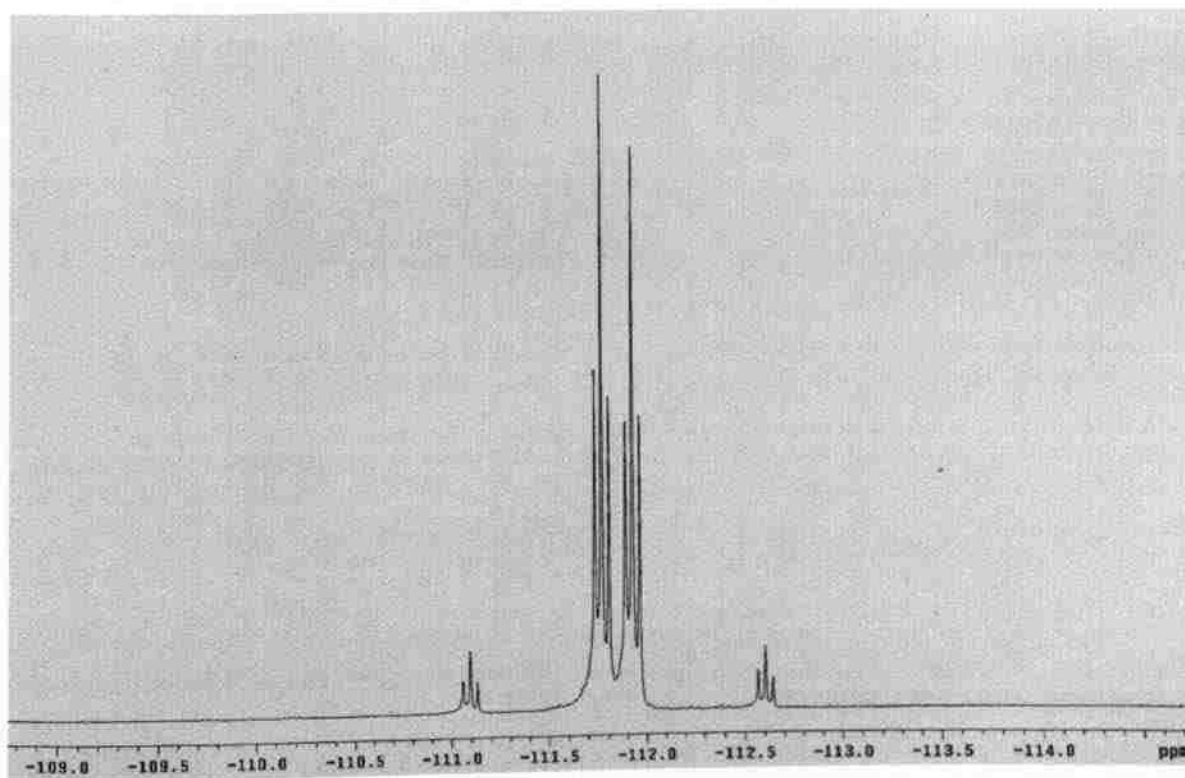


Figure 1.50: ^{19}F NMR spectra of compound 1.89

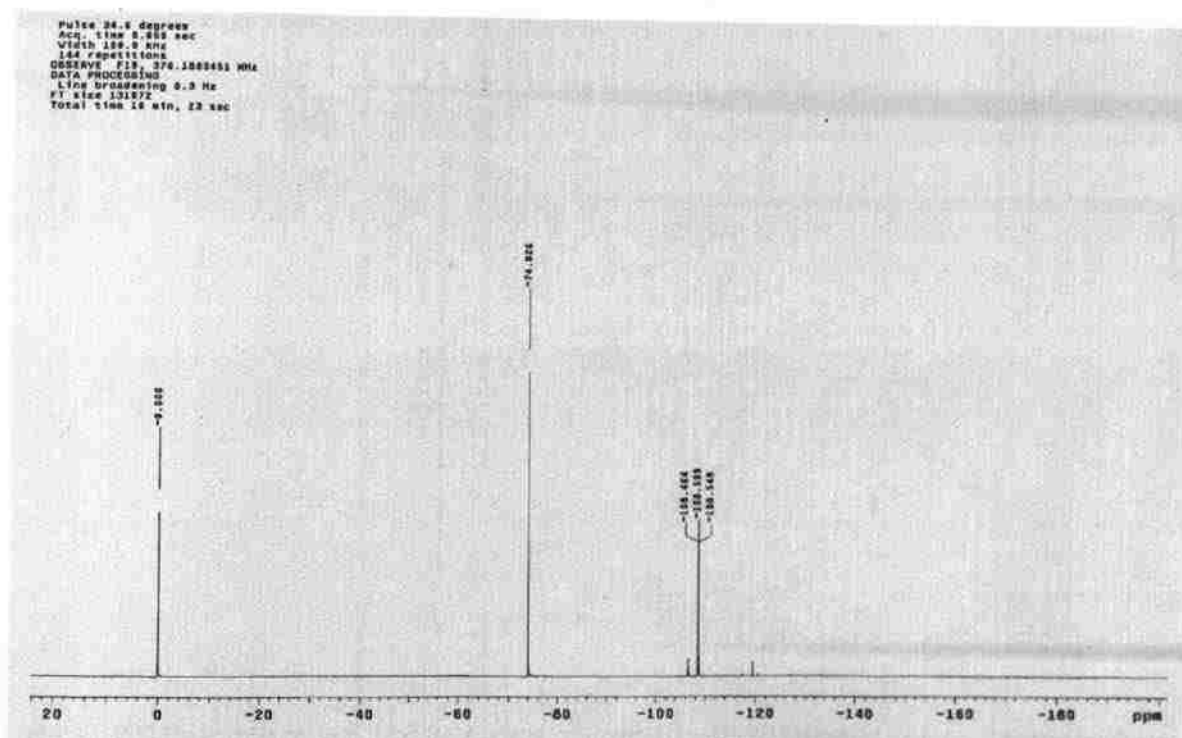


Figure 1.51: ^1H NMR spectra of compound 1.93

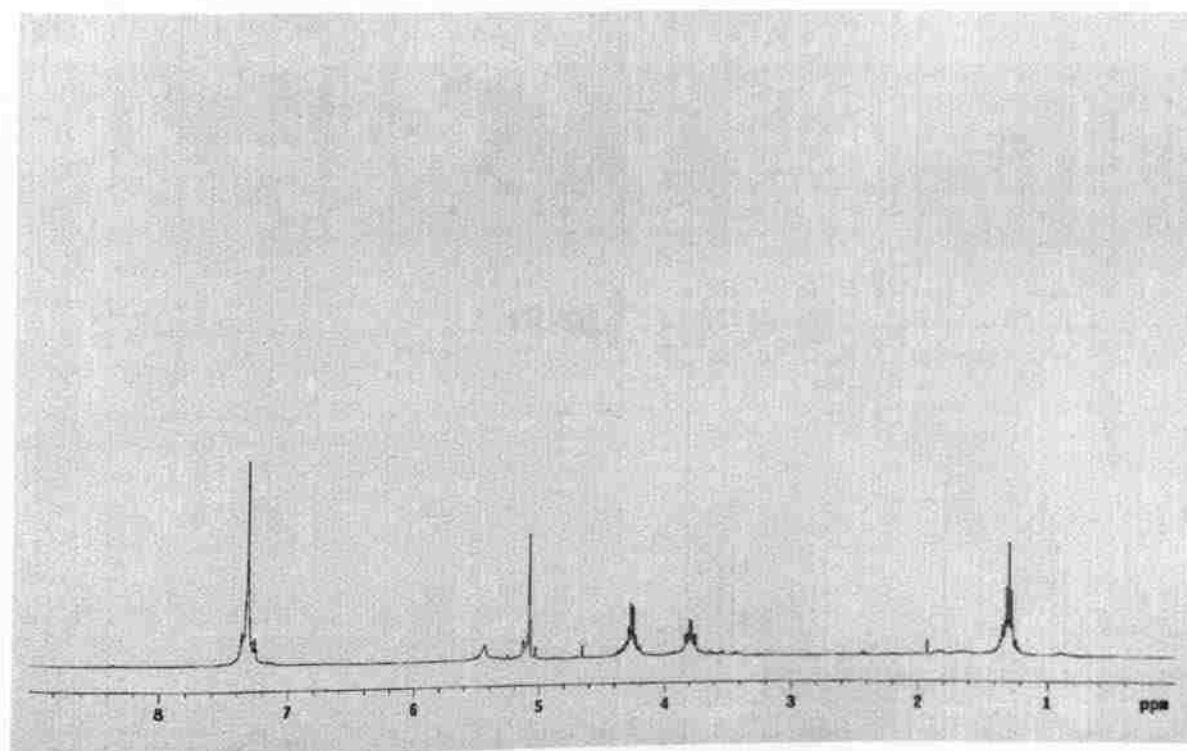


Figure 1.52: ^1H NMR spectra of compound 1.92

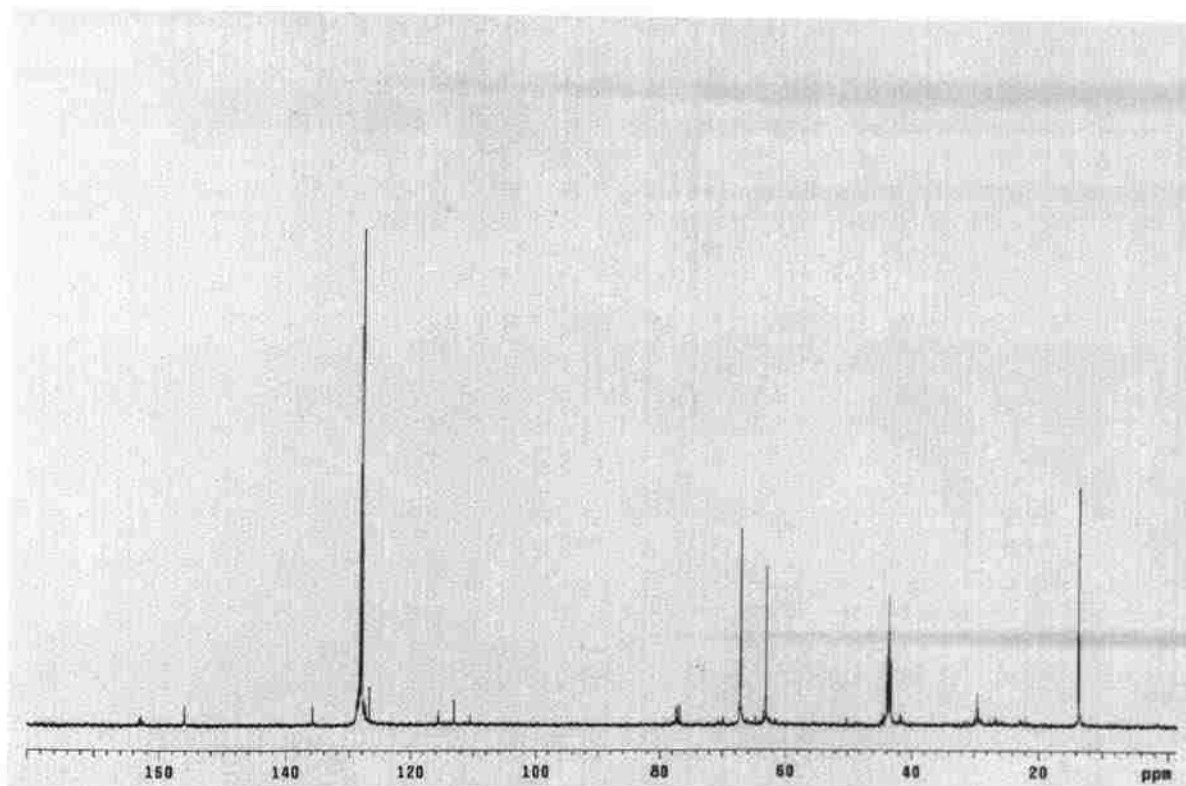


Figure 1.53: ^{13}C NMR spectra of compound 1.92

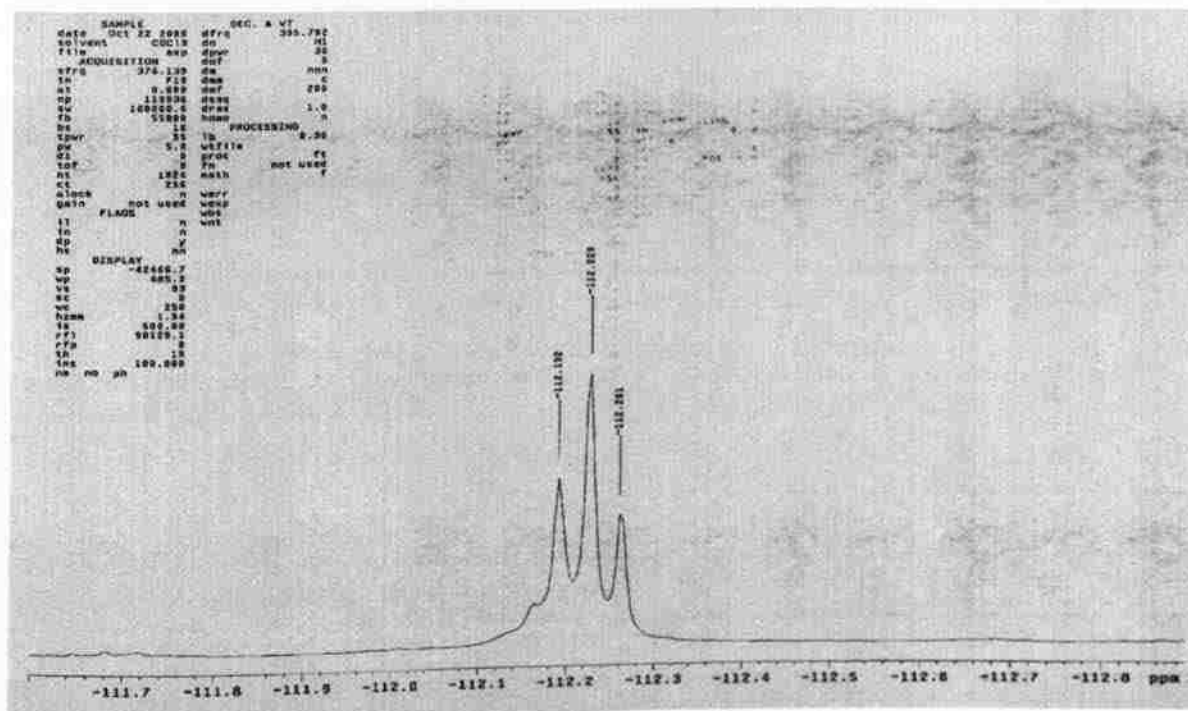


Figure 1.54: ^{19}F NMR spectra of compound 1.92

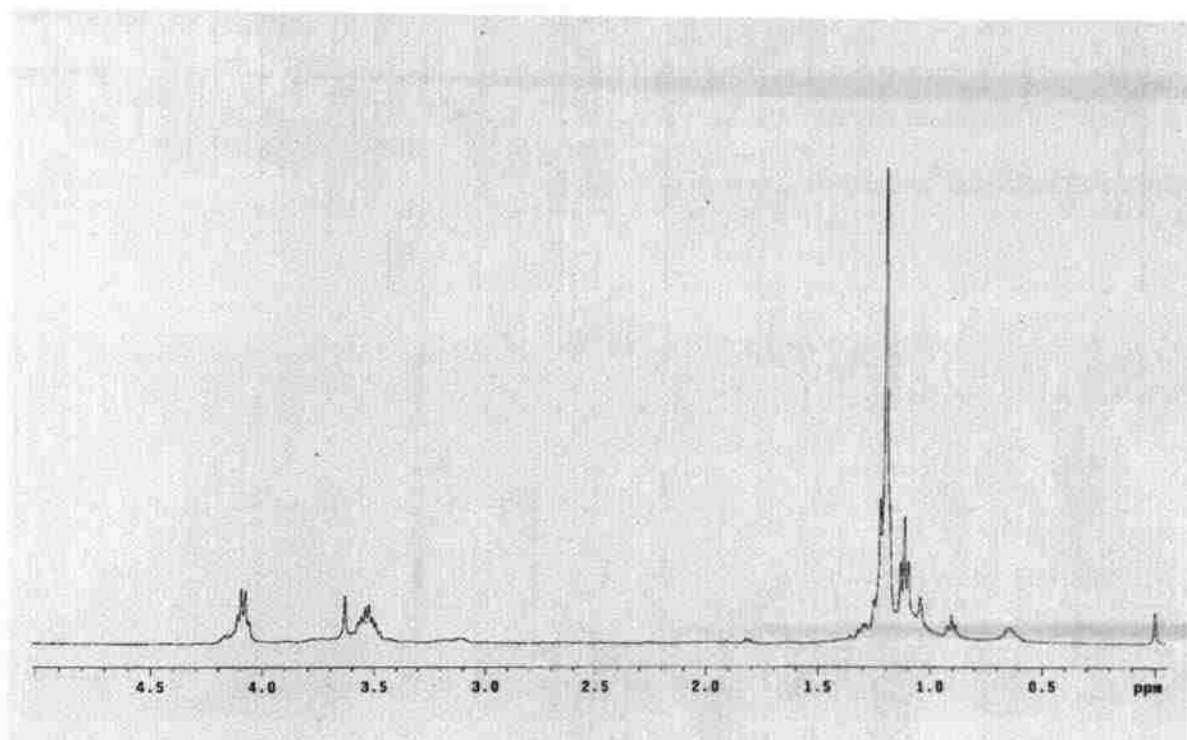


Figure 1.55: ^1H NMR spectra of compound 1.94

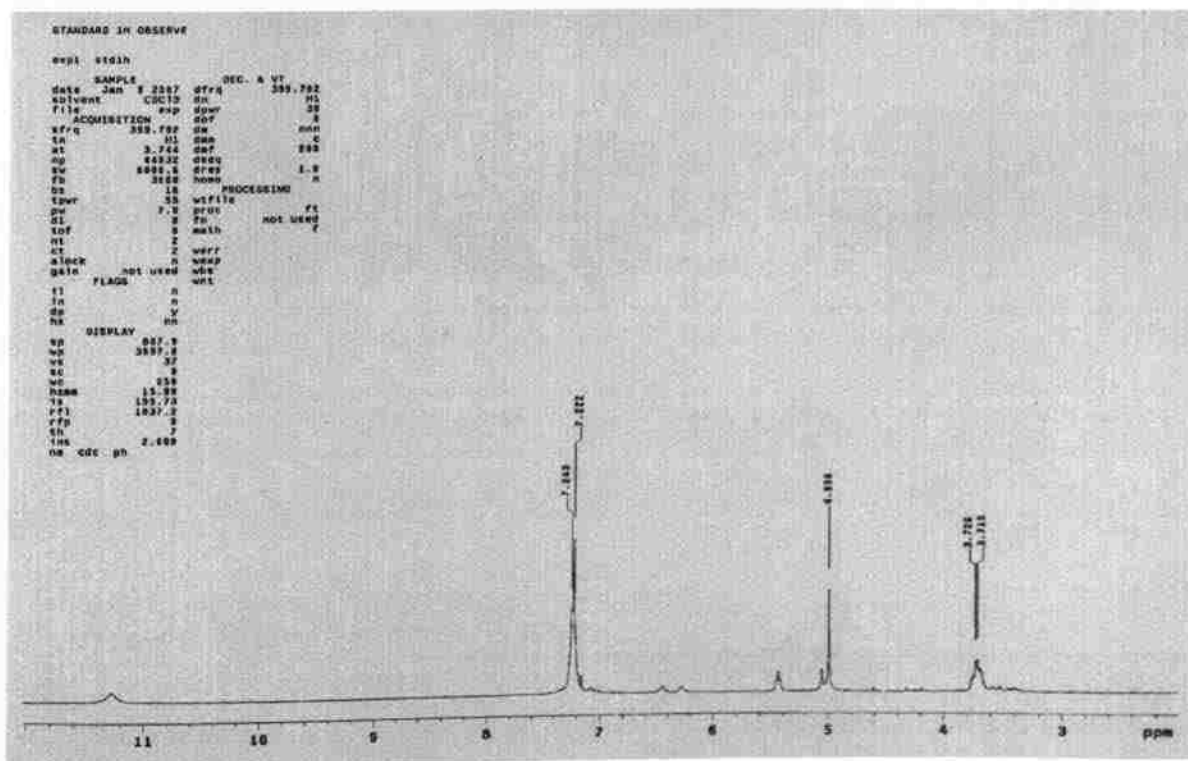


Figure 1.56: ^1H NMR spectra of compound 1.95

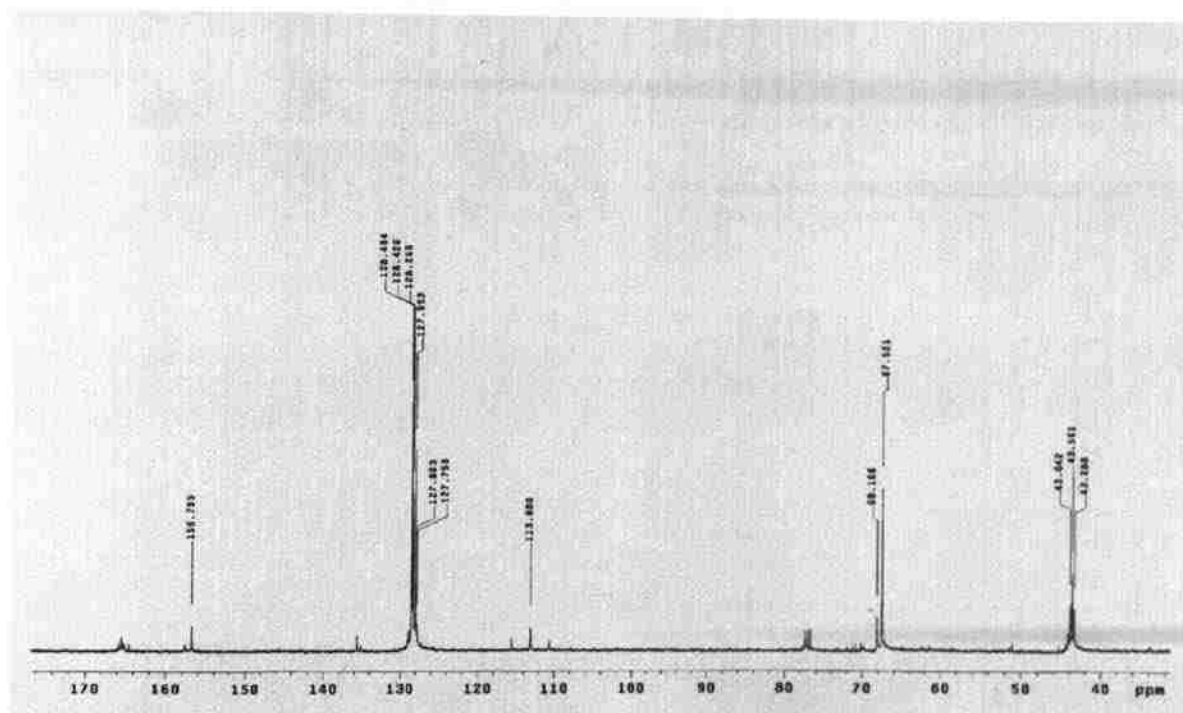


Figure 1.57: ^{13}C NMR spectra of compound 1.95

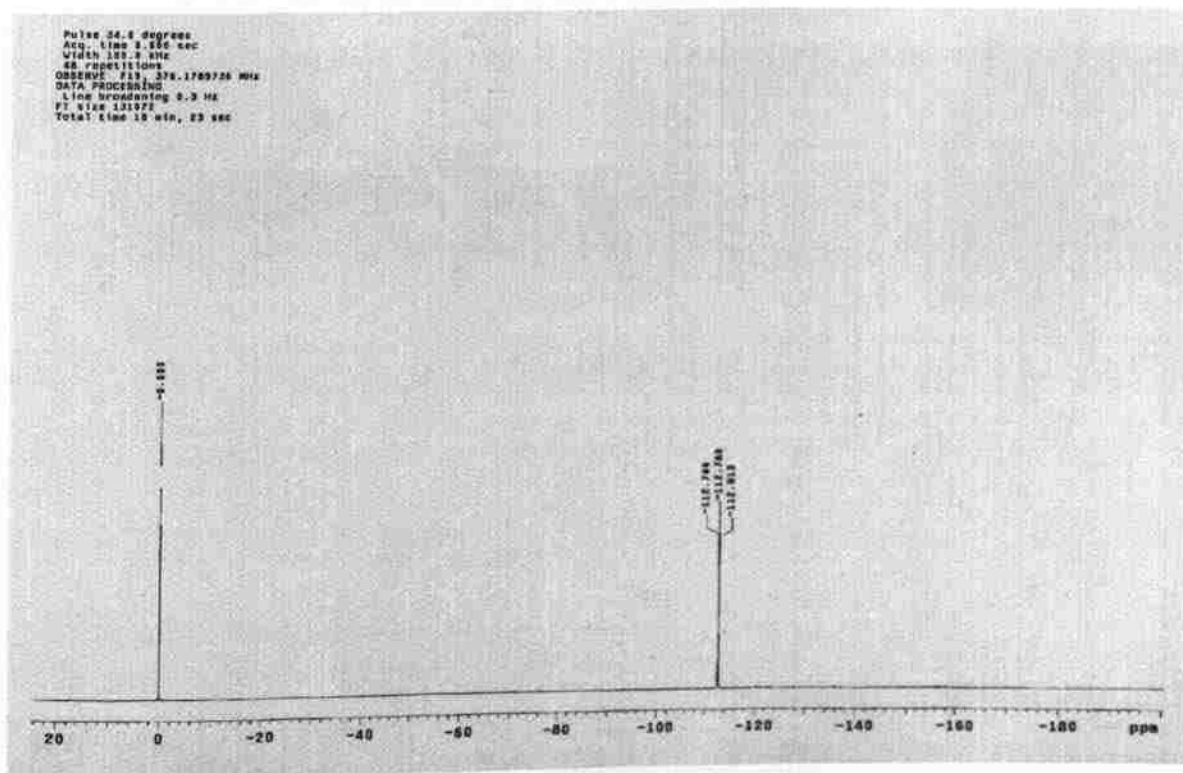


Figure 1.58: ^{19}F NMR spectra of compound 1.95

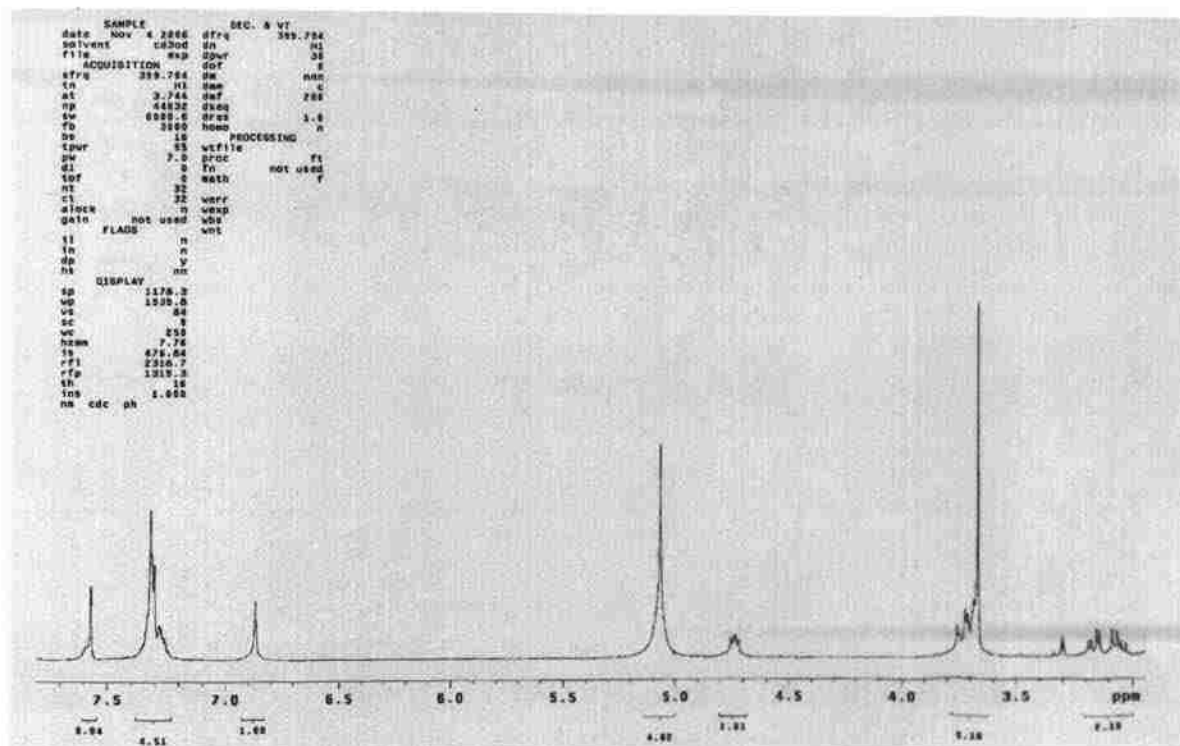


Figure 1.59: ^1H NMR spectra of compound 1.96

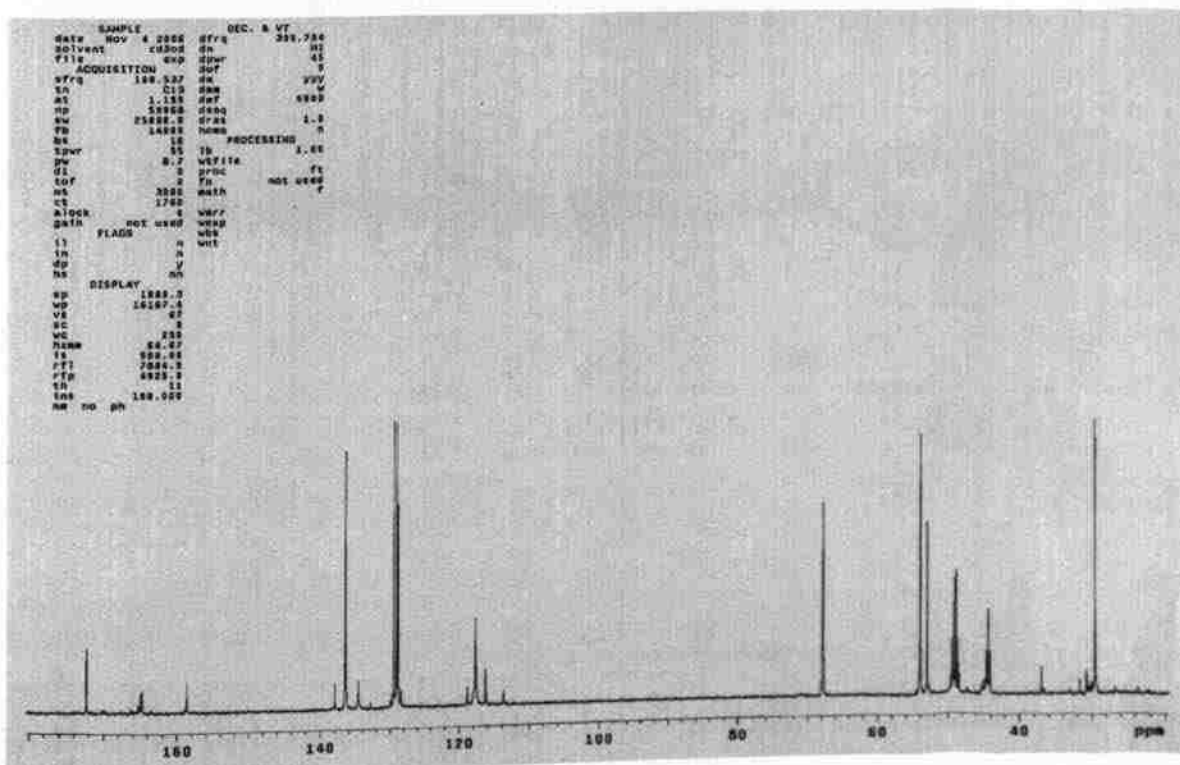


Figure 1.60: ^{13}C NMR spectra of compound 1.96

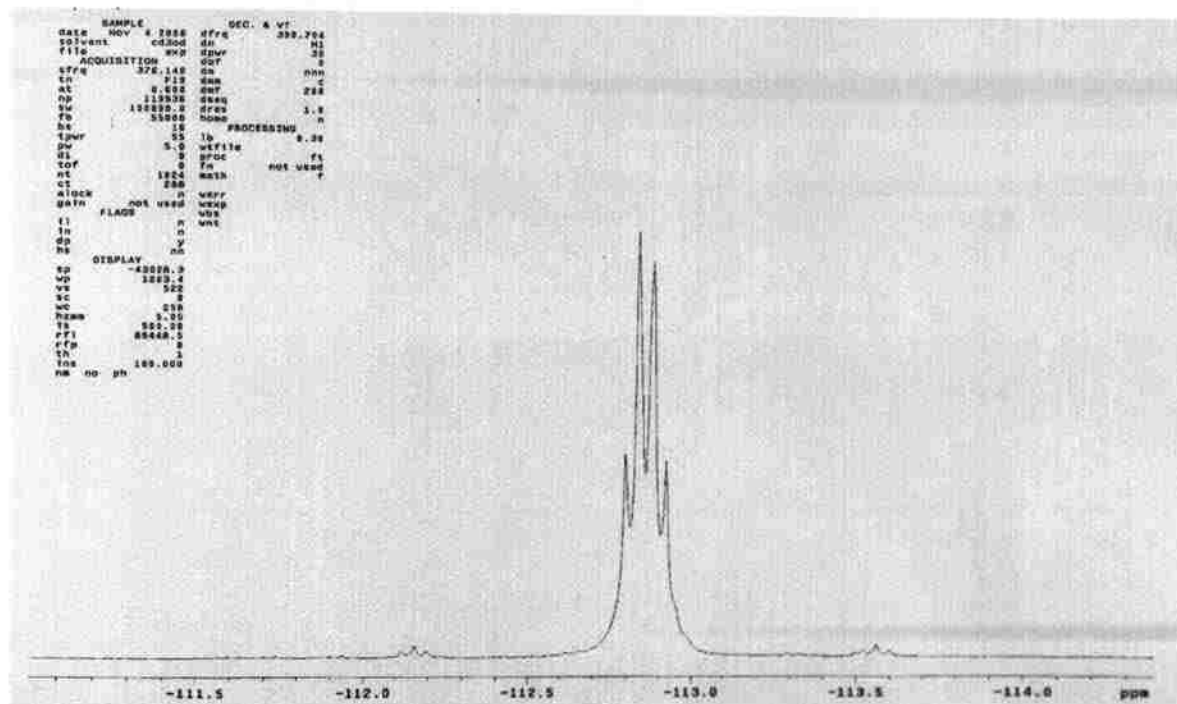


Figure 1.61: ^{19}F NMR spectra of compound 1.96

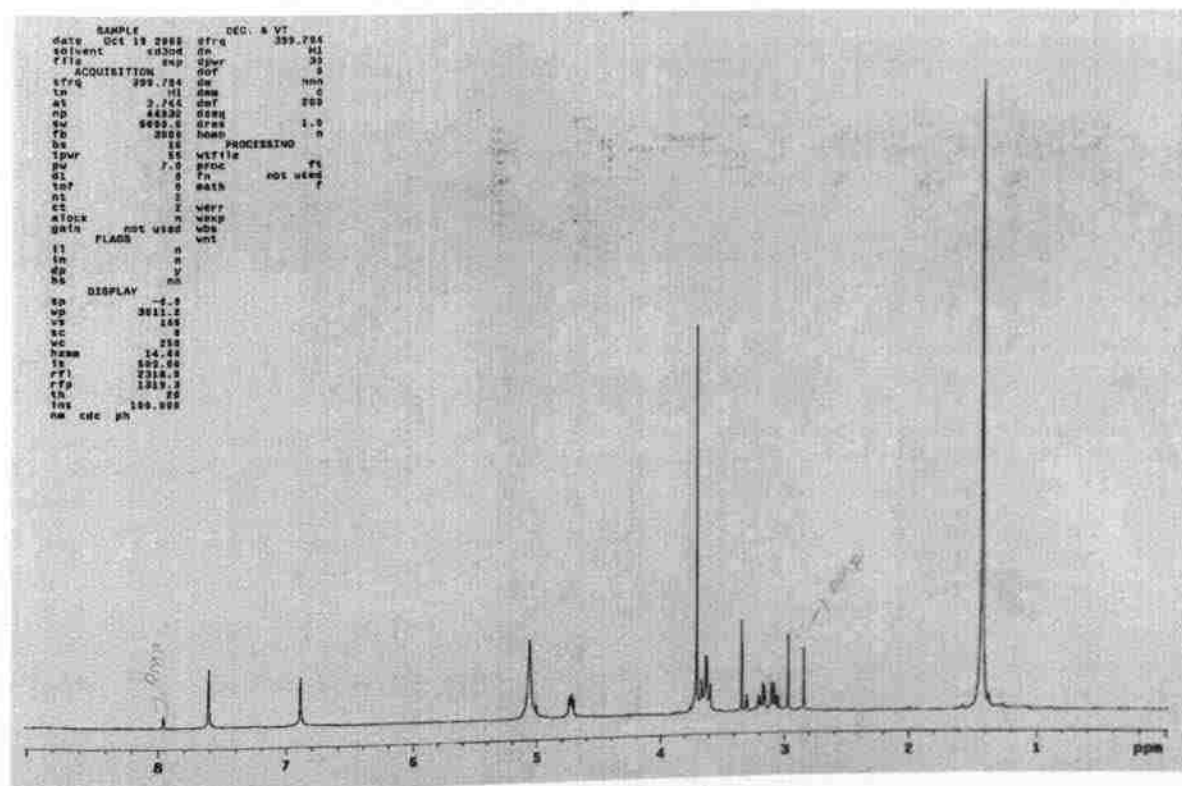
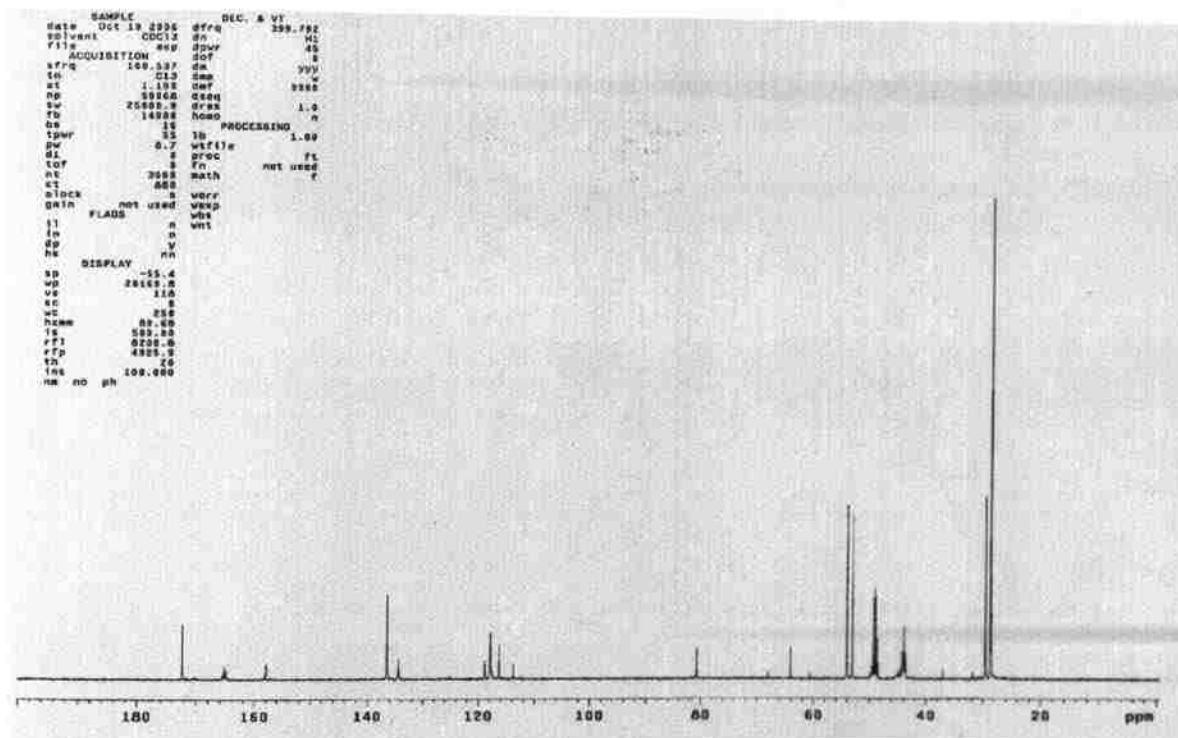
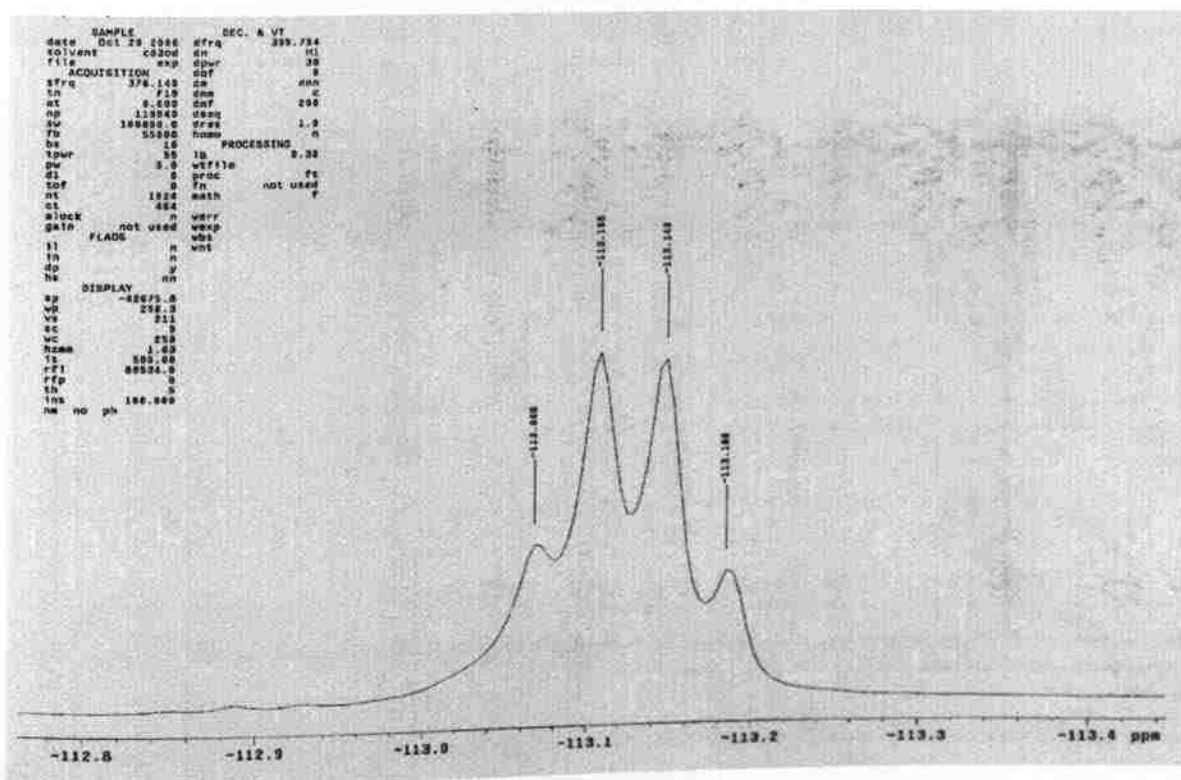


Figure 1.62: ^1H NMR spectra of compound 1.98

Figure 1.63: ^{13}C NMR spectra of compound 1.98Figure 1.64: ^{19}F NMR spectra of compound 1.98

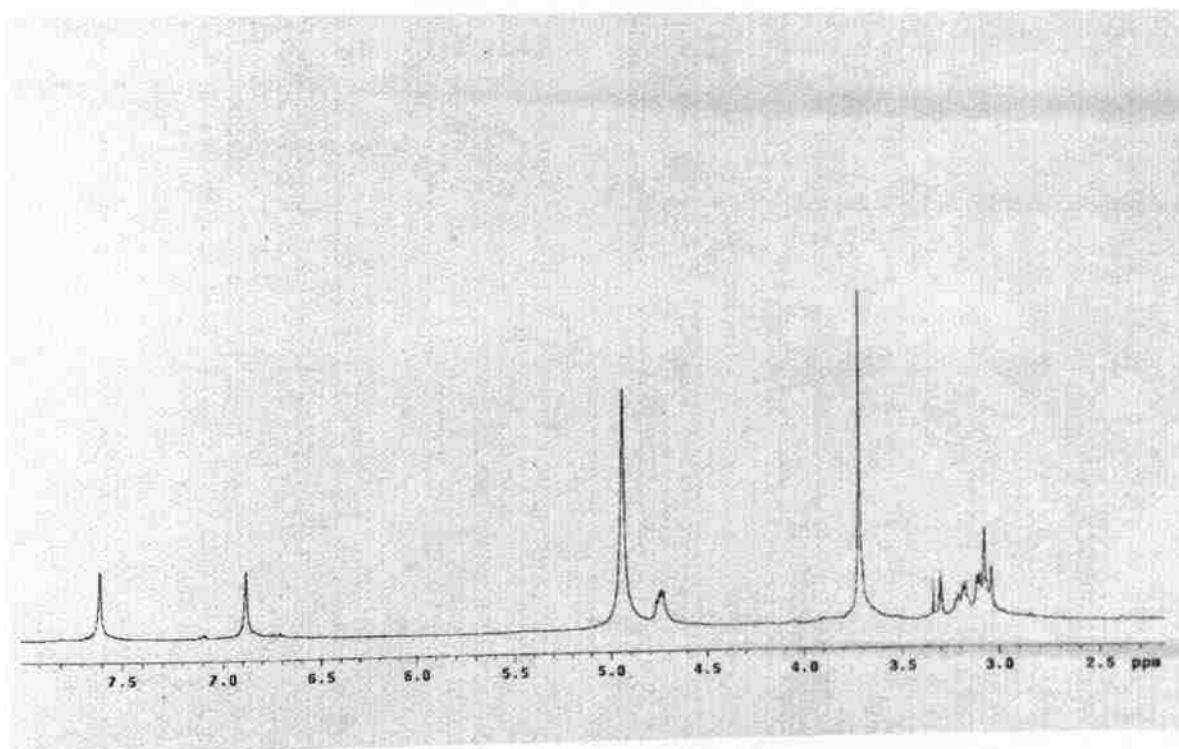


Figure 1.65: ^1H NMR spectra of compound 1.90

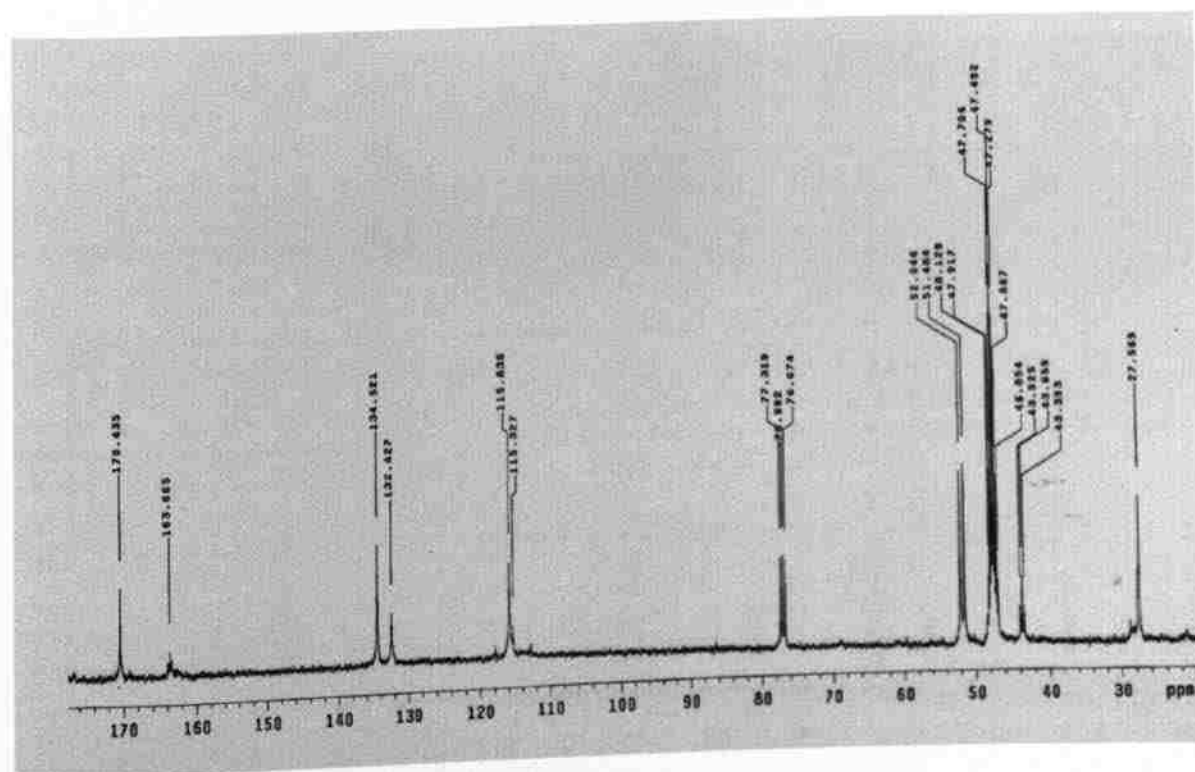


Figure 1.66: ^{13}C NMR spectra of compound 1.90

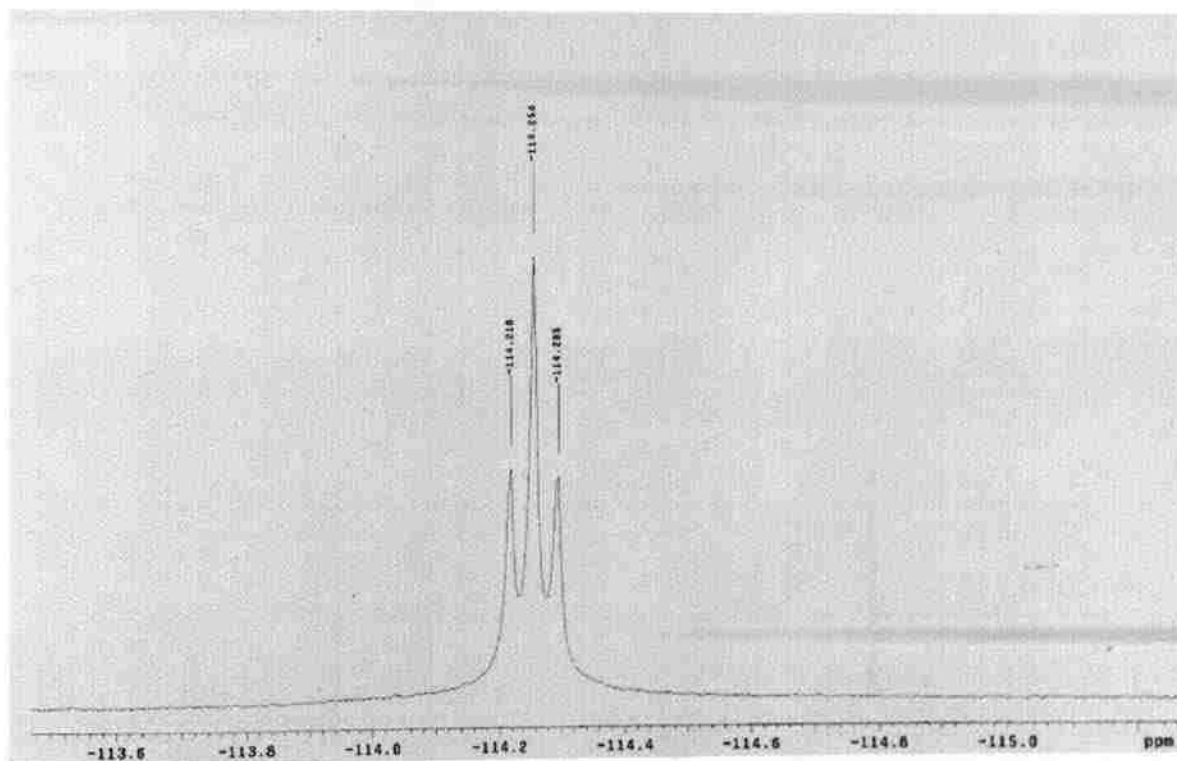


Figure 1.67: ^{19}F NMR spectra of compound 1.90

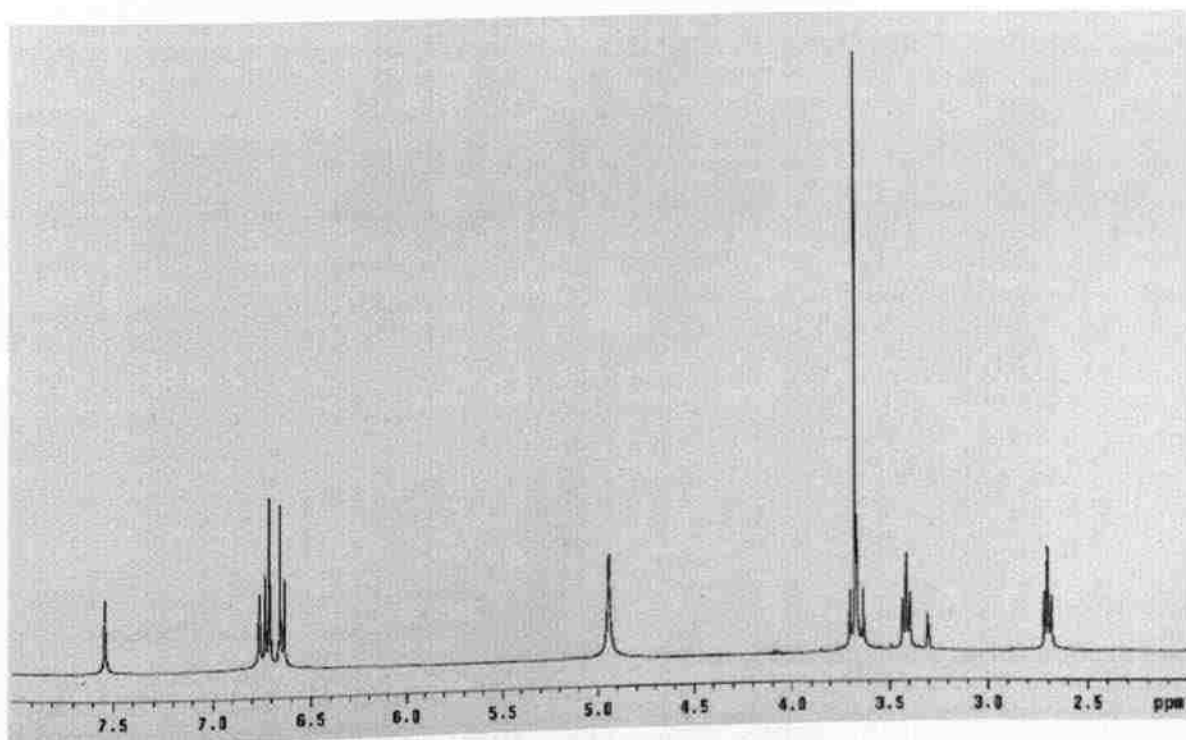


Figure 1.68: ^1H NMR spectra of compound 1.99

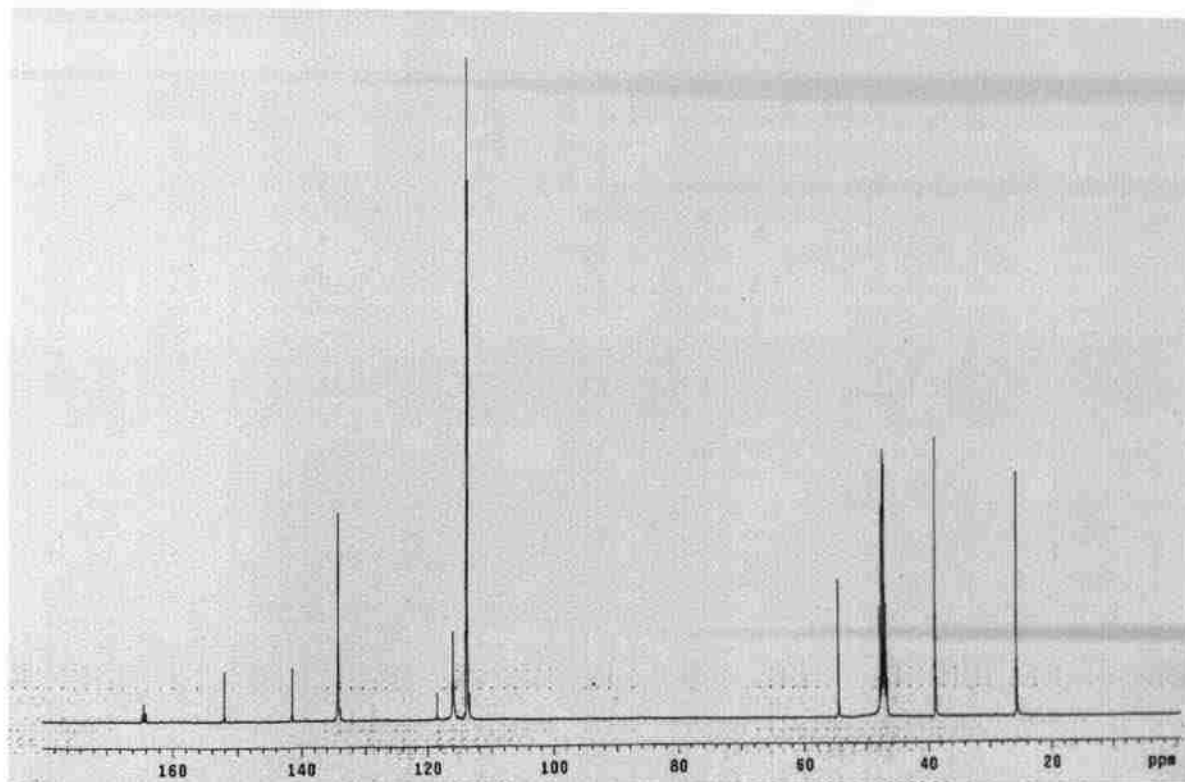


Figure 1.69: ^{13}C NMR spectra of compound 1.99

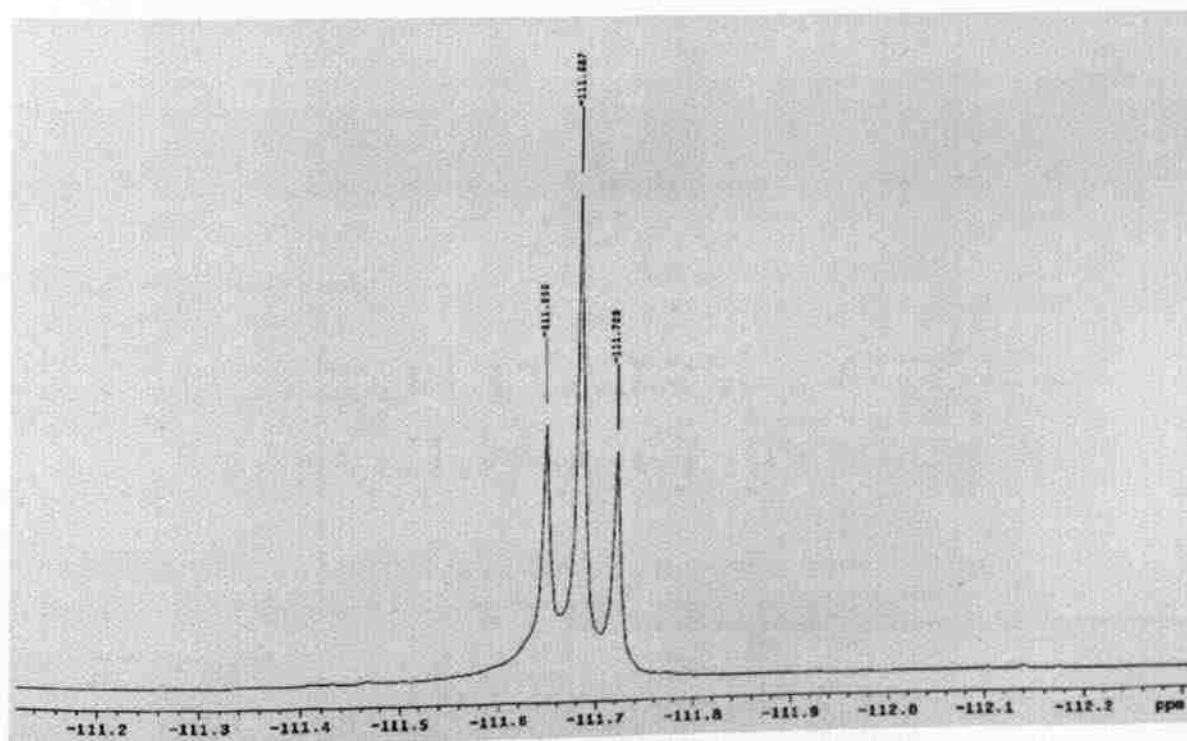
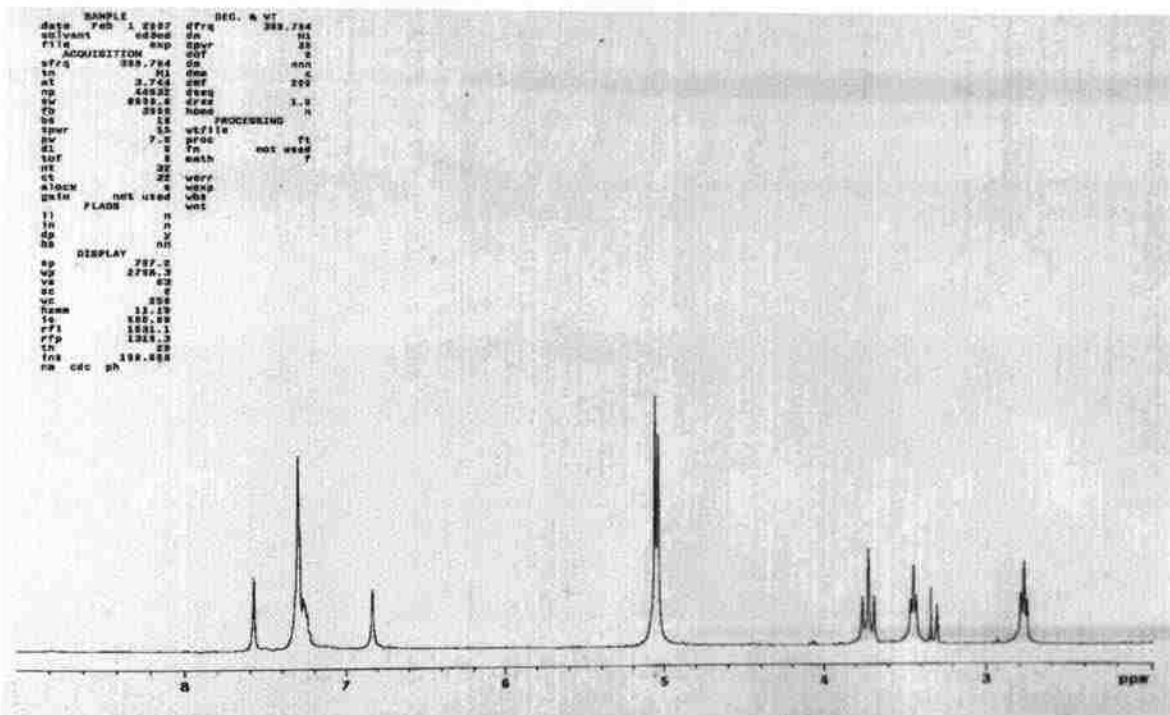
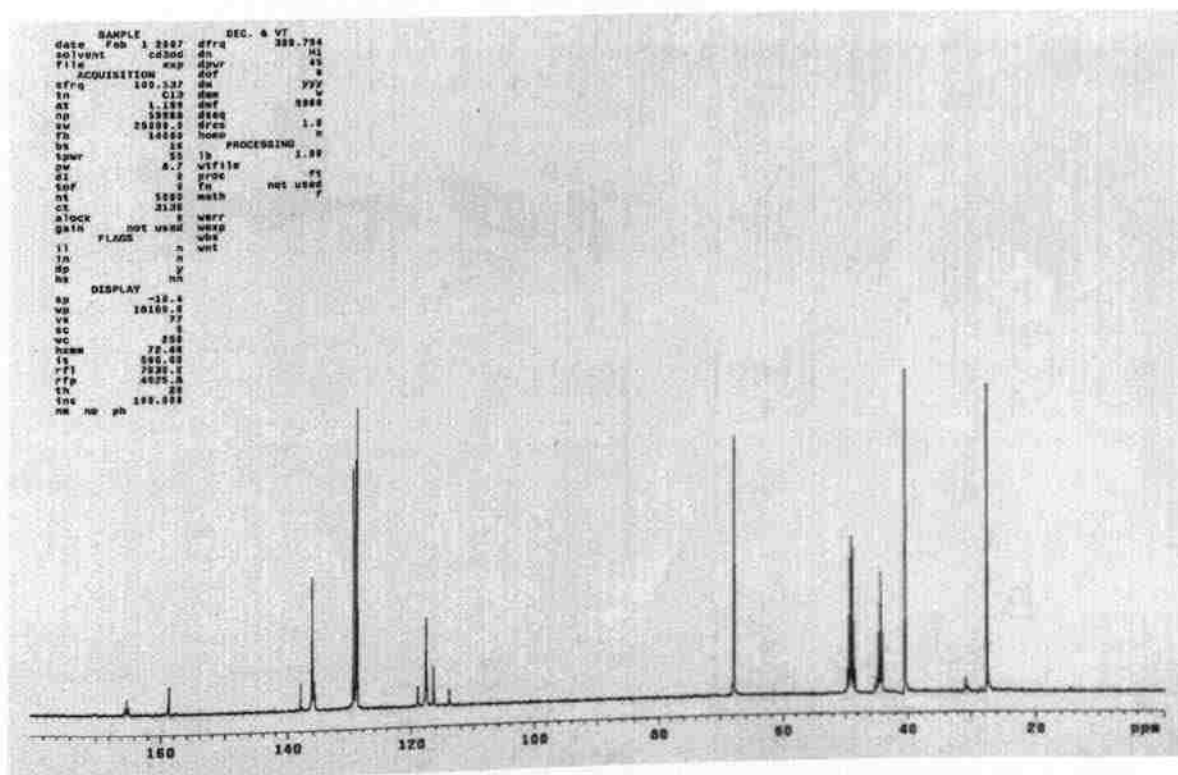


Figure 1.70: ^{19}F NMR spectra of compound 1.99

Figure 1.71: ^1H NMR spectra of compound 1.101Figure 1.72: ^{13}C NMR spectra of compound 1.101

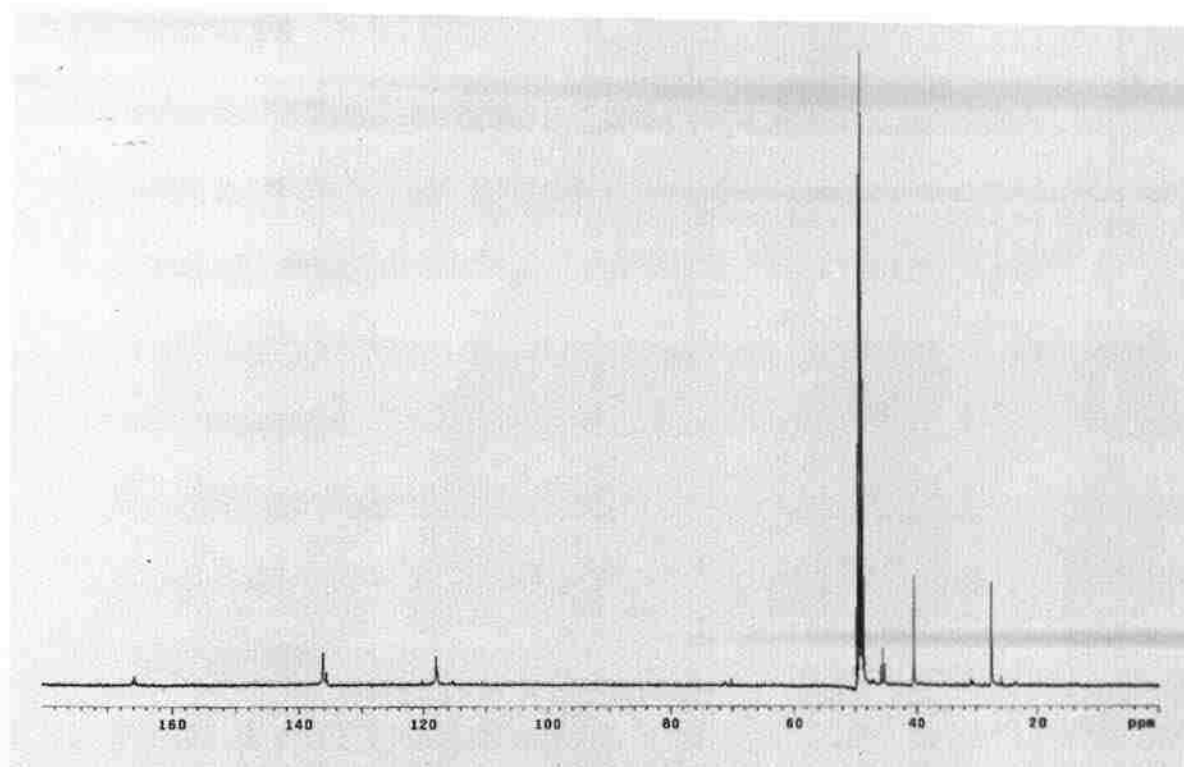


Figure 1.75: ^{13}C NMR spectra of compound 1.100

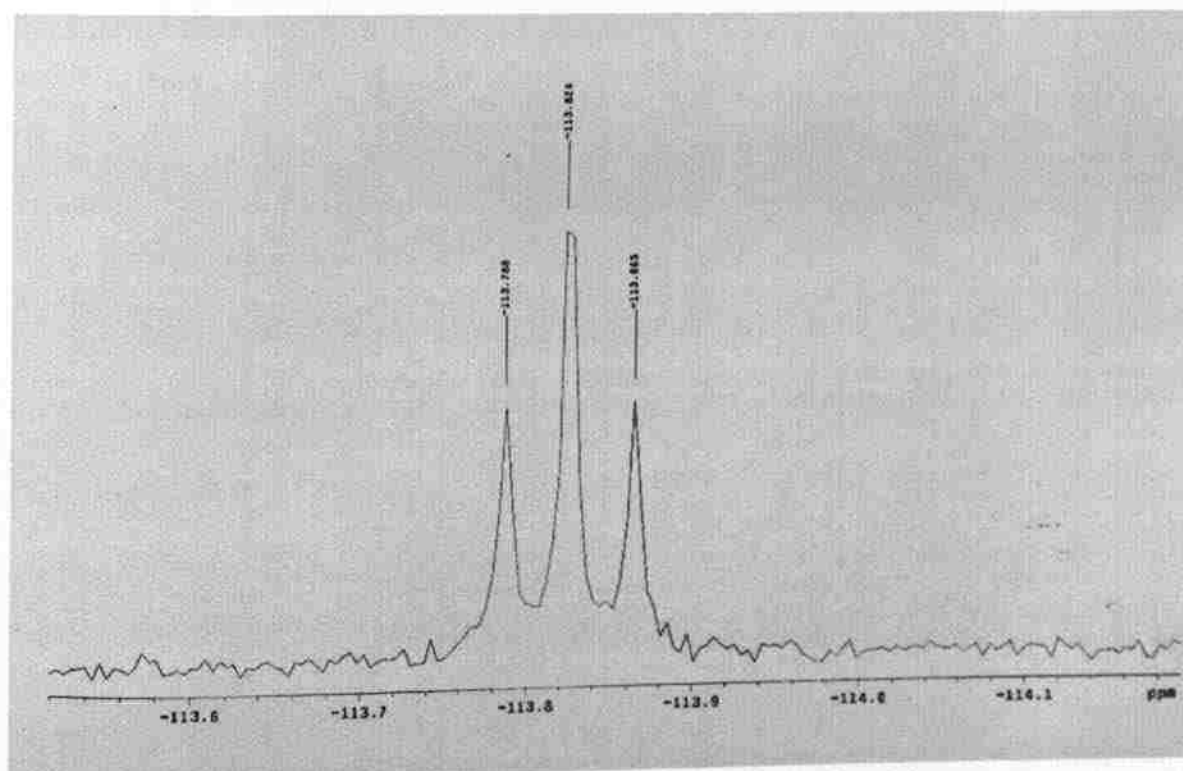


Figure 1.76: ^{19}F NMR spectra of compound 1.100

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2.1.3 Dithioacetals and dithioketals to *gem*-difluoromethylene compounds:

Dithioacetals and dithioketals derived from aldehydes and ketones were converted to the corresponding *gem*-difluoro-compounds by using 1,3-dibromo-5,5-dimethylhydantoin (DBH) or NBS with the presence of nucleophilic fluorine source PPHF in good yields.³⁷ But here in these procedures, electrophilic ring bromination has occurred in case of electron rich aromatic systems and gave side products.

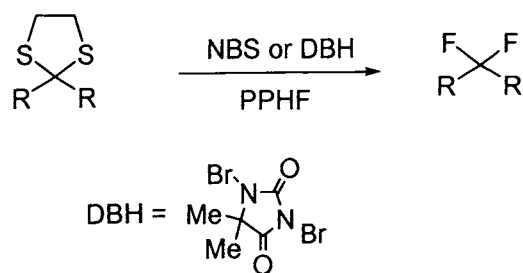
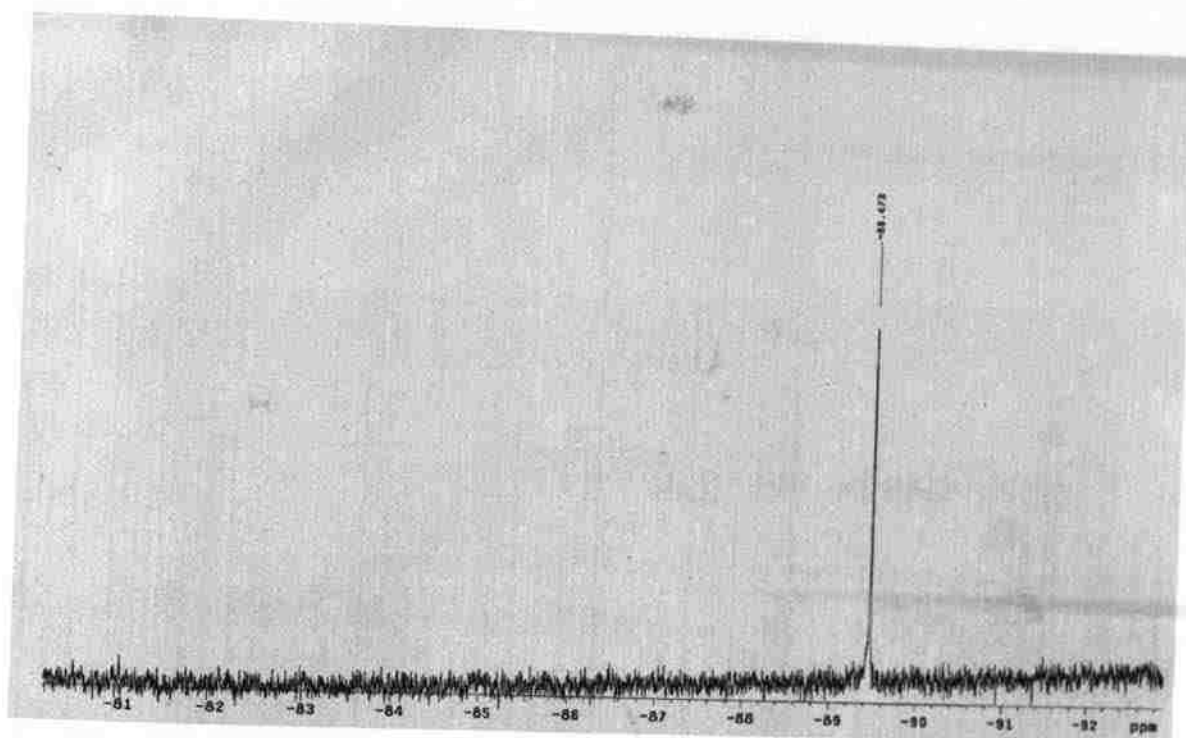
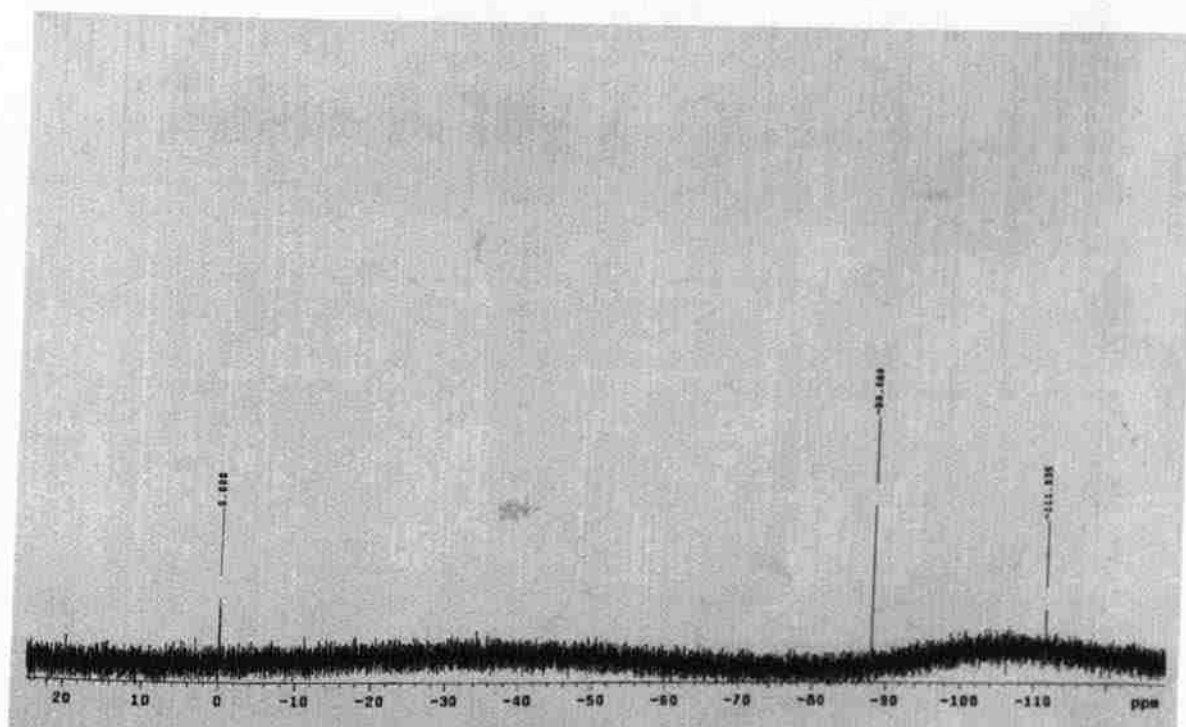


Figure 2.7: Conversion of dithiolanes to *gem*-difluoromethylene compounds using NBS/PPHF and DBH/PPHF

SO_2Cl_2 and SO_2ClF have been used as an alternative to the NBS as a source of electrophilic species in this transformation to facilitate the work up of the reaction, as the only byproducts of the reaction are the volatile SO_2 and HCl .³⁸ Nitrosonium tetrafluoroborate in the combination with PPHF has also been used for this transformation.³⁹ The desulfurative fluorination of dithiolanes was also done by using 1-diethylamino-1,1,2,2,3,3-hexafluoropropane in the presence of DBH as a source of electrophilic “ Br^+ ”.⁴⁰ 1-Diethylamino-1,1,2,2,3,3-hexafluoropropane releases HF on reaction with water and thus acts as nucleophilic fluoride ion source in this reaction. BrF_3 is also an effective reagent to bring about the transformation of dithiolanes to *gem*-difluoro compounds.⁴¹⁻⁴³

2.4.4 NMR spectra of products:

Figure 2.27: ^{19}F NMR spectra of compound 2.10Figure 2.28: ^{19}F NMR spectra of compound 2.11

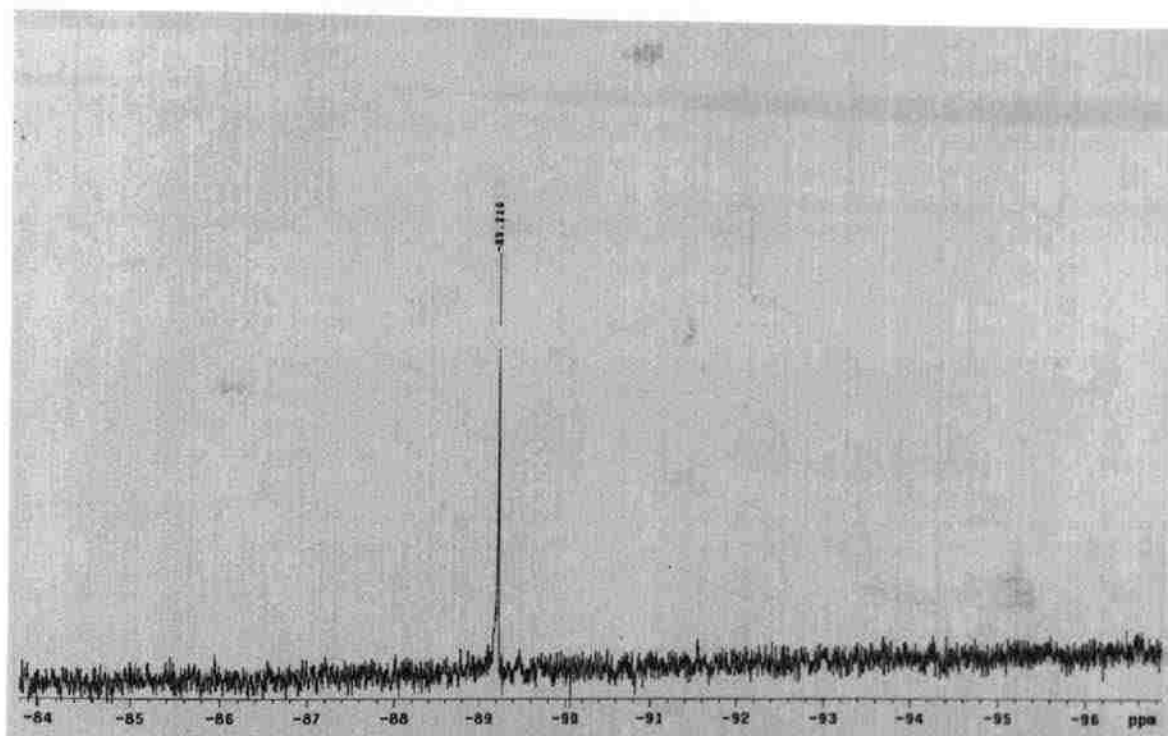


Figure 2.29: ^{19}F NMR spectra of compound 2.12

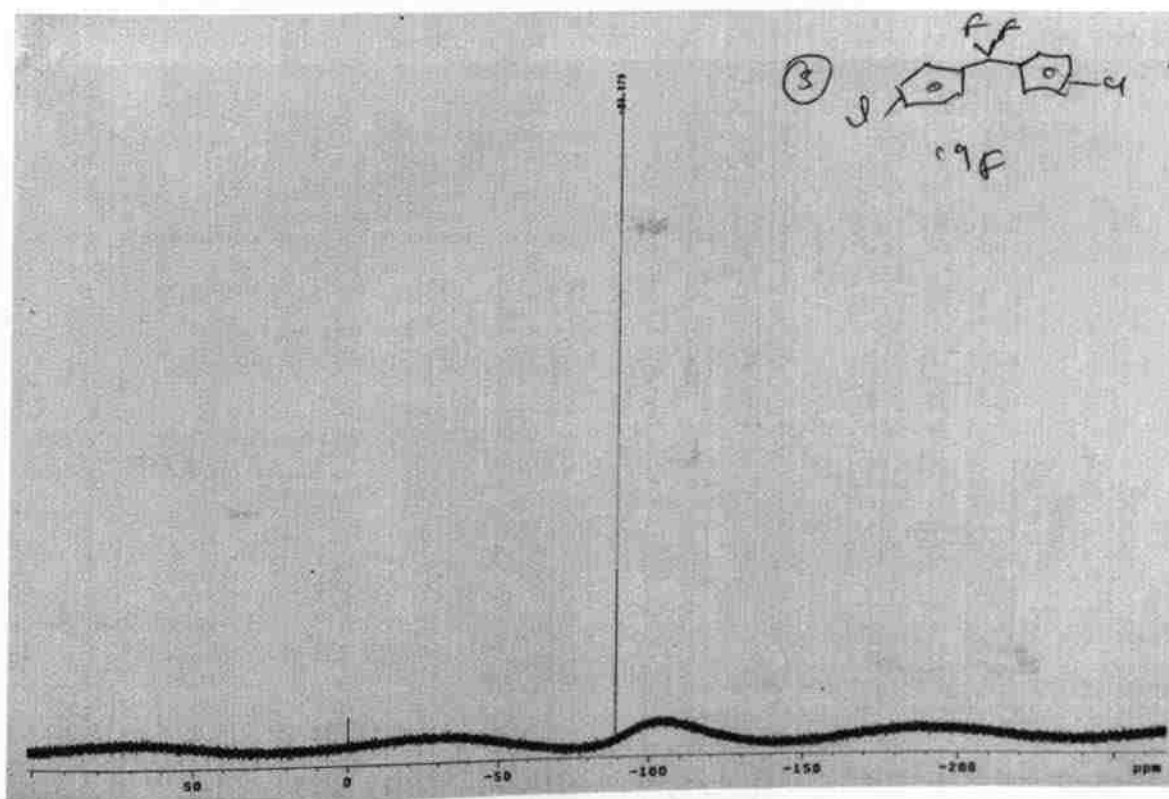


Figure 2.30: ^{19}F NMR spectra of compound 2.13

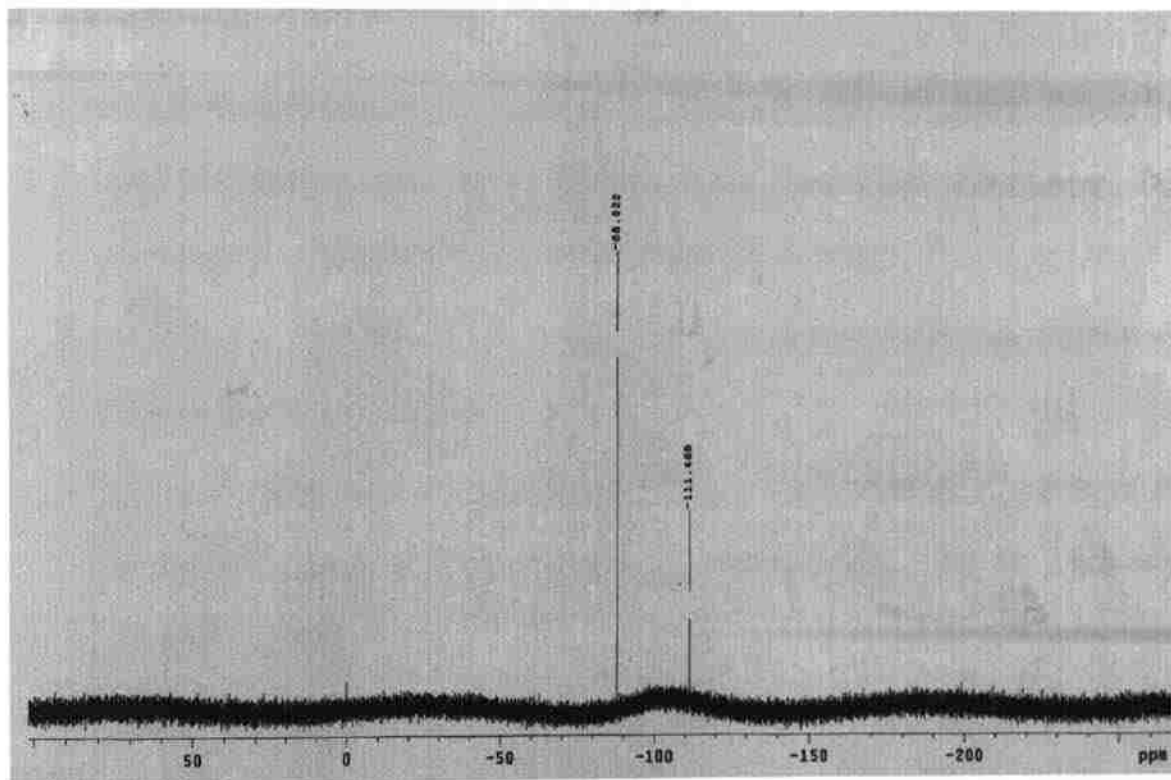


Figure 2.31: ^{19}F NMR spectra of compound 2.14

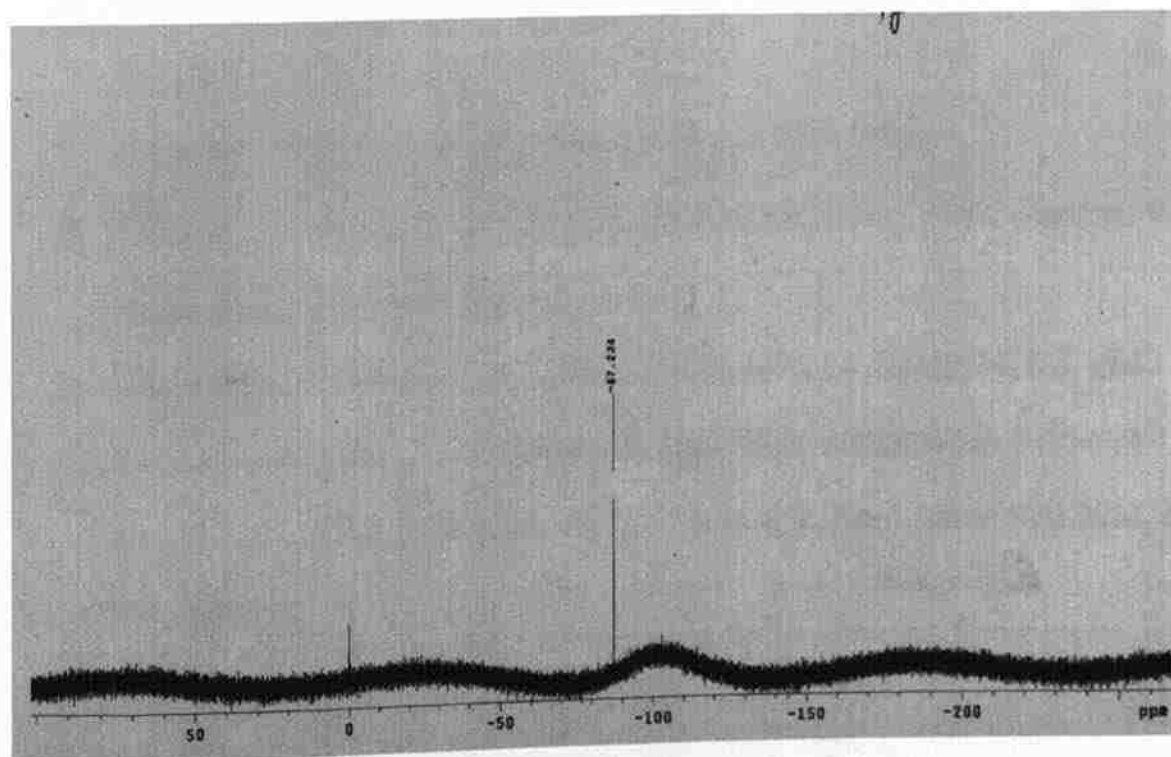
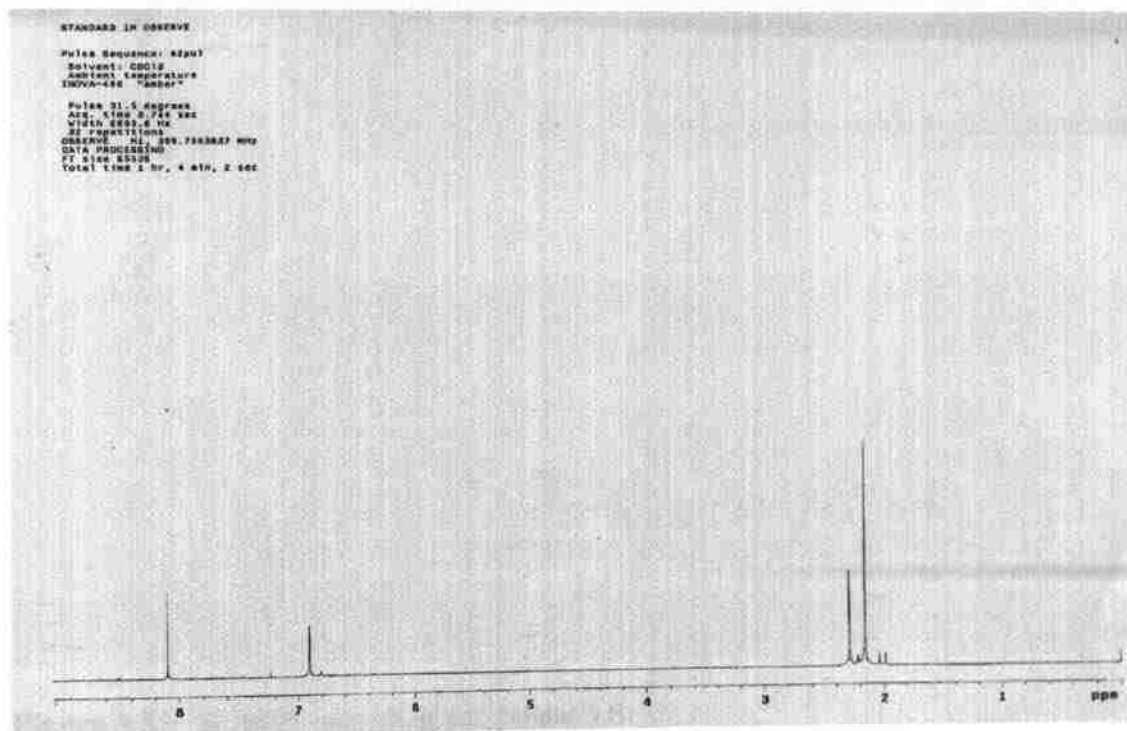
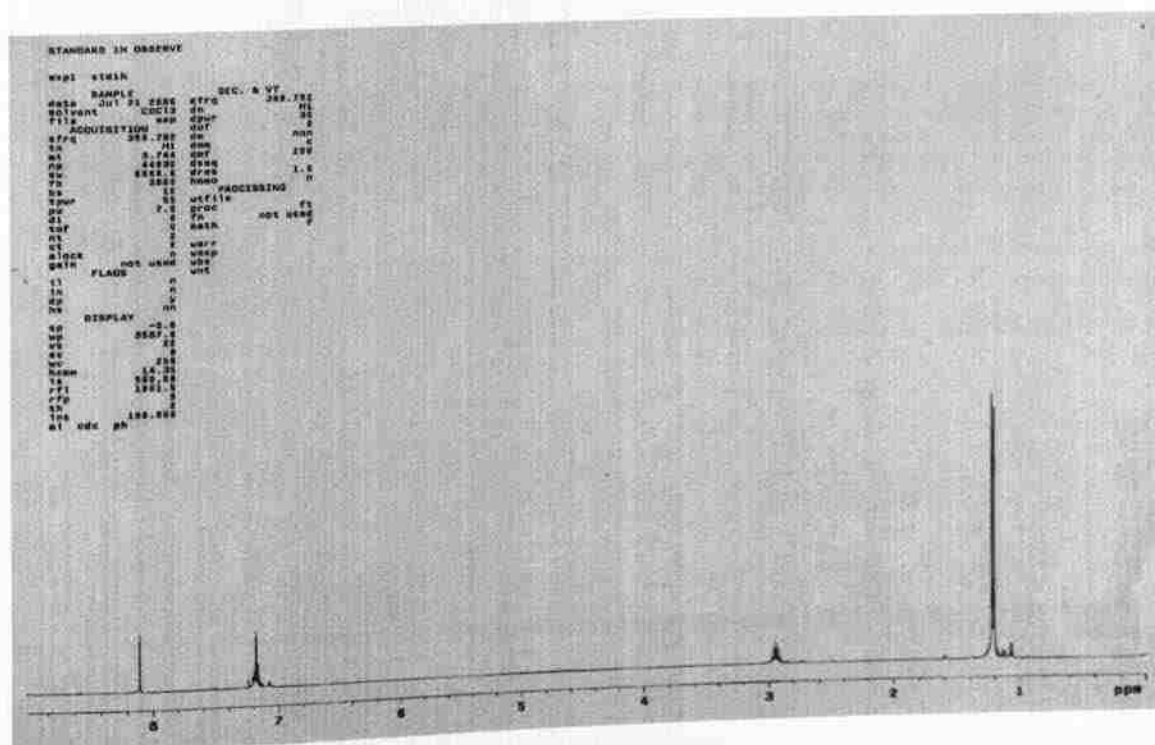
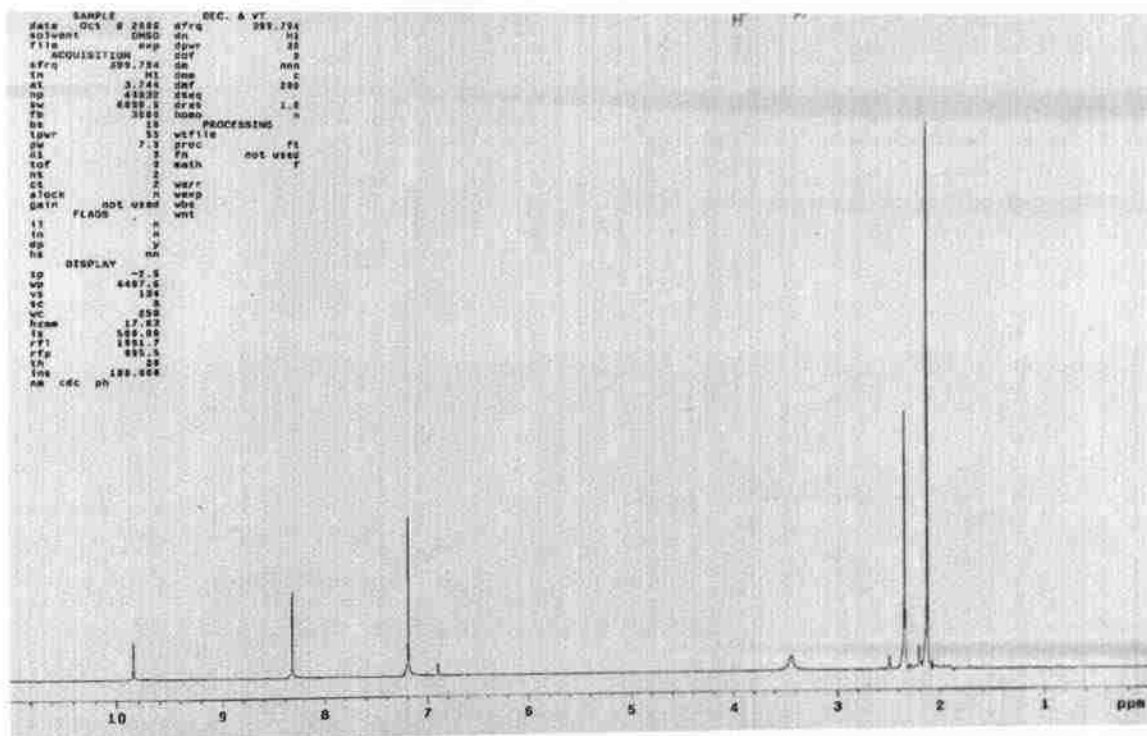
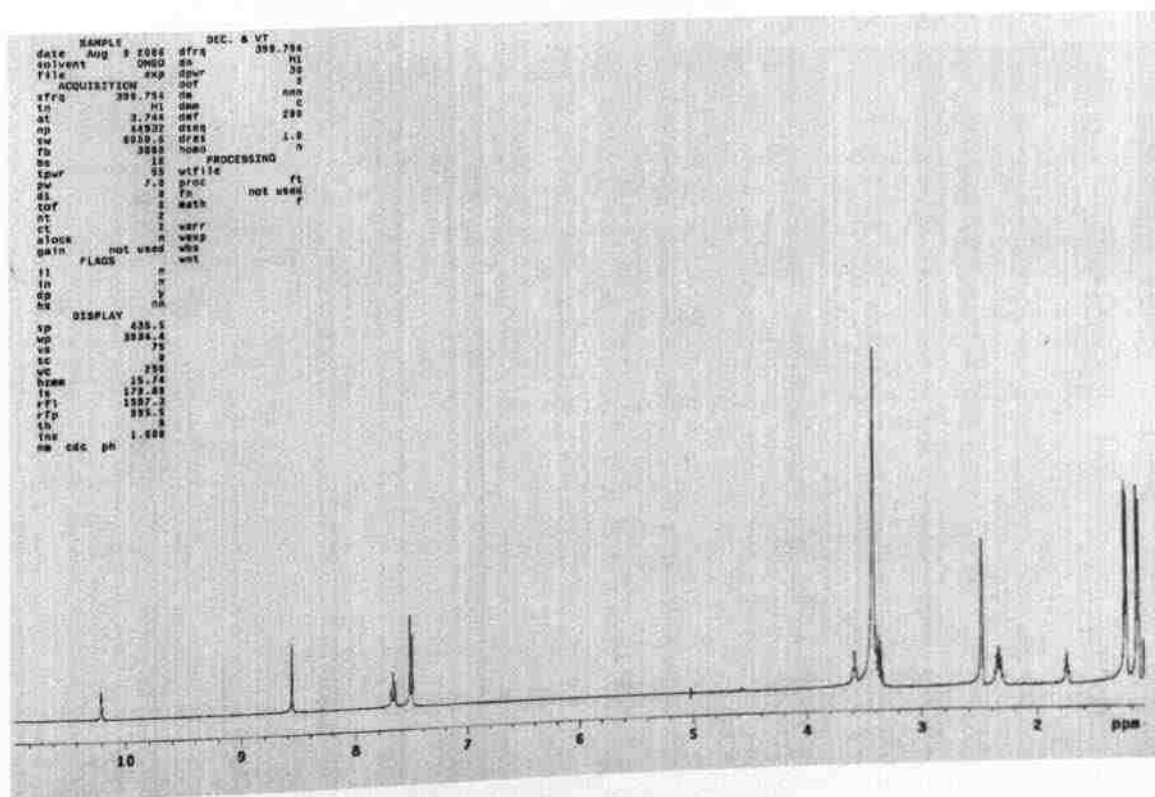


Figure 2.32: ^{19}F NMR spectra of compound 2.15

3.4.6 NMR spectra of products:

Figure 3.21: ^1H NMR spectra of compound 3.6Figure 3.22: ^1H NMR spectra of compound 3.7

Figure 3.23: ^1H NMR spectra of compound 3.8Figure 3.24: ^1H NMR spectra of compound 3.9

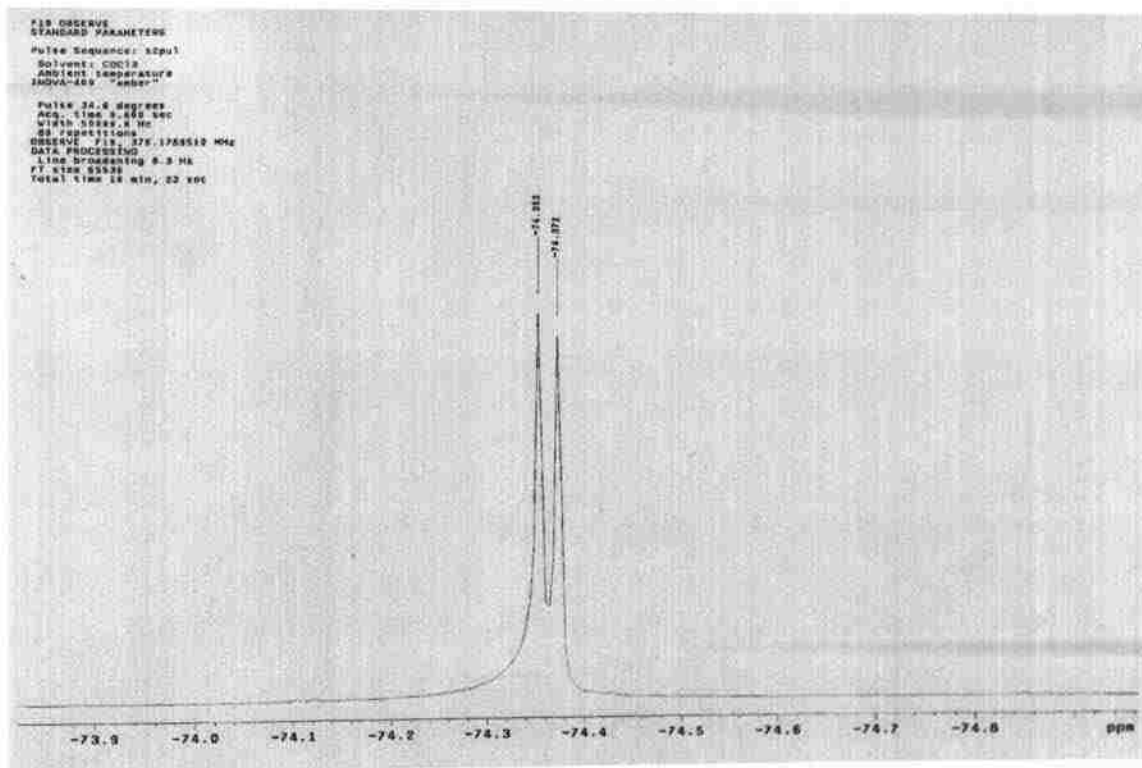


Figure 3.25: ^{19}F NMR spectra of compound 3.11

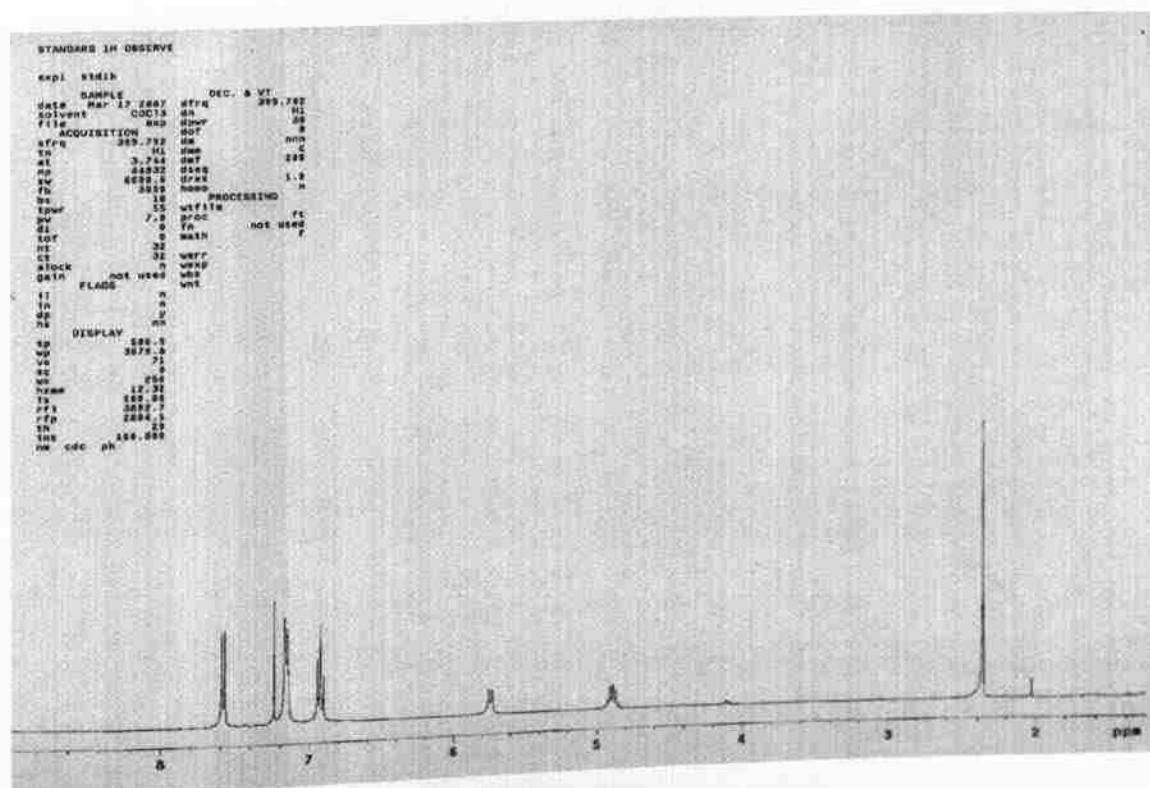


Figure 3.26: ^1H NMR spectra of compound 3.16

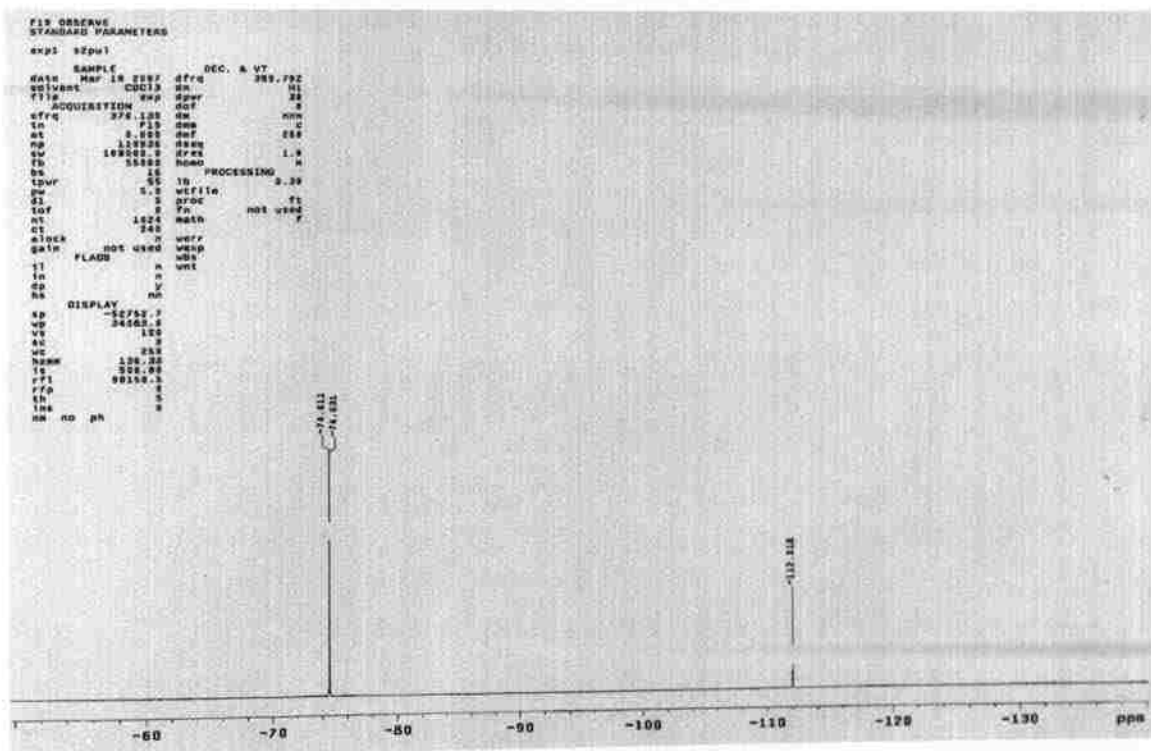


Figure 3.27: ^{19}F NMR spectra of compound 3.16

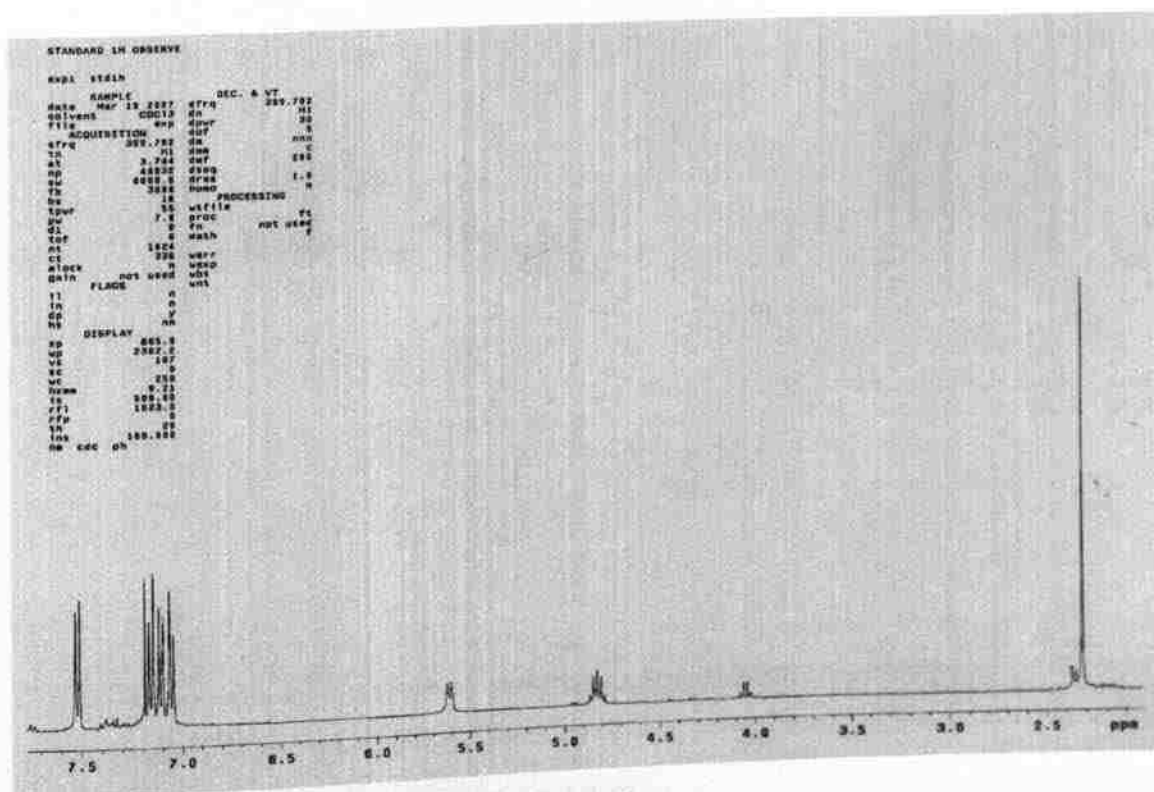


Figure 3.28: ^1H NMR spectra of compound 3.17

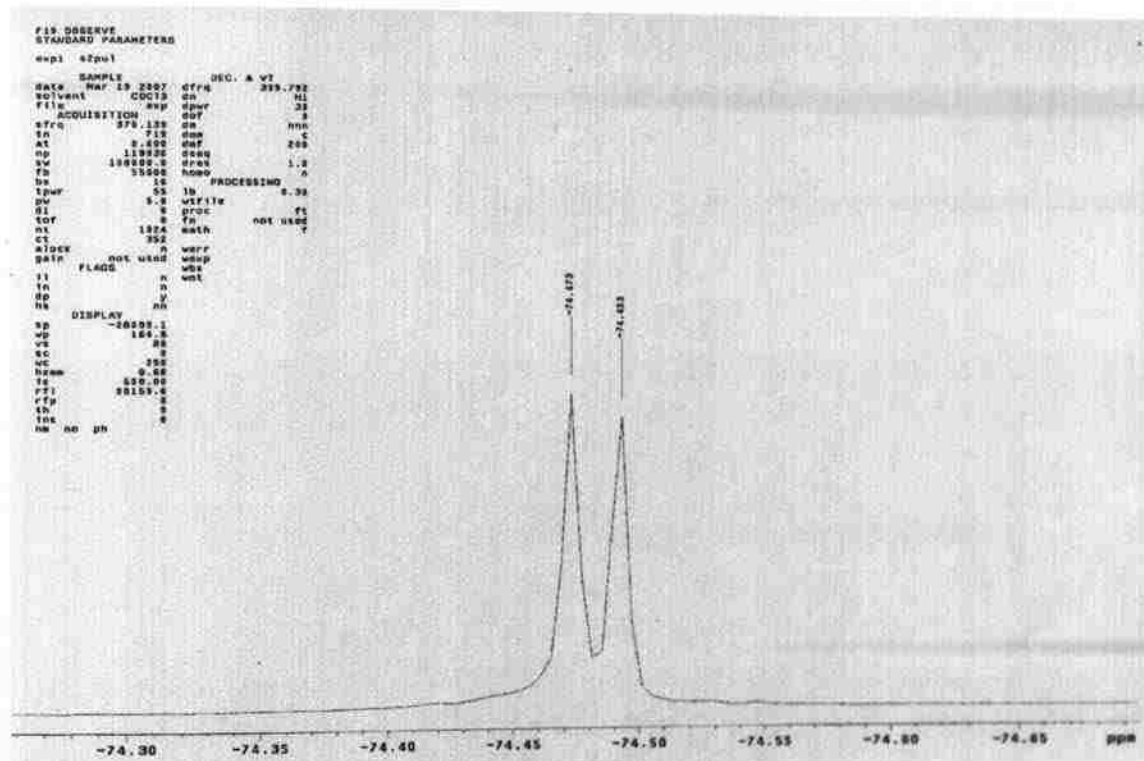


Figure 3.29: ^{19}F NMR spectra of compound 3.17

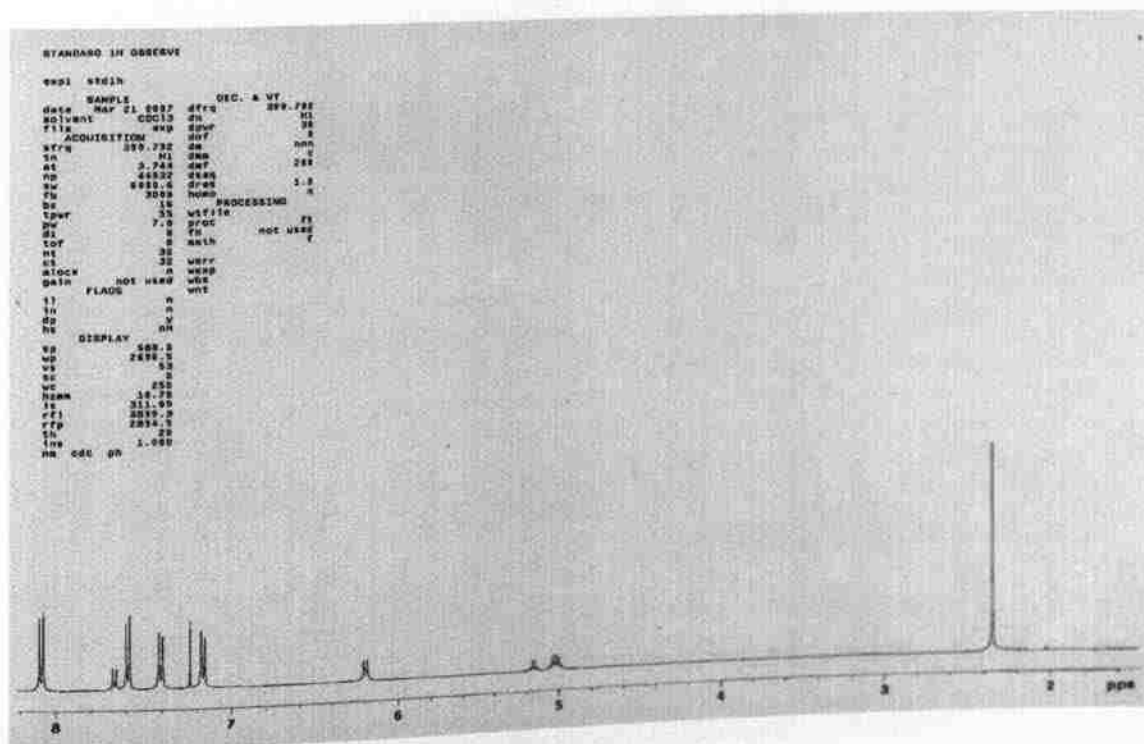


Figure 3.30: ^1H NMR spectra of compound 3.18

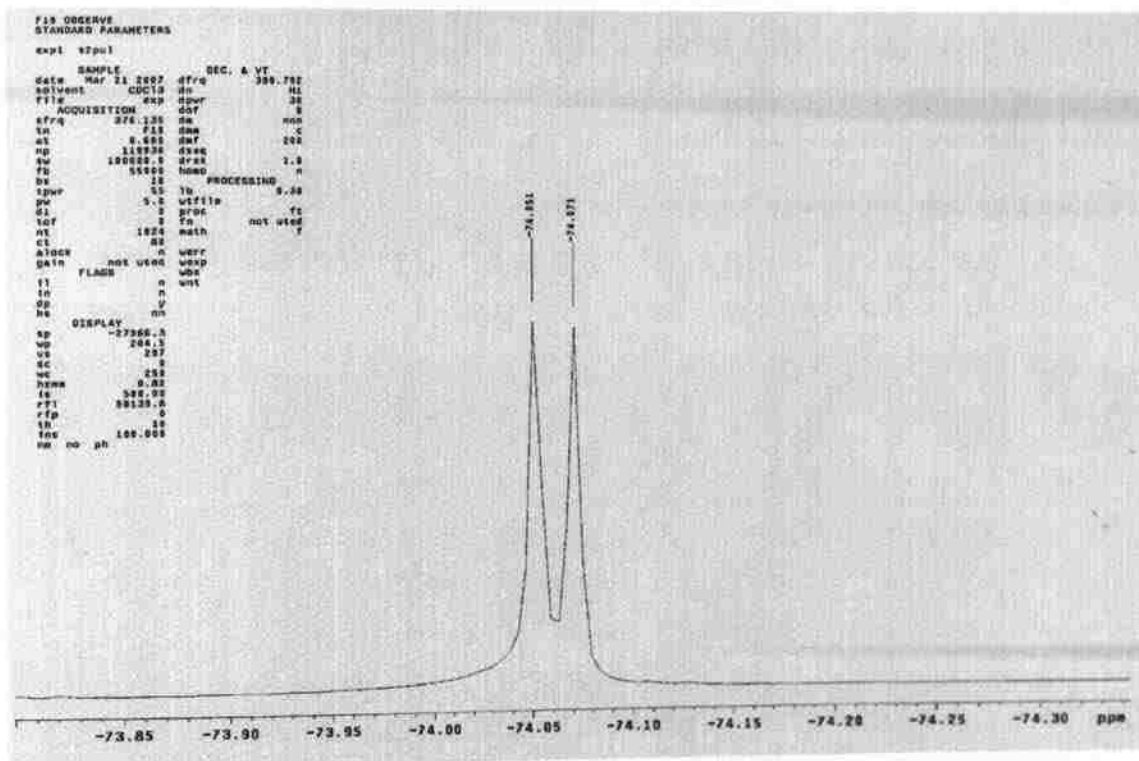


Figure 3.31: ^{19}F NMR spectra of compound 3.18

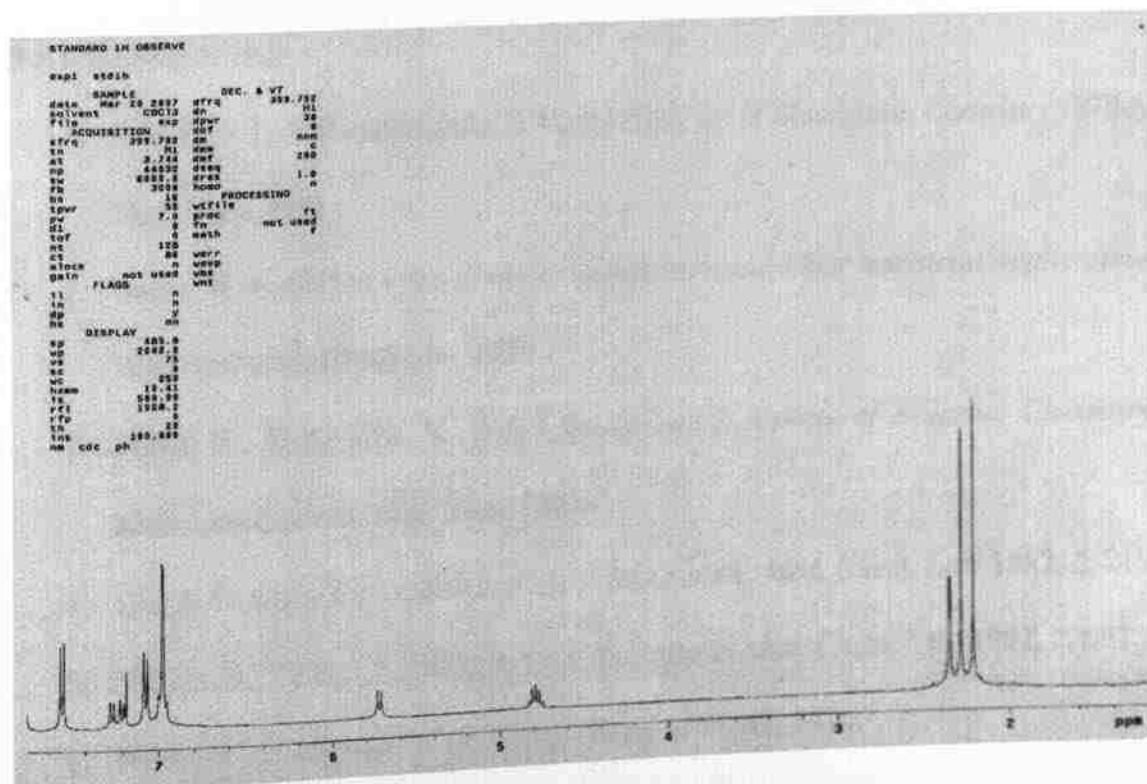


Figure 3.32: ^1H NMR spectra of compound 3.19

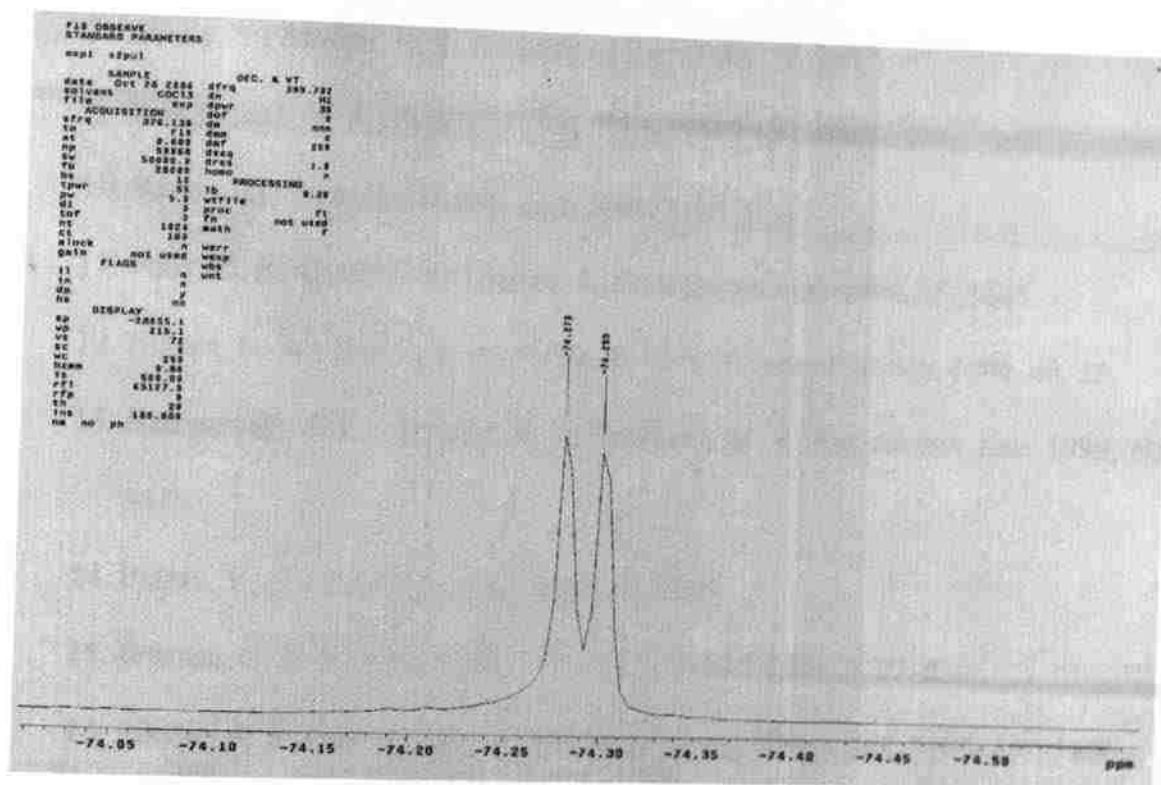


Figure 3.33: ^{19}F NMR spectra of compound 3.19

3.5 REFERENCES

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